# Rat Olfactory Mucosa Displays a High Activity in Metabolizing Methyl tert-butyl Ether and Other Gasoline Ethers

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Methyl tert-butyl ether (MTBE) is a widely used gasoline oxygenate. Two other ethers, ethyl tert-butyl ether (ETBE) and tertamyl methyl ether (TAME), are also used in reformulated gasoline. Inhalation is a major route for human exposure to MTBE and other gasoline ethers. The possible adverse effects of MTBE in humans are a public concern and some of the reported symptoms attributed to MTBE exposure appear to be related to olfactory sensation. In the present study, we have demonstrated that the olfactory mucosa of the male Sprague-Dawley rat possesses the highest microsomal activities, among the tissues examined, in metabolizing MTBE, ETBE, and TAME. The metabolic activity of the olfactory mucosa was 46-fold higher than that of the liver in metabolizing MTBE, and 37- and 25-fold higher, respectively, in metabolizing ETBE and TAME. No detectable activities were found in the microsomes prepared from the lungs, kidneys, and olfactory bulbs of the brain. The observations that the metabolic activity was localized exclusively in the microsomal fraction, depended on the presence of NADPH, and was inhibitable by carbon monoxide are consistent with our recent report on MTBE metabolism in human and mouse livers (Hong et al., 1997) and further confirm that cytochrome P450 enzymes play a critical role in the metabolism of MTBE, ETBE, and TAME. The apparent  $K_m$  and  $V_{\text{max}}$  values for the metabolism of MTBE, ETBE, and TAME in rat olfactory microsomes were very similar, ranging from 87 to 125  $\mu$ M and 9.8 to 11.7 nmol/min/mg protein, respectively. Addition of TAME (0.1 to 0.5 mM) into the incubation mixture caused a concentration-dependent inhibition of the metabolism of MTBE and ETBE. Coumarin (50  $\mu$ M) inhibited the metabolism of these ethers by approximately 87%. Further comparative studies with human nasal tissues on the metabolism of these ethers are needed in order to assess the human relevance of our present findings. © 1997 Society of Toxicology.

Addition of oxygenates to gasoline improves combustion by increasing the octane number and thereby reduces tailpipe emission of carbon monoxide (CO) and other pollutants. Currently, methyl *tert*-butyl ether (MTBE) is the most widely used gasoline oxygenate. Approximately 20% of the gasoline sold in the United States contains 2 to 15% MTBE by volume (Costantini, 1993). Two other gasoline ethers, ethyl *tert*-butyl ether (ETBE) and *tert*-amyl methyl ether (TAME), are also used in reformulated gasoline. ETBE can be made from renewable sources and TAME has a lower vapor pressure than MTBE, which results in lower evaporative emissions. Therefore, it has been proposed that they replace MTBE in the future (Health Effects Institute, 1996). The use of MTBE and other ethers as gasoline oxygenates is expected to increase over the next decade (Costantini, 1993). However, the possible adverse effects of MTBE exposure in humans are currently a public concern.

Inhalation constitutes a major route for environmental exposure to MTBE and other gasoline ethers. When refueling cars, the concentrations of MTBE range from less than 1 to 4 ppm within the breathing zone and from 0.01 to 0.1 ppm inside the cars (Clayton Environmental Consultants, 1991; Hartle, 1993; Lioy et al., 1994). Studies in rats and humans have shown that inhaled MTBE is eliminated primarily through the lungs in exhaled air and through the kidneys in urine (Health Effects Institute, 1996). Oxidative demethylation of MTBE results in the formation of tert-butyl alcohol (TBA), a major circulating metabolite and an exposure marker of MTBE (Costantini, 1993). Deethylation of ETBE also results in the formation of TBA, and demethylation of TAME leads to the formation of tert-amyl alcohol (TAA). The chemical structures of these ethers and their metabolites are shown in Fig. 1. Metabolism of MTBE in the rat liver microsomes and the involvement of cytochrome P450 enzymes (P450) were previously reported by our group (Brady et al., 1990). We have recently demonstrated that human livers are active in metabolizing MTBE to TBA and P450 enzymes play a critical role in the metabolism of MTBE in human livers (Hong et al., 1997).

