

US EPA ARCHIVE DOCUMENT

SOURCES, COMPOSITION, VARIABILITY AND TOXICOLOGICAL CHARACTERISTICS OF COARSE ($PM_{2.5-10}$) PARTICLES IN SOUTHERN CALIFORNIA

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Acknowledgements

□ Co- Investigators

□ **USC:** Payam Pakbin, Kalam Cheung, Zhi Ning, Katharine Moore

□ **University of Wisconsin Madison:** James Schauer, Martin Shafer

□ **UCLA:** Arthur Cho , Ning Li

□ **EPA Agreement Number:** RD-83374301-0

□ **Project Period:** November 1, 2008 to October 31, 2012

Outline

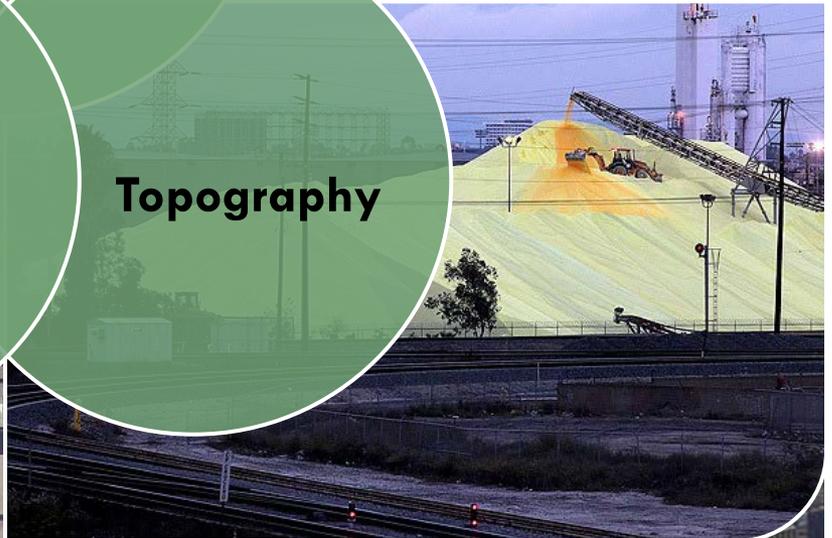
- **Introduction**
 - Rationale
 - Objectives
 - Project Methodology
- **Key Findings/Results**
 - **Physical** Characterization of **Coarse PM (CPM)**
 - **Chemical** Characterization of CPM
 - **Toxicological** Characterization CPM
 - **Trends** in CPM Concentration and Chemical Speciation
- **Conclusions**
 - Limitations
 - Recommendations

Objectives

- What are the **spatial, diurnal and seasonal** differences in coarse PM mass and chemical composition in **rural and in urban areas** of the Los Angeles Basin?
- What is the **fraction** of chemically speciated PM that **penetrates indoors**?
- How do the **chemical characteristics** of coarse PM collected in each of the above environments determine their toxicity?
- What are the **trends of coarse PM mass and composition** over the past 15+ yrs of monitoring in the LA Basin?
- What would be the most **effective ways of regulating** coarse PM?



Sources



Meteorology

Topography

USC Coarse PM Study

Comprehensive Investigation

12 months

24-hour samples

10 sampling sites

Intensive Investigation

Summer and winter

Diurnal samples

3 sampling sites

Physical Characterization

- What causes the spatial and seasonal trends in CPM mass?
- Has the mass concentration changed over the years?
- Is PM_{10} a good surrogate of CPM?

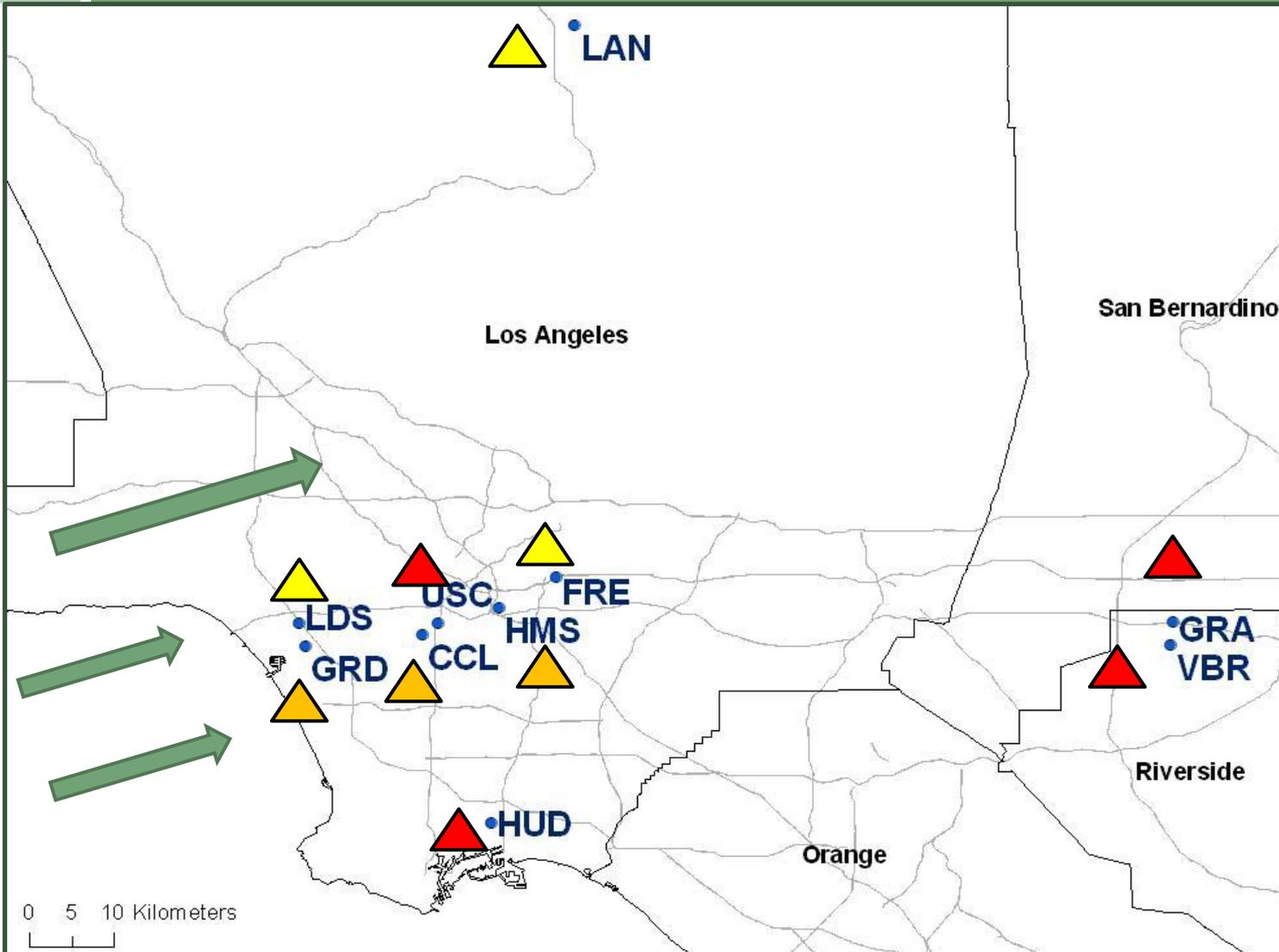
Chemical Characterization

- Is the chemical composition of CPM different from $PM_{2.5}$?
- What are the sources and formation mechanisms of CPM?
- Has the chemical profile of coarse particle changed over the years?

Toxicological Characterization

- What are driving the toxicity of coarse particles?
- Is there a link between source, composition and toxicity of CPM?

