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Freeport, Tx 77541
Brad Toups
Air Permits Section (6PD-R)


diS. EPA, Region 6 445 Ross Avenue, Suite 1200 Dallas, Texas 75202

To: Brad Toups, EPA 1/ APR 10 PA 6:03
Re: Greenhouse gas PSD air quality permit for The Dow Chemical Company Light Hydrocarbon No. 9 Plant, Proposed Permit PSD-TX-1328-GHG The proposed project consists of a new ethylene production Facility, consisting of (8) steam cracking Furnaces and related utilities.

I live in Freeport, Tx in Brazoria, county, about 6 miles from the proposed Ethylene Cracker.

The DOW LCH-9 plant is subject to PSD review for the pollutant GHG's, as the project will result in increased greenhouse gas emissions For the Facility.

EPA should address the following comments and questions re: this proposed ethylene cracker
(1) Has EPA looked closely at the multiple chemical industry projects going on in this same area and the cumulative effect of all these projects and present chemical plants operating and emitting greenhouse gases (GAGs)?

* Brazoria $c_{0}$ is currently designated Severe Non-attainment for Ozone.
To: Brad Toups, EPA
Re: GHG PSD Air Permit For the Dow Chemical Company Light Hydrocarbon No 9 Plant, PSD-TX-1328-GHG
(1) $($ continued $)-$

For example, the proposed Freeport LNG Pretreatment Plant which is to be built about $1-2$ miles down the road (Hw/ll emits about $1,5 \mathrm{M}$ tons
of $\mathrm{CO}_{2}$. Please, look at he cummulative effect of adding another project which emits $G H G$ s $-\mathrm{CO}_{2}$. What is the impact on Public Health?
(2) Ask DoW Chemical Co why they now say that it is too expensive to do carbon capure and then pipe the $\mathrm{CO}_{2}$ to Denbury at the Hasting Field for oil recovery. In the Dow Permit Application, dated $11 / 28 / 12, p .21$, stated "if a pipeline was constructed, Denbury Resources owns and operates a $\mathrm{CO}_{2}$ Pipeline that has a terminus point at Hastings Field, and is about 47 mi from a resonable tie-in to DoW Freeport

EPA must base its BACT decision on the average cost etFectivenes (3) of CCS and why Dow wont pipe it?
To: Brad Toups, EPA
Re: $P S D-T X-1328-G H G$
,however, there is no existing connection For the pipeline for Hastings Field - Denbury Green Pipeline. If Dow Freeport wants to decrease $\mathrm{CO}_{2}$ in the air, why don't they look into constructing this pipeline? and monitor all emission points at the Dow Chemical" Ethylene Cracker."?

Plus, Freepott LNG-]also needs to pipe their $\mathrm{CO}_{2}$ to Denbury 50 Why doesn't DoW Chemical and FLNG look into both Funding this Pipeline Connection? What would the Cost be for this pipeline extension? Also, what are the $\mathrm{CO}_{2}$ storage capabilities in Brazoria Co? What would the legal risks be for building infrastructure solely for CO, Storage? Please, explain, to us the citizens of Freeport, $T_{1}$, the Economic Analysis, For the Control Options For the proposed cracking Furnaces at DoW's LHC-9 facility, Thanks i!?

Brazoria County is currently designated severe nonattainment for ozoned and is currently designated attainment for all other pollutants. The nearest Class I area, at a distance of more than 500 kilometers, is Breton National Wildlife Refuge. The plot plan for the Dow Freeport LHC-9 facility is depicted in Attachment A. The facility location map is here:


### 4.1.4 Step 4: Evaluate the Most Effective Controls

### 4.1.4.1 Carbon Capture ond Storage

$\mathrm{CO}_{2}$ capture is a relatively new concept. In its March 2011 PSD permitting guidance for GHGs, EPA takes the position that, "for the purpose of a BACT analysis for GHGs, EPA classifies CCS as an add-on pollution control technology that is "available" for facilities emitting $\mathrm{CO}_{2}$ in large amounts, including fossil fuel-fired power plants, and for industrial facilities with high-purity $\mathrm{CO}_{2}$ streams (e.g., hydrogen production, ammonia production, natural gas processing, ethanol production, ethylene oxide production, cement production, and iron and steel manufacturing). For these types of facilities, CCS should be listed in Step 1 of a top-down BACT analysis for GHGs". (Footnote 1 pg 17)
These emerging carbon capture and storage (CCS) technologies generally consist of processes that separate $\mathrm{CO}_{2}$ from combustion process flue gas, compression of the separated $\mathrm{CO}_{2}$, transportation via pipeline to a site for injection and then inject it into geologic formations such as oil and gas reservoirs, un-mineable coal seams, and underground saline formations.

