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#### DATA EVALUATION RECORD

#### STUDY 2

CHEM 039003

Metam-Sodium

§t63-1

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#### CONCLUSIONS:

#### Mobility - Aged Column Leaching

- t. This study cannot be used to fulfill the Leaching-Adsorption/Desorption data requirement.
- 2. Metam-sodium residues were very mobile in a sandy clay loam, a loamy sand, and two sand solis. Methyl isothlocyanate (MiTC) was identified as the predominant compound (≥86% of radioactivity) recovered in the leachates. In each soil, approximately 60% of the applied <sup>14</sup>C leached from the 32 cm columns in one peak. The k<sub>d</sub>'s calculated by the author were: Columbia Sand, 0.02; Feiton Sand, 0.05; Sorrento Loamy Sand, 0.08; and Sunnyvale Sandy Clay, 0.08.
- This study provides <u>supplemental</u> information but does not meet Subdivision N guidelines for the following reasons:
  - a. The soils were not analyzed after aging and immediately prior to leaching, so that the composition and concentration of the aged residues at the start of leaching was unknown.

- b. Soils were sieved through a 0.5 mm mesh, not the required 2 mm. Increasing the effective surface area of the soil may have served to artificially increase the  $k_{\rm d}$  values.
- c. Total recovery was only 70 to 78% of the applied radioactivity for the four soll types.
- d. The analytical technique for identifying residues, gas chromatography, does not provide confirmatory identification. MS should have been used to confirm the identification of the residues.
- e. The residues remaining in the aqueous phase after extraction with ethyl acetate (7 to 14% of the activity recovered in the leachate) were not identified.

### METHODOLOGY:

Columbia River Basin sand and Felton sand soils were packed in glass columns (id tt mm) to depths of 32 to 37 cm and saturated with deionized water. A 0.5 cm layer of chromatography sand was placed on top of each column followed by 8.0 g of dry Columbia sand. The Columbia sand on the columns was then treated with methyl-labeled [14C)metam-sodium (radiochemical purity >96.6%, specific activity 0.0835 mCi/mmol, Stauffer) at a concentration equivalent to 2 qt/100 ft<sup>2</sup> and then the soils were aged for an unspecified period of time. After the aging period, the glass columns were leached with the equivalent of 20 inches of water, and the leachates were collected in 2- to 4-mL fractions.

Aliquots of the leachate fractions were analyzed by LSC Immediately on collection; the remainder of each leachate fraction was stored on ice until column leaching was complete. Fractions containing the most radioactivity were pooled and extracted with ethyl acetate. The aqueous phase was analyzed for metam-sodium by spectrophotometry; spectrophotometry measurements consisted of measuring absorbance at 420 nm to detect a complex formed between metam-sodium and the cupric ion in dilute acetic acid. The ethyl acetate phase was analyzed for the degradate methyl isothiocyanate (MITC) by GC with flame ionization detection. The soll from the glass columns was divided into 6-cm sections. The soil sections were frozen for up to one week prior to analysis by LSC following combustion.

In order to determine the rate of degradation of metam-sodium in the aging soils, additional subsamples of Columbia sand soil were placed in glass and stainless steel columns and treated with [14C]metam-sodium as described above. The soils were sampled at unspecified intervals and extracted with 4.0 mL of water. The water extracts were analyzed for total radioactivity by LSC and for metam-sodium by spectrophotometry.

#### **DATA SUMMARY:**

Based on column leaching experiments, aged methyl-labeled [14C]metam-sodium residues were very mobile in a sandy clay loam, a loamy sand, and two sand soils that were leached with approximately 20 inches of water. After leaching, 57.2-63.4% of the applied radioactivity was in the leachates and 9.7-15.1% remained in the soil columns.

#### Methyl isothlocyanate (MITC)

comprised ≥86% of the radioactivity recovered from the leachates. Total radioactivity in the soils following aging ranged from 60 to 78% of the applied; total radioactivity in the soil and leachate following leaching ranged from 70.5 to 78.2% of the applied.

