

US EPA ARCHIVE DOCUMENT

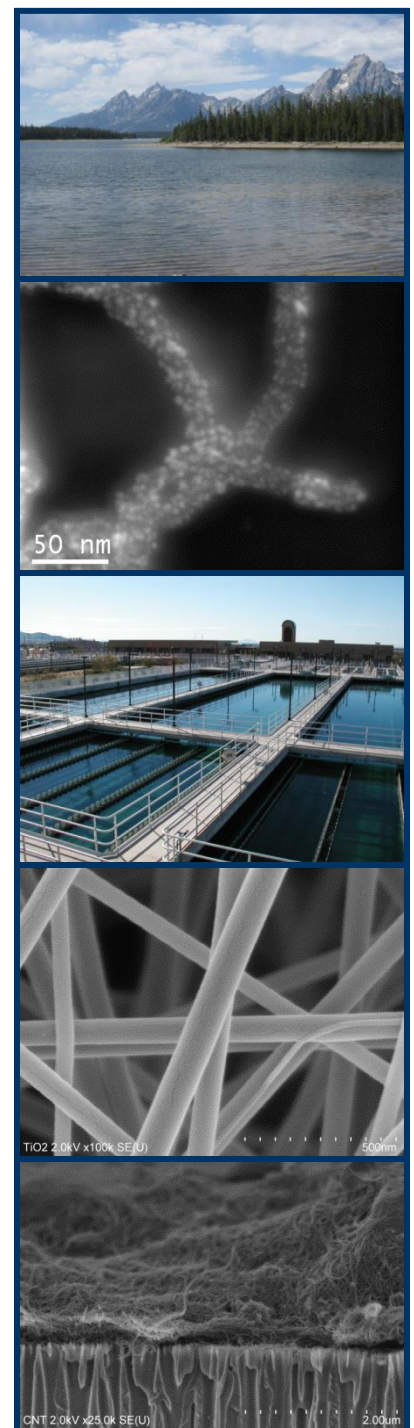
Research and Demonstration of Electrospun Nanofiber Filters: Multifunctional, Chemically Active Filtration Technologies for Small-Scale Water Treatment Systems

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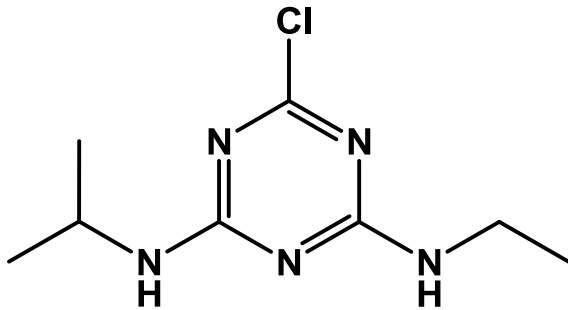
Our motivation: Water quality in Iowa small drinking water systems

Arsenic

(8% samples > MCL)

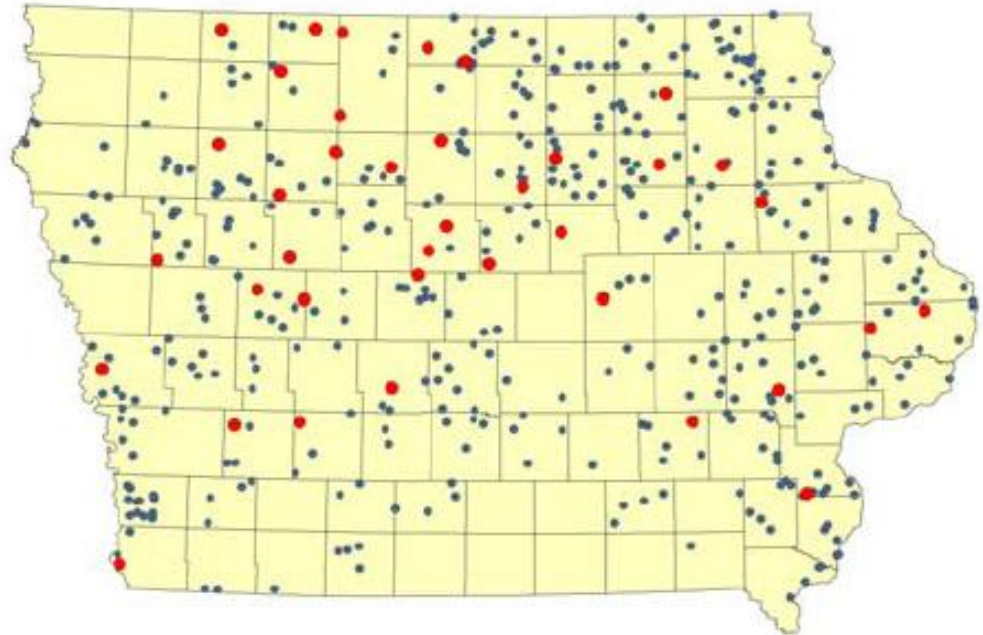
Red: ≥ 0.01 mg/L (MCL)

Blue: 0.001-0.009 mg/L



Atrazine

(8% of samples)



Iowa Statewide Rural Well Water Survey Phase 2 (SWRL2): Results and Analysis. UI Center for Health Effects of Environmental Contaminants (2009)

There is a niche for nanotechnology in small systems

A Water Treatment Engineers Introduction to Reactive Nanomaterials

Must harness	How?	Water quality benefit
High reactivity	Specific surface area & size effects	Efficiency toward historically recalcitrant pollutants
Scale	Size- & morphology-directed synthesis	“Small” promotes point-of-use & decentralized technologies
Tunable properties	Surface & bulk chemical modification	Tailored treatment to source water quality
“Bottom up” processing	Design & fabrication of hetero-nanostructures	Multi-functionality shrinks the process train

