

US EPA ARCHIVE DOCUMENT

# Environmental Technology Verification Report

Physical Removal of Microbial  
Contamination Agents in Drinking Water

EcoWater Systems, Inc.  
Sears Kenmore Ultrafilter 500 Drinking  
Water Treatment System

Prepared by



NSF International

 Under a Cooperative Agreement with  
U.S. Environmental Protection Agency

ET ✓ ET ✓ ET ✓



# THE ENVIRONMENTAL TECHNOLOGY VERIFICATION PROGRAM



U.S. Environmental Protection Agency



NSF International

## ETV Joint Verification Statement

TECHNOLOGY TYPE:	<b>POINT-OF-USE DRINKING WATER TREATMENT SYSTEM</b>	
APPLICATION:	<b>REMOVAL OF MICROBIAL CONTAMINATION AGENTS IN DRINKING WATER</b>	
PRODUCT NAME:	<b>SEARS KENMORE ULTRAFILTER 500</b>	
VENDOR:	<b>SEARS ROEBUCK, AND COMPANY</b>	
MANUFACTURER:	<b>ECOWATER SYSTEMS, INCORPORATED</b>	
ADDRESS:	<b>1890 WOODLANE DRIVE</b>	<b>PHONE: 1-800-808-9899</b>
	<b>WOODBURY, MN 55125</b>	<b>FAX: 651-739-5293</b>
EMAIL:	<b>INFO@ECOWATER.COM</b>	

NSF International (NSF) manages the Drinking Water Systems (DWS) Center under the U.S. Environmental Protection Agency's (EPA) Environmental Technology Verification (ETV) Program. The DWS Center recently evaluated the performance of the Sears Kenmore Ultrafilter 500 point-of-use (POU) reverse osmosis (RO) drinking water treatment system. NSF performed all of the testing activities and also authored the verification report and this verification statement. The verification report contains a comprehensive description of the test.

EPA created the ETV Program to facilitate the deployment of innovative or improved environmental technologies through performance verification and dissemination of information. The goal of the ETV Program is to further environmental protection by accelerating the acceptance and use of improved and more cost-effective technologies. ETV seeks to achieve this goal by providing high-quality, peer-reviewed data on technology performance to those involved in the design, distribution, permitting, purchase, and use of environmental technologies.

ETV works in partnership with recognized standards and testing organizations, stakeholder groups (consisting of buyers, vendor organizations, and permittees), and with the full participation of individual technology developers. The program evaluates the performance of innovative technologies by developing test plans that are responsive to the needs of stakeholders, conducting field or laboratory tests (as appropriate), collecting and analyzing data, and preparing peer-reviewed reports. All evaluations are conducted in accordance with rigorous quality assurance protocols to ensure that data of known and adequate quality are generated and that the results are defensible.

## ABSTRACT

The Sears Kenmore Ultrafilter 500 RO system was tested for removal of bacteria and viruses at NSF's Drinking Water Treatment Systems Laboratory. EcoWater Systems submitted ten units for testing, which were split into two groups of five. One group received 25 days of conditioning prior to challenge testing, while the second group was tested immediately. Both groups were identically challenged. The challenge organisms were the bacteriophage viruses fr, MS2, and Phi X 174, and the bacteria *Brevundimonas diminuta* and *Hydrogenophaga pseudoflava*. The test units were challenged at two different inlet pressures – 40 and 80 pounds per square inch, gauge (psig). The virus challenges were conducted at three different pH settings (6, 7.5, and 9) to assess whether pH influences the performance of the test units. The bacteria challenges were conducted only at pH 7.5.

The log<sub>10</sub> reduction data is shown in Tables 2 through 5. The test units removed all challenge organisms to less-than-detectable levels in all challenges but the pH 9, 80 psig challenge. The data does not show whether conditioning, inlet pressure or pH influenced bacteria and virus removal.

## TECHNOLOGY DESCRIPTION

The following technology description was provided by the manufacturer and has not been verified.

The Ultrafilter 500 is a three-stage POU drinking water treatment system. In addition to the RO membrane, the system employs carbon filtration. The first stage of treatment is carbon filtration to remove chlorine as well as suspended particles such as silt, dirt, and rust. The second stage is the RO membrane, which removes a wide variety of contaminants. The RO treated water is sent to the storage tank. When the user opens the faucet, the water leaves the storage tank and travels through a second carbon filter that removes any remaining tastes and odors before it is dispensed. The Ultrafilter 500 is designed to produce approximately five gallons of wastewater for every gallon of treated water.

The test units were evaluated without the carbon filters in place to eliminate the possibility that these filters could temporarily trap a portion of the challenge organisms, causing a positive bias of system performance.

## VERIFICATION TESTING DESCRIPTION

### *Test Site*

The testing site was the Drinking Water Treatment Systems Laboratory at NSF in Ann Arbor, Michigan. A description of the test apparatus can be found in the test/QA plan and verification report. The testing was conducted in January through March of 2004.

### *Methods and Procedures*

The testing methods and procedures are detailed in the Test/QA Plan for Verification Testing of the Sears Kenmore Ultrafilter 500 Point-of-Use Drinking Water Treatment System for Removal of Microbial Contamination Agents. Ten Ultrafilter 500 systems were tested for bacteria and virus removal performance using the bacteriophage viruses fr, MS2, and Phi X 174, and the bacteria *B. diminuta* and *H. pseudoflava*. The challenge organisms were chosen because they are smaller than most other viruses and bacteria, and so provide a conservative estimate of performance.

The test units were randomly split into two groups of five. One group was conditioned for 25 days by operating the units daily using the test water without challenge organisms. The second group was

challenged without receiving the 25-day conditioning period. The test units were challenged at both 40 and 80 psig inlet pressure. The test water for the bacteria challenges was set to pH  $7.5 \pm 0.5$ , while the virus challenges were conducted at pH  $6.0 \pm 0.5$ ,  $7.5 \pm 0.5$ , and  $9.0 \pm 0.5$ . The challenge schedule is shown in Table 1. The different challenge conditions were intended to evaluate whether inlet pressure or pH influenced bacteria and virus removal. However, the test water chemistry gave it little buffering capacity, which made it difficult to keep the pH below 6.5 for the pH 6.0 virus challenges. As a result, the pH was above 6.5 for three of the four pH 6.0 virus challenges.

**Table 1. Challenge Schedule**

Day	Surrogate Challenge	pH	Inlet Pressure (psig)
1	<i>H. pseudoflava</i>	$7.5 \pm 0.5$	$40 \pm 3$
2	<i>H. pseudoflava</i>	$7.5 \pm 0.5$	$80 \pm 3$
3	<i>B. diminuta</i>	$7.5 \pm 0.5$	$40 \pm 3$
4	<i>B. diminuta</i>	$7.5 \pm 0.5$	$80 \pm 3$
5	All Viruses	$*6.0 \pm 0.5$	$40 \pm 3$
6	All Viruses	$*6.0 \pm 0.5$	$80 \pm 3$
7	All Viruses	$7.5 \pm 0.5$	$40 \pm 3$
8	All Viruses	$7.5 \pm 0.5$	$80 \pm 3$
9	All Viruses	$9.0 \pm 0.5$	$40 \pm 3$
10	All Viruses	$9.0 \pm 0.5$	$80 \pm 3$

\*actual pH ranged from 6.7 – 6.9 in three of four days.

On each challenge day, the test units were operated for one tank-fill period (approximately two hours). The end of this period was evident through engagement of each system’s automatic shutoff mechanism, which causes the flow of reject water to cease. Influent water samples were collected at the beginning and end of each challenge period. After each test unit ceased operation, the entire contents of the product water storage tank were emptied into a sterile container, and a subsample was collected for microbiological analysis. All samples were enumerated in triplicate. Following each challenge period, the test units were flushed by operating them for one tank-fill period using the test water without challenge organisms.

**VERIFICATION OF PERFORMANCE**

Tables 2 and 3 show the bacteria reduction data for the unconditioned units and conditioned units, respectively. In all challenges for both sets of test units, the bacteria were removed to less than detectable levels (< 1 CFU/100mL). The predominance of non-detectable results does not allow any evaluation of whether conditioning, inlet pressure or pH influenced the bacteria reduction performance of the RO membranes.

Tables 4 and 5 show the virus reduction data for the unconditioned units and conditioned units, respectively. In all challenges but the pH 9, 80 psig challenge, both sets of test units removed all three viruses to less than detectable levels (< 1 PFU/mL). The maximum mean effluent count for the pH 9, 80 psig challenges was 11 PFU/mL, which corresponds to the 3.0 log<sub>10</sub> reduction of fr for unconditioned unit 3. As with the bacteria, the predominance of non-detectable results does not allow an evaluation of the effect of conditioning, inlet pressure, or pH on RO membrane performance. Complete descriptions of the verification testing results are included in the verification report.