Mass Concentration - Overview



Annual mean
CPM levels

 0-10 $\mu\text{g}/\text{m}^3$

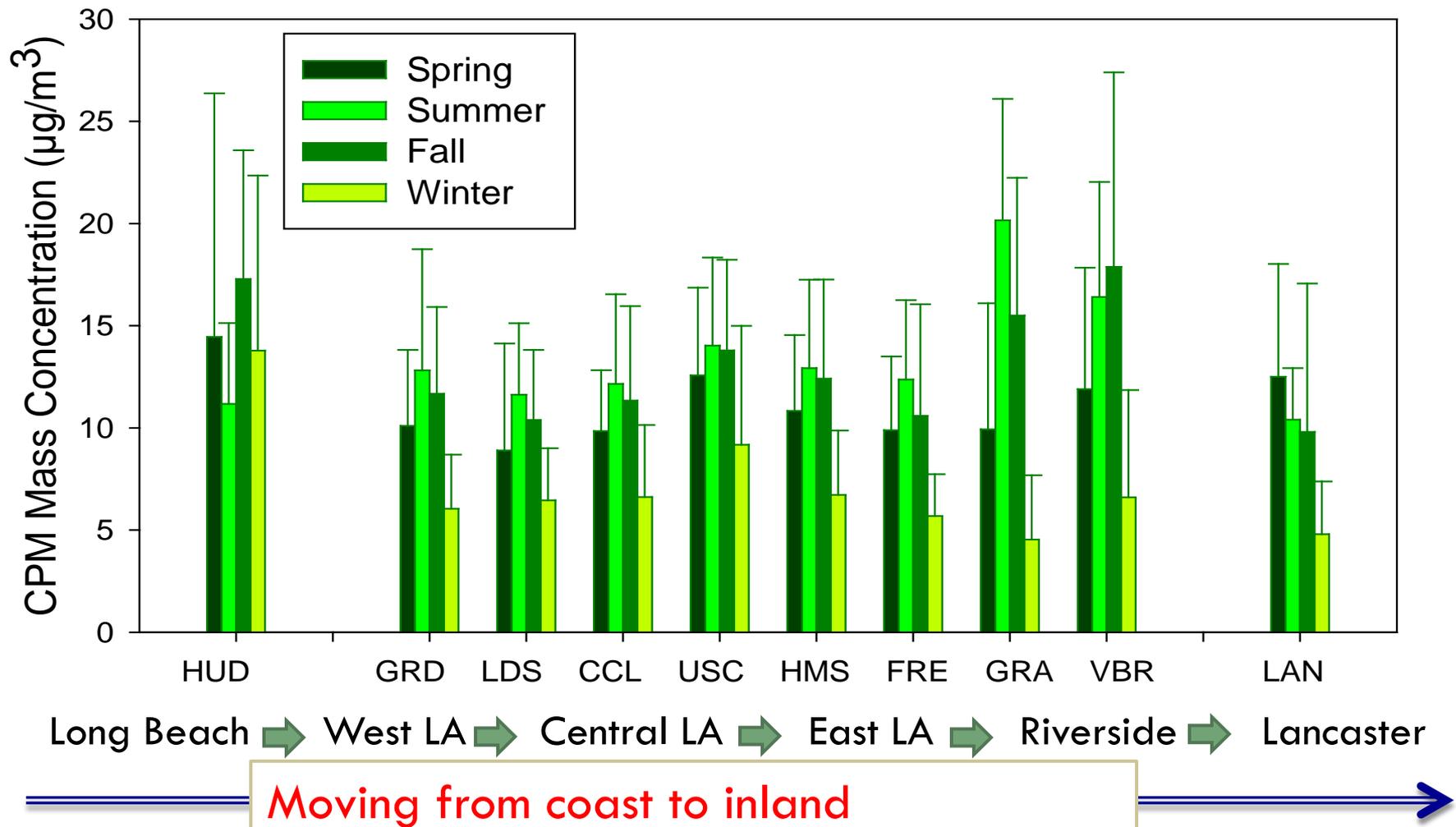
 10-12 $\mu\text{g}/\text{m}^3$

 12-15 $\mu\text{g}/\text{m}^3$

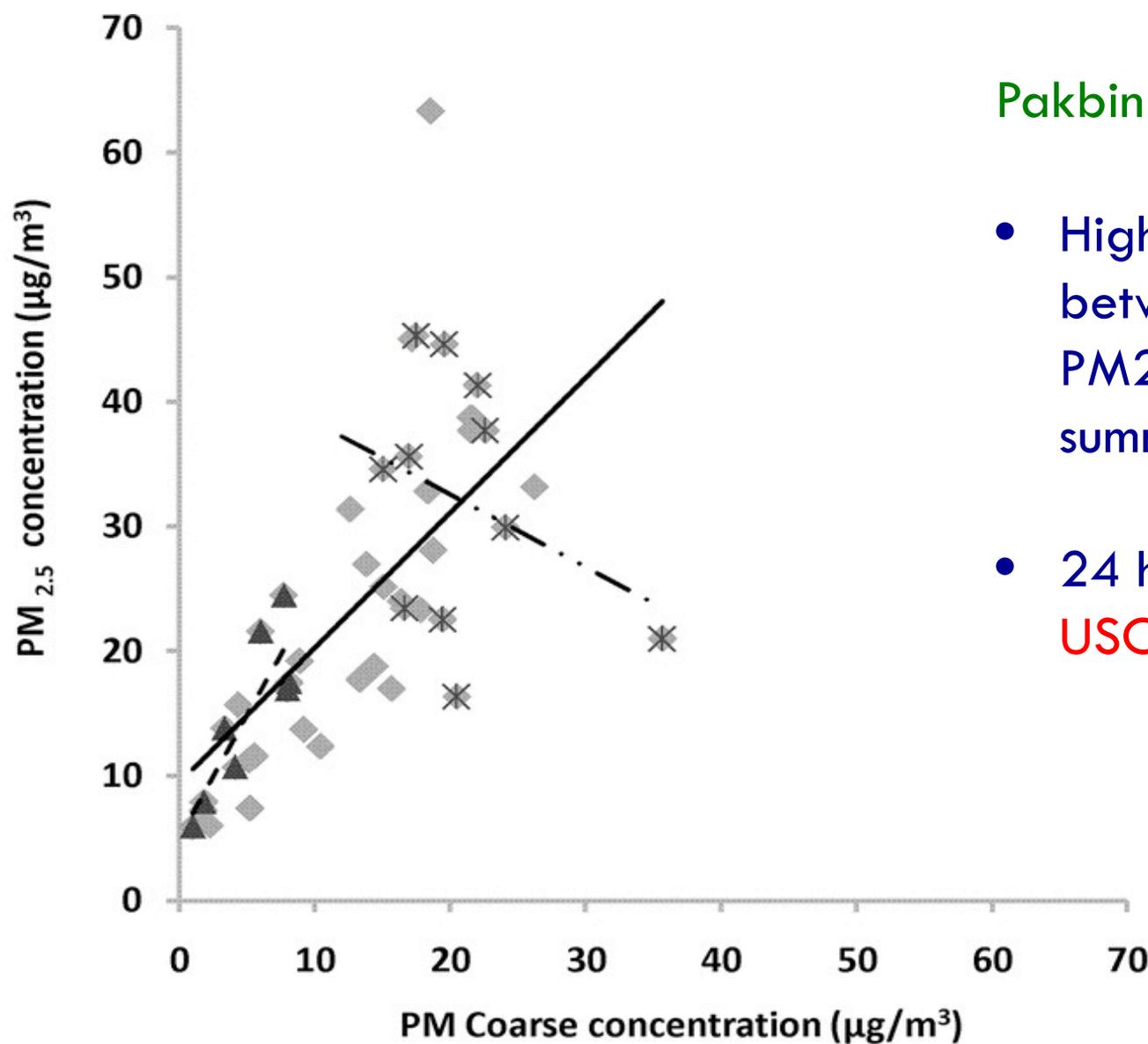
0 5 10 Kilometers

Spatial and Seasonal Trends in CPM Mass

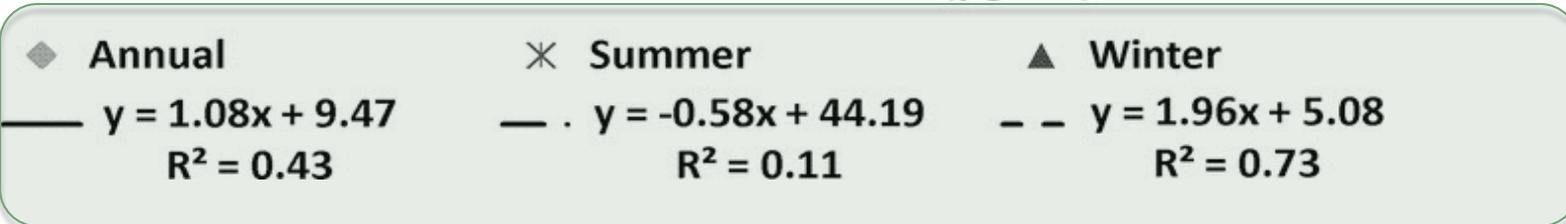
(*Cheung et al 2010*)



Pakbin et al AS&T 2010

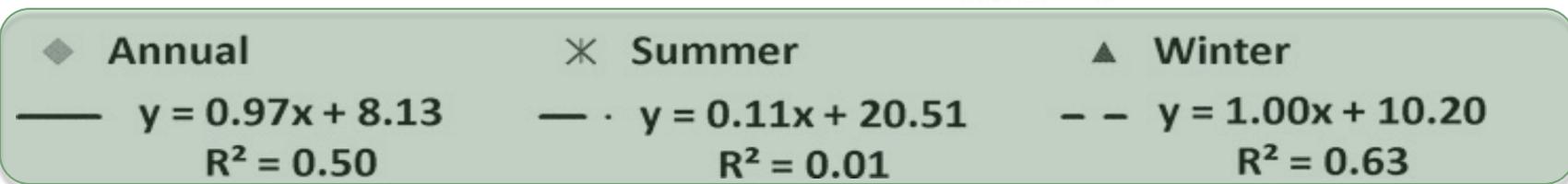
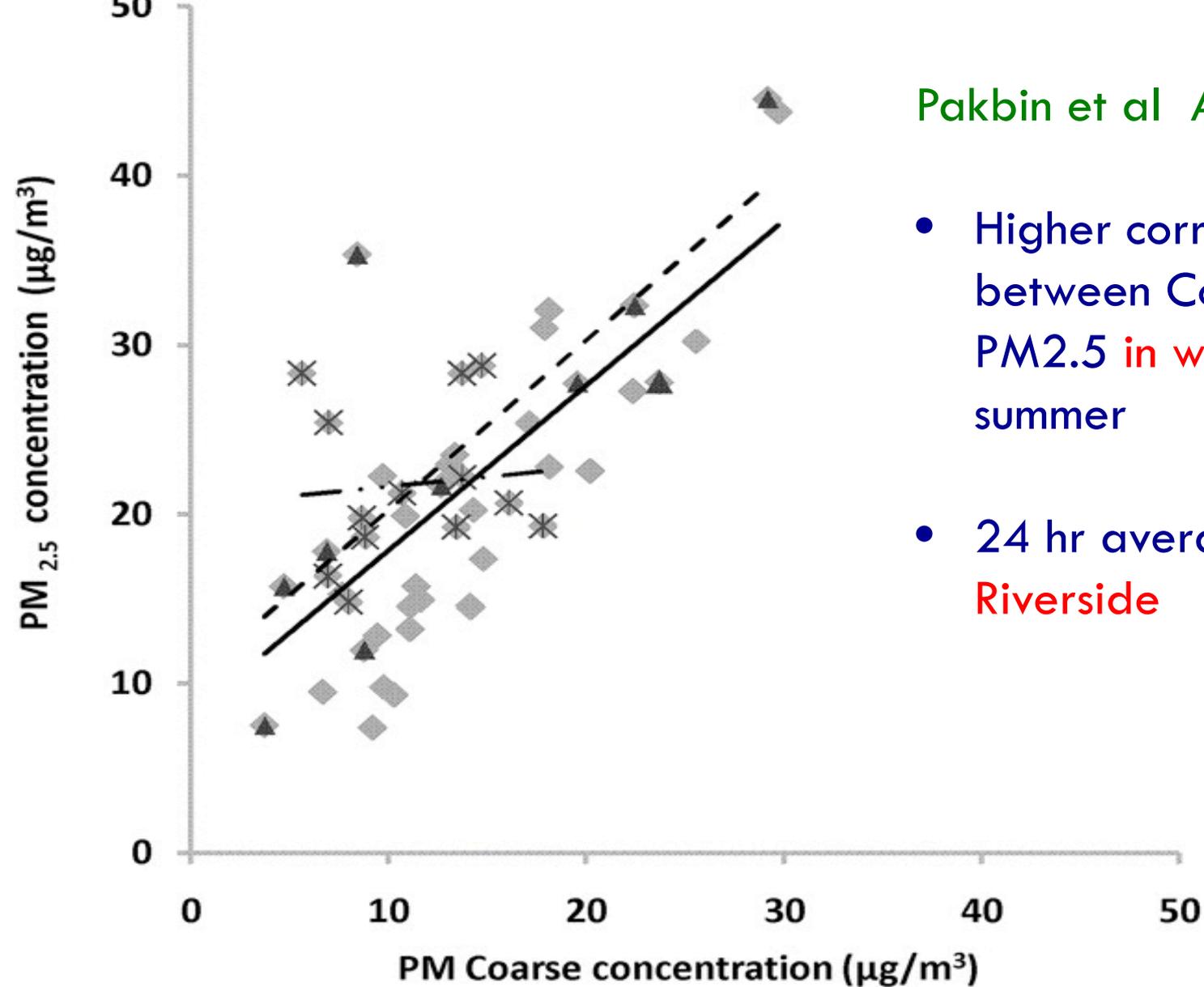


- Higher correlation between Coarse and PM_{2.5} in winter than summer
- 24 hr average Data at USC



Pakbin et al AS&T 2010

- Higher correlation between Coarse and PM_{2.5} in winter than summer
- 24 hr average Data at Riverside



Historical Trend

- PM_{10} and $PM_{2.5}$ data from CARB
- $PM_{10-2.5}$ concentrations from the subtraction method
- Year 1999 to 2009, 24-hr concentration

The screenshot shows the California Environmental Protection Agency (CalEPA) Air Resources Board website. The main navigation bar includes links for Home, Reducing Air Pollution, Air Quality, Business Assistance, Laws & Regulations, and Health. A search bar is located in the top right corner. The page title is "Air Quality Data Query Tool" and it includes a "This page last reviewed April 30, 2010" notice.

The interface features three tabs: "Daily Data", "Hourly Data", and "Special Reports". The "Daily Data" tab is active. The form consists of five steps:

- Step 1:** Select a Parameter (Ozone) and Select a Statistic (ppm). A sub-selection for "Daily Average" is also present.
- Step 2:** Select an End Date (Date: 2012, April, 26).
- Step 3:** Select One (Location: --COUNTY--, --AIR BASIN--, --PART OF STATE--).
- Step 4:** Select a Type of Report (Week-at-a-Glance).
- Step 5:** Select the Sort Order (Basin/County/Site).

A "Retrieve Data" button is located at the bottom of the form. Below the form, there is a link: "Identify Data Changes Since Last Air Quality DVD or Data Download".