The nasal tissues are exposed to the highest levels of airborne environmental chemicals. Mammalian nasal mucosa is known to be active in metabolizing xenobiotics and the activity levels are comparable to, or in some cases, even

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methyl tert-butyl ether (MTBE)

tert-amyl methyl ether (TAME)

ethyl tert-butyl ether (ETBE)

FIG. 1. Chemical structures of MTBE, ETBE, and TAME, as well as their metabolites TBA and TAA.

much higher than those of the liver when based on the per unit of microsomal protein. Our previous studies have shown that both rabbit and rat nasal mucosa are highly active in the metabolic activation of several carcinogenic nitrosamines, including a tobacco-specific carcinogen (Hong *et al.*, 1991, 1992). Efficient metabolism *in situ* resulting from this high level of xenobiotic-metabolizing activity in the nasal mucosa could be particularly important for the inhaled chemicals, such as MTBE, in determining their fate and elicited biological responses (Dahl, 1989; Morris, 1994).

In the present study, we investigated the metabolism of MTBE and two other gasoline ethers, ETBE and TAME, by rat olfactory mucosa. For comparison, the metabolic activities toward these gasoline ethers in other rat tissues, including the liver, were also determined.

## MATERIALS AND METHODS

Chemicals. MTBE, ETBE, TAME, TBA, and TAA were purchased from Aldrich Chemical Co. (Milwaukee, WI). The purity of these chemicals was >99%, except for TAME, which was 97% pure. Glucose 6-phosphate, glucose-6-phosphate dehydrogenase, and NADP+ were from Sigma Chemical Co. (St. Louis, MO). All other chemicals used were of reagent grade and were obtained from standard suppliers.

Animals and microsomes. Male Sprague-Dawley rats at 9 weeks of age were obtained from Taconic Farms (Germantown, NY) and were acclimated for 1 week at the animal facility of Rutgers University. They received lab chow (Lab Diet 5012, The Richmond Standard, PMI Feeds, Inc., St. Louis, MO) and tap water ad libitum and were maintained in air-conditioned quarters with 12-h light-dark cycles. The animals were killed by decapitation and the nasal mucosas, including olfactory and respiratory epithelium, were collected and immediately stored at -80°C according to a procedure previously described (Hong et al., 1991). The livers, lungs, kidneys, and

olfactory bulbs of these untreated animals were also immediately removed after termination and stored at -80°C prior to further processing.

Tissues were homogenized in ice-cold Tris-HCl buffer (pH 7.4, containing 1.15% KCl) with an Ultra-Turrax polytron (Janke and Kunkel Co., Germany). Microsomes and cytosolic fractions were prepared by differential centrifugation (Hong and Yang, 1985). The protein content was determined by the method of Lowry et al. (Lowry et al., 1951).

Metabolism of gasoline ethers. Incubations for the metabolism of MTBE, ETBE, and TAME were performed according to a previously described protocol (Brady et al., 1990). The incubation mixture (0.4 ml final volume) contained 50 mM Tris–HCl (pH 7.4), 10 mM MgCl<sub>2</sub>, 150 mM KCl, an NADPH-generating system (0.4 mM NADP<sup>+</sup>, 10 mM glucose 6-phosphate, 0.2 unit glucose-6-phosphate dehydrogenase), and microsomal or cytosolic proteins. Unless specified, the substrate concentration was 0.25 mM. Incubations were carried out in sealed headspace vials and the reaction was initiated by injecting the NADPH-generating system into the solution. After a 30 min incubation at 37°C, the reaction was terminated with 50  $\mu$ l each of 25% ZnSO<sub>4</sub> and saturated Ba(OH)<sub>2</sub>. The metabolism rates were linear under our assay conditions. For the blanks, ZnSO<sub>4</sub> and Ba(OH)<sub>2</sub> were added into the incubation mixture prior to the initiation of reaction with the NADPH-generating system.

For kinetic studies, the metabolizing activities were determined at different substrate concentrations and the determinations were always repeated at least two times to ensure data reproducibility. Kinetic parameters were determined by nonlinear regression analysis using the Enzymekinetics V1.11 software (Trinity Software, Plymouth, NH).

Headspace gas chromatography (GC) analysis. TBA, from the metabolism of MTBE and ETBE, and TAA, from the metabolism of TAME, were determined by a headspace GC method as previously described (Brady et al., 1990) with slight modifications. A Perkin-Elmer model 8500 gas chromatograph was used with a Carbopack B/5% Carbowax 20M stainless steel column (1/8" × 6', Supelco, Inc., Bellefonte, PA) and a HS-101 headspace autoinjector. The carrier gas was helium and the flow rate was 20 ml/min. The injector and flame ionization detector were at 160°C, while the oven temperature was 60°C. Under these analytical conditions, the retention time (min) of these chemicals on headspace GC were: MTBE, 5.2; ETBE, 10.3; TAME, 14.6; TBA, 6.77; and TAA, 19.4.