#### COMMENTS:

- Leachates and soil sections were stored frozen prior to analysis; however, no freezer storage stability data were provided. In addition, the period of time that leachate fractions were stored prior to analysis for MiTC was not reported. Since metam-sodium degrades extremely rapidly, freezer storage stability data are necessary for even very short storage intervals.
- Concentrations of MiTC detected by GC were consistently higher than the concentrations obtained by assuming that all the radiolabel detected by LSC was MiTC. The authors stated that the LSC data are presumed to be more accurate and that consequently all of the <sup>14</sup>C extracted by ethyl acetate is MiTC.
- 3. Recoveries from soil sub-samples used to determine the extent of degradation of metam-sodium after aging and immediately prior to leaching ranged from 60 to 78% of the applied. Since recoveries of total radioactivity after leaching were approximately 70%, it is probable that the loss of radioactivity took place during aging.
- Data for the concentrations of MITC and unidentified residues in the extracted aqueous phase in the leachates were not adequately expressed. These data were reported in terms of percent recovered from the pooled leachate fractions; however, all leachate fractions were not pooled, and the study authors did not report which leachate fractions were pooled and analyzed for degradates. These data should have been expressed either in terms of percent of the applied radioactivity or in terms of total radioactivity recovered in the leachates.

TABLE 1. Soil Analysis a,5

# A. Physical and Chemical Characteristics

Soi?	Half Saturation (%)	βH	ECe <sup>2</sup>	Organic Maîter (%)	Cation Exchange Capacity meg/100 g
Columbia <sup>d</sup>	14	7.3	0.2	0.3	<b>5.</b> 7
eiton <sup>d</sup>	11	ã.5	1.0	0.9	5.2
orresto <sup>d</sup>	16	6.3	0.3	1.1	9.5
unnyvale <sup>d</sup>	18	7.3	0.8	2.3	23.0

# B. Soil Fractions (%)

Soil	Sand F	raction	Siit	Clay	
	(.5-1.0 mm)	(.055 mm)	(.00205 пап)	(0002 mm)	
Columbia <sup>d</sup>	5.9	86.5	5.0	2.6	
Felton <sup>d</sup>	18.4	74.0	<b>3.</b> 5	4.1	
Sorrento <sup>d</sup>	1.8	79.3	12.0	6.9	
Sunnyvale <sup>d</sup>	3.1	46.5	23.0	27. <u>4</u>	

<sup>2.</sup> Soil and Plant Laboratory, Inc., Santa Clara, CA.

<sup>5.</sup> Soil seived to 0.5 mm.

c. Saturation extract conductivity, mmhos/cm @ 26°C.

d. References: Colúmbia, MRC-1119-37-1; Felton, MRC-1119-37-2; Sorrento, MRC-1119-117-1; Sunnyvale, MRC-1119-117-2.

TABLE 2A. Distribution of Ageo [ $^{14}\mathrm{C}$ ]-Vapam After Leaching in Columbian and Felton Soils.