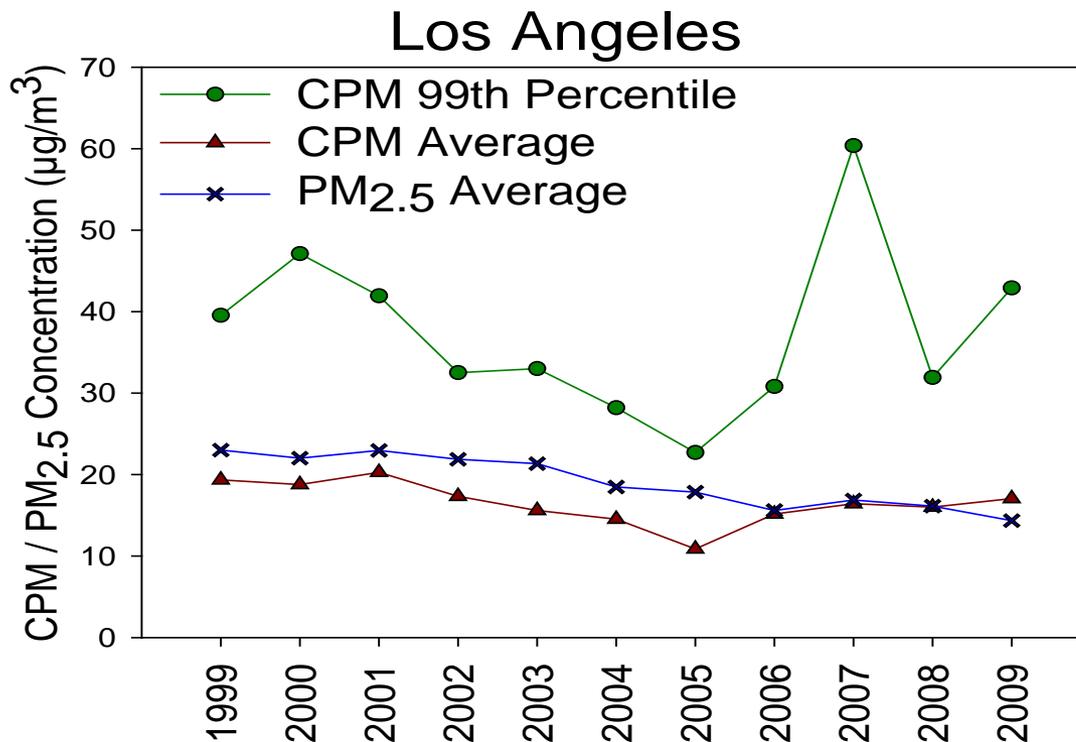
UP LINKS

- Air Quality (AQ) & Emissions
 - Air Quality and Meteorological Information System

PROGRAM LINKS

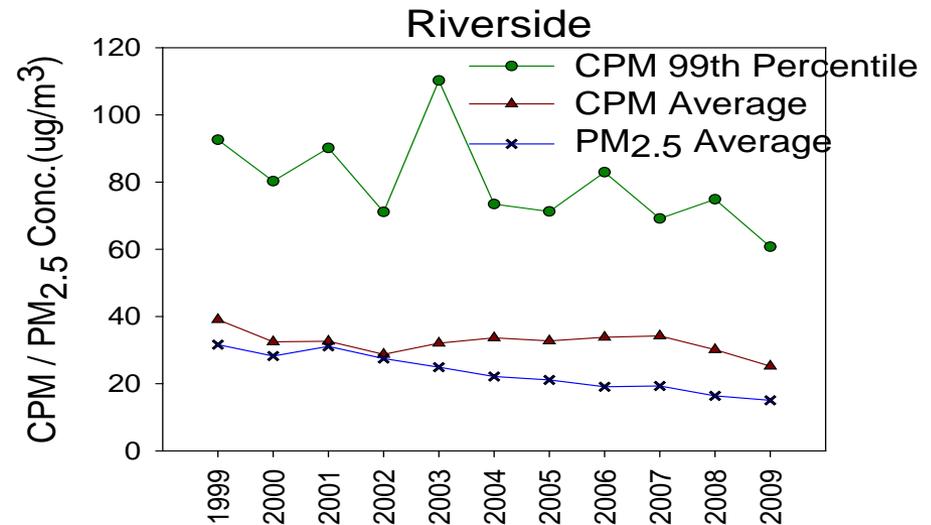
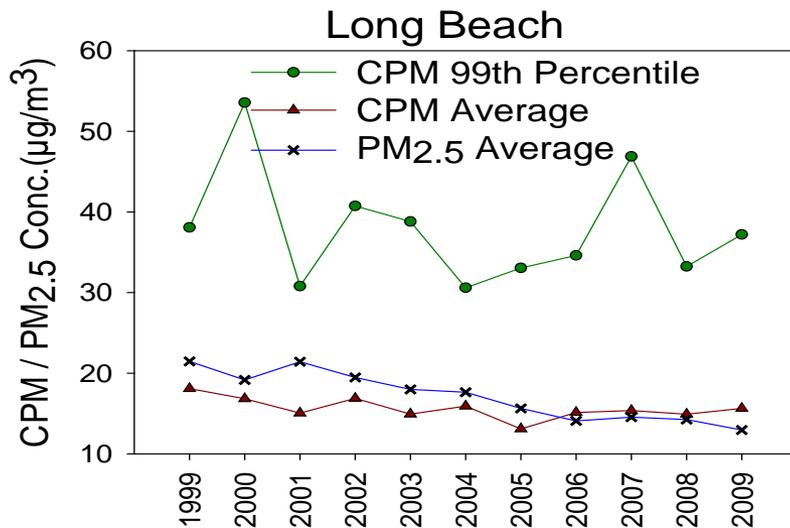
- Background
- District Responsibilities
- Maps
 - Ozone or PM2.5
 - Google
- Ozone
 - Latest Ozone
 - Latest Week's Ozone
 - Latest Year's Ozone
- Query Tools

