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Environmental Technology Verification Report

MOLECULAR ANALYTICS
IONPRO-IMS AMMONIA ANALYZER

Prepared by Battelle



In collaboration with the U.S. Department of Agriculture

Under a cooperative agreement with





Environmental Technology Verification Report

ETV Advanced Monitoring Systems Center

Molecular Analytics IonPro-IMS Ammonia Analyzer

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Notice

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Foreword

The U.S. Environmental Protection Agency (EPA) is charged by Congress with protecting the nation's air, water, and land resources. Under a mandate of national environmental laws, the Agency strives to formulate and implement actions leading to a compatible balance between human activities and the ability of natural systems to support and nurture life. To meet this mandate, the EPA's Office of Research and Development provides data and science support that can be used to solve environmental problems and to build the scientific knowledge base needed to manage our ecological resources wisely, to understand how pollutants affect our health, and to prevent or reduce environmental risks.

The Environmental Technology Verification (ETV) Program has been established by the EPA to verify the performance characteristics of innovative environmental technology across all media and to report this objective information to permitters, buyers, and users of the technology, thus substantially accelerating the entrance of new environmental technologies into the marketplace. Verification organizations oversee and report verification activities based on testing and quality assurance protocols developed with input from major stakeholders and customer groups associated with the technology area. ETV consists of seven environmental technology centers. Information about each of these centers can be found on the Internet at http://www.epa.gov/etv/.

Effective verifications of monitoring technologies are needed to assess environmental quality and to supply cost and performance data to select the most appropriate technology for that assessment. Under a cooperative agreement, Battelle has received EPA funding to plan, coordinate, and conduct such verification tests for "Advanced Monitoring Systems for Air, Water, and Soil" and report the results to the community at large. Information concerning this specific environmental technology area can be found on the Internet at http://www.epa.gov/etv/centers/center1.html.

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List of Abbreviations

AFO animal feeding operation

AMS Advanced Monitoring Systems

CI confidence interval

DL detection limit

EPA U.S. Environmental Protection Agency
ETV Environmental Technology Verification

FIA flow injection analysis/analyzer

IMS ion mobility spectroscopy

L liter

Lpm liters per minute

 μg microgram mg milligram mL milliliter

NIST National Institute of Standards and Technology

NH₃ ammonia

NH₄⁺ ammonium

ppb part per billion

%D percent difference

QA quality assurance

QC quality control

QMP quality management plan

RA relative accuracy

RPD relative percent difference RSD relative standard deviation

SD standard deviation

TSA technical systems audit

USDA U.S. Department of Agriculture

Chapter 1 Background

The U.S. Environmental Protection Agency (EPA) supports the Environmental Technology Verification (ETV) Program to facilitate the deployment of innovative environmental technologies through performance verification and dissemination of information. The goal of the ETV Program is to further environmental protection by accelerating the acceptance and use of improved and cost-effective technologies. ETV seeks to achieve this goal by providing high-quality, peer-reviewed data on technology performance to those involved in the design, distribution, financing, permitting, purchase, and use of environmental technologies.

ETV works in partnership with recognized testing organizations; with stakeholder groups consisting of buyers, vendor organizations, and permitters; and with the full participation of individual technology developers. The program evaluates the performance of innovative technologies by developing test plans that are responsive to the needs of stakeholders, conducting field or laboratory tests (as appropriate), collecting and analyzing data, and preparing peer-reviewed reports. All evaluations are conducted in accordance with rigorous quality assurance (QA) protocols to ensure that data of known and adequate quality are generated and that the results are defensible.

The EPA's National Exposure Research Laboratory and its verification organization partner, Battelle, operate the Advanced Monitoring Systems (AMS) Center under ETV. The AMS Center, in collaboration with the U.S. Department of Agriculture's (USDA) National Soil Tilth Laboratory, recently evaluated the performance of the Molecular Analytics IonPro-IMSTM ammonia (NH₃) analyzer.

Chapter 2 Technology Description

The objective of the ETV AMS Center is to verify the performance characteristics of environmental monitoring technologies for air, water, and soil. This verification report provides results for the verification testing of the IonPro-IMS. The following is a description of the IonPro-IMS, based on information provided by the vendor. The information provided below was not subjected to verification in this test.

The IonPro-IMS (Figure 2-1) is designed to continuously monitor NH_3 in ambient air and can operate outdoors in temperatures from -40°C to 50°C and within a relative humidity range of 0% to 100%. The IonPro-IM uses ion mobility spectroscopy (IMS) to detect (0.1 parts per billion [ppb]) and respond to changes in NH_3 concentration. The IonPro-IMS display updates the concentration readings once per second. Data from the IonPro-IMS can be acquired from either the serial RS-232 output or the 4-20 milliamp signal loop using a data acquisition system (not included).

The operation of the IonPro-IMS cell is similar to time-of-flight mass spectrometry except that it functions at atmospheric pressure. Ambient air samples are drawn into the cell and over a semi-permeable membrane, allowing NH₃ to enter while attenuating possible interferents. Purified dry instrument air sweeps the membrane on the inside of the cell and delivers the sample to the reaction region. There the sample is ionized by a weak plasma formed by a nickel-63 source



Figure 2-1. IonPro-IMS Ammonia Analyzer

(sealed). A patented dopant material is added to the flow to enhance the ionization process and increase specificity. The ionized sample molecules drift through the cell under the influence of an electric field. An electronic shutter grid allows periodic introduction of the ions into a drift tube where they separate based on charge, mass, and shape. Smaller ions move faster than larger ions through the drift tube and arrive at the detector. A microprocessor evaluates the spectrum for the target compound and determines the concentration based on the peak height.

The IonPro-IMS is available in rack-mountable or wall-mountable configurations, the latter of

which can be used for outdoor applications. The IonPro-IMS weighs approximately 23 kilograms. The only safety precaution necessary for the sealed nickel-63 source is a semiannual wipe test. An on-board permeation generator option enables field calibration of the IonPro-IMS. The IonPro-IMS is capable of sampling through four separate channels for multipoint sampling. The cost of the IonPro-IMS ranges from \$28,900 to \$78,900, depending on selected options.

Chapter 3 Test Design and Procedures

3.1 Test Design

Livestock agriculture is thought to be the primary source of atmospheric NH₃ in the United States and accounts for approximately 70% of NH₃ emissions in the United States.⁽¹⁾ As a result, a means to accurately quantify these emissions is needed. The objective of this verification test was to verify the IonPro-IMS's performance in measuring gaseous NH₃ in ambient air at animal feeding operations (AFOs).

This verification test was conducted according to procedures specified in the *Test/QA Plan for Verification of Ambient Ammonia Monitors at Animal Feeding Operations*, ⁽²⁾ with the exception of six deviations that are addressed later in this report. The verification test was conducted in two phases, each at separate AFOs. The first phase of testing was conducted between September 8 and October 3, 2003, at a swine finishing farm near Ames, Iowa. The second phase was conducted between October 20 and November 14, 2003, at a cattle feedlot in Carroll, Iowa. These sites were selected to provide realistic testing conditions, which were expected to exhibit a wide range of NH₃ concentrations during the test periods.

The verification test was designed to evaluate the following performance parameters:

- Relative accuracy
- Linearity
- Precision
- Response time
- Calibration/zero drift
- Interference effects
- Comparability
- Ease of use
- Data completeness.

The IonPro-IMS was not available during Phase I of the verification test. However, during Phase II of the verification test, the IonPro-IMS response to a series of NH₃ gas standards of known concentration was used to quantify relative accuracy (RA), linearity, precision (repeatability), and calibration/zero drift. The IonPro-IMS response time, the time to reach 95% of the stable signal, was also assessed during the delivery of the NH₃ standards. During Phase II, interference effects were quantified from the IonPro-IMS response to various chemical species that may be

present at AFOs; the potential interferent gases were delivered both in the presence and absence of NH₃. The IonPro-IMS response to ambient air also was evaluated during Phase II as the comparability to simultaneous determinations by an ambient NH₃ reference method (acid-coated denuders). Additionally, the ease of use of the IonPro-IMS was evaluated based on operator observations. Data completeness was determined based on the amount of data collected as a percentage of the amount of data that could have been collected.