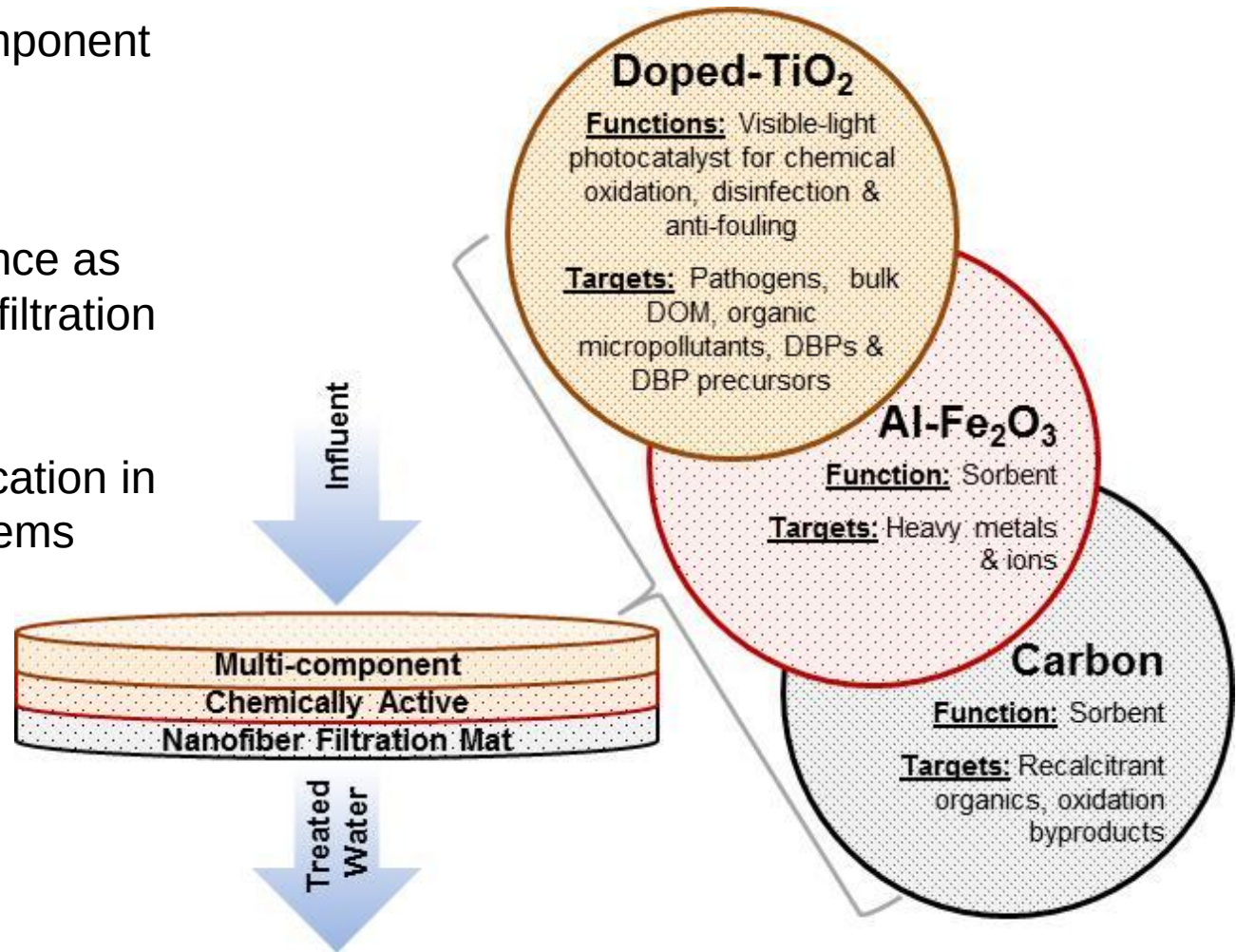
Caveats:

1. Nanomaterials will not fulfill promise in treatment applications as suspensions
2. The most reactive nanomaterial may not be optimal for application

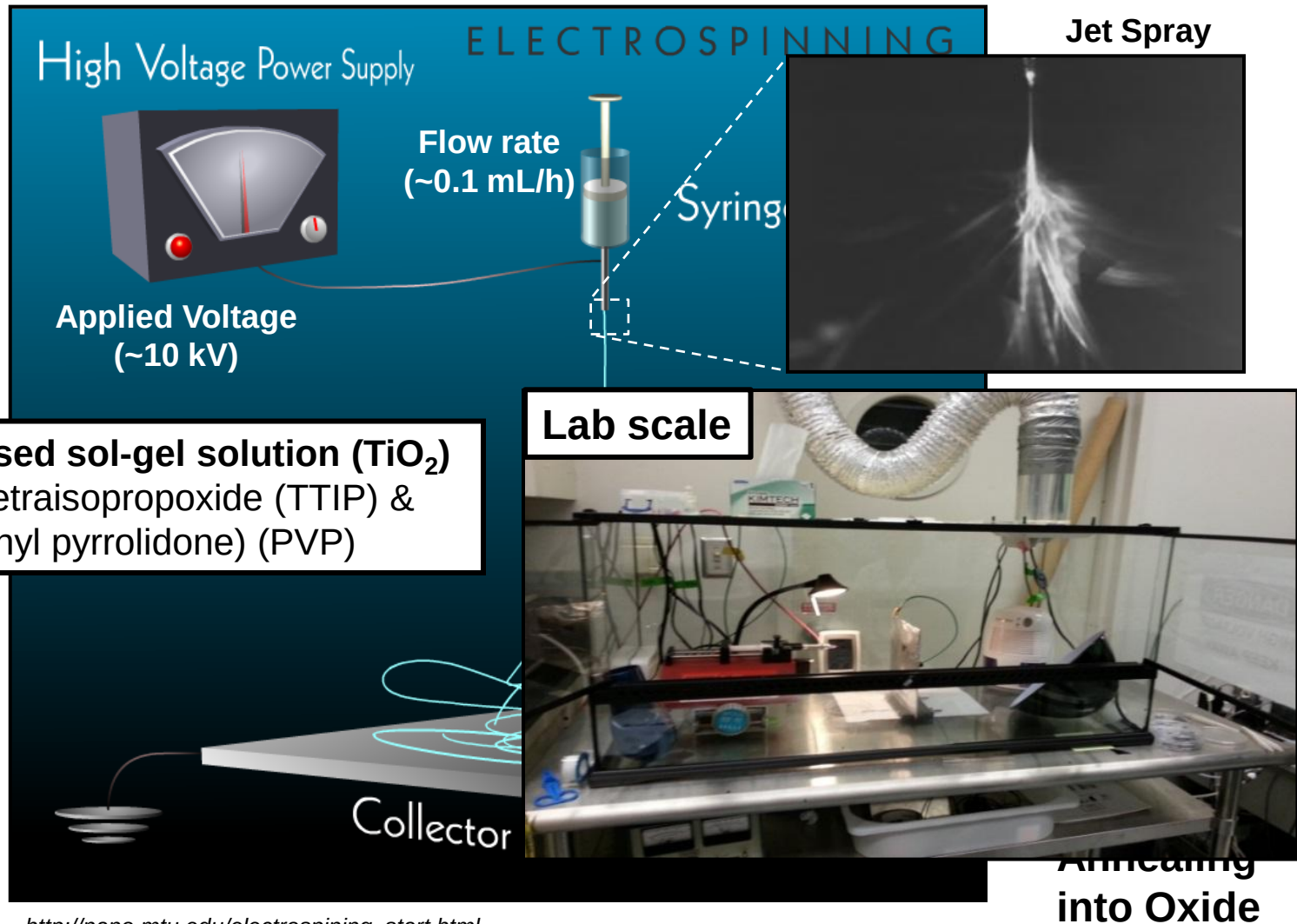
EPA STAR Small Systems Project: Chemically reactive filtration technologies