Of the emerging $\mathrm{CO}_{2}$ capture technologies that have been identified, only amine absorption is currently commercially used for state-of-the-art $\mathrm{CO}_{2}$ separation processes. Amine absorption has been applied to processes in the petroleum refining and natural gas processing industries and for exhausts from furnaces. Other potential absorption and membrane technologies are currently considered developmental.

Dow is evaluating CCS for the proposed project based on technological, environmental, and economic feasibility.

Table 4-1 Technical Feasibility of CCS Technologies

| CCS Component | CCSTEChnology | Techinical Feasibility |
| :---: | :---: | :---: |
| Capture and Compression | Post-Combustion | Y |
|  | Pre-Combustion | N |
|  | Oxy-Fuel Combustion | N |
|  | Industrial Separation (natural gas processing, ammonia production) | N |
| Transportation | Pipeline | Y |
|  | Shipping | Y |
| Geological Storage | Enhanced Oil Recovery (EOR) | Y |
|  | Gas or Oil Fields | N |
|  | Saline Formations | N |
|  | Enhanced Coal Bed Methane Recovery (ECBM) | N |
| Ocean Storage | Direct Injection (Dissolution Type) | N |
|  | Direct Injection (Lake Type) | N |
| Mineral Carbonation | Natural Silicate Minerals | N |
|  | Waste Minerals | N |
| Large Scale $\mathrm{CO}_{2}$ Utilization/Application |  | N |

and the high volume stream would need to be transported via pipeline to a geologic formation capable of long-term storage.

The capabilities for $\mathrm{CO}_{2}$ storage in the vicinity around Freeport are early in development, and there is tenuous commercial viability and demonstration of large-scale, long-term $\mathrm{CO}_{2}$ storage; therefore, the capital and legal risks of building infrastructure solely for $\mathrm{CO}_{2}$ storage from this LHC-9 project are unreasonable. (However if a pipeline was constructed, Danbury Resources owns and operates a $\mathrm{CO}_{2}$ pipeline that has a terminus point at Hastings Field ${ }^{4}$, and is in reasonable proximity for a tie-in to Dow Freeport. The Denbury Green Pipeline that crosses the Galveston Bay area is located approximately 60 miles from Dow Freeport and the Hastings Field EOR site is approximately 47 miles from Dow Freeport; however, there is no existing connection to the pipeline for Hastings Field and currently the level of anthropogenic sources of $\mathrm{CO}_{2}$ in the Green Pipeline being sent to Hastings Field is minimal.