Historical Trends



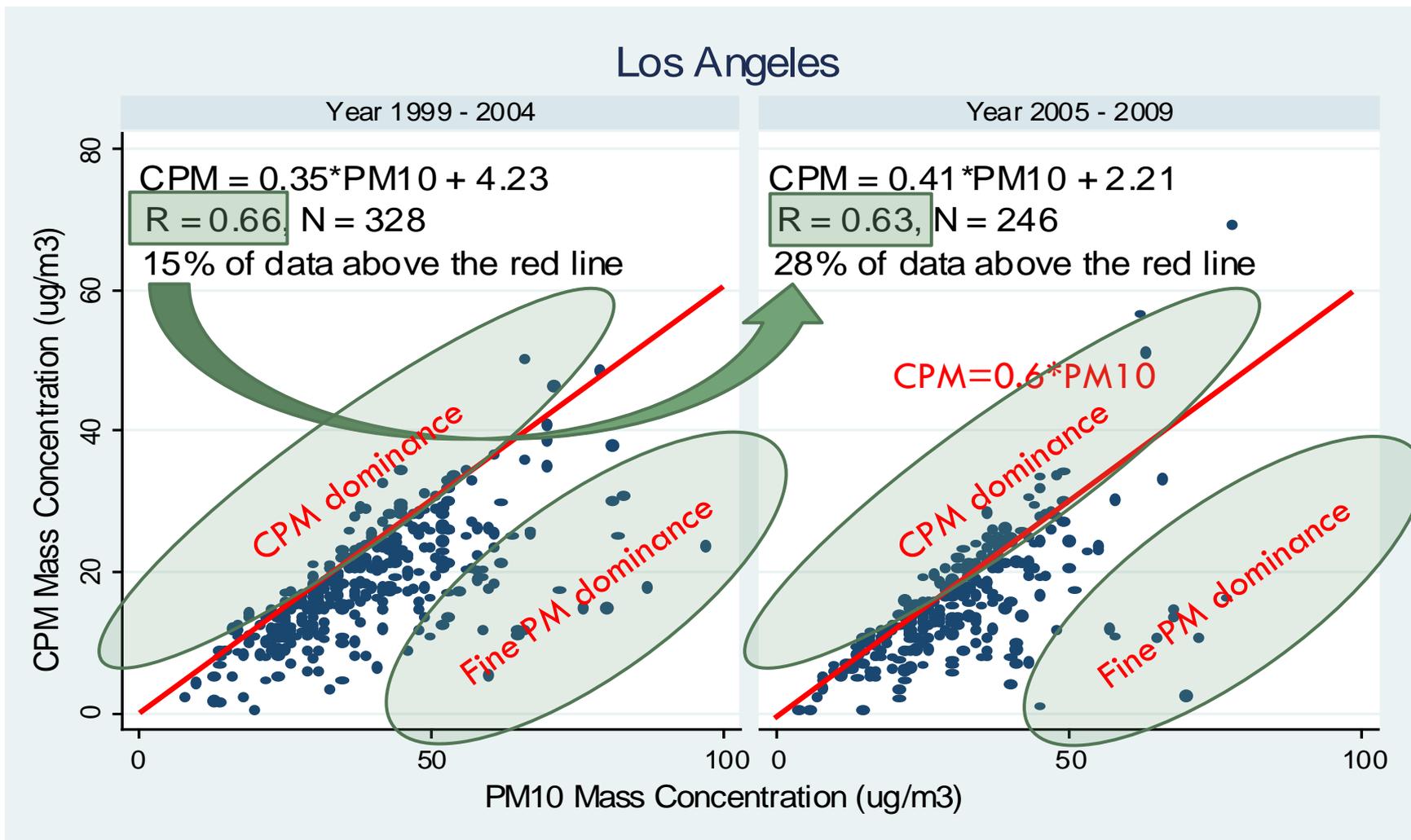
Downtown LA		Linear Regression	R ²	p
PM ₁₀	Average	$y = -1.33(\pm 0.12) x + 57.4(\pm 1.6)$	0.86	<0.001
PM _{10-2.5}	Average	$y = -0.39(\pm 0.23) x + 18.9(\pm 1.5)$	0.25	0.11
PM _{2.5}	Average	$y = -0.92(\pm 0.09) x + 24.6(\pm 0.64)$	0.91	<0.001

Historical Trends in Other LAB Areas



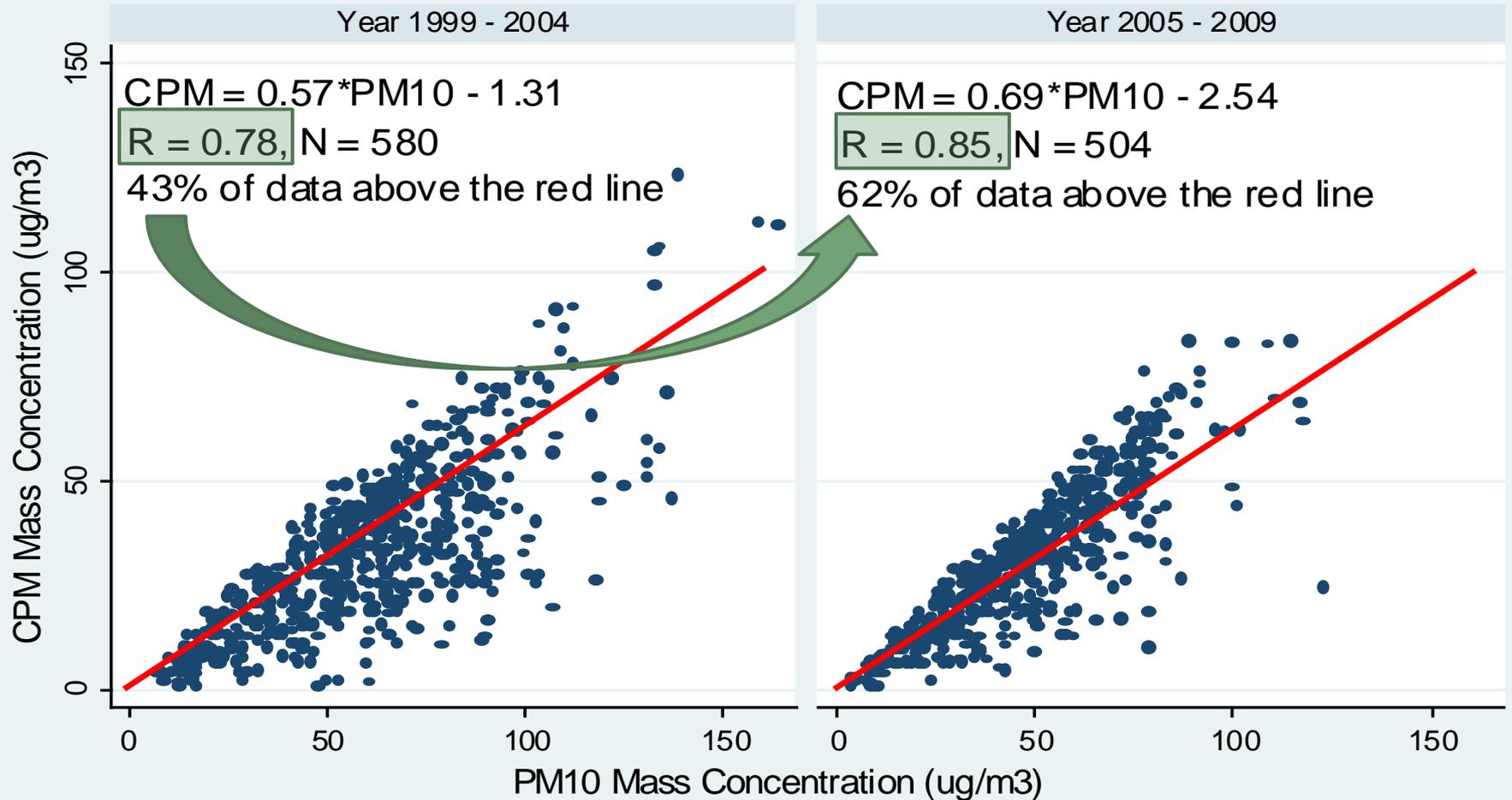
Long Beach		Linear regression	R ²	p
PM ₁₀	Average	$y = -0.76(\pm 0.11) x + 44.8(\pm 1.4)$	0.72	<0.001
PM _{10-2.5}	Average	$y = -0.22(\pm 0.11) x + 17.0(\pm 0.74)$	0.31	0.07
	Average	$y = -0.87(\pm 0.09) x + 22.4(\pm 0.61)$	0.91	<0.001
PM _{2.5}	Average	$y = -0.87(\pm 0.09) x + 22.4(\pm 0.61)$	0.91	<0.001
Riverside		Linear regression	R ²	p
PM ₁₀	Average	$y = -1.76(\pm 0.23) x + 84.6(\pm 3.0)$	0.75	<0.001
PM _{10-2.5}	Average	$y = -0.57(\pm 0.29) x + 35.7(\pm 2.0)$	0.30	0.08
	Average	$y = -1.69(\pm 0.12) x + 33.5(\pm 0.84)$	0.96	<0.001

PM₁₀ as a Surrogate of CPM



PM₁₀ as a Surrogate of CPM

Riverside



Summary

- Seasonal and spatial trend
- PM_{10} is a generally good surrogate for CPM
- The significant decrease in PM_{10} is driven mostly by fine PM

Physical Characterization

- What causes the spatial and seasonal trends in CPM mass?
- Has the mass concentration changed over the years?
- Is PM_{10} a good surrogate of CPM?