**Table 2. Bacteria Log Reduction Data for Unconditioned Units**

pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub> Influent Challenge	Log <sub>10</sub> Reduction				
				Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
7.5	40	<i>H. pseudoflava</i>	6.6	All effluents non-detect				
		<i>B. diminuta</i>	6.4	Log reductions equal to influents				
7.5	80	<i>H. pseudoflava</i>	6.0	All effluents non-detect				
		<i>B. diminuta</i>	6.6	Log reductions equal to influents				

**Table 3. Bacteria Log Reduction Data for Conditioned Units**

pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub> Influent Challenge	Log <sub>10</sub> Reduction				
				Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
7.5	40	<i>H. pseudoflava</i>	6.6	All effluents non-detect				
		<i>B. diminuta</i>	7.1	Log reductions equal to influents				
7.5	80	<i>H. pseudoflava</i>	6.0	All effluents non-detect				
		<i>B. diminuta</i>	6.8	Log reductions equal to influents				

**Table 4. Virus Log<sub>10</sub> Reduction Data for Unconditioned Units**

Challenge Conditions					Log <sub>10</sub> Reduction				
Target pH	Measured pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub> Influent Challenge	Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
6.0 ± 0.5	6.86	40	fr	5.0	All effluents non-detect				
			MS2	4.8	Log reductions equal to influents				
			Phi X 174	4.5					
6.0 ± 0.5	6.88	80	fr	5.4	All effluents non-detect				
			MS2	5.2	Log reductions equal to influents				
			Phi X 174	4.0					
7.5 ± 0.5	7.69	40	fr	4.3	All effluents non-detect				
			MS2	5.0	Log reductions equal to influents				
			Phi X 174	5.3					
7.5 ± 0.5	7.91	80	fr	4.0	All effluents non-detect				
			MS2	4.9	Log reductions equal to influents				
			Phi X 174	4.4					
9.0 ± 0.5	8.71	40	fr	5.3	All effluents non-detect				
			MS2	5.1	Log reductions equal to influents				
			Phi X 174	4.4					
9.0 ± 0.5	8.67	80	fr	4.1	3.8	3.6	3.0	4.1	4.1
			MS2	3.9	3.9	3.6	2.9	3.9	3.9
			Phi X 174	3.7	3.7	3.7	3.7	3.7	3.7

**Table 5. Virus Log<sub>10</sub> Reduction Data for Conditioned Units**

Challenge Conditions			Log <sub>10</sub> Reduction						
Target pH	Measured pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub> Influent Challenge	Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
6.0 ± 0.5	6.48	40	fr	4.8	All effluents non-detect Log reductions equal to influents				
			MS2	4.5					
			Phi X 174	3.8					
6.0 ± 0.5	6.69	80	fr	4.5	All effluents non-detect Log reductions equal to influents				
			MS2	4.4					
			Phi X 174	4.2					
7.5 ± 0.5	7.45	40	fr	5.3	All effluents non-detect Log reductions equal to influents				
			MS2	5.0					
			Phi X 174	4.3					
7.5 ± 0.5	7.56	80	fr	4.9	All effluents non-detect Log reductions equal to influents				
			MS2	4.7					
			Phi X 174	3.9					
9.0 ± 0.5	8.73	40	fr	5.6	All effluents non-detect Log reductions equal to influents				
			MS2	5.4					
			Phi X 174	3.8					
9.0 ± 0.5	8.73	80	fr	5.1	5.1	4.6	5.1	5.1	5.1
			MS2	4.8	4.5	4.3	4.8	4.8	4.5
			Phi X 174	4.5	4.5	4.5	4.5	4.5	4.5

**Quality Assurance/Quality Control (QA/QC)**

NSF provided technical and quality assurance oversight of the verification testing as described in the verification report, including an audit of nearly 100% of the data. NSF personnel also conducted a technical systems audit during testing to ensure the testing was in compliance with the test plan. A complete description of the QA/QC procedures is provided in the verification report.

*Original signed by*  
*E. Timothy Oppelt*

*09/20/04*

E. Timothy Oppelt  
Director  
National Homeland Security Research Center  
United States Environmental Protection  
Agency

Date

*Original signed by*  
*Gordon Bellen*

*09/23/04*

Gordon Bellen  
Vice President  
Research  
NSF International

Date

NOTICE: Verifications are based on an evaluation of technology performance under specific, predetermined criteria and the appropriate quality assurance procedures. EPA and NSF make no expressed or implied warranties as to the performance of the technology and do not certify that a technology will always operate as verified. The end-user is solely responsible for complying with any and all applicable federal, state, and local requirements. Mention of corporate names, trade names, or commercial products does not constitute endorsement or recommendation for use of specific products. This report is not an NSF Certification of the specific product mentioned herein.



**Availability of Supporting Documents**

Copies of the test protocol, the verification statement, and the verification report (NSF report # NSF 04/14/EPADWCTR) are available from the following sources:

(NOTE: Appendices are not included in the verification report. Appendices are available from NSF upon request.)

1. ETV Drinking Water Systems Center Manager (order hard copy)  
NSF International  
P.O. Box 130140  
Ann Arbor, Michigan 48113-0140
2. NSF web site: <http://www.nsf.org/etv> (electronic copy)
3. EPA web site: <http://www.epa.gov/etv> (electronic copy)

## **Environmental Technology Verification Report**

### **Removal of Microbial Contamination Agents in Drinking Water**

#### **EcoWater Systems Incorporated Sears Kenmore Ultrafilter 500 Drinking Water Treatment System**

Prepared by:

NSF International  
Ann Arbor, Michigan 48105

Under a cooperative agreement with the U.S. Environmental Protection Agency

Jeffrey Q. Adams, Project Officer  
National Risk Management Research Laboratory  
U.S. Environmental Protection Agency  
Cincinnati, Ohio 45268

## Notice

The U.S. Environmental Protection Agency (EPA), through its Office of Research and Development, has financially supported and collaborated with NSF International (NSF) under Cooperative Agreement No. R-82833301. This verification effort was supported by the Drinking Water Systems (DWS) Center, operating under the Environmental Technology Verification (ETV) Program. This document has been peer-reviewed, reviewed by NSF and EPA, and recommended for public release.

## Foreword

The U.S. Environmental Protection Agency (EPA) is charged by Congress with protecting the Nation's land, air, and water resources. Under a mandate of national environmental laws, the Agency strives to formulate and implement actions leading to a compatible balance between human activities and the ability of natural systems to support and nurture life. To meet this mandate, EPA's research program is providing data and technical support for solving environmental problems today and building a science knowledge base necessary to manage our ecological resources wisely, understand how pollutants affect our health, and prevent or reduce environmental risks in the future.

The National Risk Management Research Laboratory (NRMRL) is the Agency's center for investigation of technological and management approaches for preventing and reducing risks from pollution that threaten human health and the environment. The focus of the Laboratory's research program is on methods and their cost-effectiveness for prevention and control of pollution to air, land, water, and subsurface resources; protection of water quality in public water systems; remediation of contaminated sites, sediments and ground water; prevention and control of indoor air pollution; and restoration of ecosystems. NRMRL collaborates with both public and private sector partners to foster technologies that reduce the cost of compliance and to anticipate emerging problems. NRMRL's research provides solutions to environmental problems by: developing and promoting technologies that protect and improve the environment; advancing scientific and engineering information to support regulatory and policy decisions; and providing the technical support and information transfer to ensure implementation of environmental regulations and strategies at the national, state, and community levels.

This publication has been produced as part of the Laboratory's strategic long-term research plan. It is published and made available by EPA's Office of Research and Development to assist the user community and to link researchers with their clients.