3.2 Site Description—Phase II

The IonPro-IMS was not available during Phase I of the verification test. During Phase II, the IonPro-IMS was installed by a vendor representative and was set to record NH₃ measurements every 30 seconds. Battelle and USDA staff worked with the vendor representative to establish procedures for operating the IonPro-IMS during this verification test. The vendor representative trained Battelle and USDA staff to check several instrument parameters to verify the operation of the IonPro-IMS and identify signs of malfunction, which was done on a daily basis. A checklist, provided by the vendor representative and included as Appendix A, was completed by Battelle and USDA staff during daily monitor checks. In the event of an instrument malfunction, Battelle and/or USDA staff could contact the vendor representative and conduct minor troubleshooting procedures upon request as necessary, but were not expected to make any major repairs. The vendor representative remained on-site until the installation was complete. All the

testing activities were conducted by Battelle and USDA staff.

Figure 3-1 shows a schematic diagram of the cattle feedlot during Phase II of the verification test. A temperature-regulated instrument trailer was used to house the monitoring equipment and to provide a sheltered workspace. The IonPro-IMS was installed in this instrument trailer with a Teflon sample line used to supply outside air to the inlet of the IonPro-IMS. Outside the trailer, the inlet of the sample line was positioned approximately 1.5 meters from ground level. The instrument trailer was in a harvested corn field surrounded on three sides by cow pens. The farm was surrounded on all sides by corn fields, most of which had been harvested. Approximately 2,000 to 3,000 head of cattle were on the farm during the verification test.

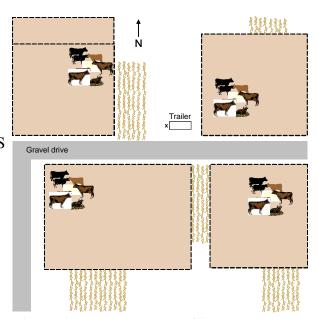


Figure 3-1. Phase II Test Site

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3.3 Test Procedures

3.3.1 Accuracy, Linearity, Precision, and Response Time

During the first week of Phase II, the IonPro-IMS was independently supplied with compressed NH₃ gas standards to achieve NH₃ concentrations over a range from 0 to 2,000 ppb to simulate the approximate range expected in ambient air during Phase II. The gases delivered to the IonPro-IMS were prepared by diluting higher-concentration NH₃ standard gases (i.e., 100 to 500 parts per million) in zero air using a calibrated dilution system provided by the USDA.

The NH₃ gas was supplied to the IonPro-IMS for approximately 30 minutes at each concentration level. Accuracy, linearity, and precision were established based on the continuous digital data set recorded by the IonPro-IMS during the periods when the NH₃ gas was supplied. Data were used for the calculations once the signal had stabilized at a constant concentration (i.e., the signal did not appear to be increasing or decreasing with time). The time required to reach 95% of the change in the stable reading for each concentration was also recorded for the IonPro-IMS. These data were used to assess the response time of the IonPro-IMS.

3.3.2 Calibration and Zero Drift

On Monday, Wednesday, and Friday of the first and last (fourth) week of testing during Phase II, the IonPro-IMS was supplied with an NH₃ gas standard at nominally 1,000 ppb and zero air to check the calibration and zero drift of the IonPro-IMS, respectively. The 1,000-ppb NH₃ standard and zero air were each supplied to the IonPro-IMS for between 30 and 60 minutes, during which time the measured concentrations were recorded by the IonPro-IMS.

3.3.3 Interference Effects

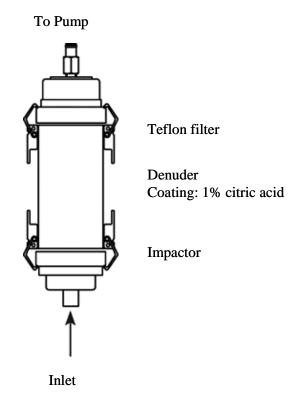
Once during Phase II, the IonPro-IMS was independently supplied with a series of potential interference gases (hydrogen sulfide, nitrogen dioxide, 1,3-butadiene, and diethylamine) to assess any impact the gases have on the IonPro-IMS response. The interferent gases were supplied from diffusion tubes (VICI Metronics, Poulsbo, Washington) at concentrations of approximately 100 to 300 ppb in zero air and a 500-ppb NH₃ standard as carrier gases.

The process for supplying the interferent gases was as follows: zero air was supplied to the IonPro-IMS until a stable reading was achieved. The interferent gas was added to the zero air flow and supplied to the IonPro-IMS until a stable reading was observed (at least 2 minutes). The IonPro-IMS was flushed for at least 2 minutes with zero air, and the next interferent gas was delivered. This process was repeated for the four interferent gases. A 500-ppb NH₃ standard was then supplied to the IonPro-IMS until a stable reading was achieved. The interferent gas was added to the NH₃ standard for delivery to the IonPro-IMS, and the process outlined above was repeated, delivering the 500-ppb NH₃ standard for at least 2 minutes between each interferent gas.

3.3.4 Comparability

The comparability of the IonPro-IMS with a standard reference method was established by comparing the average IonPro-IMS readings with time-integrated NH₃ samples collected using citric-acid-coated denuders. The reference samples were collected based on procedures described in the EPA Compendium Method IO-4.2, *Determination of Reactive Acidic and Basic Gases and Acidity of Fine Particles* ($< 2.5 \mu m$).⁽³⁾

For this test, NH₃ samples were collected using a ChemComb Model 3500 Speciation Sampling Cartridge (Rupprecht & Patashnick Co., East Greenbush, New York). Figure 3-2 shows a schematic illustration of the ChemComb sampling cartridge. Samples were collected by drawing ambient air through an impactor at a nominal rate of 10 liters per minute (Lpm) to remove particulate matter with aerodynamic diameters greater than 2.5 micrometers. The air was passed through two or more citric-acid-coated denuders to collect gaseous NH₃. A single Teflon filter was used to collect the particulate matter that passed through the denuder. During Phase II, automated Partisol Model 2300 speciation samplers (Rupprecht & Patashnick Co., East Greenbush, New York) were used. The Partisol samplers were equipped with mass-flow controlled sampling systems that were pressure- and temperaturecorrected.



The procedures that were used for preparing and coating the denuders were based on the procedures given in the ChemComb Operating

Figure 3-2. Reference Method Sampling Cartridge

Manual⁽⁴⁾ and the test/QA plan.⁽²⁾ The denuders were coated in an NH₃-free glove box at a USDA National Soil Tilth Laboratory facility in Ames, Iowa, and stored in an NH₃-free glove box until they were installed in the ChemComb sampling cartridge and transported to the test site. Cartridges were assembled in the laboratory and transported to the test site. All denuders were used within 72 hours of being coated and within 24 hours of being transported to the field.