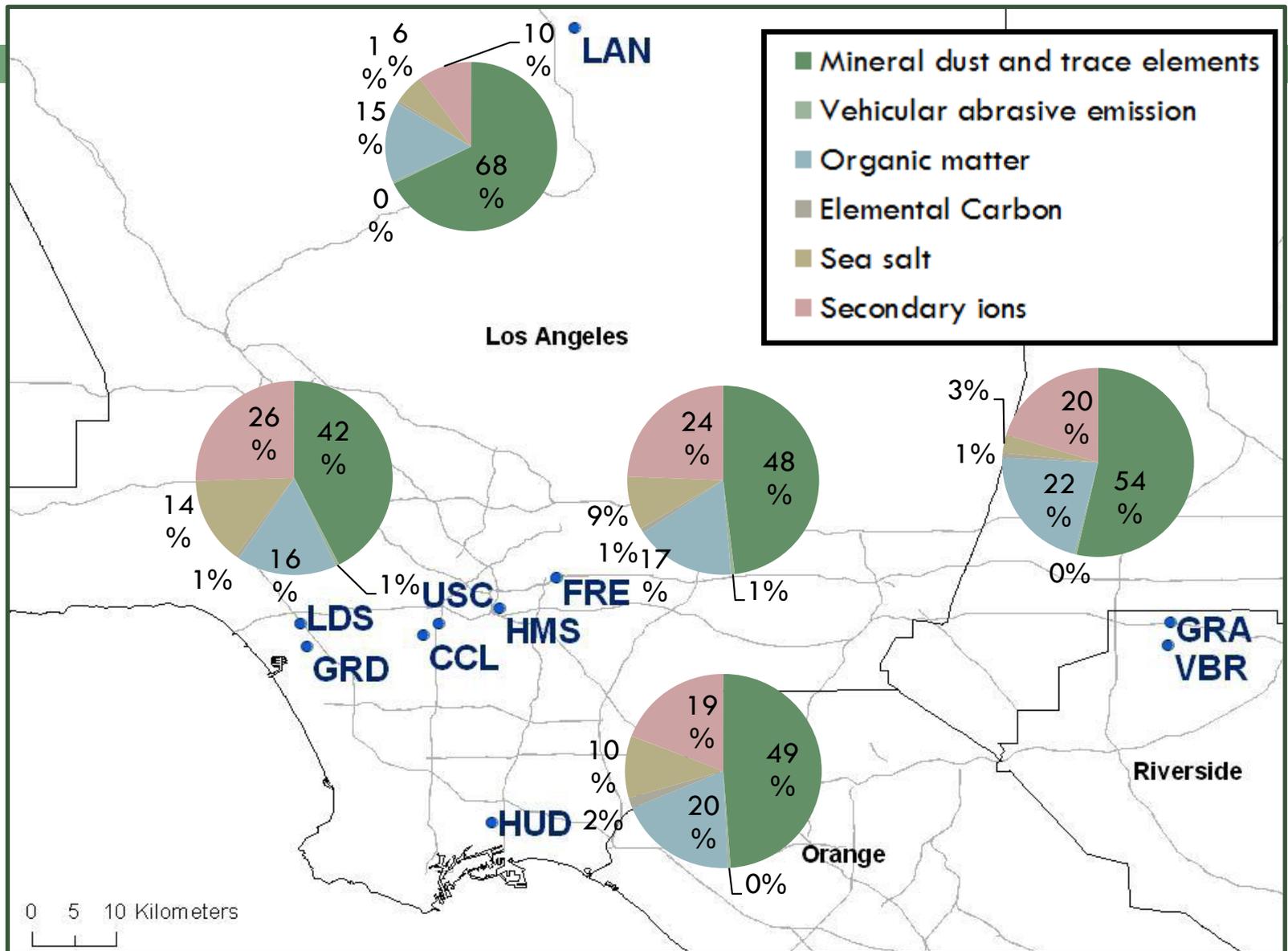
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- Is the chemical composition of CPM different from $PM_{2.5}$?
- What are the sources and formation mechanisms of CPM?
- Has the chemical profile of coarse particle changed over the years?

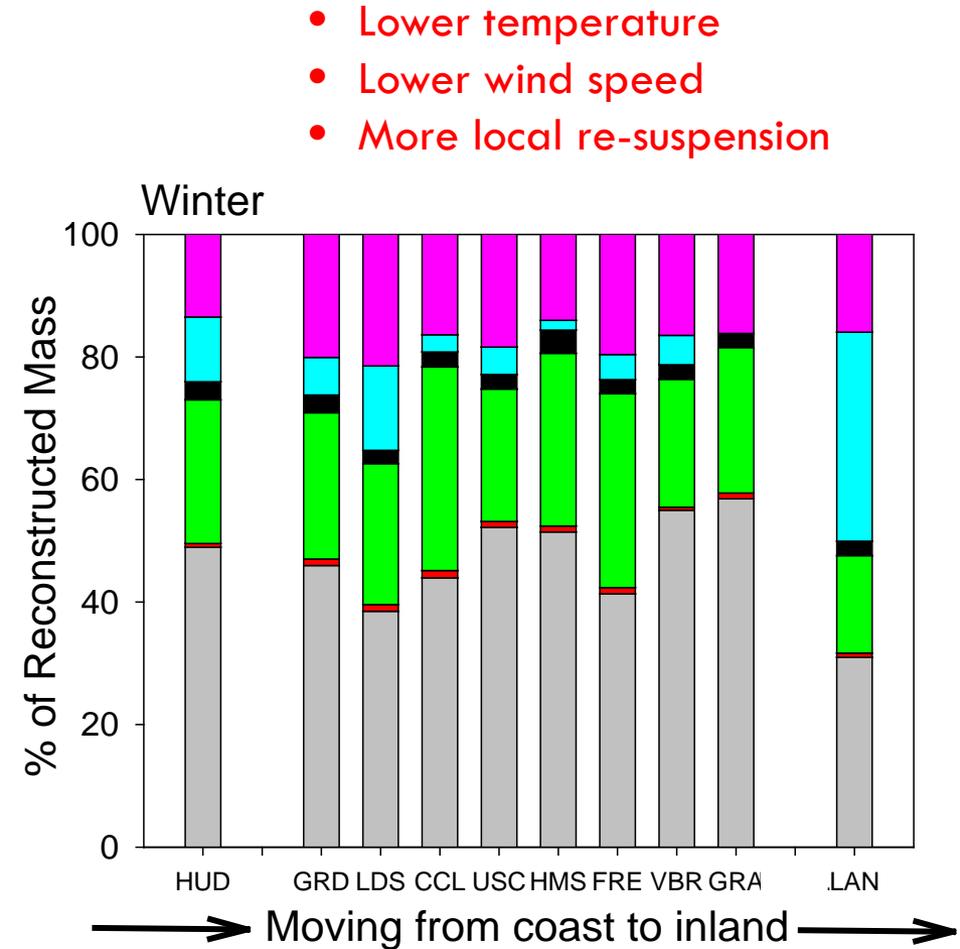
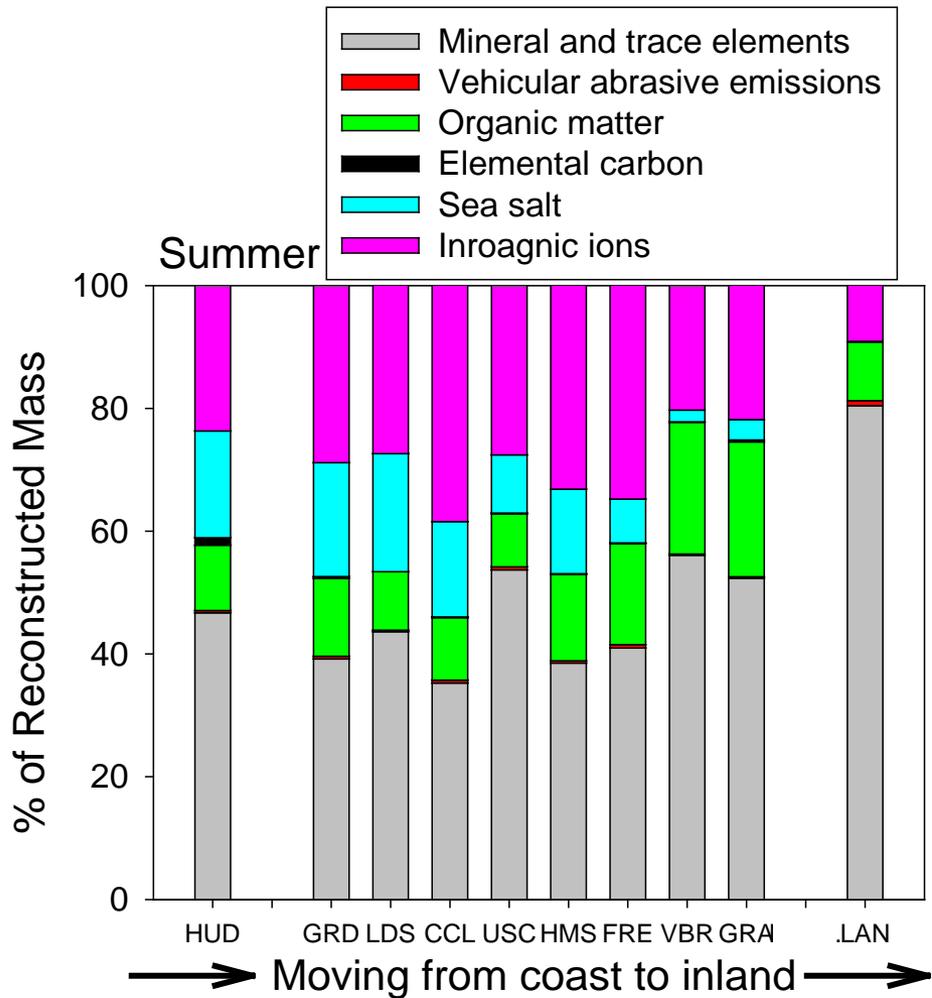
Toxicological Characterization