Overarching objectives:

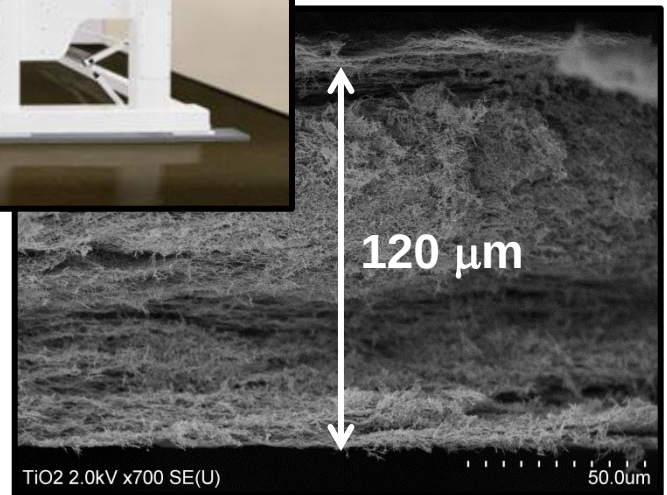
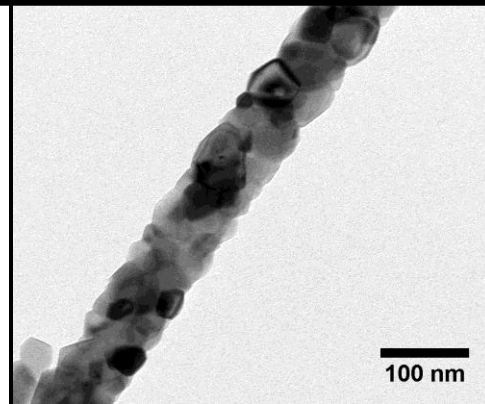
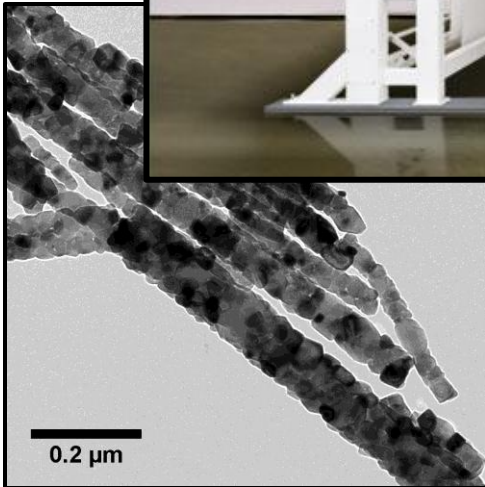
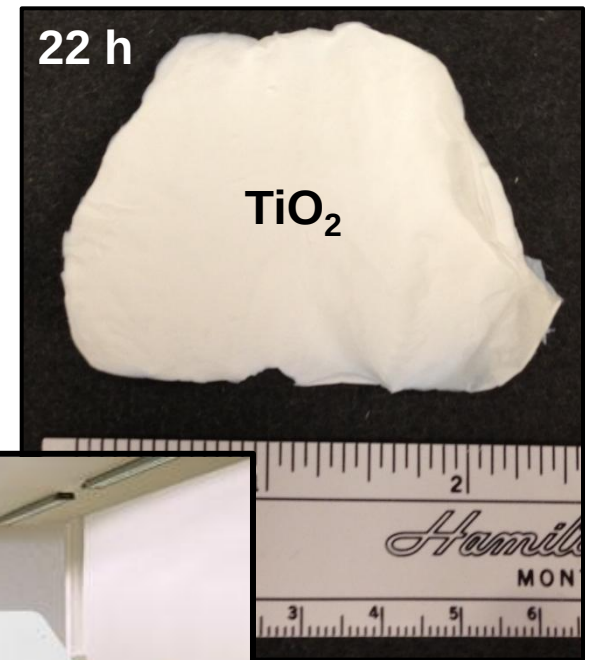
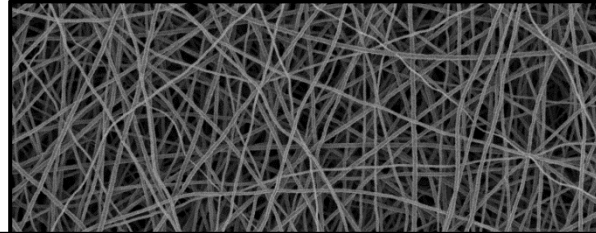
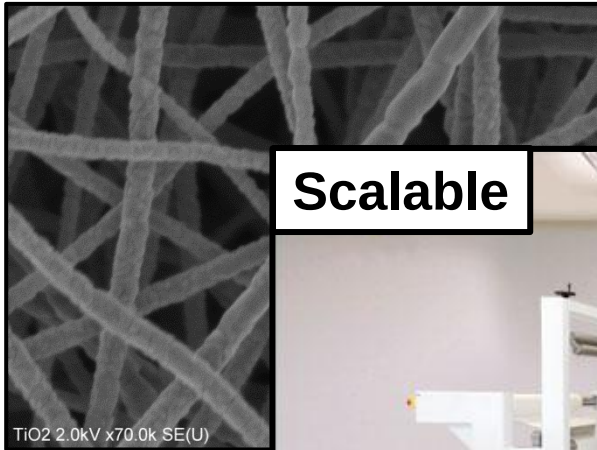
- 1) **Fabricate** multi-component nanofiber mats via electrospinning,
- 2) **Optimize** performance as chemically reactive filtration materials
- 3) **Demonstrate** application in POU and POE systems