### **RESULTS**

A representative chromatograph for the metabolic profile of MTBE in the microsomes prepared from the rat liver and olfactory mucosa is shown in Fig. 2. Although the metabolic profile is the same, the activity in metabolizing MTBE to TBA is dramatically higher in the olfactory mucosa microsomes than that in the liver microsomes under the same incubation conditions. Further analysis with more samples and using ETBE and TAME as substrates showed that rat olfactory mucosa microsomes possess the highest activities in metabolizing all three ethers. The microsomal activity level of the olfactory mucosa was 46-fold higher than the liver in metabolizing MTBE, and 37- and 25-fold higher, respectively, in metabolizing ETBE and TAME (Table 1). The activity level of the olfactory mucosa microsomes was approximately 5- to 9-fold higher than that of the respiratory mucosa microsomes, depending on which substrate was used. No detectable activities were found in the microsomes prepared from the lungs, kidneys, and olfactory bulbs of these untreated rats (Table 1).

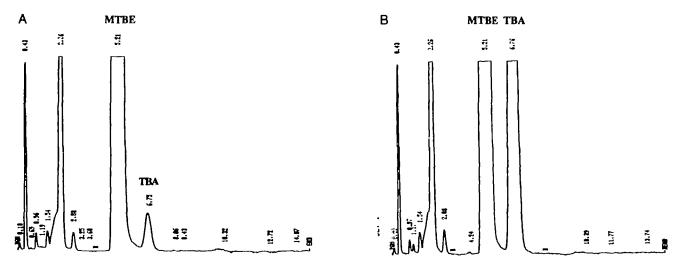


FIG. 2. Headspace GC chromatogram of MTBE metabolism in the rat liver (A) and olfactory mucosa (B) microsomes. Microsomes (600  $\mu$ g protein) were incubated with 1 mM MTBE at 37°C for 30 min. Formation of TBA was determined by headspace GC as described under Materials and Methods. The sensitivity of the detector (absorbance units full scale) was the same for both chromatograms.

For all the rat tissues examined, there was no detectable TBA or TAA formation when the cytosolic fractions were used for the incubation with MTBE, ETBE, and TAME, even though a larger amount of proteins (2.5-fold of the microsomes) were used in the incubation (data not shown). We further determined the effects of NADPH, a cofactor required for P450-catalyzed reactions, and carbon monoxide, an inhibitor of P450, on the metabolism of these gasoline ethers in the rat olfactory microsomes. When the incubation was carried out in the absence of an NADPH-generating system, the activities in metabolizing MTBE, ETBE, and TAME became undetectable (Table 2). Bubbling the microsomal incubation

TABLE 1

Metabolism of MTBE, ETBE, and TAME in the Microsomes

Prepared from Various Rat Tissues

		Activity (nmol metabolite/min/mg protein)				
	n	мтве	ЕТВЕ	TAME		
Nasal mucosa						
Olfactory	6	$9.28 \pm 1.33$	$8.78 \pm 0.55$	$13.15 \pm 0.64$		
Respiratory	6	$1.84 \pm 0.26$	$0.95 \pm 0.18$	$2.08 \pm 0.37$		
Liver	6	$0.20 \pm 0.01$	$0.24 \pm 0.01$	$0.52 \pm 0.03$		
Lung	5	n.d.	n.d.	n.d.		
Kidney	5	n.d.	n.d.	n.d.		
Olfactory bulb	3	n.d.	n.d.	n.d.		

Note. Microsomes were incubated with 1 mM substrate in a final volume of 400  $\mu$ l at 37°C for 30 min. The microsomal protein used was 600  $\mu$ g for liver, lung, kidney, and olfactory bulb, and 100  $\mu$ g for olfactory or respiratory mucosa microsomes. Values are shown as mean  $\pm$  SE. n.d., not detectable. Tissues from three to five rats were pooled to make one sample. n, number of samples.

mixture with 95% CO for 3 min prior to initiation of the reaction caused 85 to 87% reductions in the formation of TBA or TAA (Table 2). These results are consistent with our recent studies on human and mouse livers (Hong *et al.*, 1997), providing further evidence to support the role of P450 enzymes in the metabolism of these gasoline ethers.