- What are driving the toxicity of coarse particles?
- Is there a link between source, composition and toxicity of CPM?

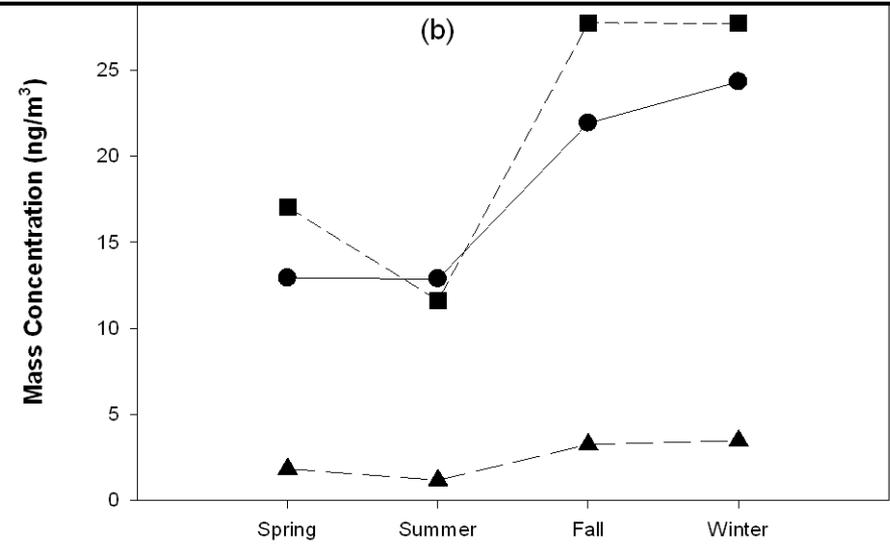
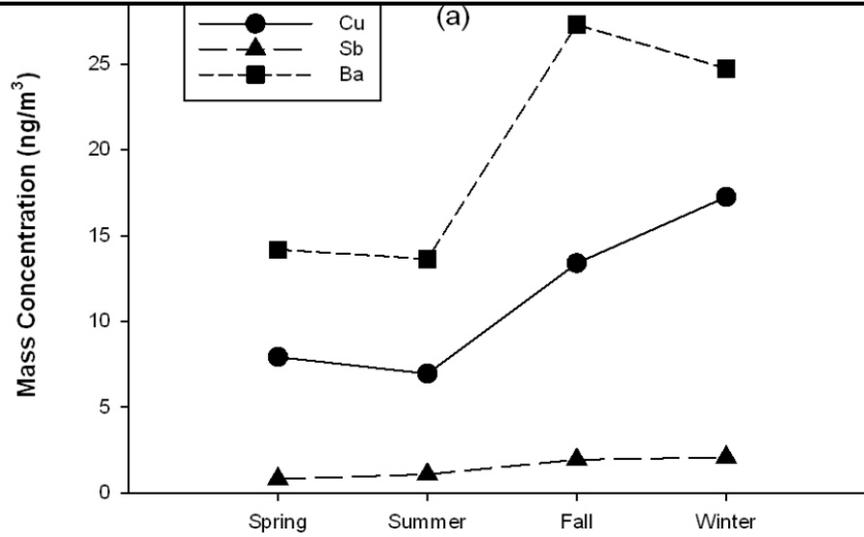
Chemical Composition - Overview



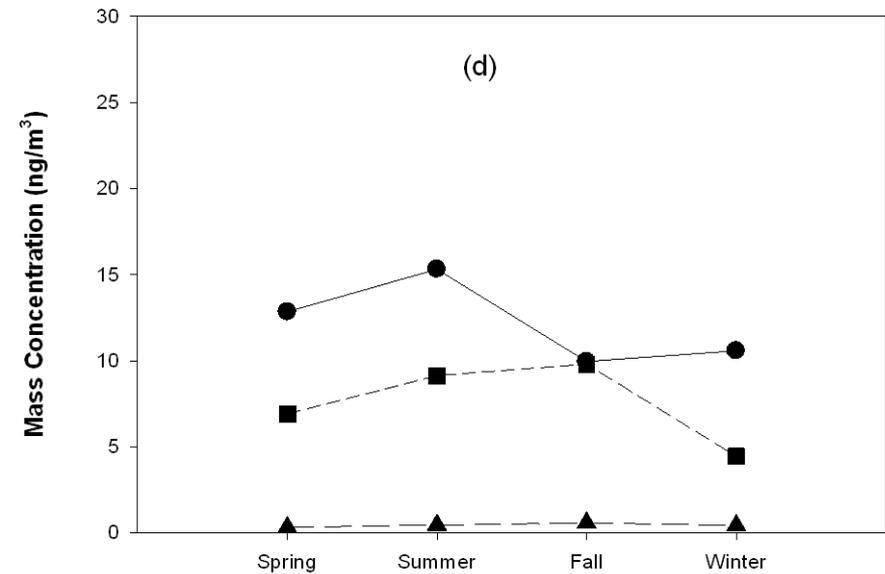
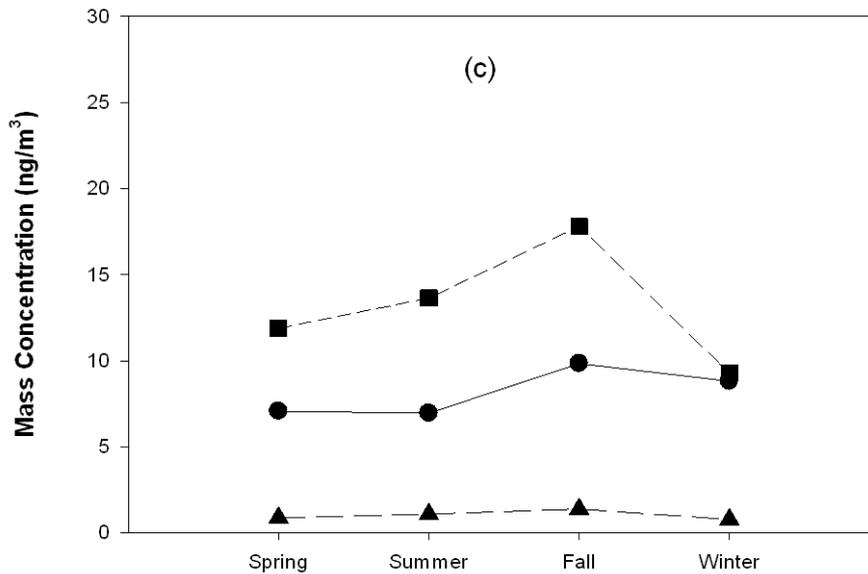
Seasonal and Spatial Trend