**Table of Contents**

Verification Statement ..... VS-i  
 Title Page ..... i  
 Notice..... ii  
 Foreword..... iii  
 Table of Contents..... iv  
 List of Tables ..... vi  
 List of Figures ..... vi  
 Abbreviations and Acronyms ..... vii  
 Acknowledgements..... viii

Chapter 1 Introduction..... 1  
 1.1 Environmental Technology Verification (ETV) Program Purpose and Operation..... 1  
 1.2 Development of Test/Quality Assurance (QA) Plan..... 1  
     1.2.1 Bacteria and Virus Surrogates ..... 2  
     1.2.2 Inlet Pressure ..... 3  
     1.2.3 Long-Term Conditioning..... 3  
 1.3 Testing Participants and Responsibilities ..... 3  
     1.3.1 NSF International..... 4  
     1.3.2 EcoWater Systems Incorporated..... 4  
     1.3.3 U.S. Environmental Protection Agency..... 4

Chapter 2 Equipment Description ..... 6  
 2.1 RO Membrane Operation ..... 6  
 2.2 Equipment Capabilities..... 6  
 2.3 System Components ..... 6  
 2.4 System Operation..... 8  
 2.5 Equipment Operation Limitations ..... 8  
 2.6 Operation and Maintenance Requirements..... 8

Chapter 3 Methods and Procedures ..... 9  
 3.1 Test Equipment ..... 9  
     3.1.1 Equipment Selection..... 9  
     3.1.2 Test Unit Configuration..... 9  
 3.2 Verification Test Procedure ..... 9  
     3.2.1 Test Rig..... 9  
     3.2.2 Test Rig Sanitization..... 9  
     3.2.3 Test Water..... 10  
         3.2.3.1 Base Water ..... 10  
         3.2.3.2 Bacteria and Virus Challenges..... 11  
     3.2.4 Test Unit Operation ..... 12  
         3.2.4.1 Test Unit Installation ..... 12  
         3.2.4.2 TDS Reduction System Check ..... 12  
         3.2.4.3 Long-Term Conditioning..... 12

3.2.4.4	Challenge Testing .....	13
3.3	Analytical Methods.....	14
3.3.1	Water Quality Analytical Methods .....	14
3.3.2	Microbiology Analytical Methods.....	14
3.3.2.1	Sample Processing, and Enumeration of Viruses .....	14
3.3.2.2	Bacteria Cultivation.....	14
3.3.2.3	Preparation of Bacteria Challenge Suspensions .....	15
3.3.2.4	Sample Processing and Enumeration of Bacteria.....	15
Chapter 4	Results and Discussion.....	16
4.1	TDS Reduction .....	16
4.2	Bacteria Reduction.....	16
4.3	Virus Reduction.....	17
Chapter 5	QA/QC .....	19
5.1	Data Review.....	19
5.2	Test Procedure QA/QC.....	19
5.3	Water Chemistry Analytical Methods QA/QC.....	19
5.4	Microbiology Laboratory QA/QC.....	19
5.4.1	Growth Media .....	19
5.4.2	Bacteria Cell Size .....	20
5.4.3	Sample Processing and Enumeration.....	20
5.5	Sample Handling .....	20
5.6	Documentation.....	20
5.7	Data Quality Indicators .....	20
5.7.1	Representativeness.....	21
5.7.2	Accuracy.....	21
5.7.3	Precision.....	21
5.7.4	Statistical Uncertainty.....	22
5.7.5	Completeness .....	22
5.7.5.1	Completeness Measurements.....	22
5.8	Measurements Outside of the Test/QA Plan Specifications .....	23
5.8.1	Total Chlorine .....	23
5.8.2	pH .....	23
Chapter 6	References.....	24

**Appendices**

Appendix A Virus and Bacteria Reduction Data  
 Appendix B QA/QC Measurements  
 Appendix C NSF Drinking Water Treatment Systems Laboratory and Chemistry Laboratory Bench Sheets  
 Appendix D Microbiology Laboratory Bench Sheets  
 Appendix E NSF Testing Laboratory Reports

**List of Tables**

Table 1-1. Virus and Host ATCC Designations .....2  
 Table 3-1. Challenge Schedule ..... 13  
 Table 4-1. Short-Term TDS Reduction Test Results..... 16  
 Table 4-2. Bacteria Log Reduction Data for Unconditioned Units ..... 17  
 Table 4-3. Bacteria Log Reduction Data for Conditioned Units ..... 17  
 Table 4-4. Virus Log Reduction Data for Unconditioned Units ..... 18  
 Table 4-5. Virus Log Reduction Data for Conditioned Units ..... 18  
 Table 5-1. Completeness Requirements .....22

**List of Figures**

Figure 2-1. Photograph of the Ultrafilter 500 .....7  
 Figure 2-2. Schematic Diagram of the Ultrafilter 500.....7  
 Figure 3-1. Schematic Diagram of Test Rig ..... 10  
 Figure 3-2. Test Units Installed on Test Rig..... 11

## Abbreviations and Acronyms

ANSI	American National Standards Institute
ASTM	American Society of Testing Materials
ATCC	American Type Culture Collection
°C	Degrees Celsius
CFU	Colony Forming Unit
cm	Centimeter
DWS	Drinking Water Systems
EPA	U. S. Environmental Protection Agency
ETV	Environmental Technology Verification
°F	Degrees Fahrenheit
L	Liter
mg	Milligram
mL	Milliliter
nm	Nanometer
NRMRL	National Risk Management Research Laboratory
NSF	NSF International (formerly known as National Sanitation Foundation)
PBDW	Phosphate-Buffered Dilution Water
PFU	Plaque Forming Unit
POU	Point-of-Use
psig	Pounds per Square Inch Gauge
QA	Quality Assurance
QC	Quality Control
QA/QC	Quality Assurance/Quality Control
RO	Reverse Osmosis
SOP	Standard Operating Procedure
TDS	Total Dissolved Solids
TSA	Tryptic Soy Agar
TSB	Tryptic Soy Broth
µg	Microgram
µl	Microliter
µm	Micrometer
µmho	Micromho
µS	MicroSieman



## Acknowledgments

NSF International was responsible for all elements in the testing sequence, including collection of samples, calibration and verification of instruments, data collection and analysis, data management, data interpretation and the preparation of this report.

The Manufacturer of the Equipment was:

EcoWater Systems Incorporated  
1890 Woodlane Drive  
Woodbury, MN 55125

NSF wishes to thank the members of the expert technical panel for their assistance with development of the test plan.

## Chapter 1 Introduction

### 1.1 Environmental Technology Verification (ETV) Program Purpose and Operation

The U.S. Environmental Protection Agency (EPA) has created the ETV Program to facilitate the deployment of innovative or improved environmental technologies through performance verification and dissemination of information. The goal of the ETV Program is to further environmental protection by accelerating the acceptance and use of improved and more cost-effective technologies. ETV seeks to achieve this goal by providing high-quality, peer-reviewed data on technology performance to those involved in the design, distribution, permitting, purchase, and use of environmental technologies.

ETV works in partnership with recognized standards and testing organizations; with stakeholder groups consisting of buyers, vendor organizations, and permittees; and with the full participation of individual technology developers. The program evaluates the performance of innovative technologies by developing test plans that are responsive to the needs of stakeholders; by conducting field or laboratory testing, collecting and analyzing data; and by preparing peer-reviewed reports. All evaluations are conducted in accordance with rigorous quality assurance protocols to ensure that data of known and adequate quality are generated and that the results are defensible.

The EPA has partnered with NSF International (NSF) under the ETV Drinking Water Systems (DWS) Center to verify performance of drinking water treatment systems that benefit the public and small communities. It is important to note that verification of the equipment does not mean the equipment is “certified” by NSF or “accepted” by EPA. Rather, it recognizes that the performance of the equipment has been determined and verified by these organizations under conditions specified in ETV protocols and test plans.

### 1.2 Development of Test/Quality Assurance (QA) Plan

As part of the national Homeland Security effort, NSF has developed a test/QA plan under the EPA ETV Program for evaluating point-of-use (POU) reverse osmosis (RO) drinking water treatment systems for removal of biological contamination agents. This test/QA plan uses surrogate bacteria and viruses in place of testing with the actual agents of concern. The test organisms serve as surrogates not only for bacteria and viruses, but also protozoa, such as *Cryptosporidium* oocysts. Please note that this test plan does not cover chemical agents derived from microorganisms, such as ricin or botulism toxin.

To assist in this endeavor, NSF assembled an expert technical panel, which recommended the experimental design and surrogate choices prior to the initiation of testing. Panel members included experts from the EPA, United States Army, and United States Centers for Disease Control and Prevention, Division of Parasitic Diseases, as well as a water utility microbiologist,

a university professor, and an independent consultant in the POU drinking water treatment systems industry.

By participating in this ETV, vendors obtain EPA and NSF verified third-party test data indicating potential user protection against intentional biological contamination of potable water. POU RO systems are not typically marketed as water purifiers that remove bacteria and viruses from drinking water, but they may still remove significant numbers of the microorganisms, thus offering the user a significant level of protection. The verifications serve to notify the public of the possible level of protection against biological contamination agents afforded to them by the use of verified systems.

The test/QA plan called for testing ten Ultrafilter 500 units with a standard test water set to pH 6, 7.5, and 9, containing bacterial or viral surrogates. The systems were also challenged at both 40 and 80 pounds per square inch gauge (psig). The test units were subjected to challenge scenarios that were unique combinations of the challenge organisms, pH, and inlet water pressure. Five units were challenged immediately after completion of the manufacturer’s installation and conditioning instructions, while the other five underwent a 25-day conditioning period prior to being challenged with the surrogates.