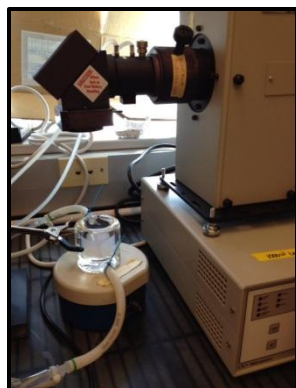
Electrospinning of chemically active (inorganic) nanofibers for treatment



Electrospinning yields mats ideal as reactive coatings

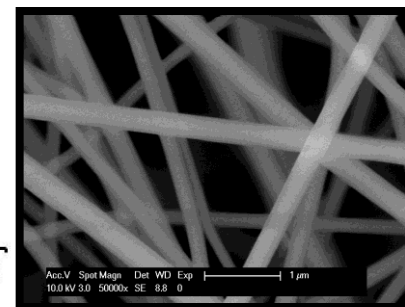
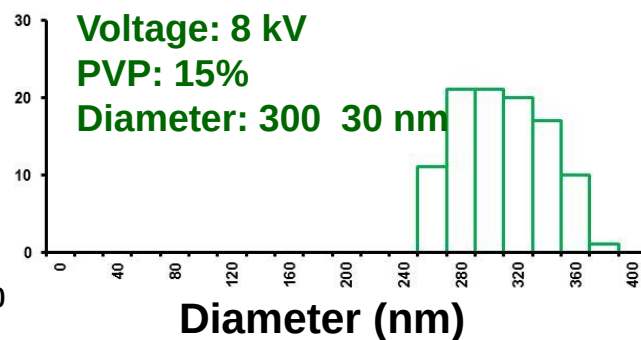
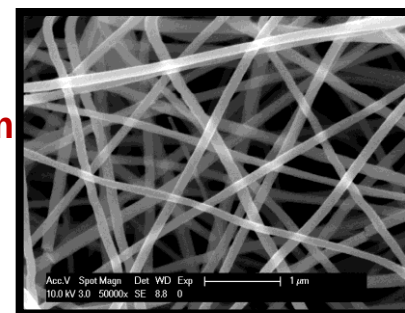
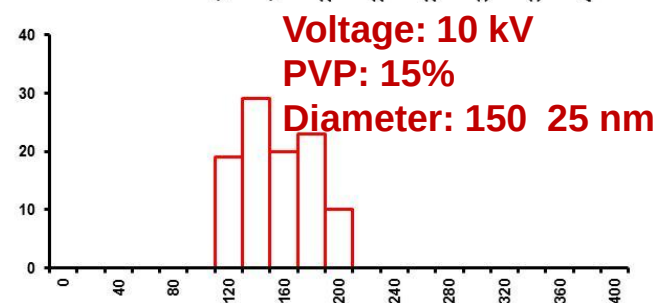
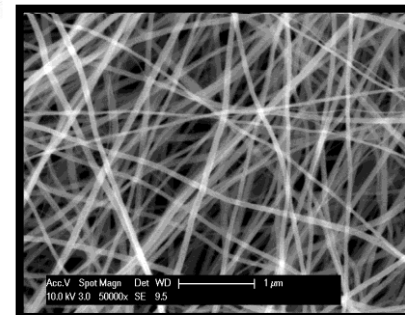
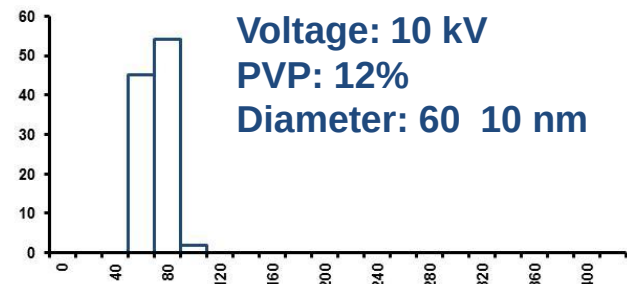
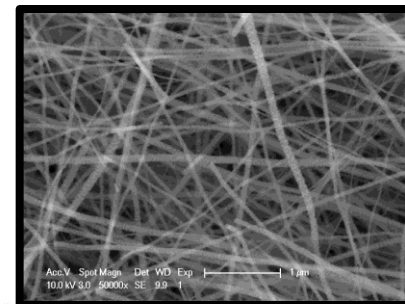
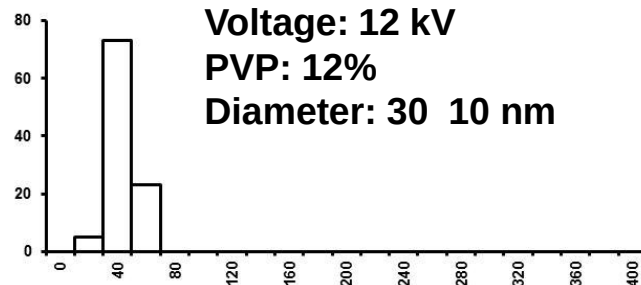
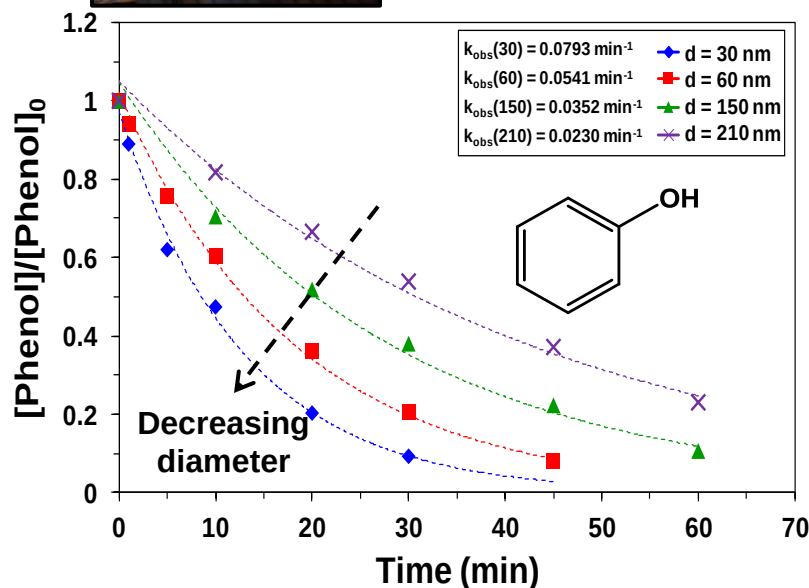