		Columbia <sup>a</sup>			Feiton <sup>5</sup>	
SAMPLE	<sup>14</sup> C (dom)	Recovery <sup>C</sup> (%)	Size (mL or cm)	<sup>14</sup> 5 (dom)	Recovery <sup>c</sup>	Size (mL or cm)
Trestment	10442000	100.90	(mL) 0.84	10724300	100.00	0.34 (mL)
Leachata and the Francis and t	50 7319 2763790 1921370 818216 567930 210646 94501 41074 19192 16332 10198 3249 7658 6731 639926	<pre>&lt;0.01     0.07 26.46 18.40     7.83 6.39 2.01 0.90 0.39 0.18 0.07 0.07 0.07 0.06 63.20</pre>	(mL) 4.35 4.95 4.07 1.55 4.09 4.03 4.07 4.97 4.97	0 5313 390225 1284620 1502970 1736160 738059 270387, 289691 162526 106718 50791 36735 23529 19308 13012 5631044	0.053711882 0.053711882 0.197144633.370197744183 0.183 0.183 0.183	(4.52.13.2.13.34.4.4.4.4.4.4.4.4.4.4.4.4.4.4.4.4.4
Soil Segment (bottom)			(cm)			(=m)
A B C D E (top) F Total	9574 13486 19205 31980 32761 107025 214032	0.09 0.12 0.18 0.30 0.31 1.02 2.06	6.0 6.1 6.0 5.8 4.8 2.9	24091 25622 29274 44590 62419 59540 245536	0.22 0.23 0.27 0.41 0.58 0.55 2.29	6.0 6.0 5.0 5.0 2.7
Aging soil <sup>d</sup> Total	803773	7.69	· ·	900565	8.39	
TOTAL 14C RECOVERED	7617731	72.95	<b>~</b>	7777145	72.52	~ ·*

a. MRC-1119-121-#2 b. MRC-1157-7-#3

dpm's recovered relative to the total applied dpm's, or "treatment".
 d. Columbia sand (MRC-1119-37-1)

Distribution of Aged [ $^{14}\mathrm{Cl} ext{-Vapam}$  After Leaching in Sorranto and Sunnyyale Soils.

		Sorrento Soil <sup>a</sup>			Sunnyvale Soii <sup>b</sup>		
SAMPLE	<sup>14</sup> C (dom)	Recovery <sup>C</sup> (%)	Size (mL or cm	14 <sub>0</sub>	Recovery <sup>c</sup> (%)		
Treatment	4307050	100.00	(mL) 0.58	4164440	100.00	(mL; 0.58	
Leachale Fractions 1231 = 67 8 9 0 112 He NASHSI Tatal	772513 1567700 197771 76460 38103 27971 19618 12541 11598 921  5489	17.93 36.39 4.59 1.77 0.54 0.54 0.29 0.25 0.02 0.12	(mL) 2.00 2.37 2.08 2.77 3.37 4.29 4.17 3.71 4.02 0.40	31069 838819 1144440 231930 50101 45129 12072 4439 7256 3763 2339 1364 7942 2934 2383602	0.74 20.14 27.48 5.36 1.20 1.08 0.28 0.10 0.17 0.09 0.03 0.03 0.19 0.07 57.23	(mL) 0.57 0.57 0.50 0.76 0.10 0.10 0.10 1.70 1.70 1.70 1.70 1.70 1.70	
Soil Segment (bottom)  A  B  C  D  E  (top) F  Total	19307 18669 21460 20391 30771 115409 225007	0.44 0.43 0.49 0.47 0.71 2.67 5.25	(cm) 5.6 5.2 5.4 4.8 5.9 4.6	20458 21278 17810 16262 18612 75081 169501	0.49 0.51 0.42 0.39 0.44 1.80 4.07	(cm) 5.1 4.6 4.3 5.3	
Aging Soil Tatal	409977	9.49		383364	9.20	<b></b>	
TOTAL 14C RECOVERED	3366669	78.16		2936469	70.51		

a. MRC-1157-39-#1

b. MRC-1167-61-#1

dom's recovered relative to the total dpm's applied, or "treatment"

d. After removal of soil, the column tube was washed with H<sub>2</sub>O to recover any TC that might have remained in the column tube.

e. The liquid head space found in the column above the top of the soil. Only the Sunnyvale column had significant dpm's and volume since the soil was forced out of the tube by remoins water into the soil the religion. forced out of the tube by bumping water into the top of the column.

TABLE 3. Distribution Coefficients for /spam and its Degradates in Soil.