**1.2.1 Bacteria and Virus Surrogates**

The expert technical panel recommended that NSF and the EPA use the bacteria *Brevundimonas diminuta* (American Type Culture Collection (ATCC) strain 19146, formerly *Pseudomonas diminuta*), and *Hydrogenophaga pseudoflava* (ATCC strain 33668) as surrogates for bacterial agents. These surrogates were chosen based on their small sizes, as the smallest identified bacterium of concern can be as small as 0.2 μm in diameter. *H. pseudoflava* has a minimum diameter of 0.1 to 0.2 μm, while *B. diminuta* has a minimum diameter of 0.2 to 0.3 μm (please note that these minimum diameters were not obtained during this study. See section 5.4.2 for discussion). *B. diminuta* is the accepted bacteria of choice for testing filters and membranes designed to remove bacteria. It is used in the American Society of Testing Materials (ASTM) “Standard Test Method for Retention Characteristics of 0.2-μm Membrane Filters Used in Routine Filtration Procedures for the Evaluation of Microbiological Water Quality” (2001).

The virus surrogates were the bacteriophages MS2, Phi X 174, and fr. The ATCC designation and host *E. coli* strain for each virus is given Table 1-1.

**Table 1-1. Virus and Host ATCC Designations**

Virus	ATCC Designation	Host Bacteria ATCC Strain
MS2	ATCC 15597-B1	<i>E. coli</i> ATCC 15597
Phi X 174	ATCC 13706-B1	<i>E. coli</i> ATCC 13706
fr	ATCC 15767-B1	<i>E. coli</i> ATCC 19853

The expert technical panel recommended these viruses based on their small sizes and isoelectric points. The isoelectric point is the pH at which the virus surface is neutrally charged. MS2 is 24 nm in diameter with an isoelectric point at pH 3.9, Phi X 174 is 27 nm in diameter with an isoelectric point at pH 6.6, and fr is 19 nm in diameter with an isoelectric point at pH 8.9. With varying isoelectric points, the viruses have different surface charges, or different strengths of negative or positive charge, depending on the pH. In solutions above the isoelectric point, the virus is negatively charged. Below the isoelectric point, the virus is positively charged. Using different pH settings for the virus challenges allowed an evaluation of whether electrostatic forces enhance virus retention in mechanical filtration scenarios. The pH 6 and 9 settings were chosen because they just are beyond the upper and lower boundaries for allowable pH in the EPA National Secondary Drinking Water Regulations. The pH 7.5 setting was chosen because it is the midpoint between the boundaries.

The bacteria reduction challenges were performed only at pH 7.5, because the expert panel believed that bacteria cell size and mass are too large for electrostatic interactions to play a significant role.

### **1.2.2 Inlet Pressure**

The bacteria and virus challenge tests were performed at dynamic inlet pressures of both 40 and 80 psig to evaluate whether inlet pressure affects microorganism rejection by RO membranes. Forty psig is a worse case scenario for ionic rejection mechanisms, while 80 psig represents a poorer mechanical filtration scenario. In a traditional mechanical filtration scenario, the higher pressure could push suspended particles further into, and perhaps all the way through, the filtration media, and it could also distort seals to the point that they leak. However, this may or may not be the case with RO membranes, since they operate by a different principle.

### **1.2.3 Long-Term Conditioning**

The expert technical panel was presented with anecdotal evidence that RO membrane performance could be erratic for approximately the first month of operation, so they recommended that NSF split the test units into two groups, one group to be tested immediately after installation and completion of the manufacturer's conditioning instructions (hereafter referred to as "unconditioned units"), and a second group to be tested after a 25 working day conditioning period (hereafter referred to as "conditioned units").

## **1.3 Testing Participants and Responsibilities**

The ETV testing of the Sears Kenmore Ultrafilter 500 was a cooperative effort between the following participants:

NSF  
EcoWater Systems, Inc.  
EPA

The following is a brief description of each of the ETV participants and their roles and responsibilities.

### **1.3.1 NSF International**

NSF is a not-for-profit organization dedicated to public health and safety, and to protection of the environment. Founded in 1946 and located in Ann Arbor, Michigan, NSF has been instrumental in the development of consensus standards for the protection of public health and the environment. The EPA partnered with NSF to verify the performance of drinking water treatment systems through the EPA's ETV Program.

NSF performed all verification testing activities at its Ann Arbor location. NSF prepared the test/QA plan, performed all testing, managed, evaluated, interpreted, and reported on the data generated by the testing, and reported on the performance of the technology.

#### Contact Information:

NSF International  
789 N. Dixboro Road  
Ann Arbor, MI 48105  
Phone: 734-769-8010  
Fax: 734-769-0109  
Contact: Bruce Bartley, Project Manager  
Email: bartley@nsf.org

### **1.3.2 EcoWater Systems Incorporated**

The verified system is manufactured by EcoWater Systems, a manufacturer of residential and commercial water treatment products.

The manufacturer was responsible for supplying the RO systems in accordance with the equipment selection criteria given in section 3.1.1, and for providing logistical and technical support as needed.

#### Contact Information:

EcoWater Systems Incorporated  
1890 Woodland Drive  
Woodbury, MN 55125  
Contact Person: Ann Baumann  
Phone: 1-800-808-9899

### **1.3.3 U.S. Environmental Protection Agency**

The EPA, through its Office of Research and Development, has financially supported and collaborated with NSF under Cooperative Agreement No. R-82833301. This verification effort

was supported by the DWS Center operating under the ETV Program. This document has been peer-reviewed, reviewed by the EPA, and recommended for public release.

## Chapter 2 Equipment Description

### 2.1 RO Membrane Operation

Membrane technologies are among the most versatile water treatment processes with regard to their ability to effectively remove the widest variety of contaminants at the lowest costs. Reverse osmosis membranes operate by the principal of cross-flow filtration. In this process, the influent water flows over and parallel to the filter medium and exits the system as reject water. Under pressure, a portion of the water diffuses through the membrane becoming “permeate”. Membrane pore sizes are small enough to reject bacteria and viruses, but they may still pass through imperfections in the membrane, or go around the membrane due to microscopic seal leaks.

### 2.2 Equipment Capabilities

The Ultrafilter 500 is certified by NSF International to NSF/ANSI Standard 58 – *Reverse Osmosis Drinking Water Treatment Systems*. It has a certified production rate of 12 gallons per day, and produces five gallons of wastewater per gallon of treated water. These measurements are based on system operation at 50 psig inlet pressure, a water temperature of 77 °F, and a total dissolved solids (TDS) level of 750 mg/L. The amount and quality of treated water produced varies depending on the inlet pressure, water temperature, and level of TDS. These measurements were not subject to verification during this study.

### 2.3 System Components

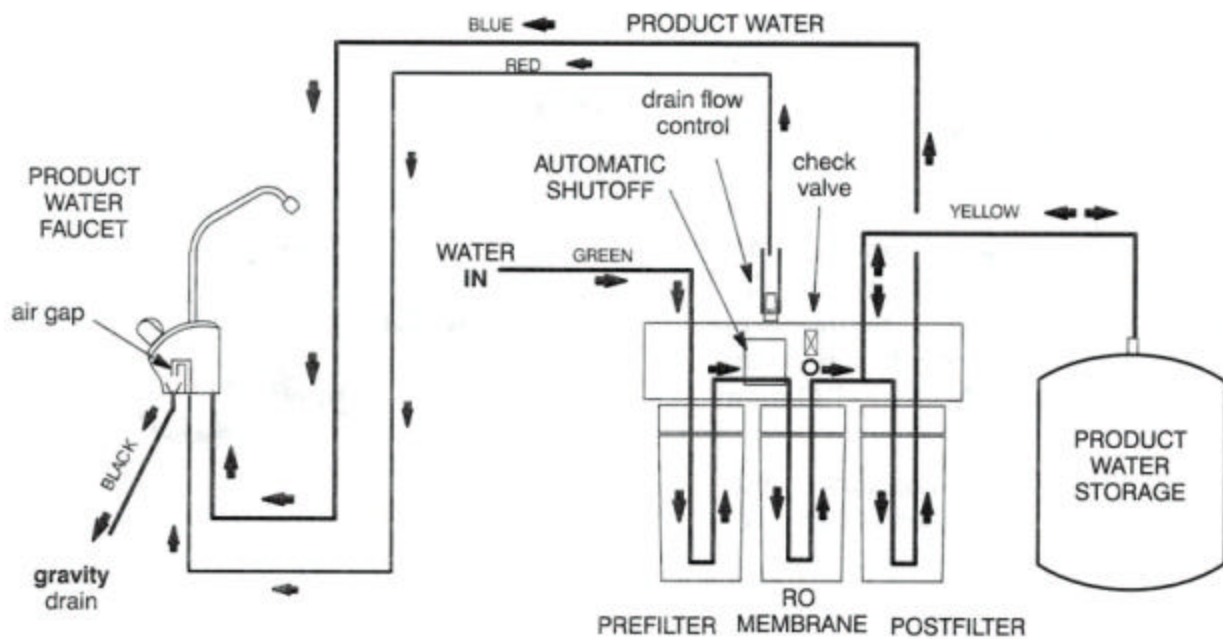
The Ultrafilter 500 is a three-stage treatment system. Incoming water first passes through a carbon filter designed to remove chlorine and particulate matter such as dirt, silt, and rust. The second stage of treatment is the reverse osmosis membrane, which reduces a wide variety of contaminants. The permeate water is sent to a 2.3-gallon maximum capacity storage tank. Upon leaving the storage tank, the water passes through a second carbon filter to remove any remaining tastes and odors, then out through the faucet. Figure 2-1 is a photograph of the system, and Figure 2-2 is a schematic diagram of the system showing the path of water flow.