Reference samples were collected during the second and third weeks of testing during Phase II. To capture diurnal variations in NH₃ concentrations, sampling was conducted on the following schedule: 8:00 a.m. to 12:00 p.m., 12:00 p.m. to 2:00 p.m., 2:00 p.m. to 4:00 p.m., 4:00 p.m. to 8:00 p.m., and 8:00 p.m. to 8:00 a.m., so that five sets of samples were collected in each 24-hour period. The short-term (2-hour and 4-hour) sampling captured the midday peaks in NH₃ concentrations, whereas the 12-hour sampling captured overnight, generally low, concentrations. The ChemComb sampling cartridges for a full day of sampling were installed in the Partisol speciation samplers before the first sampling period. The Partisol samplers automatically

switched the ambient air flow to each cartridge according to the schedule defined above. As a result, the ChemComb samplers were exposed to the ambient environment for approximately 24 hours. After the final sampling period, all of the sampling media were retrieved and transported to the USDA laboratory for extraction and analysis. The reference sampling for single-point monitors was conducted at one location near the monitor inlets at the instrument trailer. Duplicate samples were also obtained at this site. Two additional sampling locations were positioned approximately 44 and 74 meters from the instrument trailer for use in the verification testing of open-path monitors, but duplicate samples could not be obtained at these locations due to limitations of the Partisol samplers. The sampling schedule for Phase II deviated from the test/QA plan⁽²⁾ in that sampling was conducted every other day, including weekends, during the two-week sampling period. The schedule allowed sufficient time for sample transportation and processing between sampling days. The test/QA plan⁽²⁾ called for sampling every day, Monday through Friday, during the sampling period.

Extraction and analysis of the denuders were performed as described in the test/QA plan, (2) with one exception. The water volume used to extract the denuders was increased from 10 milliliters (mL), as specified in the test/QA plan, (2) to 20 mL. The volume was increased to accommodate the sample volume requirements of the analysis method described below. A deviation was filed to address this change, which does not impact the quality of the reference data. Samples were extracted in an NH₃-free glove box and stored in acid-washed scintillation vials to prevent contamination. The samples were analyzed by USDA by flow injection analysis (FIA) using a Lachat QuikChem Automated Flow Injection Ion Analyzer (Lachat Company, Loveland, Colorado) according to QuikChem Method No. 10-107-06-2-A. This method involves heating the NH₃ sample with salicylate and hypochlorite in an alkaline phosphate buffer, which produces an emerald green color proportional to the NH₃ concentration. The color was intensified by adding sodium nitroprusside and monitored photometrically.

When possible, samples were analyzed within 24 hours of extraction, as specified in the test/QA plan. When analysis within 24 hours of extraction was not possible, the samples were stored frozen until the analysis could be performed, in accordance with the test/QA plan.

Chapter 4 Quality Assurance/Quality Control

QA/quality control (QC) procedures were performed in accordance with the quality management plan (QMP) for the AMS Center⁽⁵⁾ and the test/QA plan for this verification test.⁽²⁾

Six deviation reports were filed during this test and have been addressed in this report. In summary, a change was made in the reference sampling schedule and equipment for Phase II (Section 3.3.4), the denuder extraction volume was increased (Section 3.3.4), some percent difference values measured for duplicate reference samples exceeded 10% (Section 4.2.3), laboratory blank thresholds were redefined (Section 4.2.4), the order in which laboratory blanks and calibration check standards were submitted for analysis was changed (Section 4.2.4 and 4.2.5), and not all of the test data were reviewed within two weeks of the end of the test phase (Section 4.5). None of these deviations impacted the quality of this verification test.

4.1 Equipment Calibrations

4.1.1 Reference Method Sampling Equipment

Reference method sampling was conducted based on the procedures described in the EPA method⁽³⁾ and the ChemComb operating manual.⁽⁴⁾ A single-point calibration of the flow rate through each of the sampling systems (i.e., pump, flow controller, filter pack, denuder, impactor) was performed prior to starting Phase II using a flow meter with a National Institute of Standards and Technology (NIST)-traceable calibration. For Phase II, flows were controlled by the pressure- and temperature-corrected mass flow controllers used in the USDA's Partisol samplers. These samplers shut off automatically if the flow deviated by \pm 5% from the 10 Lpm setpoint for more than 5 minutes, and the data were flagged. Actual sample volumes were recorded by the samplers.

4.1.2 Analytical Equipment

The reference samples were analyzed in the USDA laboratory using FIA. A five-point calibration was measured on the FIA for the reference sample analysis prior to each analytical session by the USDA staff performing the analysis. The calibration was conducted according to the manufacturer's recommendations and included concentrations of NH₃ standard solutions throughout the operating range of the FIA. The calibration was acceptable if the coefficient of determination (r²) of the calibration curve was greater than 0.99. The FIA detection limit (DL)

was 0.03 milligrams per liter (mg/L) and was determined as three times the standard deviation of repeated measurements of a low-level NH₃ standard. Any analytical results that fell below the FIA DL were used without any further adjustment.

Calibration check standards were analyzed after every fifteenth sample in the batch. These calibration checks were considered acceptable if the measured concentration agreed within 10% of the standard solution concentration. If a calibration check failed to agree within 10% of the standard concentration, the FIA was recalibrated; all analyses since the last acceptable calibration check were repeated. All calibration results were documented for inclusion in the verification test data files.

4.1.3 Meteorological Equipment

The sensors used for meteorological monitoring had been calibrated by the manufacturer (Met One Instruments, Inc., Grants Pass, Oregon) within one year of their use in this verification test. The calibration results were included in the verification test data files.

4.1.4 Ammonia Dilution System

The USDA NH₃ dilution system (Environics, Tolland, Connecticut) employs three heated mass flow controllers and valves dedicated for the dilution of compressed NH₃ mixtures. The output flow rates were verified using an independent, NIST-traceable flow meter and agreed to within 10%.

4.2 QC Samples

4.2.1 Field Blanks

At least 10% of all reference samples collected were field blanks. The field blanks were collected by installing the sampling media (i.e., denuder and filters) in the sampling train without drawing any air through the train. The media were recovered and handled as normal samples. Field blanks were collected at each of the sampling locations and during each of the sampling periods (e.g., 8:00 a.m. to 12:00 p.m.). Field blank results were used to detect potential sample contamination (defined in the test/QA plan⁽²⁾ as field blank values greater than 5% of any reference samples for that day) and also to determine the reference method DL.

The reference method DL was determined from the field blank results and reported in terms of an NH_3 mass corresponding to three times the standard deviation of the NH_3 mass collected on the field blanks. The reference method DL was more than six times higher than the equivalent FIA DL (0.6 microgram [μ g] NH_3 per 20-mL sample).

The reference method DL, reported as an NH₃ mass, was used to determine the minimum detectable NH₃ concentrations for Phase II. Since the mass of NH₃ collected by the reference method is a function of the sampling time, flow rate, and the ambient NH₃ concentration, the minimum (time-integrated) ambient NH₃ concentration detectable by the reference method

varies depending on the sample period duration. (This assumes a constant flow rate.) For example, to collect $100~\mu g~NH_3$, the time-integrated ambient NH_3 concentration must be 20~ppb for a 12-hour sample and 120~ppb for a 2-hour sample. Accordingly, the minimum ambient NH_3 concentrations that could be detected from the collection of 2-, 4-, and 12-hour samples at a nominal flow rate of 10~Lpm were calculated from the reference method DL for Phase II.

A total of 14 field blanks were collected in Phase II. The average NH_3 mass collected on these blanks was 2.5 μ g NH_3 , and the range was 0.5 to 4.6 μ g NH_3 . The mass collected on the field blanks ranged from 1.2% to 55.0% of the smallest reference sample mass collected on the same day, with an average of 19.2%. These percentages are not indicative of unusually high levels of contamination, but rather are a result of relatively low ambient NH_3 levels at the AFO. The impact of these blank levels on the results of this verification test may be manifested as a small positive bias of the reference method results relative to the readings of the technologies being verified. This bias would be most pronounced on days with low ambient NH_3 concentrations. The highest field blank percentages were measured on days when the integrated ambient NH_3 levels were as low as 6 ppb, which is approaching the 4.9-ppb minimum detectable ambient NH_3 concentration for a 2-hour sample. Assuming an ambient air sample volume of 0.2 cubic meters, the smallest volume collected during Phase II, the maximum field blank value corresponds to an ambient concentration of 5.5 ppb. Thus, the sample handling may account for up to 5.5 ppb of the measured values.