Kinetic studies on the metabolism of MTBE, ETBE, and TAME in the rat olfactory microsomes were conducted and the results are summarized in Fig. 3. At substrate concentrations ranging from 0.01 to 2.5 mM, Michaelis-Menton kinetics were observed in the formation of TBA from MTBE or ETBE and the formation of TAA from TAME. The apparent  $K_m$  and  $V_{\text{max}}$  values for the metabolism of MTBE, ETBE,

TABLE 2
Effects of NADPH and CO on the Metabolism of MTBE,
ETBE, and TAME in Rat Olfactory Mucosa Microsomes

	Ac	Activity (nmol/min/mg protein)				
	Control	Without NADPH	With CO pretreatment			
МТВЕ	7.1	n.d.	0.9 (87)			
ETBE	9.5	n.d.	1.2 (88)			
TAME	11.4	n.d.	1.8 (85)			

Note. Rat olfactory mucosa microsomes were pooled from six samples (each sample was pooled from five rats). Microsomes (50  $\mu$ g) were incubated with 0.25 mM substrate at 37°C for 30 min. To determine the effects of NADPH, the NADPH-generating system was omitted from the incubation. The effect of CO was determined by bubbling the microsomal mixture with 95% CO for 3 min before the addition of substrate. The activity values are the average of two separate determinations. The percentage of inhibition is shown in parentheses. n.d., not detectable.

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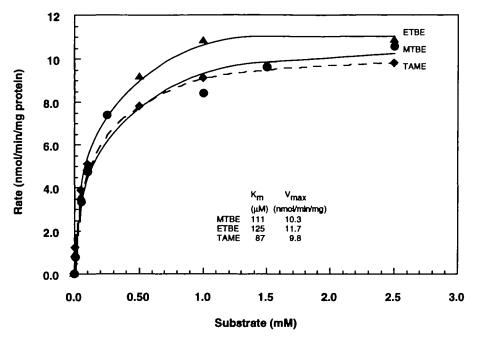


FIG. 3. Kinetics of the metabolism of MTBE, ETBE, and TAME in rat olfactory mucosa microsomes. Microsomes (50  $\mu$ g protein) were incubated with the substrates at different concentrations and the formation of TBA or TAA was determined by headspace GC.

and TAME were very close, ranging from 87 to 125  $\mu$ M and 9.8 to 11.7 nmol/min/mg protein, respectively.

Because of the structure similarity, MTBE, ETBE, and TAME are probably metabolized by the same P450 enzyme(s). We, therefore, examined the effects of different concentrations of TAME on the metabolism of MTBE and ETBE in the rat olfactory mucosa microsomes. When TAME (0.1 to 0.5 mM) was added into the incubation mixture, a concentration-dependent inhibition was observed in the metabolism of MTBE and ETBE (Table 3). On the other hand, the metabolism of TAME was also inhibited by the presence of MTBE or ETBE in the incubation mixture. At 0.25 mM TAME, the presence of MTBE or ETBE in the incubation mixture caused a 24 or 32% decrease (7.0 and 6.2 vs 9.2 nmol/min/mg) in the formation of TAA (Table 3). The metabolism of MTBE, ETBE, and TAME in the olfactory mucosa microsomes was also significantly inhibited to a similar extent by addition of coumarin. At 50  $\mu$ M, coumarin caused approximately an 87% inhibition in the metabolism of these ethers (data not shown).

# DISCUSSION

The present study clearly demonstrates that rat nasal mucosa, in particular the olfactory mucosa, possesses a high activity towards the oxidative metabolism of MTBE, ETBE, and TAME. The results that the metabolizing activities for these gasoline ethers were localized exclusively in the microsomal fraction, depended on the existence of NADPH, and were inhibitable by CO are consistent with our recent report on the metabolism of MTBE in human and mouse livers (Hong *et al.*, 1997) and further confirm that P450 enzymes catalyze the metabolism of MTBE, ETBE, and TAME.