Please note that this description, and the system operation description in section 2.4 are given for informational purposes only. This information was not subject to verification.

Figure 2-1. Photograph of the Ultrafilter 500



Figure 2-2. Schematic Diagram of the Ultrafilter 500





## 2.4 System Operation

When the flow of water into the system is started, treated water will be continually produced until the storage tank is nearly full. At that time, the water pressure in the tank causes an automatic shut-off device to activate, stopping the flow of water through the system. After approximately ten percent of the treated water is dispensed from the storage tank, the shut-off device deactivates, allowing water to again flow into the system until the storage tank is nearly full. The operational storage tank capacity will vary slightly from unit to unit, and is also affected by the inlet water pressure, but is approximately two gallons under normal use conditions.

The Ultrafilter 500 has a combination volume and TDS level meter that measures the volume of treated water produced and the amount of TDS in the treated water. The faucet has a three colored indicator light to tell the user when to replace the carbon filters and RO membrane. Under normal operation, the indicator light is green. After six months have passed, or 750 gallons of treated water has been produced, the light changes to amber, indicating that the carbon filters need to be replaced. The light turns red when the TDS level in the treated water has risen to above a certain level. At this point the RO membrane should be replaced. The user must reset the meter each time any treatment elements are replaced.

## 2.5 Equipment Operation Limitations

EcoWater Systems gives the following limitations for the drinking water to be treated by the system:

- temperature of 40 – 100 °F;
- pressure of 40 – 100 psig;
- pH of 4 – 10;
- maximum TDS level of 2,000 mg/L;
- maximum water hardness of 10 grains per gallon (1 grain per gallon equals 17.1 mg/L of TDS, expressed as calcium carbonate equivalent);
- no detectible iron, manganese, or hydrogen sulfide; and
- maximum chlorine level of 2 mg/L.

## 2.6 Operation and Maintenance Requirements

The following are the operation and maintenance requirements specified in the product owner's manual:

- Replacement of the carbon filters when indicated by the meter (every six months or 750 gallons);
- Replacement of the RO membrane cartridge when indicated by the meter; and
- Sanitization of the system when the carbon filters or RO membrane are replaced (instructions included in the owner's manual.)

## **Chapter 3 Methods and Procedures**

### **3.1 Test Equipment**

#### **3.1.1 Equipment Selection**

Equipment selection criteria were developed to ensure that the test units were representative of product variability. The test/QA plan called for EcoWater Systems to supply ten units from three different production runs, with RO membranes from three different lots, if possible. At the time of testing, EcoWater Systems had units and RO membranes from only one lot in inventory, so the test units were randomly split into two groups of five.

#### **3.1.2 Test Unit Configuration**

The Ultrafilter 500 was tested with only the RO membrane in place. The carbon filters do not have pore sizes small enough to remove bacteria or viruses, but they could temporarily retain significant numbers of the organisms through electrostatic interactions, giving a positive bias to the performance data. Empty filter cartridges were used in place of the carbon filters. Otherwise the systems were operated as sold to the consumer.

### **3.2 Verification Test Procedure**

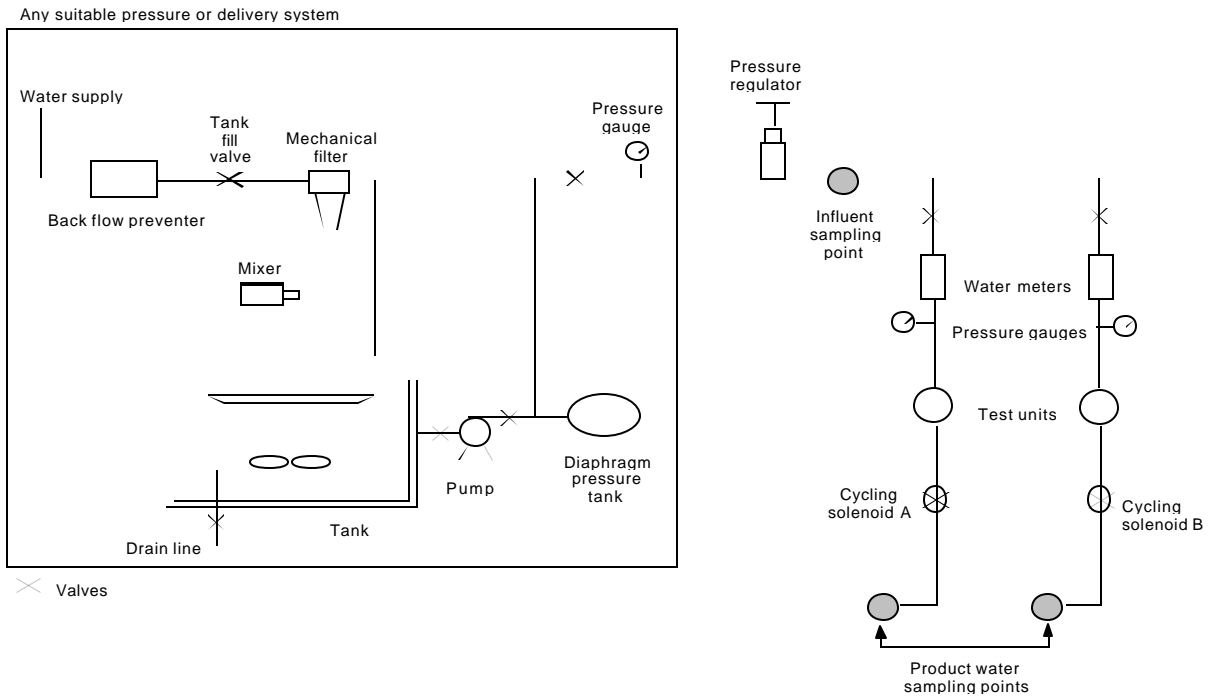
#### **3.2.1 Test Rig**

Each group of five test units was plumbed to a single test station. The test station used a 500-gallon polyethylene tank to hold the influent challenge water. See Figure 3-1 for a schematic diagram of the test rig. Please note that the units of each group of five were attached to the rig, such that all were plumbed to the same influent feed line. Figure 3-2 shows one group of the test units installed on the test rig.

#### **3.2.2 Test Rig Sanitization**

The test apparatus was sanitized with a sanitization agent prior to the beginning of each test to keep the heterotrophic bacteria population to a minimum. After sanitization, the test apparatus was flushed until a less-than-detectable concentration of sanitizing agent was present.

**Figure 3-1. Schematic Diagram of Test Rig**



### 3.2.3 Test Water

#### 3.2.3.1 Base Water

Ann Arbor, Michigan municipal drinking water was deionized to make the base water for the tests. The base water had the following constraints:

- Conductivity  $\leq 2 \mu\text{S}/\text{cm}$  at  $25 \text{ }^\circ\text{C}$ ;
- TOC  $< 100 \mu\text{g}/\text{L}$ ; and
- Heterotrophic bacteria plate count  $< 100$  colony forming units (CFU)/mL.

The base water was then adjusted to meet the following characteristics:

- Total chlorine =  $0.05 \text{ mg}/\text{L}$ ;
- Addition of sodium bicarbonate to achieve an alkalinity (expressed as calcium carbonate) of  $100 \pm 5 \text{ mg}/\text{L}$  prior to pH adjustment;
- pH adjustment with hydrochloric acid or sodium hydroxide to reach a value of  $6.0 \pm 0.5^*$ ,  $7.5 \pm 0.5$ , or  $9.0 \pm 0.5$  as required by challenge protocol; and
- Temperature of  $20 \pm 2.5 \text{ }^\circ\text{C}$ .

\*Note that the lab technicians experienced difficulty maintaining the pH below 6.5. As a result, the challenge water pH for three of the four pH 6.0 challenges was between 6.5 and 7.0. See Section 5.8.2 for more discussion.

The test water was made daily in 200-gallon volumes. In addition to the above characteristics, total hardness, TDS, and turbidity were measured daily.

**Figure 3-2. Test Units Installed on Test Rig**



### 3.2.3.2 Bacteria and Virus Challenges

The viruses were purchased from Biological Consulting Services of North Florida, and the bacteria from ATCC. The viruses were purchased in adequate volumes so that the suspensions received were added directly to the test water. The bacteria were cultivated at NSF to obtain the challenge suspensions. Section 3.3.2.3 describes the method used to create the bacteria challenges.