Soil	Average K <sub>d</sub>		
Columbia <sup>b</sup>	0.021		
Felton <sup>C</sup>	0.045		
Sorrento <sup>0</sup>	0.084		
Sunnyvale <sup>e</sup>	0.080		

## a. Calculations:

The following example illustrates the calculation of  $K_{\rm d}$  for the two Columbia sand columns (MRC-1119-121) as described by Swoboda and Thomas (7).

For column # 2,

$$V_p$$
 = volume required to leach 50% of  $^{14}C$  = 14.00 mL  
 $V_c$  = void volume of leaching column = 13.30 mL  
 $W$  = dry weight of soil = 45.33 mL

$$K_{d} (coi # 2) = \frac{14.00}{-1} - 1 \times \frac{13.30}{-1} = 0.016$$
13.30 45.33

Similarly, for column # 3,

$$K_{d}$$
 (col # 3) = 0.025

Average  $K_{d} = (0.016 \pm 0.025)/2 = 0.021$ 

b. MRC-119-121

a. MRC-1197-7

a. MRC-1157-39

<sup>≘.</sup> MRC-1157-61

TABLE 4. Partial Identification of 140 in Pooled Leachate Fractions.

Soil (col ≠)ª		y 14	MITC in EtOAc (bom)		
		Aqueous <sup>b</sup>	EtOACC	LSC <sup>d</sup>	SC <sup>e</sup>
Columbia	# 2 # 3	14 11	66 39	32 46	102 57
Felton	<del>=</del> 1 2	13	39 67	141 133	172 167
Sprrento	≠ <u>1</u> ≠ <u>2</u>	7 8	93 92	40 85	76 108
Sunnyvale	<del>†</del>	9 12	91 68	148 164	185 186

a. References for LSC data for ethyl acetate (EtOAc) extraction of pobled leachate fractions: Columbia, Felton and Sunnyvale Soils: (MRC-1187-70), Sorrento: (MRC-1187-48).

d. MITC concentration calculated by assuming all dpm's in the ETOAc extract to be due to MITC. The concentration calculated automatically by the section's computer was based on the specific radioactivity of 1434 dpm/ug of metam. Therefore, the ppm of MITC is obtained by multiplying ppm "metam" by the mole ratio of MITC to metam:

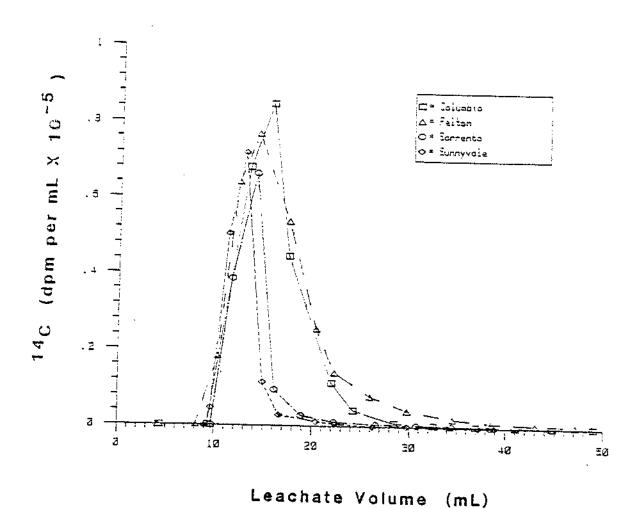
ppm MITC = ppm "metam" 
$$X = \frac{73.1 \text{ (molec. wt. MITC)}}{129.2 \text{ (molec. wt. metam)}}$$

e. MITC concentration in EtOAc measured by GC (MRC-1167-97).

b. %  $^{14}\mathrm{C}$  remaining in the appeaus phase after extraction with StOAc.

c.  $3^{-14}$ C extracted into the EtOAc.

Figure 3. Leaching of Vapam Degradates From Soil Columns. a



a. References: Columbia, MRC-1119-121-#2; Felton, MRC-1157-7-#3; Sorrento, MRC-1157-39-#1; Sunnyvale, MRC-1157-61-#1.