The standard deviation of the NH_3 collected from field blanks for Phase II was 1.4 µg, which resulted in a 6.6 µg NH_3 Phase II reference method DL. The minimum detectable ambient NH_3 concentrations for 2-, 4-, and 12-hour samples (at a nominal flow rate of 10 Lpm) are shown in Table 4-1. During Phase II, one measured NH_3 concentration in ambient air fell below the minimum detectable NH_3 concentration, as summarized in Table 4-1.

Table 4-1. Minimum Detectable Ambient NH, Concentrations During Phase II

	2-Hour Sample	4-Hour Sample	12-Hour Sample
Minimum detectable NH ₃ concentration	7.9 ppb	4.0 ppb	1.3 ppb
Number of reference samples collected	56	56	29
Number less than minimum detectable NH ₃ concentration	2	0	0

4.2.2 Denuder Breakthrough Checks

Use of backup denuders is called for in the test/QA plan⁽²⁾ during periods when breakthrough greater than 10% of the front denuder is observed or expected. These backup denuders were used to check the degree of NH₃ breakthrough. The breakthrough checks were conducted at each of the sampling locations and included checks during each of the five sampling periods (i.e., 8:00 p.m. to 8:00 a.m., 8:00 a.m. to 12:00 p.m., etc.). Figure 4-1 shows the percentage of NH₃

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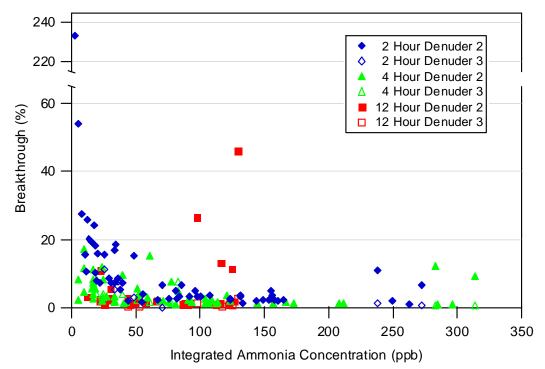


Figure 4-1. Denuder Breakthrough During Phase II as a Function of Integrated Ammonia Concentration

collected on the backup denuders relative to the front denuder (i.e., breakthrough) as a function of the average NH₃ concentration for each of the sampling period lengths (combined data from both sampling locations). The solid symbols in this figure represent the first backup denuder (identified as Denuder 2 in the legend), and the open symbols represent the second backup denuder (identified as Denuder 3 in the legend). Data for all three Phase II sampling locations are included here.

In general, breakthrough onto the first backup denuder (Denuder 2 in the figure legend) was low, with an average breakthrough of 8.6%. As shown in the figure, many of the high breakthrough values (i.e., greater than 10%) observed on the first backup denuder occurred at very low NH₃ concentrations where the mass of NH₃ collected was similar to that collected for field blanks. The high values do not indicate that breakthrough occurred, but rather that the measurements were near the DL of the reference method. High breakthrough of the first backup denuder also occurred at higher NH₃ concentrations and/or long sample durations. Although these high breakthrough values may indicate that breakthrough of the first backup denuder occurred, the second backup denuder (Denuder 3 in the figure legend) was in place to collect the remaining NH₃. With the exception of one sample that occurred at a low ambient NH₃ concentration, breakthrough observed on the second backup denuder was always less than 10% of the amount collected on the front denuder. Thus, it is unlikely that NH₃ was lost as a result of breakthrough of the first or second backup denuders. Table 4-2 summarizes the results of the breakthrough checks for Phase II.

Table 4-2. Denuder Breakthrough Checks During Phase II

	2-Hour	Samples	4-Hour	Samples	12-Hour Samples		
	1st Backup Denuder (%)	2nd Backup Denuder (%)	1st Backup Denuder (%)	2nd Backup Denuder (%)	1st Backup Denuder (%)	2nd Backup Denuder (%)	
Percent of reference samples with denuder	100	18	100	18	100	24	
Average concentration as % of concentration on front denuder	8.6	4.1	4.4	2.8	5.2	1.1	
Maximum concentration as % of concentration on front denuder	[233.3] ^(a) 53.8	11.3	17.2	7.5	45.9	2.5	
Percent of samples with breakthrough greater than 10% of front denuder	29	10	10.7	0	17.2	0	

^(a) Suspect value rejected based on Q-test and not included in other calculations. This value corresponded to an NH₃ concentration that was less than the minimum detectable NH₃ concentration.

4.2.3 Duplicate Samples

For at least 10% of the reference samples, duplicates were collected using a collocated sampling train (within 1 meter). The relative percent difference (RPD) between the duplicate samples was calculated by dividing the absolute difference of the sample concentrations by the average of the sample concentrations.

Table 4-3 summarizes the results of the duplicate sampling for Phase II. During Phase II, duplicate samples were collected during every sampling period at the sampling location next to the trailer, resulting in a total of 35 duplicate measurements. The absolute RPD varied between 0.7% and 32%, with an average of 7%. The absolute RPD for 7 of the duplicate samples exceeded the QA limit of 10% specified in the test/QA plan. To verify the quality of the reference method, NH₃ gas standards were delivered to the reference method. Repeated delivery of the same concentration standard gave an average RPD of 1.3%. Thus, it is probable that the exceedences were caused by non-uniformity in the air sampled and did not impact the quality of the reference method itself. However, some contributions may result from small variations in sampling flow rates and analytical uncertainties.

4.2.4 Laboratory Blanks

Laboratory blank solutions were prepared for the FIA using distilled, deionized water. In each analytical batch, at least 10% of the number of reference samples analyzed were laboratory blanks and were submitted to the laboratory as blind samples. The analysis of the laboratory

Table 4-3. Duplicate Sampling During Phase II

	RPD (%)	Absolute Difference (ppb)
Average	7	5
Maximum	32	18
Minimum	0.7	0.6
Number of duplicate samples		35
Number with RPD >10%		7

blanks deviated from the test/QA plan⁽²⁾ in that, rather than submitting the blanks routinely (e.g., every tenth sample), the blanks were interspersed among the other samples and submitted as blind samples. (Note: The test/QA plan⁽²⁾ indicates that laboratory blanks should not exceed 5% of any concentration measured on that day. As written, this threshold includes field blanks and backup denuder samples. A deviation report has been filed to change this threshold so that it applies only to composite reference samples and does not include samples that would be expected to have low concentrations, such as field blanks.) During Phase II, a total of 27 laboratory blank samples were analyzed. The analytical results from the laboratory blanks indicated no apparent drift in the baseline of the FIA, and none of the blank values was greater than 5% of the lowest measured reference sample on that day.

4.2.5 Calibration Checks

In addition to analyzing every 15th calibration check samples, as described in Section 4.1.2, at least 10% of the samples were submitted to the laboratory as blind calibration check samples. These blind calibration check samples were prepared by diluting NIST-traceable NH₄⁺ standard stock solution.

During Phase II, 24 calibration check samples were prepared from four different standard solutions. Measured concentrations for six of these calibration check samples differed from the delivered standard concentration by more than 10%, and the full set of measured values was on average 4.4% lower than the delivered concentration. Of the six calibration check samples that failed, five were prepared from two of the four standard solutions. It is possible that the failures may be attributable to inadvertent dilution or degradation of the standard solutions used, since these standards were prepared prior to submission of the first samples and failed consistently only near the end of the analysis period. The sixth calibration check sample that failed may be associated with a transcription error in the submission log.

4.2.6 Gas Standard Dilution Checks

At each of the nominal NH₃ levels to be used for the accuracy and linearity checks, at least one sample of the dilution of the NH₃ gas standard was collected using the reference method. These

samples were analyzed as regular samples and used to check the accuracy of the dilution system. Figure 4-2 shows the measured NH_3 captured by the sampling cartridges versus the NH_3 delivered during the dilution checks.