The targeted influent challenge concentrations for the bacteria were  $1 \times 10^5$  CFU of bacteria per 100 milliliters, or greater. This target was exceeded for the *B. diminuta* challenges, but not for the *H. pseudoflava* challenges during the first attempt. The influent sample analyses from the *H. pseudoflava* challenges for both the unconditioned and conditioned groups were less than  $1 \times 10^5$

CFU/100mL. This may have been due to the use of a bad batch of growth media, or a non-viable or stressed challenge suspension. The *H. pseudoflava* challenges were conducted again after all other challenges were complete. The influent CFU counts for the second challenges were above the  $1 \times 10^5$  CFU/100mL target.

The target influent concentration for the viruses was  $1 \times 10^4$  plaque forming units (PFU) of virus per milliliter, or greater. Phi X 174 is more difficult to cultivate, and so was sometimes supplied at lower concentrations than the other viruses. As a result, four of the virus challenge influents did not meet the target concentration for Phi X 174. Assuring Phi X 174 influents greater than  $1 \times 10^4$  PFU/mL would have been prohibitively expensive, due to the high cost of the virus per liter.

See Appendix A for all influent challenge levels.

The test units were challenged with each bacteria separately, but all three viruses were mixed together for each virus challenge. After addition of the challenge organism to the base test water, the resultant challenge water was mixed for a minimum of 30 minutes using a recirculation pump prior to beginning the test.

### **3.2.4 Test Unit Operation**

#### **3.2.4.1 Test Unit Installation**

All test units were installed on the test rigs by a laboratory technician. Immediately after installation, the units were conditioned according to the vendor's instructions using the base test water at  $\text{pH } 7.5 \pm 0.5$ . The conditioning instructions call for operation for six tank-filling periods. At the end of the conditioning procedure, an effluent sample was collected from each unit as a negative control and analyzed for the challenge organisms.

#### **3.2.4.2 TDS Reduction System Check**

After completion of the vendor's conditioning procedure, the test units underwent a one-day TDS reduction test using the test protocol in NSF/ANSI Standard 58. The Standard 58 test protocol was modified so that the units were operated continuously for one tank-fill period. Product water samples were then collected from each storage tank and analyzed for TDS. This test ensured that the products undergoing verification testing were representative of the expected performance of the system, and that there were no membrane integrity or membrane seal problems.

#### **3.2.4.3 Long-Term Conditioning**

After the TDS reduction system check test, the five units receiving long-term conditioning were operated using the test water without surrogate organisms for a period of 25 working days prior to challenge testing. On each day the units were operated continuously at a dynamic inlet pressure of  $80 \pm 3$  psig for one tank-fill period. The units then sat idle overnight under pressure, and the tanks were emptied the next morning prior to resumption of unit operation.

### 3.2.4.4 Challenge Testing

Following the conditioning period, the conditioned units were challenged according to the schedule in Table 3-1. Prior to the start of challenge testing for this group, the test rig was sanitized again as described in section 3.2.2. The test units were taken off-line to prevent sanitizer from entering them, and the test rig was flushed free of sanitizer before they were reconnected to the rig.

**Table 3-1. Challenge Schedule**

Day	Surrogate Challenge	pH	Inlet Pressure (psig)
1	<i>H. pseudoflava</i>	7.5 ± 0.5	40 ± 3
2	<i>H. pseudoflava</i>	7.5 ± 0.5	80 ± 3
3	<i>B. diminuta</i>	7.5 ± 0.5	40 ± 3
4	<i>B. diminuta</i>	7.5 ± 0.5	80 ± 3
5	All Viruses	6.0 ± 0.5	40 ± 3
6	All Viruses	6.0 ± 0.5	80 ± 3
7	All Viruses	7.5 ± 0.5	40 ± 3
8	All Viruses	7.5 ± 0.5	80 ± 3
9	All Viruses	9.0 ± 0.5	40 ± 3
10	All Viruses	9.0 ± 0.5	80 ± 3

Challenge testing for the unconditioned units began the day after the TDS system check test. Testing for this group also followed the schedule in Table 3-1.

At the end of the workday before each challenge, the base test water was prepared as described in section 3.2.3.1. The morning of the challenge, the pH was checked and adjusted, if necessary, and the bacteria or viruses were added as described in section 3.2.3.2.

The dynamic inlet water pressure for operation was set at either 40 ± 3 or 80 ± 3 psig according to the challenge schedule.

An influent sample was collected each day at the time test unit operation started. Each test unit was then operated continuously for one tank-fill period. At 40 psig, approximately 1.5 gallons of treated water were produced, while at 80 psig, approximately 2 gallons were produced. The time of operation was approximately two hours at both pressures.

After each unit shut off, its storage tank was emptied into a sterile container, and a sub-sample was collected for challenge organism enumeration. The sub-sample volumes were 1.0 L for the bacteria challenges, and 150 mL for virus challenges. A second influent sample was collected after all units ceased operation. All samples were collected in sterile polypropylene bottles, and were enumerated in triplicate.

Following each day's challenge period, the systems were operated for one tank-fill period using the test water without any test organisms present. This served to flush the systems in-between

challenge periods. The units rested overnight under pressure, and the storage tanks were emptied the next morning prior to initiation of that day's challenge period.

### 3.3 Analytical Methods

#### 3.3.1 Water Quality Analytical Methods

The following are the analytical methods used during verification testing. All analyses followed procedures detailed in NSF Standard Operating Procedures (SOPs).

- pH – All pH measurements were made with an Orion Model SA 720 meter. The meter was operated according to the manufacturer's instructions, which are based on Standard Method 4500-H<sup>+</sup>.
- Temperature – Water temperature was measured using an Omega model HH11 digital thermometer.
- TDS – TDS for the TDS reduction system check test was measured through conductivity according to Standard Method 2510 using a Fisher Scientific Traceable<sup>TM</sup> Conductivity Meter.
- Total Chlorine – Total chlorine was measured according to Standard Method 4500-Cl G with a Hach Model DR/2010 spectrophotometer using AccuVac vials.

#### 3.3.2 Microbiology Analytical Methods

##### 3.3.2.1 Sample Processing, and Enumeration of Viruses

The viruses were enumerated using a double agar layer method published in *NSF/ANSI Standard 55 – Ultraviolet Microbiological Water Treatment Systems* for enumerating MS2. This method is similar to the double agar layer method in EPA Method 1601.

Four to eighteen hours prior to sample processing, 100 µL of the appropriate host *E.coli* suspension was pipetted into tubes containing 10 mL of fresh Tryptic Soy Broth (TSB), and incubated at 35 °C. After incubation, 100 µL volumes of the resulting *E. coli* culture were transferred to sterile, capped test tubes.

All samples were enumerated in triplicate. All samples were serially diluted for enumeration, and the effluent samples were also enumerated directly. One milliliter volumes of the sample or dilution were pipetted into the *E. coli* suspension test tubes. The tubes were vortexed for a minimum of 30 seconds to “mate” the bacteria and virus, and then 4 mL of molten, tempered TSB plus 1% agar was added to each tube. These mixtures were then poured over Tryptic Soy Agar (TSA) plates, and allowed to solidify. The plates were incubated at 35 °C for 18-24 hours. Viral plaques were counted using a Quebec Colony Counter.

##### 3.3.2.2 Bacteria Cultivation

The bacteria were purchased from ATCC and rehydrated with nutrient broth. After 48 hours of incubation at 30 °C, tubes containing 10 mL of TSB were inoculated with 100 µL of the nutrient broth suspension. These tubes were incubated for 48 hours at 30 °C. After this incubation

period, 100 µL of these suspensions were pipetted into new tubes containing 10 mL of fresh TSB. These tubes were then also incubated for 48 hours at 30 °C. This process was repeated at least three times, up to a maximum of 30 times.

### 3.3.2.3 Preparation of Bacteria Challenge Suspensions

To obtain the challenge suspensions, 1 mL of a 48-hour TSB culture was pipetted onto an appropriate number of TSA slants. The slants were inoculated at 30 °C for 48 hours. When a challenge suspension was needed, 5 mL of sterile phosphate-buffered dilution water (PBDW) was pipetted onto the slants, and the agar surfaces were scraped to suspend the cells. The overlying water was then pipetted out of the slants into an appropriate volume of PBDW. The resulting challenge suspension was vortexed for approximately 30 seconds to disperse the cells. The challenge suspensions were refrigerated and added to the tank of test water within one hour. Samples of the challenge suspension were collected and enumerated according to the method in 3.3.2.4.