Nanofiber size can be tailored during synthesis

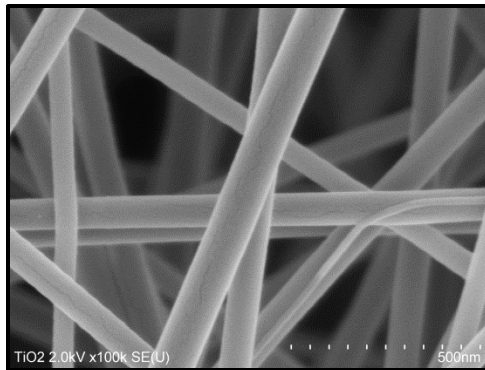


Light source:
Suntest Solar Simulator
or
1000W Xe ($\lambda > 305$ nm)

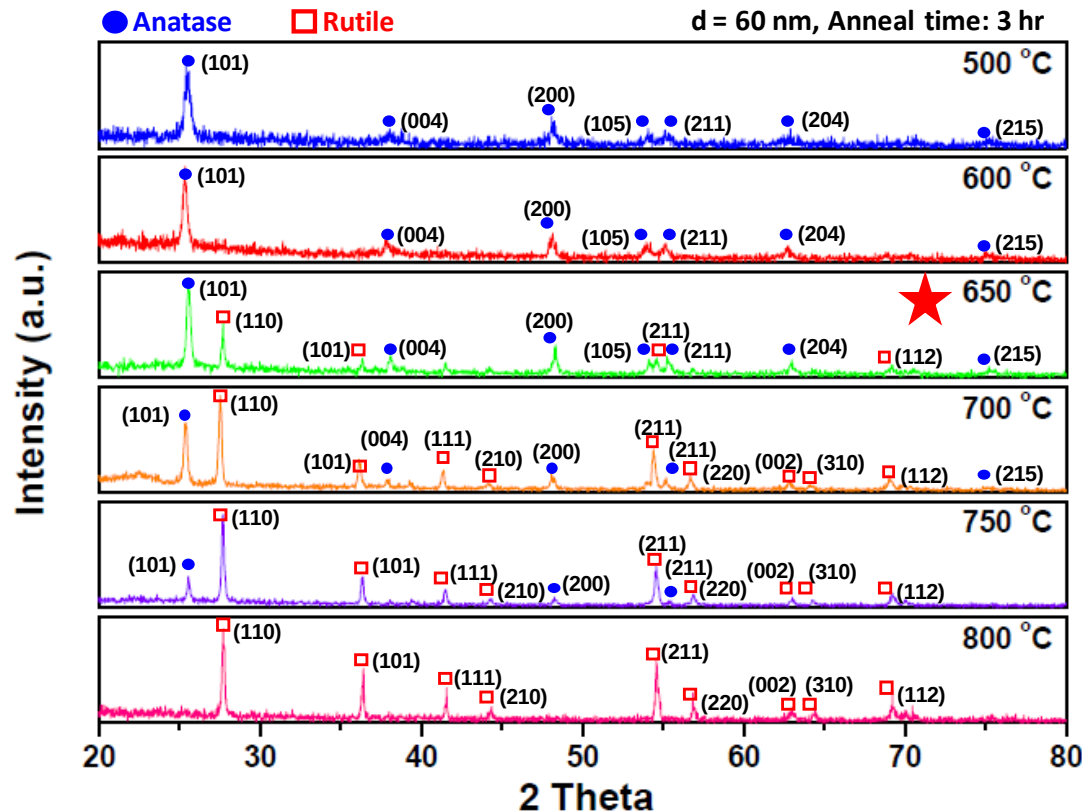
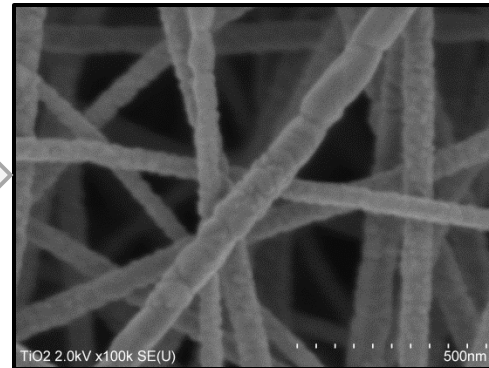
Solution:
pH 7
0.1 g/L TiO₂