The dilution check was conducted before Week 2 of Phase I. However, the sampling line was thought to have not been flushed with the diluted NH₃ sample prior to collecting the check samples, and the measured concentrations did not agree within 10% of the expected concentration. Consequently, the dilution check was repeated prior to Phase II, and the results are shown in Figure 4-2. The average RA of the measured concentrations was 4% and indicates that the NH₃ gas standards as delivered by the dilution system were accurate with respect to the reference method.

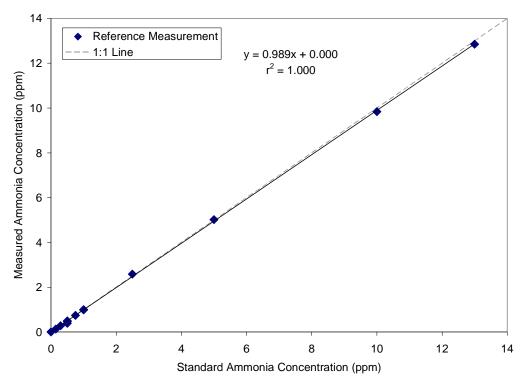


Figure 4-2. Analysis of Diluted Ammonia Standards Using the Denuder Reference Method

4.3 Audits

4.3.1 Performance Evaluation Audit

A performance evaluation audit was conducted to assess the quality of the measurements made in this verification test. This audit addressed only those measurements that factor into the data used for verification, i.e., the sample flow rate and the analytical laboratory measurements. This audit was performed once during the verification test by analyzing a standard or comparing a reading to a reference that was independent of standards used during the testing.

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The flow rates of the reference method sampling assemblies were audited once during testing using a flow meter independent of the meter used to calibrate the flow rate. During Phase I, agreement between the audit flow rate and the nominal flow rate indicated a bias in the calibrated flow rates. The flow rates were recalibrated. The bias was later attributed to a faulty audit flow meter, and the original flow calibrations were verified against a second audit flow meter.

The performance of the FIA was audited by analyzing an NH_4^+ standard independent of those used for the calibration, but were the same as those used for the calibration checks described in Section 4.2.5. These samples were provided as blind audit samples, and the operator of the FIA was not aware of the concentrations of the samples. In several cases, agreement between the measured concentration and the standard concentration was not within \pm 10% (ranged from -43% to 64%). The cause of the discrepancy was investigated but could not be identified. It is possible that some of the discrepancy is attributable to uncertainties associated with dilution of the stock 1,000 mg/L NH_4^+ standard solution. Multiple solutions were prepared, and only some of those solutions showed discrepancies with the analytical results. The relative agreement between the reference samples collected during the gas standard dilution check (performed between Phases I and II) and their expected values provide additional verification of the accuracy of the FIA.

4.3.2 Technical Systems Audit

Battelle's ETV Quality Manager performed a technical systems audit (TSA) of the performance of this verification test during the test. The purpose of this TSA was to ensure that the verification test was being performed in accordance with the test/QA plan⁽²⁾ and that all QA/QC procedures were implemented. As part of the audit, Battelle's ETV Quality Manager reviewed the reference sampling and analysis methods used, compared actual test procedures to those specified in the test/QA plan,⁽²⁾ and reviewed data acquisition and handling procedures. Observations and findings from this audit were documented and submitted to the Battelle Verification Test Coordinator for response. The records concerning the TSA are permanently stored with the Battelle Quality Manager.

4.3.3 Audit of Data Quality

At least 10% of the data acquired during the verification test was audited. Battelle's Quality Manager traced the data from the initial acquisition, through reduction and statistical analysis, to final reporting, to ensure the integrity of the reported results. All calculations performed on the data undergoing the audit were checked during the technical review process.

4.4 QA/QC Reporting

Each audit was documented in accordance with Sections 3.3.4 and 3.3.5 of the QMP for the ETV AMS Center. Once the audit report was prepared, the Battelle Verification Test Coordinator ensured that a response was provided for each adverse finding or potential problem

and implemented any necessary follow-up corrective action. The Battelle Quality Manager ensured that follow-up corrective action was taken. The results of the TSA were sent to the EPA.

4.5 Data Review

Records generated in the verification test were reviewed before these records were used to calculate, evaluate, or report verification results. Table 4-4 summarizes the types of data recorded. The review was performed by a technical staff member involved in the verification test, but not the staff member who originally generated the record. The person performing the review added his/her initials and the date to a hard copy of the record being reviewed. In some cases, the entries in the laboratory record books or on field data sheets were not reviewed within two weeks. A deviation report was filed to address this.

Table 4-4. Data Recording Process

Data to be Recorded	Responsible Party	Where Recorded	How Often Recorded	Disposition of Data ^(a)
Dates, times of test events (site activities, etc.)	USDA/ Battelle staff	Laboratory record books/field data sheet.	Start/end of test, and at each test activity.	Used to organize/ check test results; manually incorporated in data spreadsheets as necessary.
Reference method sampling data	USDA/ Battelle staff	Laboratory record books, chain-of- custody forms, or file data sheets as appropriate.	At least at start/end of reference sample, and at each change of a test parameter.	Used to organize/ check test results; manually incorporated in data spreadsheets as necessary.
Meteorological conditions	Battelle	Meteorological station data logger.	Continuously.	Used to assess meteorological conditions during testing as necessary.
Ammonia analyzer readings	Vendor or designee	Data acquisition system (data logger, personal computer, laptop, etc.).	Continuously at specified acquisition rate throughout analyzer operation.	Electronically transferred to spreadsheets.
Reference sample analysis and results	USDA/ Battelle staff	Laboratory record books, data sheets, or data acquisition system, as appropriate.	Throughout sample handling and analysis process.	Transferred to spreadsheets.

⁽a) All activities subsequent to data recording were carried out by Battelle.

Chapter 5 Statistical Methods and Reported Parameters

The statistical methods presented in this chapter were used to verify the performance parameters listed in Section 3.1. Beginning on November 6, 2003 (during Phase II), the IonPro-IMS data were recorded once every second instead of every 30 seconds. (The change in data recording frequency was made inadvertently when the data acquisition computer was switched to one with data output capabilities.) In order to make the Phase II data set consistent, the higher-frequency data collected starting November 6 were averaged over 30-second intervals using IgorPro software, Version 4.09 (Wavemetrics, Lake Oswego, Oregon), so that they were comparable to the data recorded before November 6. The 30-second interval data were used in the calculations described in this section and reported in Chapter 6.

5.1 Relative Accuracy

The percent difference (%D) of the average IonPro-IMS response to each NH₃ gas standard was calculated according to Equation 1:

$$\%D = \frac{\overline{x} - x_n}{x_n} \times 100 \tag{1}$$

where \bar{x} is the average IonPro-IMS response to an NH₃ gas standard of nominal concentration x_n . During Phase II, the RA with respect to all of the gas standards (n) delivered to the IonPro-IMS within its reported measurement range was calculated using Equation 2:

Average RA =
$$\frac{1}{n} \left(\sum_{i=1}^{n} |\% \mathbf{D}_{i}| \right) \times 100$$
 (2)

5.2 Linearity

Linearity was assessed by a linear regression analysis using the compressed gas standard concentrations as the independent variable and average results from the IonPro-IMS as the dependent variable. Linearity was expressed in terms of slope, intercept, and r² value. The 95% confidence interval (CI) for the slope and intercept was also calculated.

5.3 Precision

Precision was calculated in terms of the percent relative standard deviation (RSD) of the IonPro-IMS measurements of several NH₃ gas standards. The mean and standard deviations of those readings were calculated. The RSD was then determined as:

$$RSD = \frac{SD}{\overline{x}} \times 100$$
 (3)

where SD is the standard deviation of the IonPro-IMS readings and \bar{x} is the mean of the IonPro-IMS readings.