### 3.3.2.4 Sample Processing and Enumeration of Bacteria

All samples were enumerated in triplicate using a membrane filtration method based on Standard Method 9215 D. All samples were serially diluted for enumeration with sterile PBDW, and the effluent samples were also enumerated directly. One-milliliter volumes of either the sample or dilution were pipetted into sterile vacuum filtration apparatuses, 25 mL of PBDW added, and the suspension vacuum filtered through sterile 0.1 µm membrane filters. The funnels were then rinsed three times with approximately 5 mL of PBDW, and the rinse water also suctioned through the filters. The membrane filters were aseptically removed from the apparatuses and placed onto R2A agar plates. The plates were incubated at 30 °C for 48 hours. Characteristic *B. diminuta* or *H. pseudoflava* colonies were counted with a Quebec Colony Counter.



## Chapter 4 Results and Discussion

### 4.1 TDS Reduction

The performance data from the TDS reduction system check test described in 3.2.4.2 are presented in Table 4-1. EcoWater Systems’s reported average TDS reduction performance for the Ultrafilter 500 is 97%, so the units tested are representative of expected membrane performance.

**Table 4-1. Short-Term TDS Reduction Test Results**

Unconditioned Units			Conditioned units		
	TDS (mg/L)	Percent Reduction		TDS (mg/L)	Percent Reduction
Influent	796		Influent	795	
Effluents:			Effluents:		
Unit 1	64	92	Unit 1	51	94
Unit 2	61	92	Unit 2	44	94
Unit 3	62	92	Unit 3	34	96
Unit 4	63	92	Unit 4	44	94
Unit 5	58	93	Unit 5	20	97

### 4.2 Bacteria Reduction

Presented in Tables 4-2 and 4-3 are the log<sub>10</sub> reduction data for the bacteria challenge portion of the verification test. The influent and effluent bacteria count and log<sub>10</sub> reduction data for each individual test unit is given in Appendix A. The triplicate influent and effluent counts in Appendix A were averaged by calculating geometric means. The means were then log<sub>10</sub> transformed and log<sub>10</sub> reduction values were calculated for each test unit. The tables below give the log<sub>10</sub> reduction performance data for each pH and inlet pressure combination.

In all challenges for both sets of test units the bacteria were removed to less than detectable levels (< 1 CFU/100mL). The variables of pH and pressure had no discernible impact on the performance of the test units. There was also no measurable difference in performance between the unconditioned units and conditioned units.

**Table 4-2. Bacteria Log Reduction Data for Unconditioned Units**

pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub>	Geometric Mean Log <sub>10</sub> Reduction				
			Influent Challenge	Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
7.5	40	<i>H. pseudoflava</i>	6.6	6.6	6.6	6.6	6.6	6.6
		<i>B. diminuta</i>	6.4	6.4	6.4	6.4	6.4	6.4
7.5	80	<i>H. pseudoflava</i>	5.9	5.9	5.9	5.9	5.9	5.9
		<i>B. diminuta</i>	6.6	6.6	6.6	6.6	6.6	6.6

**Table 4-3. Bacteria Log Reduction Data for Conditioned Units**

pH	Pressure (psig)	Challenge Organisms	Log <sub>10</sub>	Geometric Mean Log <sub>10</sub> Reduction				
			Influent Challenge	Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
7.5	40	<i>H. pseudoflava</i>	6.6	6.6	6.6	6.6	6.6	6.6
		<i>B. diminuta</i>	7.1	7.1	7.1	7.1	7.1	7.1
7.5	80	<i>H. pseudoflava</i>	5.9	5.9	5.9	5.9	5.9	5.9
		<i>B. diminuta</i>	6.8	6.8	6.8	6.8	6.8	6.8

### 4.3 Virus Reduction

The virus log<sub>10</sub> reduction data for each challenge scenario are presented in Tables 4-4 and 4-5. The influent and effluent virus PFU count and log<sub>10</sub> reduction data for each individual test unit are given in Appendix A. As was done for the bacteria, the triplicate influent and effluent counts were averaged by calculating geometric means. The means were then log<sub>10</sub> transformed and log<sub>10</sub> reduction values calculated for each test unit.

In all challenges but the pH 9, 80 psig challenge, both sets of test units removed all three viruses to less than detectable levels (< 1 PFU/mL). The maximum mean effluent count for the pH 9, 80 psig challenges was 11 PFU/mL (3.0 log<sub>10</sub> fr reduction for unconditioned unit 3). As with the bacteria, the prevalence of undetectable virus effluent counts does not allow an evaluation of the effect of conditioning, inlet pressure, or pH on RO membrane performance.

**Table 4-4. Virus Log Reduction Data for Unconditioned Units**

Challenge Conditions			Challenge Organisms	Log <sub>10</sub> Influent Challenge	Geometric Mean Log <sub>10</sub> Reduction				
Target pH	Actual pH	Pressure (psig)			Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
6.0 ± 0.5	6.86	40	fr	5.0	5.0	5.0	5.0	5.0	5.0
			MS2	4.8	4.8	4.8	4.8	4.8	4.8
			Phi X 174	4.5	4.5	4.5	4.5	4.5	4.5
6.0 ± 0.5	6.88	80	fr	5.4	5.4	5.4	5.4	5.4	5.4
			MS2	5.2	5.2	5.2	5.2	5.2	5.2
			Phi X 174	4.0	4.0	4.0	4.0	4.0	4.0
7.5 ± 0.5	7.69	40	fr	4.3	4.3	4.3	4.3	4.3	4.3
			MS2	5.0	5.0	5.0	5.0	5.0	5.0
			Phi X 174	5.3	5.3	5.3	5.3	5.3	5.3
7.5 ± 0.5	7.91	80	fr	4.0	4.0	4.0	4.0	4.0	4.0
			MS2	4.9	4.9	4.9	4.9	4.9	4.9
			Phi X 174	4.4	4.4	4.4	4.4	4.4	4.4
9.0 ± 0.5	8.71	40	fr	5.3	5.3	5.3	5.3	5.3	5.3
			MS2	5.0	5.0	5.0	5.0	5.0	5.0
			Phi X 174	4.4	4.4	4.4	4.4	4.4	4.4
9.0 ± 0.5	8.67	80	fr	4.1	3.8	3.6	3.0	4.1	4.1
			MS2	3.9	3.9	3.6	2.9	3.9	3.9
			Phi X 174	3.7	3.7	3.7	3.7	3.7	3.7

**Table 4-5. Virus Log Reduction Data for Conditioned Units**

Challenge Conditions			Challenge Organisms	Log <sub>10</sub> Influent Challenge	Geometric Mean Log <sub>10</sub> Reduction				
Target pH	Actual pH	Pressure (psig)			Unit 1	Unit 2	Unit 3	Unit 4	Unit 5
6.0 ± 0.5	6.48	40	fr	4.8	4.8	4.8	4.8	4.8	4.8
			MS2	4.5	4.5	4.5	4.5	4.5	4.5
			Phi X 174	3.8	3.8	3.8	3.8	3.8	3.8
6.0 ± 0.5	6.69	80	fr	4.5	4.5	4.5	4.5	4.5	4.5
			MS2	4.4	4.4	4.4	4.4	4.4	4.4
			Phi X 174	4.2	4.2	4.2	4.2	4.2	4.2
7.5 ± 0.5	7.45	40	fr	5.3	5.3	5.3	5.3	5.3	5.3
			MS2	4.9	4.9	4.9	4.9	4.9	4.9
			Phi X 174	4.3	4.3	4.3	4.3	4.3	4.3
7.5 ± 0.5	7.56	80	fr	4.9	4.9	4.9	4.9	4.9	4.9
			MS2	4.7	4.7	4.7	4.7	4.7	4.7
			Phi X 174	3.9	3.9	3.9	3.9	3.9	3.9
9.0 ± 0.5	8.73	40	fr	5.6	5.6	5.6	5.6	5.6	5.6
			MS2	5.4	5.4	5.4	5.4	5.4	5.4
			Phi X 174	3.8	3.8	3.8	3.8	3.8	3.8
9.0 ± 0.5	8.73	80	fr	5.1	5.1	4.6	5.1	5.1	5.1
			MS2	4.8	4.5	4.3	4.8	4.8	4.5
			Phi X 174	4.5	4.5	4.5	4.5	4.5	4.5

## Chapter 5 QA/QC

### 5.1 Data Review

NSF QA/QC staff reviewed the raw data records for compliance with QA/QC requirements and checked 100% of the data against the reported results in the official laboratory reports.

### 5.2 Test Procedure QA/QC

The test procedure followed an NSF SOP created specifically for this ETV.