MTBE, ETBE, and TAME are similar in structure and

TABLE 3
Effects of TAME on the Metabolism of MTBE and ETBE in Rat Olfactory Mucosa Microsomes

	Activity (nmol/min/mg protein)				
	МТВЕ	TAME	ЕТВЕ	TAME	
TAME (mM)					
0	6.7		9.2		
0.10	2.8	4.2	4.8	3.7	
	(58)		(48)		
0.25	2.0	7.0	3.7	6.2	
	(71)		(60)		
0.50	1.2	8.5	2.2	8.5	
	(82)		(76)		
TAME alone (mM)	. ,		• •		
0.25		9.2		9.2	

Note. The microsomes were pooled from six samples (each sample was pooled from five rats). Microsomal protein (50  $\mu$ g) was incubated with the substrates in a final volume of 400  $\mu$ l at 37°C for 30 min. The substrate concentrations for MTBE and ETBE were 0.25 mM. Metabolism of MTBE and ETBE was assayed as TBA formation, and metabolism of TAME was assayed as TAA formation. The activity values are the average of duplicate determinations. The percentage of inhibition is shown in parentheses.

physical-chemical properties. Several lines of evidence appear to support our hypothesis that in rat olfactory mucosa, the metabolism of these ethers is catalyzed by the same P450 enzyme(s). These include: (1) The apparent  $K_m$  values for oxidative dealkylation of MTBE, ETBE, and TAME are nearly the same; (2) the inhibition of MTBE and ETBE metabolism by TAME appears to be competitive in nature. and the metabolism of TAME was also inhibited by MTBE and ETBE; (3) the metabolism of these ethers were all inhibited to the same extent by coumarin. In addition, our recent study showed that heterologously expressed human P450 2A6 was efficient in catalyzing the metabolism of MTBE, ETBE, and TAME (Hong et al., 1997). Whereas the exposure, metabolism, and health effects of MTBE have been under extensive investigation, corresponding information for ETBE and TAME is lacking. If further studies demonstrate that the metabolism of ETBE and TAME is indeed catalyzed by the same P450 enzymes metabolizing MTBE, the obtained MTBE data could be useful in predicting or interpreting the pharmacokinetics and health effects of ETBE and TAME.

The present finding is consistent with our previous findings and the reports by others that rat nasal mucosa is highly active in metabolizing many environmental chemicals, and the olfactory mucosa is usually more active than the respiratory mucosa (Bond, 1983; Bond and Li, 1983; Brittebo and Ahlman, 1984; Hadley and Dahl, 1983; Hong et al., 1991, 1992). Several P450 enzymes have been demonstrated to be present in rat nasal mucosa (Ding and Coon, 1993). At present, the particular P450 forms responsible for the metabolism of MTBE, ETBE, and TAME are not known. Significant inhibition of the metabolism of these ethers by coumarin suggests that the P450 enzymes responsible for the ether metabolism are probably also involved in coumarin metabolism. In human livers, coumarin hydroxylation is mainly catalyzed by P450 2A6, which has been demonstrated by us to be active in metabolizing MTBE, ETBE, and TAME (Hong et al., 1997). Preliminary results showed that baculovirus-expressed rat P450 2A3, a predominant P450 form expressed in rat olfactory mucosa and an ortholog of human P450 2A6 (Liu et al., 1996), is active in the metabolism of MTBE, ETBE, and TAME (Hong and Ding, unpublished results). The roles of P450 2A3 and other P450 enzymes in the metabolism of these ethers in rat nasal mucosa remain to be determined.

The presence of a high level of P450 activities in rat olfactory mucosa is intriguing. Little is known about the physiological and toxicological significance of these nasal P450 enzymes. An even more important question is how relevant to the human situation is this high level of P450 activity in rat olfactory mucosa. In the case of MTBE, the reported symptoms attributed to MTBE exposure include headache, nausea or vomiting, burning sensation of the nose

or mouth, coughing, dizziness, disorientation, and eye irritation (Middaugh, 1992, 1993). Some of them may be related to olfactory sensation. In addition to TBA and TAA, formaldehyde (from MTBE and TAME) and acetaldehyde (from ETBE) are produced in the oxidative metabolism of these ethers. Both of these aldehydes have been reported to be nasal carcinogens in rodents (Swenberg et al., 1980; Woutersen et al., 1986). However, the kinetics of their production in the metabolism of gasoline ethers must be considered in addressing the relevant cancer risk of ether inhalation. To assess the importance of nasal metabolism in the health effects of these gasoline ethers in humans, it is critical to conduct a comparative study on the metabolism of MTBE, ETBE, and TAME in human nasal tissues.

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