Other potential sequestration sites in Texas, which are presently commercially viable, such as
the SACROC enhanced oil recovery unit in the Permian Basin, are more than 500 miles from the
proposed project site. The closest site that is currently being field-tested to demonstrate its
capacity for large-scale geological storage of $\mathrm{CO}_{2}$ is the Southeast Regional Carbon Sequestration
Partnership's (SECARB) Cranfield test site located in Adams and Franklin Counties, Mississippi
and is over 400 miles away from the proposed project site. Therefore, assuming that it is
eventually demonstrated to indefinitely store a substantial portion of the large volume of $\mathrm{CO}_{2}$
generated by the proposed project, a very long and sizable pipeline would need to be
constructed to transport the large volume of high-pressure $\mathrm{CO}_{2}$ from the plant to the potential
storage facility. Typical costs for installation of a pipeline for flat, dry areas can be estimated at
$\$ 50,000$ (Footnote 4) per inch-diameter per mile. Thus, the high cost of $\mathrm{CO}_{2}$ transport via
pipelines 50 miles or greater in length renders it infeasible for the proposed project.
$\mathrm{CO}_{2}$ Storage -_Even if it is assumed that $\mathrm{CO}_{2}$ capture and compression could feasibly be achieved for the proposed project and that the $\mathrm{CO}_{2}$ could be transported economically, it must be stored in a suitable sequestration site. A suitable reservoir or geologic formation is not located within a reasonable proximity to the proposed site.
Potential storage sites, including enhanced oil recovery (EOR) sites and saline formations exist in Texas, Louisiana, and Mississippi. The Southeast Texas enhanced oil recovery (EOR) reservoir and other geologic formation sites are all early in development and there is tenuous commercial viability and demonstration of large-scale, long-term $\mathrm{CO}_{2}$ storage; therefore the capital cost and legal risks of building infrastructure solely for $\mathrm{CO}_{2}$ storage from this LHC-9 project are economically challenging. There are salt dome caverns near the site; however, these limestone formations have not been demonstrated to safely store acid gases such as $\mathrm{CO}_{2}$, nor is there adequate availability of space. Instead, these domes are used for cyclical storage of liquefied petroleum gases (LPGS) for use in the Gulf Coast as well as for shipment throughout the United States via pipeline. To replace this critical active storage with long-term $\mathrm{CO}_{2}$ sequestration would necessarily jeopardize energy supplies locally and nationally. Other potential sequestration sites in Texas that are presently commercially viable, such as the SACROC enhanced oil recovery unit in the Permian Basin, are more than 500 miles from the proposed project site. The closest site that is currently being field-tested to demonstrate its capacity for large-scale geological storage of $\mathrm{CO}_{2}$ is the Southeast Regional Carbon Sequestration Partnership's (SECARB) Cranfield test site located in Adams and Franklin Counties. Mississippi

[^0]The Dow Chemical Company $\quad$ Page 21 $\quad$ PSD Greenhouse Gas Permit
Ethylene Production Facility (LHC-9) November 28, 2012
cracking feed, except for the steam utilized in downstream processes. The steam produced by the cracking furnaces will be sufficient to cover any increased energy needs.
Process off-gas from LHC-9 operations will be used as fuel in LHC-9 furnaces, distributed within the site low pressure fuel gas system, or used for off-site hydrogen recovery. Electricity and steam will be provided to the proposed facility from existing production units, 3rd party facilities, and existing tie-lines.

## Downstream Impacts

The primary products produced at the LHC-9 facility (ethylene and propylene) will be used as feed stock for other existing units at the Dow Freeport site or transported via pipeline to existing underground storage caverns and exported off-site to other consumers.
By-product streams as well as off-gas from the LHC-9 unit may be routed to existing facilities at the site for product recovery and energy recovery. The Dow Freeport site is a highly integrated chemical manufacturing complex. This integration allows product and by-product streams to be processed by downstream plants resulting in efficient and low-cost production capability.

Wastewater generated by the unit will be routed to an existing on-site wastewater treatment facility. The wastewater discharged from the site wastewater treatment plant will not vary from other discharges already managed by this facility; therefore, no new pollutants will be treated or discharged.
Sources of GHG emissions at the LHC-9 facility
While there are over 40 individually listed emissions units at the site, only 15 of those are potential sources of GHG emissions. Therefore, the remainder of this review addresses only these 15 sources. The sources (Facility Identification Numbers, FINs) and their corresponding Emissions Point Numbers (EPNs) are listed in Table 1.

Table 1. GHG emissions sources of LHC-9

| LPN | FIN | Description |
| :--- | :--- | :--- | :--- |
| OC2H121 | OC2L9H121 | Ethane Cracking Furnace, F-121 |
| OC2H122 | OC L9H122 | Ethane Cracking Furnace, F-122 |
| OC2H123 | OCD L9H123 | Ethane Cracking Furnace, F-123 |
| OC2H124 | OC L9H124 | Ethane Cracking Furnace, F-124 |
| OC2H125 | OC L9H125 | Ethane Cracking Furnace, F-125 |
| OC2H126 | OC L9H126 | Ethane or Propane Cracking Furnace, F-126 |
| OC2H127 |  |  |
| OC2H128 | OC2 L9H127 |  |
| OC L9H128 |  |  | ( \(\left.\begin{array}{l}Ethane or Propane Cracking Furnace, F-127 <br>

Ethane or Propane Cracking Furnace, F-128\end{array}\right)\)


[^0]:    ${ }^{4}$ Denbury, Green Pipeline Projects, available at htpy/www.denbury.com/Corporate-Responsibility/Pipeline-Project/sogeen-pipelineprojectdefault.aspx (last visited October 10, 2012).