### 5.3 Water Chemistry Analytical Methods QA/QC

- pH – Three point calibration at pH 4, 7, and 10 was conducted daily using traceable buffers. The calibration was checked with a pH 8 buffer. The precision of the instrument was checked by collecting a sample of municipal drinking water and splitting it into two samples for pH measurement. The relative percent deviation (RPD) was calculated using the equation in section 5.7.3. The acceptable RPD limit was 10%. The daily pH 8 buffer readings and results of the duplicate analyses are given in Table B-1 of Appendix B.
- Temperature – The digital thermometer is calibrated every six months using a Hart Scientific Model 9105 Dry Well Calibrator.
- Total Chlorine – The spectrophotometer was calibrated daily according to the manufacturer's instructions. The precision of the instrument was checked daily by analyzing a sample of municipal drinking water in duplicate. The samples were diluted by approximately 50% with deionized water, and then split into subsamples for analysis. The RPD for the two samples was then calculated, with an acceptable RPD limit of 10%. The results of the duplicate analyses are given in Table B-3 of Appendix B.
- TDS – Two potassium chloride standards were used for instrument calibration. A third QC standard was then used to check the calibration. Ten percent of samples were analyzed in duplicate, and RPDs were calculated. The acceptable RPD limit was 10%. The calibration check standard measurements and results of the duplicate analyses are given in Table B-2 of Appendix B.

### 5.4 Microbiology Laboratory QA/QC

#### 5.4.1 Growth Media

All media were checked for sterility and positive growth response when prepared and when used for microorganism enumeration. The media was discarded if growth occurred on the sterility check media, or if there was an absence of growth in the positive response check. All three *E. coli* hosts for the viruses were plated on TSA and incubated with the virus enumeration plates during sample enumeration as a second positive growth control. *B. diminuta* and *H. pseudoflava*

from the stock cultures were plated on R2A agar and incubated with the bacteria enumeration plates as positive controls.

#### **5.4.2 Bacteria Cell Size**

The theoretical minimum size for *B. diminuta* and *H. pseudoflava* cells is 0.2 to 0.3  $\mu\text{m}$  in diameter, however, the NSF Microbiology Laboratory was not able to achieve that size. The stock culture was examined microscopically using a stage micrometer, and the observed diameters were approximately 0.5  $\mu\text{m}$ . To achieve the smallest cell size, the bacteria need to be grown in a medium such as Saline Lactose Broth that keeps the cells small due to osmotic pressure constraints. However, this medium is low in nutrients, so the Microbiology Laboratory had difficulty cultivating the bacteria in high titers. The bacteria were instead cultivated in TSB. TSB is more nutrient rich, and as a result yielded larger cells.

The larger cell size may have enhanced the bacteria reduction performance of the test units, so the bacteria reduction data cannot be used to predict expected performance against bacterial agents smaller than 0.5  $\mu\text{m}$ . However, the viruses used in this study are much smaller than any bacteria, so the virus results could be considered indicative of the system's minimum bacteria reduction performance.

#### **5.4.3 Sample Processing and Enumeration**

All samples were enumerated in triplicate. For each sample batch processed, an unused membrane filter and a blank with 100 mL of PBDW filtered through the membrane were also placed onto the appropriate media and incubated with the samples as negative controls. No growth was observed on any blanks.

#### **5.5 Sample Handling**

All samples analyzed by the NSF Microbiology and Wet Chemistry Laboratories were labeled with unique ID numbers. These ID numbers appear on the NSF laboratory reports for the tests. All water chemistry samples were analyzed within allowable hold times. All samples for bacteria and virus analysis were processed within one hour of collection.

#### **5.6 Documentation**

All laboratory activities were documented using laboratory bench sheets and NSF laboratory reports. This documentation can be found in the appendices.

#### **5.7 Data Quality Indicators**

The quality of data generated for this ETV can be established through five indicators of data quality: representativeness, accuracy, precision, statistical uncertainty, and completeness.

### 5.7.1 Representativeness

Representativeness refers to the degree to which the data accurately and precisely represent the expected performance of the RO system under normal use conditions. The test protocol was designed to be a conservative evaluation of product performance. The test water was of very low turbidity to minimize the potential of microbial adhesion to suspended particles, which could enhance apparent log reduction. The surrogates were chosen because of their small size. The virus surrogate challenges were carried out at pH 6, 7.5, and 9 to assess whether pH affects the performance of the RO membrane.

### 5.7.2 Accuracy

Accuracy of the pH meter was evaluated with a pH 8.0 check standard after the daily calibrations. The calibration check measurements were all in the range 7.91 to 8.10.

Accuracy of the conductivity meter used for TDS analysis was measured through the use of QC samples with every batch of samples analyzed. Two batches of samples were analyzed, one for each set of test units. The percent recovery of the QC samples analyzed with these batches was 102% and 103%.

During most of the testing period, the chlorine meter's accuracy was checked by measuring the chlorine level of deionized water samples. The calibration was acceptable if the measured chlorine level was 0.05 mg/L or less. Deionized water was chosen to be the calibration check because the test plan called for the use of deionized water for the test water, and that this water have a chlorine level less than or equal to 0.05 mg/L. Thus, the calibration check measured the accuracy of the meter for the range in which the test water samples fell. Toward the end of the testing period, the testing laboratory began also using three different check standards in addition to deionized water. The readings for the check standards had to be within 10% of the true value for acceptable calibration.

Accuracy check results for these parameters are given in Appendix B.

### 5.7.3 Precision

Precision refers to the degree of mutual agreement among individual measurements and provides an estimate of random error. One sample per batch was analyzed in duplicate for the TDS measurements. Duplicate municipal drinking water samples were analyzed for pH and total chlorine as part of the daily calibration process. Precision of the duplicate analyses was measured by use of the following equation to calculate relative percent deviations (RPD):

$$RPD = \frac{|S_1 - S_2|}{|S_1 + S_2|} \times 200$$

where:

$S_1$  = sample analysis result; and

$S_2$  = sample duplicate analysis result.

The RPD calculations for individual duplicate pairs are given in the tables in Appendix B. The duplicate measurements for the two TDS sample batches gave RPD values of 0.5% and 1.5%. The RPD values for the pH measurements ranged from 0% to 0.88%. The RPD values for the total chlorine measurements ranged from 0% to 5.96%.

**5.7.4 Statistical Uncertainty**

Statistical Uncertainty is expressed using 95% confidence intervals. No confidence interval calculations were made for the performance data because most of the effluent samples contained undetectable concentrations of the challenge organisms, and the sample sizes (triplicate counts) were too small to give very meaningful results.

**5.7.5 Completeness**

Completeness is the proportion of valid, acceptable data generated using each method as compared to the requirements of the test/QA plan. The completeness objective for data generated during verification testing is based on the number of samples collected and analyzed for each parameter and/or method.

**Table 5-1. Completeness Requirements**

Number of Samples per Parameter and/or Method	Percent Completeness
0 – 10	80%
11 – 50	90%
> 50	95%

Completeness is defined as follows for all measurements:

$$\%C = (V/T) \times 100$$

where:

- %C = percent completeness;
- V = number of measurements judged valid; and
- T = total number of measurements.

**5.7.5.1 Completeness Measurements**

- Ten units were tested, as called for in the test/QA plan, giving a completeness measurement of 100% for this category.
- All conditioning water and challenge water samples for pH, temperature, total chlorine, and TDS (by conductivity) were collected as scheduled and analyzed with acceptable results.

- All scheduled bacteria and virus samples were collected and analyzed with acceptable results. As described in Section 3.2.3.2, the influent samples for the *H. pseudoflava* challenges for both groups of test units were considerably less than the target levels. However, these challenges were conducted again at the end of the testing period, with acceptable challenge levels.

## 5.8 Measurements Outside of the Test/QA Plan Specifications

### 5.8.1 Total Chlorine

The test/QA plan called for the test water to have a total chlorine level at or below 0.05 mg/L. On day 24 of the conditioning period for the conditioned units, the influent chlorine level was measured at 0.06 mg/L. This is not a significant deviation from the test plan.

### 5.8.2 pH

The test water chemistry provided little buffering capacity, which made it difficult to keep the pH of the test water within the allowable range ( $\pm 0.5$ ) for the pH 6 and pH 9 challenges. The influent pH readings for the unconditioned units' pH 6, 40 psig, and pH 6, 80 psig challenges, and the conditioned units pH 6, 80 psig challenge were all above the allowable upper limit of pH 6.5. See Tables 4-4 and 4.5 for the pH measurements. These deviations are significant, however, since there is not any effluent count data with which to compare test unit performance at the three different pH values, these deviations do not affect any data analysis.



## Chapter 6 References

American Society of Testing Materials (2001). D 3862-80, Standard Test Method for Retention Characteristics of 0.2- $\mu\text{m}$  Membrane Filters Used in Routine Filtration Procedures for the Evaluation of Microbiological Water Quality.

NSF International (2002). *NSF/ANSI 55-2002, Ultraviolet microbiological water treatment systems*. Ann Arbor, NSF International.

NSF International (2002). *NSF/ANSI 58 – 2002, Reverse osmosis drinking water treatment systems*. Ann Arbor, NSF International.

APHA, AWWA and WPCF (1998). *Standard Methods for Examination of Water and Wastewater*. 20th ed. Washington, D.C. APHA.