

TEXT SEARCHABLE DOCUMENT

Data Evaluation Report of Degradation Kinetics

PMRA Submission Number {.....}

EPA MRID Number 46936101

Test material: Fipronil **IUPAC name**: 5-amino-1-(2,6-dichloro- α,α,α -trifluoro-*p*-tolyl)-4-trifluoromethylsulfinylpyrazole-3-carbonitrile

CAS name: 5-amino-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-4-[(trifluoromethyl)sulfinyl]-1*H*-pyrazole-3-carbonitrile

Primary Reviewer: James Hetrick, Ph.D. EPA

)amn G. Hetrich 3/12/08 Signature: Date:

Secondary Reviewer: Thuy Nguyen EPA

Signature: Date:

2080442

CITATION: Tang, Z. and T.S. Ramanarayanan. 2006. Degradation of Fipronil and Its Major Metabolites Following Application of Chipco TopChoice® Leachate to Outdoor Simulated Ponds: Kinetics modeling. Sponsored by Bayer Crop Science, RTP, NC. Performed by Bayer Crop Science, Stillwell, KS, MRID 46936101.

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EXECUTIVE SUMMARY:

The non-guideline study provides ancillary data on the estimation of degradation rates for fipronil and its degradation products in an outdoor pond study (Hoberg, J. 2005. Chipco® TopChoice[™] Effects on Aquatic Fauna in Outdoor Simulated Ponds). Because the modeling <u>assumed</u> the presence of the sulfide degradation product (MB 45950) as the primary biological degradation product, it implies the model water sediment system is anaerobic. Such conditions are not expected to be present in water/sediment systems capable of supporting a viable population of invertebrates.

First-order half-lives for fipronil and its degradation products were estimated for an outdoor pond study. Degradation rates were determined using ModelMaker 4.0 (2) with the Marquardt method of optimization with ordinary least squares. The modeling assumed the MB 45950 and MB 46513 were the primary degradation products of fipronil. MB 46136 occurrence was attributed as a by-product in the fipronil formulation rather than an in-situ degradation product. The summed biolysis and photolysis degradation fipronil half-life was 3.4 day (0.204 day⁻¹). The photolysis half-life of fipronil was estimated at 9.9 days (0.070 days⁻¹). The half-life of fipronil degradation products was 26.7 days (0.026 day⁻¹) for MB 46513, 43.3 days (0.016 days⁻¹) for MB 45950, and 27.7 days (0.025 days⁻¹) for MB 46136.

I. MATERIALS AND METHODS

GUIDELINE FOLLOWED:

The SETAC-Europe: Procedures for Assessing the Environmental Fate and Ecotoxicity of Pesticides (March 1995; pp. 1, 34) is not applicable.

COMPLIANCE:

This study was not conducted in compliance with USEPA FIFRA Good Laboratory Practices (40 CFR Part 160).

A. Material and Methods:

The objective of the study was to determine first-order degradation rates of fipronil and its degradation products in an outdoor sediment-water system.

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The registrant used ModelMaker Version 4.0 (2) to estimate simultaneous first-order degradation rates for fipronil and its degradation products from a simulated outdoor pond study. Data were taken from Hoberg, J. R., Chipco TopChoiceTM Effects on Aquatic Fauna in Outdoor Simulated Ponds (**Appendix 2**). The modeling strategy employed known degradation pathways of fipronil (**Figure 1, pp 11 and Figure 2, pp 13**). The formation and decline of MB 46136 was not considered because the registrant believes MB 46136 formation was associated with the fipronil formulation rather than in situ formation in the pond. Estimation of first-order degradation rates was conducted using the Marquardt method of optimization with ordinary least squares. Chemical rate constants were estimated used to estimate first-order half-lives.

The modeling employed a mass balance approach where the concentration of residue in sediment and water was used to calculate the mass at each time interval (Appendix 3).

B. Results and Discussion

First-order half-lives for fipronil and its degradation products were estimated (**Table 1, Figures 4** and **5**). The summed biolysis and photolysis degradation fipronil half-life was 3.4 day (0.204 day⁻¹). The photolysis half-life of fipronil was estimated at 9.9 days (0.070 days⁻¹). The half-life of fipronil degradation products was 26.7 days (0.026 day⁻¹) for MB 46513, 43.3 days (0.016 days⁻¹) for MB 45950, and 27.7 days (0.025 days⁻¹) for MB 46136. [The reviewer notes the fipronil degradation half-life in aerobic aquatic metabolism is 33 days. The half-lives for fipronil degradation products are estimated from aerobic soil metabolism half-lives (2x aerobic soil metabolism half-life). The estimated half-live is 1320 days for MB46513, 1400 days for MB 45950 and MB 46136.]