5.4 Response Time

Response time was assessed in terms of both the rise and fall times of the IonPro-IMS when sampling NH₃ gas standards or zero air. Rise time (i.e., 0% to 95% response time for the change in NH₃ concentration) was determined from the IonPro-IMS response to a rapid increase in the delivered NH₃ concentration. Once a stable response was achieved with the gas standard, the fall time (i.e., the 100% to 5% response time for the change in NH₃ concentration) was determined in a similar way, switching from the NH₃ standard back to zero air or a lower concentration NH₃ gas standard. Response times are reported in terms of seconds.

5.5 Calibration and Zero Drift

Calibration and zero drift are reported in terms of the mean, RSD, and range (maximum and minimum) of the readings obtained from the IonPro-IMS in the repeated sampling of the same NH₃ standard gas and of zero air. For zero drift, the SD is reported instead of the RSD since dividing the SD by a value approximately equal to zero is not meaningful. The calibration and zero drift were calculated during Phase II of testing so that up to six NH₃ standard and zero air readings (on Monday, Wednesday, and Friday of Week 1 and Week 4) were used for this calculation. The results of these checks indicate the day-to-day variation in standard and zero readings.

5.6 Interference Effects

The extent of interference was calculated in terms of the ratio of the response of the IonPro-IMS to the interfering species, relative to the actual concentration of the interfering species. For example, if 100 ppb of an interfering species resulted in a 1-ppb increase in the NH₃ reading of the IonPro-IMS, the interference effect was reported as 1% (i.e., 1 ppb/100 ppb). The interference effect was reported separately for each interferent, both in the absence and in the presence of NH₃.

5.7 Comparability

Comparability between the IonPro-IMS results and the reference method results with respect to ambient air was assessed by linear regression using the reference method NH₃ concentrations as the independent variable and results from the IonPro-IMS as the dependent variable. The IonPro-IMS measurement data were averaged over 2-, 4-, and 12-hour time intervals corresponding to the reference method sampling schedule. Comparability was expressed in terms of slope, intercept, and r².

Chapter 6 Test Results

The results of the verification test of the IonPro-IMS are presented in this section. The values presented in this section are based on 30-second average readings. The IonPro-IMS was not available for testing during Phase I of the verification test. Therefore, the results presented here are from Phase II only.

Meteorological conditions collected using the meteorological station during Phase II are presented in Figure 6-1. The average ambient temperature was 4.5°C (range: -10 to 29°C), and the average relative humidity was 75%. Winds were predominantly from the northwest and quite variable in speed, averaging 7 miles per hour (30 miles per hour maximum). Figure 6-1 shows the Phase II wind direction, wind speed, and ambient temperature data and the ambient NH₃ data set collected by the IonPro-IMS (bottom panel). The shaded region shows the period during which NH₃ reference measurements were conducted. The two gaps in the ambient data set were caused by computer-related failures, which are discussed in Section 6.8. The reported IonPro-IMS measurements ranged from 1 to 1,597 ppb during Phase II and averaged 209 ppb.

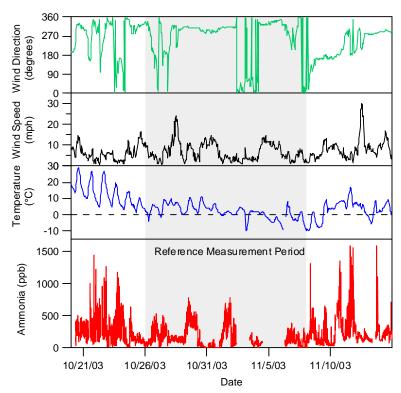


Figure 6-1. Phase II Meteorological Conditions and Ion Pro-IMS Ambient $\mathrm{NH_3}$ Measurements

6.1 Relative Accuracy

During the first week of Phase II, the IonPro-IMS was supplied with compressed NH₃ gas standards at a variety of concentrations to assess RA. The compressed NH₃ gas standards were diluted in zero air and delivered to the inlet of the IonPro-IMS at a flow rate of 5 Lpm.

Figure 6-2 presents the NH₃ concentrations recorded by the IonPro-IMS during the RA checks, along with the nominal NH₃ concentration levels supplied to the IonPro-IMS. The averages of the measurements at each nominal NH₃ concentration are presented in Table 6-1, along with the calculated %D and the number of data points used in the calculations. During delivery of the 1,500-ppb and 2,000-ppb NH₃ standards, an instrument alarm status was recorded in the data set, indicating that the measured concentrations were greater than the measurement range of the IonPro-IMS. As such, readings at 1,500 ppb and 2,000 ppb were not used to calculate the RA, but are shown in Figure 6-2. The %Ds of the IonPro-IMS ranged from -18.6% to -18.0% over the range of concentration levels measured (between 300 and 1,000 ppb), and the average RA (i.e., the average of the absolute values of the %Ds) was 18.3%.

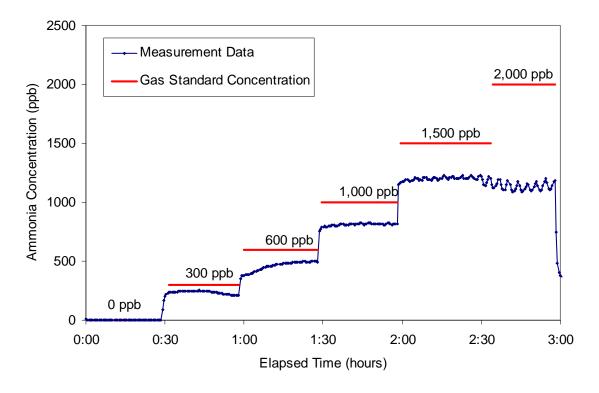


Figure 6-2. Phase II Accuracy Results for the IonPro-IMS

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Table 6-1. Relative Accuracy Results

NH ₃ Gas Standard Concentration (ppb)	Average Measured Concentration (ppb)	Number of Data Points	%D (%)
0	1	52	NA
300	244	34	-18.6
600	492	25	-18.0
1,000	816	43	-18.4
Average RA			18.3%

 $\overline{NA} = \text{not applicable.}$

6.2 Linearity

Figure 6-3 shows the results of the linearity check for Phase II. A linear regression of the average IonPro-IMS response versus the gas standard concentration, over the range of 0 to 1,000 ppb, showed a slope of $0.815~(\pm~0.009)$, an intercept of $1.08~(\pm~5.20)$ ppb, and a coefficient of determination (r^2) of 1.000, where the numbers in parentheses represent the 95% CI. It should be noted that the linearity check was conducted between the Week 1 Monday and Wednesday drift checks. Analysis of the drift check data indicates that the IonPro-IMS calibration changed by approximately 20% during this period (see Section 6.5).

6.3 Precision

Table 6-2 presents the calculated precision of the IonPro-IMS measured during the accuracy and linearity checks. The precision of the IonPro-IMS readings varied from 0.6 to 1.5% RSD at the NH₃ levels measured in the accuracy/linearity checks, with an average precision of 1.0% RSD.

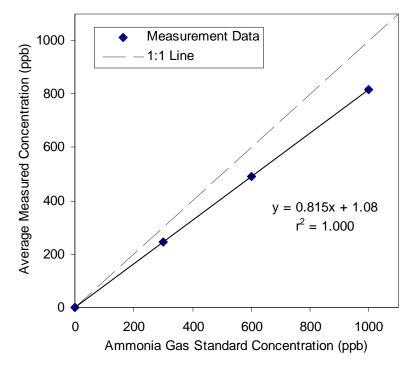


Figure 6-3. Results of Linearity Check of the IonPro-IMS During Phase II

Table 6-2. Calculated Precision of the IonPro-IMS

NH ₃ Gas Standard Concentration (ppb)	Average Measured Concentration (ppb)	Number of Data Points	RSD (%)
300	244	34	1.5
600	492	25	1.0
1,000	816	43	0.6
Average RSD			1.0

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6.4 Response Time

Response time was determined during Phase II from the amount of time required for the IonPro-IMS to achieve 95% of the change in the stable concentrations during the accuracy/linearity checks. Table 6-3 presents a summary of the response time determinations for the IonPro-IMS. Measured rise times were between 150 and 1,020 seconds (approximately 2 to 17 minutes); the fall time from a 2,000-ppb NH₃ standard to ambient levels (approximately 365 ppb) was 90 seconds. It should be noted that response times include the time associated with equilibration of NH₃ on the tubing and inlet surfaces during delivery of the gas standards.