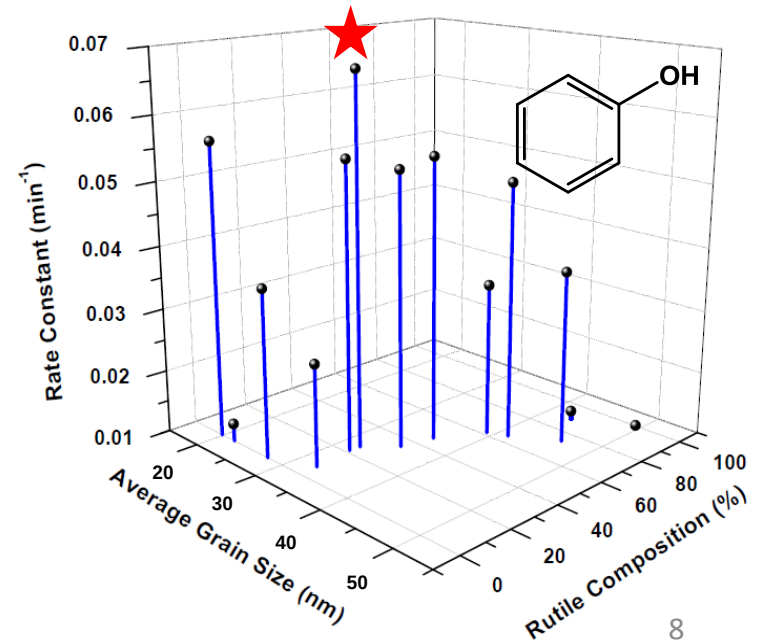
Nanofiber crystallinity and grain size tunable during annealing



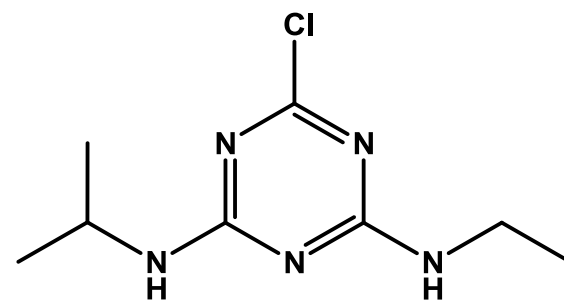
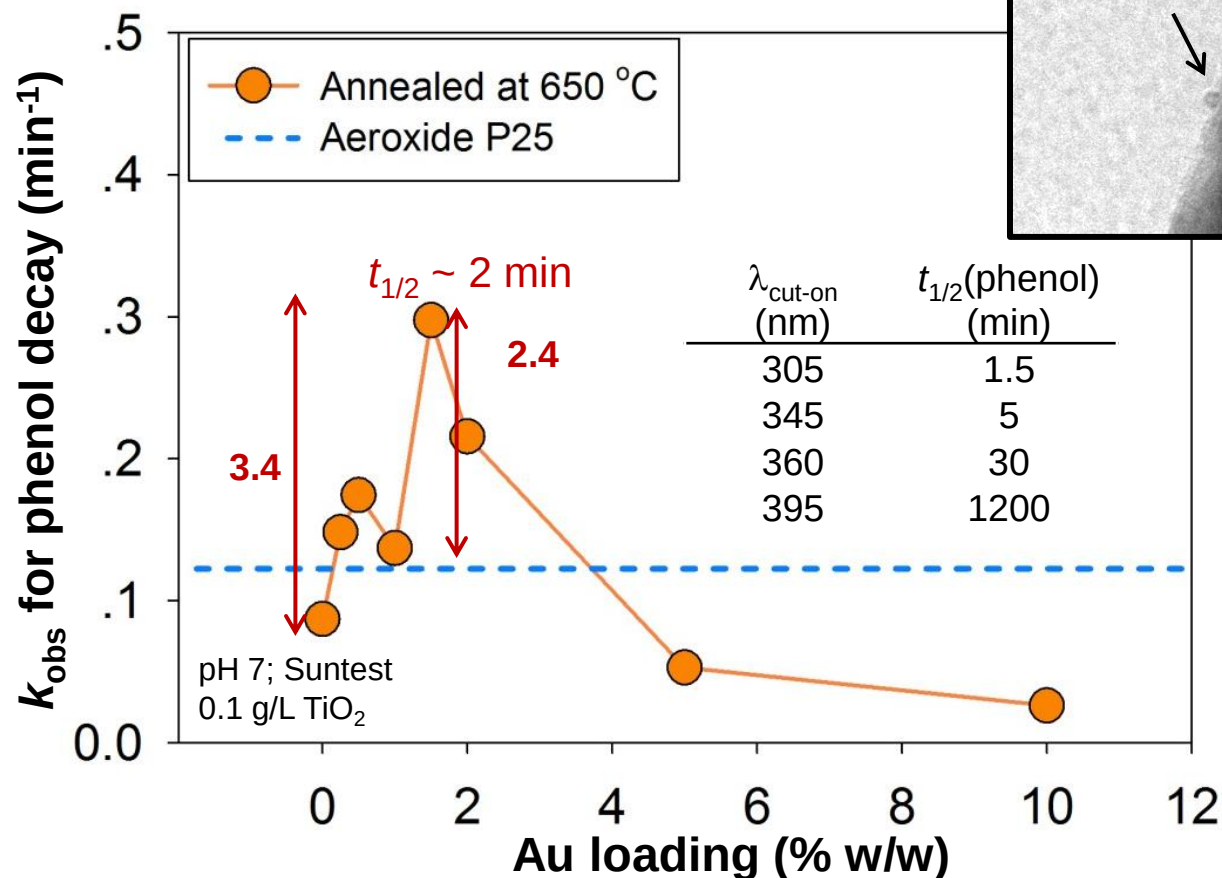
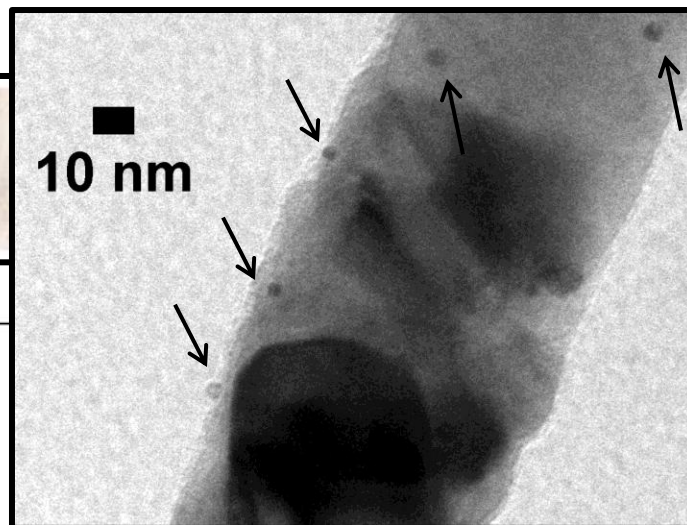
650 °C
3 h