C. Review Comments

1. The registrant did not discuss the implications of redox potentials on the formation of MB 45950 and MB 46136. The modeling strategy assumes the total system redox potential is anaerobic because MB 45950 is the major degradation product from biological degradation. Under aerobic conditions, MB 46136 is expected to be the major degradation product. The reviewer does not understand how the outdoor pond could be anaerobic and still maintain any viable population of invertebrates for toxicity testing.

2. The half-lives estimated from Model Maker are not unique values because they were derived using numerical integration methods. Estimates of half-life values may vary due different numerical techniques and convergence conditions.



Figure 1 Fipronil Degradation Pathway in the Environment



Page 11 BASF Reg. Doc. # 2006/7010025 Page 11 of 25



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(1)

 $F6 = k'_{12} * MB46136$

[6]









4.3 Goodness of Fit

The goodness of fit for the kinetics model and its parameters to fit the observed data was evaluated using the coefficient of determination (r^2) . The coefficient of determination was calculated as:

$$r^{2} = \left\{ \frac{\sum_{i=1}^{n} (O_{i} - \overline{O}) (C_{i} - \overline{C})}{\sqrt{\sum_{i=1}^{n} (O_{i} - \overline{O})^{2} \sum_{i=1}^{n} (C_{i} - \overline{C})^{2}}} \right\}^{2}$$

where:

n=total number of observations O_i = i^{th} observed value (with i = 1, 2, ..., n) C_i = i^{th} value calculated with selected model (with i = 1, 2, ..., n) \overline{O} =mean of all observed values



Table 1

Parameters describing the degradation of fipronil and its metabolites in the simulated ponds

Compound	First-Order Rate Constant (d ⁻¹)	Half-life (d) (Equation 2)
Fipronil	· · · · · · · · · · · · · · · ·	
k ₁₃	0.013	
k ₁₄	0.121	5 0
k _{13 +} k ₁₄	0.134	J.4
k ₁₂	0.070	9.9*
$\sum k_{1i}$	0.204	3.4**
MB 46513	<u>en 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 19</u>	26.7
k ₂₄	0.026	
MB 45950		43.3
k ₃₄	0.016	
Overall r ²	0.99	
MB 46136		27.7
k′ ₁₂	0.025	
r^2	0.76	

* photolysis half-life

** overall half-life: biolysis and photolysis





Comparison of observed and calculated residue concentrations of fipronil, MB 46513, and MB 45950 in the simulated ponds



Note: The symbols in the above graph are observed and the solid lines are calculated from kinetics modeling









6. <u>CONCLUSIONS</u>

The kinetics model developed in this study adequately characterized the degradation of fipronil in the sediment-water system in the simulated ponds. Using the fitted parameters of the model, simple first-order degradation rates of fipronil and its metabolites MB 46513, MB 45950, MB 46136 were estimated.

BASF Reg. Doc. # 2006/7010025 Page 20 of 25

Page 20

^a From Table 6 and Table 8 of the simulated pond study report (1). ^b Pond water volume: 2800L; Total sediment mass: 58.19 kg.

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Appendix 2

Concentration^a (ng/L)

Mass^b(ng)

Total Mass (water + sed.) (ng)

Measured concentrations and calculated mass for fipronil and metabolites in the simulated ponds

MEFIX0



<u>Appendix 3</u> <u>Percent of the total applied mass for fipronil and metabolites (fipronil equivalent)</u> used in the kinetic models

Day	Fipronil	MB 46136	MB 45513	MB 46950
0	102.8	3.2	0.0	0.0
0	92.4	2.4	0.0	0.0
0	98.0	2.4	0.0	0.0
0	97.5	2.3	0.0	0.0
7	23.3	2.8	27.3	8.1
7	28.0	2.0	20.4	1.8
7	22.4	2.1	24.8	7.0
7	24.3	2.4	26.5	5.1
28	0.0	0.6	15.2	2.3
28	0.0	2.2	17.0	5.1
28	0.0	0.7	17.0	3.5
28	0.0	0.5	15.7	2.3
56	0.0	0.4	9.6	1.4
56	0.0	0.5	5.8	1.9
56	0.0	0.8	12.5	3.0
56	0.0	0.6	11.8	2.1
84	0.0	0.6	6.0	3.1
84	0.0	1.1	6.7	5.3
84	0.0	0.5	6.0	1.7
84	0.0	0.5	4.3	2.4