Table 6-3. Response Time Determinations

Change (ppb)	Rise Time (seconds)	Fall Time (seconds)
0 - 300	150	_
300 - 600	1,020 ^(a)	_
600 - 1,000	330	_
1,000 - 1,500	210	_
2,000 - ambient ^(b)	_	90

⁽a) This transition was characterized by a rapid initial change in concentration followed by a gradual increase to a stable concentration. This behavior was not observed for other transitions.

6.5 Calibration and Zero Drift

The calibration/zero drift checks were conducted by supplying NH₃ gas and zero air to the IonPro-IMS for between 30 and 60 minutes each on Monday, Wednesday, and Friday during Weeks 1 and 4 of Phase II. The results of the Phase II calibration and zero drift checks are summarized in Table 6-4, providing the mean response, RSD, minimum, and maximum values for each check. The values reported in this table are based on the average readings during each calibration and zero check when the readings of the IonPro-IMS had stabilized (i.e., the signal was neither visibly increasing nor decreasing); thus, the calculations for each check span somewhat different time periods, which ranged from 5 to 51 minutes. As a result, the number of data points used for each calculation, shown in the table, varied for each check period. The zero check means ranged from 0.1 to 3.9 ppb, indicating no apparent drift in the response of the IonPro-IMS to zero air. The first calibration check showed a mean signal that was approximately 20% below the other calibration check results. No cause for this difference could be determined, and no other indications of calibration drift were apparent. The other calibration check means ranged from 966 to 1,013 ppb.

⁽b) Ambient NH₃ concentration was approximately 365 ppb.

Table 6-4. Calibration and Zero Checks During Phase II

	Zero Check						Cal	ibration (Check ^(a)	
Check Number	Mean (ppb)	SD ^(b) (ppb)	Min- imum (ppb)	Max- imum (ppb)	Number of Data Points	Mean (ppb)	RSD (%)	Min- imum (ppb)	Max- imum (ppb)	Number of Data Points
Week 1 Monday	3.9	0.7	3	5	24	784	0.6	775	795	82
Week 1 Wednesday	2.0	1.5	-1	4	91	979	0.5	971	989	32
Week 1 Friday	0.1	0.7	-1	1	101	983	0.4	974	988	12
Week 4 Monday	1.1	0.1	1	2	18	966	0.7	957	982	10
Week 4 Wednesday	0.1	0.2	0	1	22	1,013	0.4	1,004	1,019	55
Week 4 Friday	0.3	0.5	0	1	35	971	0.3	966	976	11

⁽a) 1,000 ppb NH₃ nominal concentration.

6.6 Interference Effects

The effect of potential interferent gases on the response of the IonPro-IMS was assessed by supplying the IonPro-IMS with a series of four gases (hydrogen sulfide, nitrogen dioxide, 1,3-butadiene, and diethylamine) in zero air and a 500-ppb NH $_3$ standard. The response of the IonPro-IMS during the introduction of these gases is summarized in Table 6-5. The interference gas concentrations carry an uncertainty of approximately \pm 15% (as reported by the manufacturer for uncertified permeation tubes). The response of the IonPro-IMS to hydrogen sulfide, nitrogen dioxide, and 1,3-butadiene was negligible both in zero air and NH $_3$. In the 500-ppb NH $_3$ matrix, the IonPro-IMS interference effect to diethylamine was -388%. No response to diethylamine was apparent in zero air.

⁽b) SD reported for zero drift since the RSD is not meaningful for near-zero values.

Table 6-5. Interference Effect Evaluation

		Interference Effect (%)				
Gas	Interferent Gas Concentration (ppb)	Zero-Air Matrix	500-ppb NH ₃ Matrix			
Hydrogen sulfide	285	-0.1 ^(a)	1.3 ^(a)			
Nitrogen dioxide	95	-0.7 ^(a)	$2.2^{(a)}$			
1,3-Butadiene	95	-0.8 ^(a)	$2.0^{(a)}$			
Diethylamine	96	$0.9^{(a)}$	-388			

⁽a) Signal not significantly different from baseline without interferent gas.

6.7 Comparability

Figure 6-4 shows the NH $_3$ concentrations measured using the reference method, along with the corresponding average readings of the IonPro-IMS for the reference sampling periods during Phase II. In general, the IonPro-IMS measurements appeared to track changes in NH $_3$ concentrations measured with the reference method, although the IonPro-IMS measurements were consistently higher than the reference method. These data are also presented in Figure 6-5 as a scatter plot to illustrate the correlation between the reference and IonPro-IMS data. A linear regression of the IonPro-IMS response versus the reference method concentration showed a slope of 1.565 (\pm 0.047), an intercept of -16.5 (\pm 6.4) ppb, and an r 2 of 0.9941, where the numbers in parentheses represent the 95% CI.

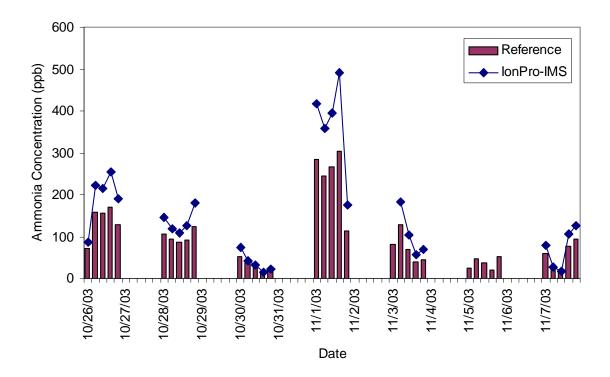


Figure 6-4. Comparison of Ambient Reference Measurements with Averages from the IonPro-IMS During Phase II

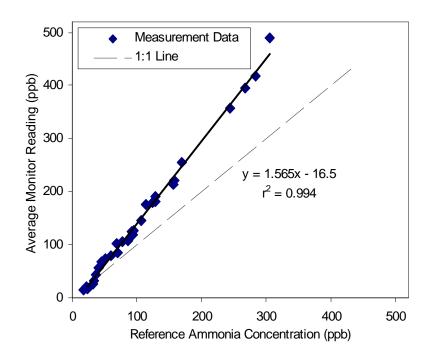


Figure 6-5. Scatter Plot of the IonPro-IMS Results versus Ambient Reference Measurements During Phase II

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6.8 Ease of Use

The IonPro-IMS was installed at the Phase II testing location by a vendor representative, who completed the installation in several hours. The IonPro-IMS was allowed to equilibrate for at least one day before the vendor representative tested the zero and span using instrument zero air and the internal calibration device. The vendor was on-site for several days, which allowed more than enough time to complete the installation and train Battelle and USDA staff to use the IonPro-IMS and check its performance. Checklists, shown in Appendix A, were completed daily by Battelle/USDA staff. The checks were quick and straightforward and could be completed by an inexperienced user. The air compressor that supplied the monitor's zero air generator required that the condensed water be drained daily. Data from the IonPro-IMS were acquired by a laptop computer using Hyper Terminal software (Hilgraeve, Inc.). As part of the daily checks, Battelle/USDA staff saved the current data acquisition file and started a new file. This was done at least once daily in an effort to minimize data loss in the event of computer malfunction or power loss. Nonetheless, it was discovered on two occasions that data acquisition from the IonPro-IMS had terminated; thus, some data were not collected although the IonPro-IMS was conducting measurements during those periods. One of the data acquisition failures appeared to be caused by a lock-up of the Hyper Terminal software. The second appeared to be associated with the change in the computer used to acquire data from the IonPro-IMS (discussed in Chapter 5). Neither failure appeared to be caused by the IonPro-IMS itself. During Phase II of testing, no corrective maintenance was performed on the IonPro-IMS. Table 6-6 presents a summary of activities involving the IonPro-IMS during Phase II. The IonPro-IMS did not produce any waste during the test.