Optimal ratio:
26:74 (0.5)% rutile:anatase



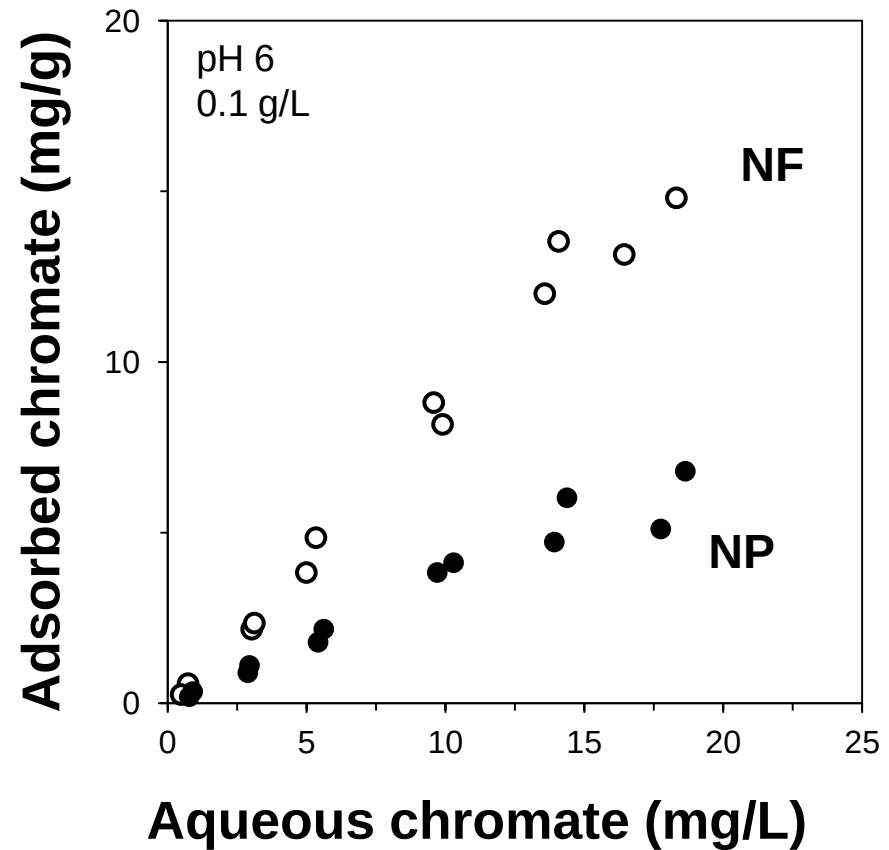
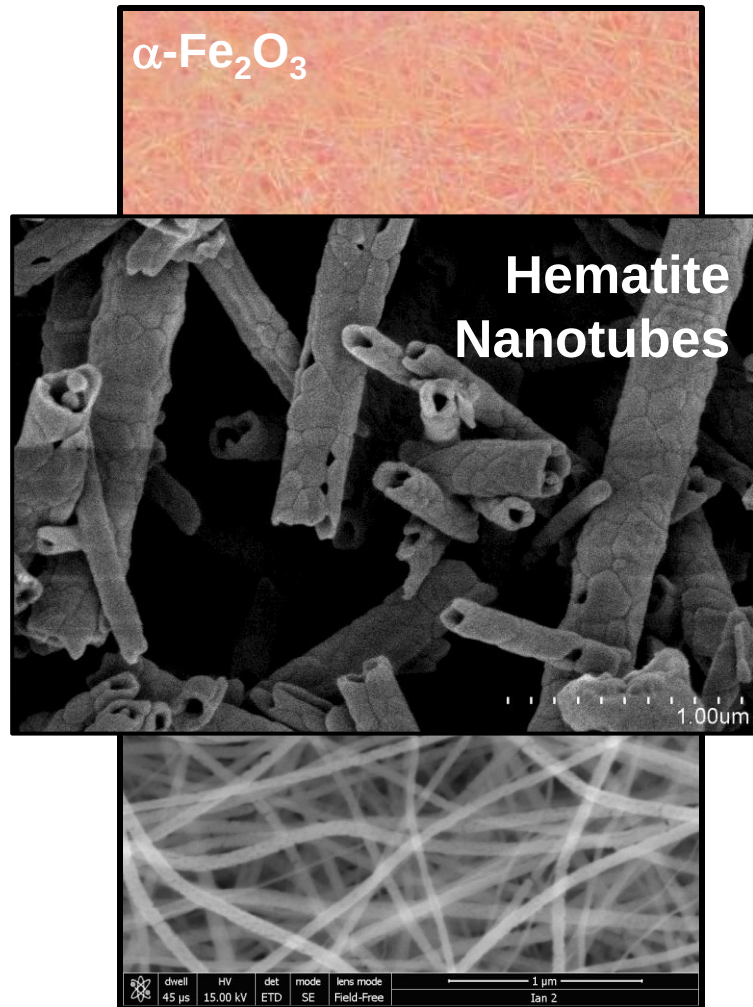
Surface-doped Au/TiO₂ is highly photoactive nanofiber catalyst for organic oxidation



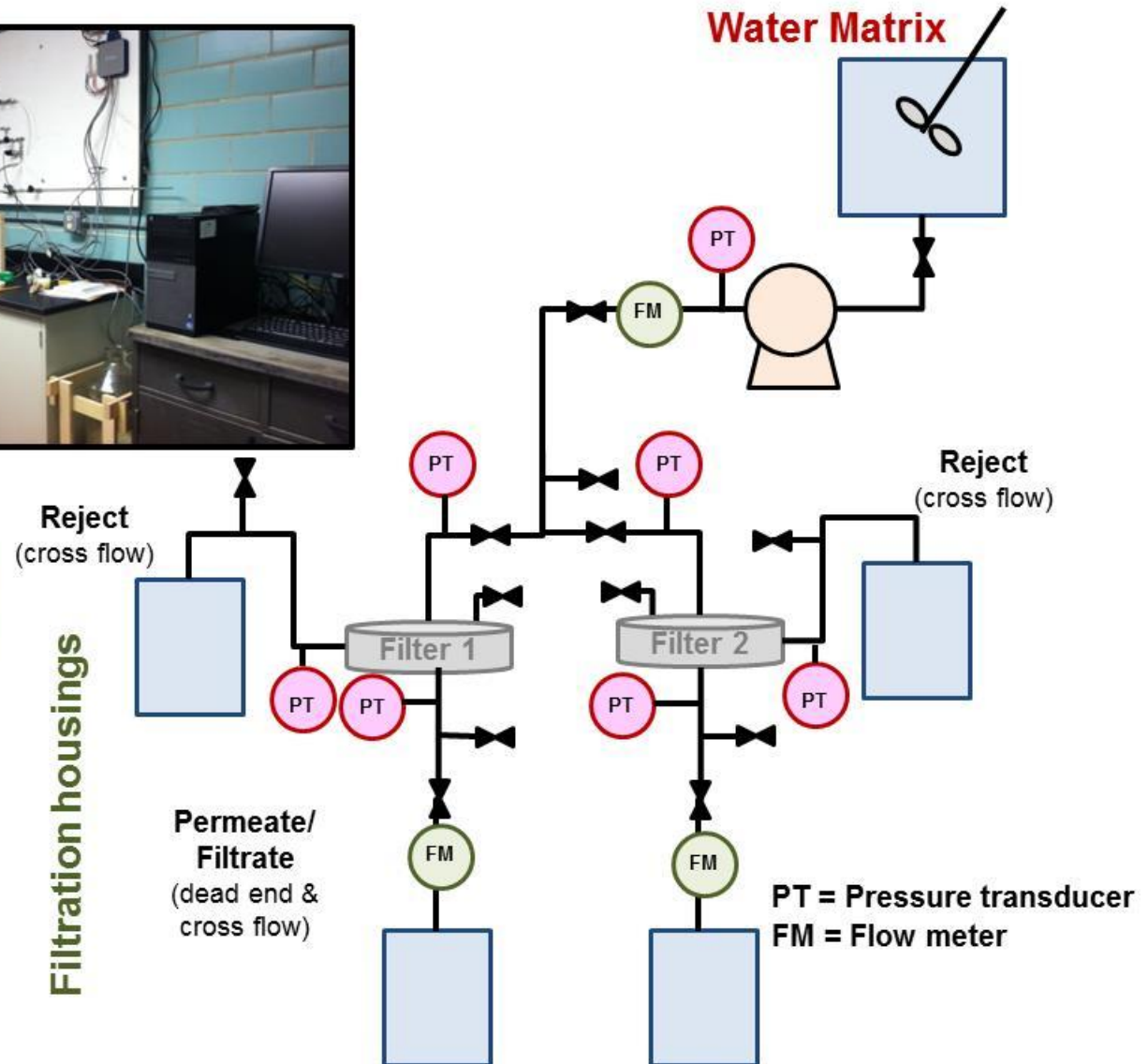
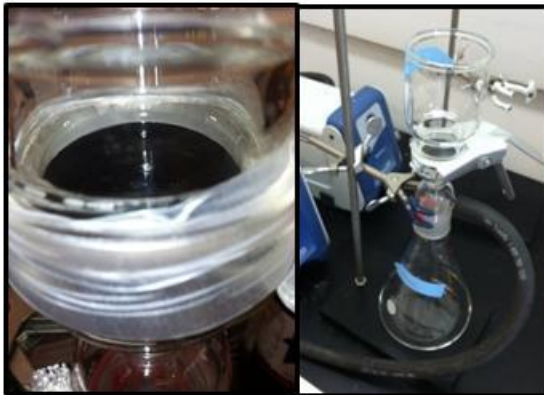
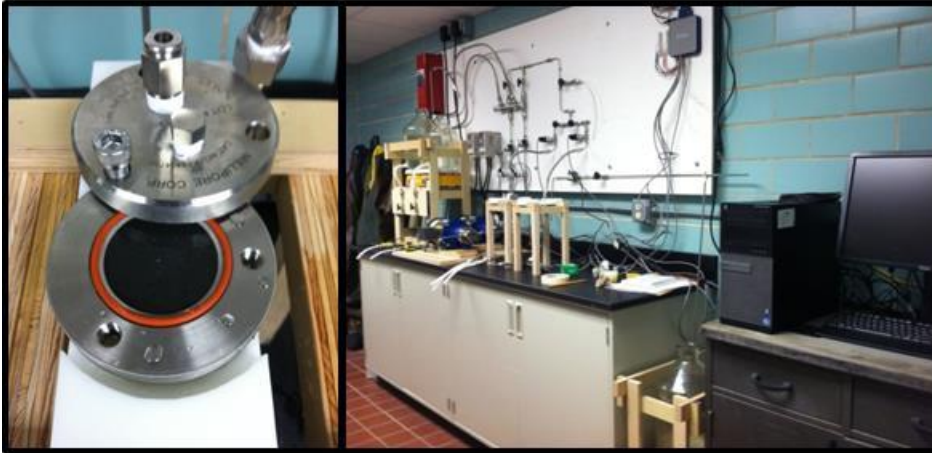
Atrazine

$t_{1/2}(\text{P25}): 10 \text{ min}$
 $t_{1/2}(1.5\% \text{ Au/TiO}_2): 15 \text{ min}$

Ongoing work: Iron oxide and carbon sorbent nanofibers



Ongoing: performance testing in flow through systems



Take home points on progress to date

- Electrospinning is highly versatile and scalable route to reactive nanofiber mats
- Nanofiber routes may be a responsible route to environmental applications of nanotechnology
- Challenges remain in optimizing reactivity (e.g., surface area) and robustness
- Future work will explore long term performance testing, reactive longevity and regeneration

Acknowledgements

Cwiertny Group members

Dr. Danmeng Shuai: Electrospinning

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Katie Greenstein: Electrospinning

Jason Haase: CNT microfiltration

Kathy Peter: Electrospinning

Caylyn Lanzl: Nano iron oxide reactivity

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Dr. Tim Mattes: Iowa

Dr. Vicki Grassian: Iowa

Jon Durst: Iowa

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WHERE DISCOVERIES BEGIN