The data collected by Hyper Terminal were saved as comma-delimited text files. The files contained the date and time for each data point, the measured NH₃ concentration, and a number of instrument parameters. Each text file containing approximately 12 hours of data was approximately 160 kilobytes in size for the lower frequency data collection (one data point reported every 30 seconds). The file sizes increased to approximately 5 megabytes for data recorded every second. The entire data set, including data collected at both frequencies, recorded by the IonPro-IMS was 111 megabytes.

6.9 Data Completeness

Although it was operating and conducting measurements during 100% of Phase II, the IonPro-IMS collected only 92% of the data. The data loss of 8% was caused by computer-related failures during data acquisition, as described in Section 6.8.

Table 6-6. Activities Performed During Phase II

	Time Offline	Down Time (b)	Service Time (c)	
Date	(minutes)	(minutes)	(minutes)	Activity
10/20/03	105			Delivered zero air and NH ₃ standard ^(d)
10/21/03	220			Delivered zero air and NH ₃ standards ^(d)
10/22/03	120			Delivered zero air and NH3 standards(d)
10/24/03	120			Delivered zero air and NH3 standards(d)
10/26/03				Instrument time did not change automatically for daylight savings
10/31/03	30		30	Performed internal zero/span check
11/2/03		1,620 ^(e)		Data not saved to computer
11/4/03		1,260 ^(e)		Data not saved to computer
11/8/03			30	Moved air compressor into instrument trailer to prevent freezing
11/10/03	135			Delivered zero air and NH ₃ standard ^(d)
11/12/03	135			Delivered zero air and NH ₃ standard ^(d)
11/12/03	45		45	Performed internal zero/span check
11/13/03	360			Performed interference tests
11/14/03	120			Delivered zero air and NH ₃ standard ^(d)
10/20 - 11/14/03			130	Drained water from air compressor daily (approximately 5 minutes per day)
Totals	1,390 (23.2 hour)	2,880 (48 hour)	235	92% data collected, (e) and 235-minute service time.

⁽a) Time Offline = time that the IonPro-IMS was taken offline for zero or standard gas measurements. The period over which time offline was evaluated began at 8:00 a.m. on 10/20/03 and ended at the conclusion of testing at 5:00 p.m. on 11/14/03. The amount of time was rounded to the nearest 5 minutes.

⁽b) Down Time = time that the IonPro-IMS was not operating or was operating but not reporting reliable measurements. The period over which down time was evaluated began at 8:00 a.m. on 10/20/03 and ended at the conclusion of testing at 5:00 p.m. on 11/14/03. The amount of time was rounded to the nearest 5 minutes.

⁽c) Service Time = time spent conducting routine operation and maintenance activities and troubleshooting problems. The period over which service time was evaluated began at 8:00 a.m. on 10/20/03 and ended at the conclusion of testing at 5:00 p.m. on 11/14/03. The amount of time was rounded to the nearest 5 minutes.

⁽d) Testing activity performed by Battelle/USDA operator.

⁽e) Data output by the IonPro-IMS was collected on a laptop computer using Hyper Terminal. However, no data were collected during these periods although the IonPro-IMS was operating during 100% of Phase II. The loss of data did not appear to be caused by the IonPro-IMS itself. The total time that was available for monitoring during Phase II was 35,220 minutes or 587 hours.

Chapter 7 Performance Summary

The performance of the IonPro-IMS was evaluated only in Phase II of this verification test. Table 7-1 presents a summary of the performance of the IonPro-IMS during Phase II of this verification test.

Table 7-1. Performance Summary of the IonPro-IMS

Parameter	Phase I	Phase II
Relative accuracy ^(a)		Average RA = 18.3% Percent difference range = -18.6% to -18.0 %
Linearity		Range = 0 to 1,000 ppb Slope = $0.815 (\pm 0.009)$ Intercept = 1.08 ppb (± 5.20) $r^2 = 1.000$
Precision		Average RSD = 1.0% Range = 0.6 to 1.5%
Response time		Rise time = 150 to 1,020 seconds Fall time = 90 seconds
Calibration/ zero drift	The IonPro-IMS was not available in Phase I	 No apparent drift in response to zero air. No apparent drift in response to a 1,000-ppb NH₃ gas standard, although the response to the first calibration check was ~20% lower than other checks.
Interference effects ^(b)		Hydrogen sulfide (285 ppb): no apparent effect Nitrogen dioxide (95 ppb): no apparent effect 1,3-Butadiene (95 ppb): no apparent effect Diethylamine (96 ppb): -388% effect in 500-ppb NH ₃ ; no apparent effect in zero air
Comparability		Slope = 1.565 (\pm 0.047) Intercept = -16.5 ppb (\pm 6.4) $r^2 = 0.994$
Ease of use		 Daily checks were simple and quick Little skill required to operate Drain condensed water from compressor daily No corrective maintenance required Data loss of approximately 48 hours resulting from data collection failures^(c)
Data completeness		92% data collected ^(c)

⁽a) Relative accuracy is expressed as an average absolute value of the percent difference from NH₃ gas standards.

⁽b) Calculated as the change in signal divided by the interferent gas concentration, expressed as a percentage.

⁽c) Although the IonPro-IMS was operating during 100% of Phase II, 8% of the data was not recovered due to failures during data collection. The failures did not appear to be caused by the IonPro-IMS.

Chapter 8 References

- 1. *National Air Pollutant Trends*, 1900-1998. EPA-454/R-00-02, U.S. Environmental Protection Agency, Office of Air Quality Planning and Standards, Research Triangle Park, North Carolina, 27711.
- 2. Test/QA Plan for Verification of Ambient Ammonia Monitors at Animal Feeding Operations, Battelle, Columbus, Ohio, September 2003.
- 3. Determination of Reactive Acidic and Basic Gases and Acidity of Fine Particles (<2.5 μm), Environmental Protection Agency Compendium Method IO-4.2, EPA/625/R-96/010A, U.S. Environmental Protection Agency, Office of Research and Development, Cincinnati, Ohio, 45268.
- 4. *Operating Manual, ChemComb Model 3500 Speciation Sampling Cartridge*, Revision A, January 2000, Rupprecht & Patashnick Co., Inc. East Greenbush, New York, 12061.
- 5. Quality Management Plan (QMP) for the ETV Advanced Monitoring Systems Center, U.S. EPA Environmental Technology Verification Program, prepared by Battelle, Columbus, Ohio, Version 4.0, December 2002.

Appendix A

IonPro-IMS Checklist



ETV Verification of Ambient Ammonia Monitors at Animal Feeding Operations Daily Checklist

$86^{\circ}\text{C} \pm 0.5^{\circ}\text{C}$	Value =	
$45^{\circ}\text{C} \pm 5^{\circ}\text{C}$	Value =	
$psi \pm 0.1 psi$ Va	lue =	
day.		
automatic draining.		
		50 psi, the
nd the compressor is	cycling.	
the checklist, but was co	nducted daily).	
re		
file name		
	<u></u>	
		
	45°C ± 5°C osi ± 0.1 psi Variation day. automatic draining. age and verify that the compressor is the checklist, but was core of file name	day. automatic draining. age and verify that the tank pressure is at 1 and the compressor is cycling. the checklist, but was conducted daily). The distribution of the checklist of the checkli