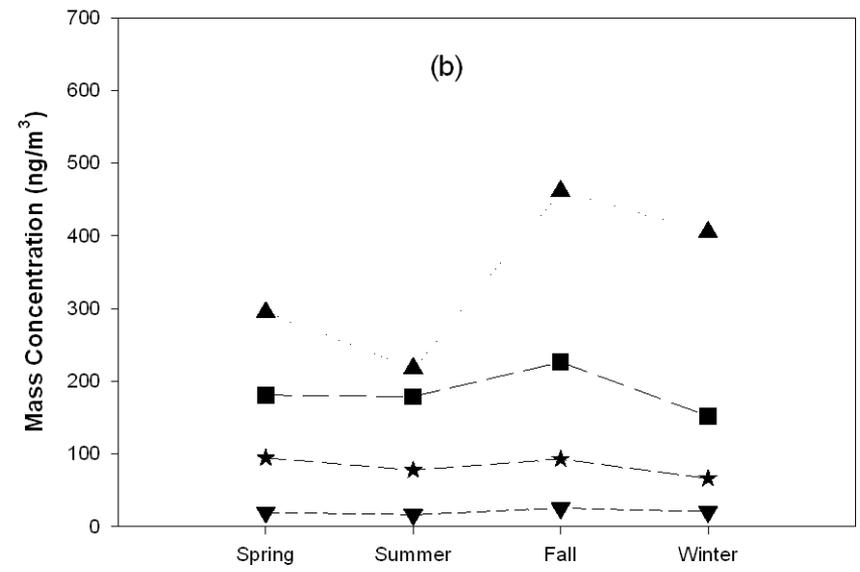
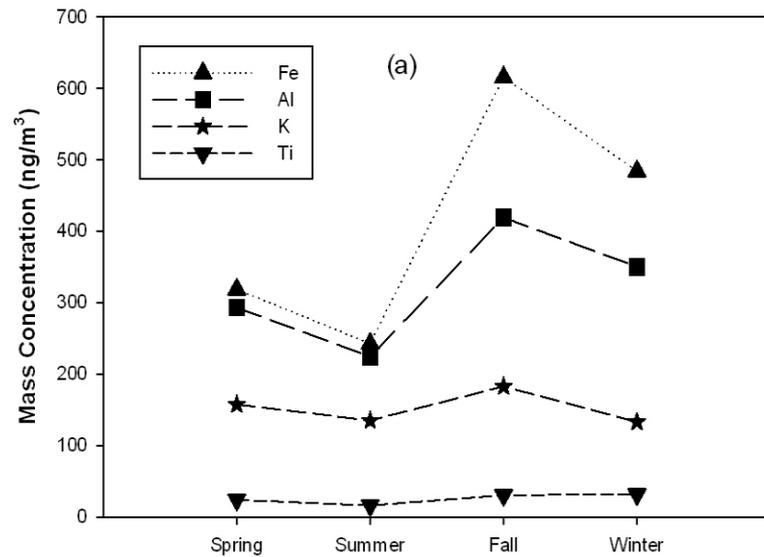
Temporal variation of **abrasive vehicular tracers (Cu, Sb, Ba)** in different sampling site clusters: (a) Los Angeles cluster, (b) Long Beach, (c) Riverside cluster, and (d) Lancaster.



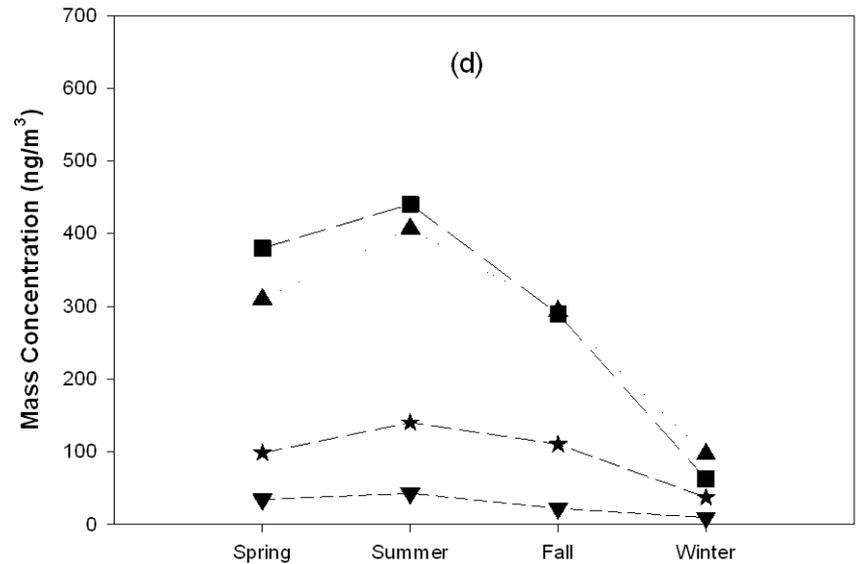
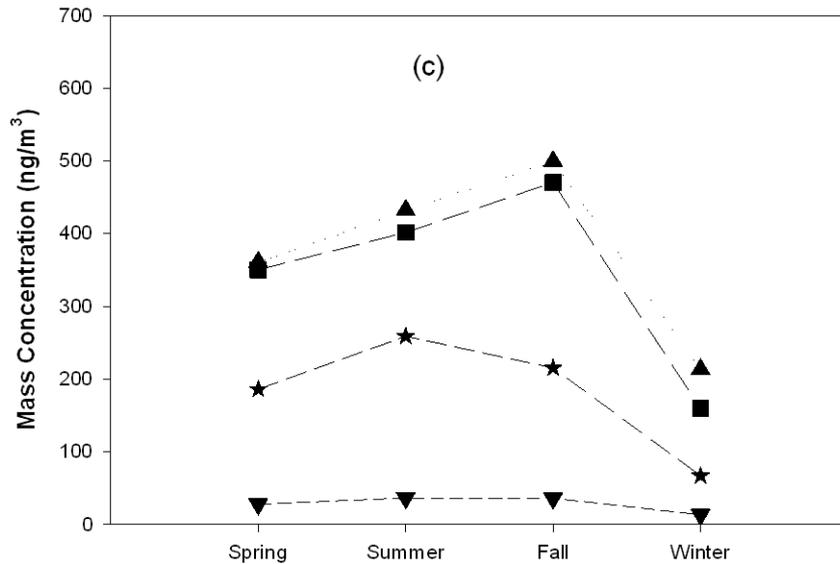
Pakbin et al, AS&T 2011



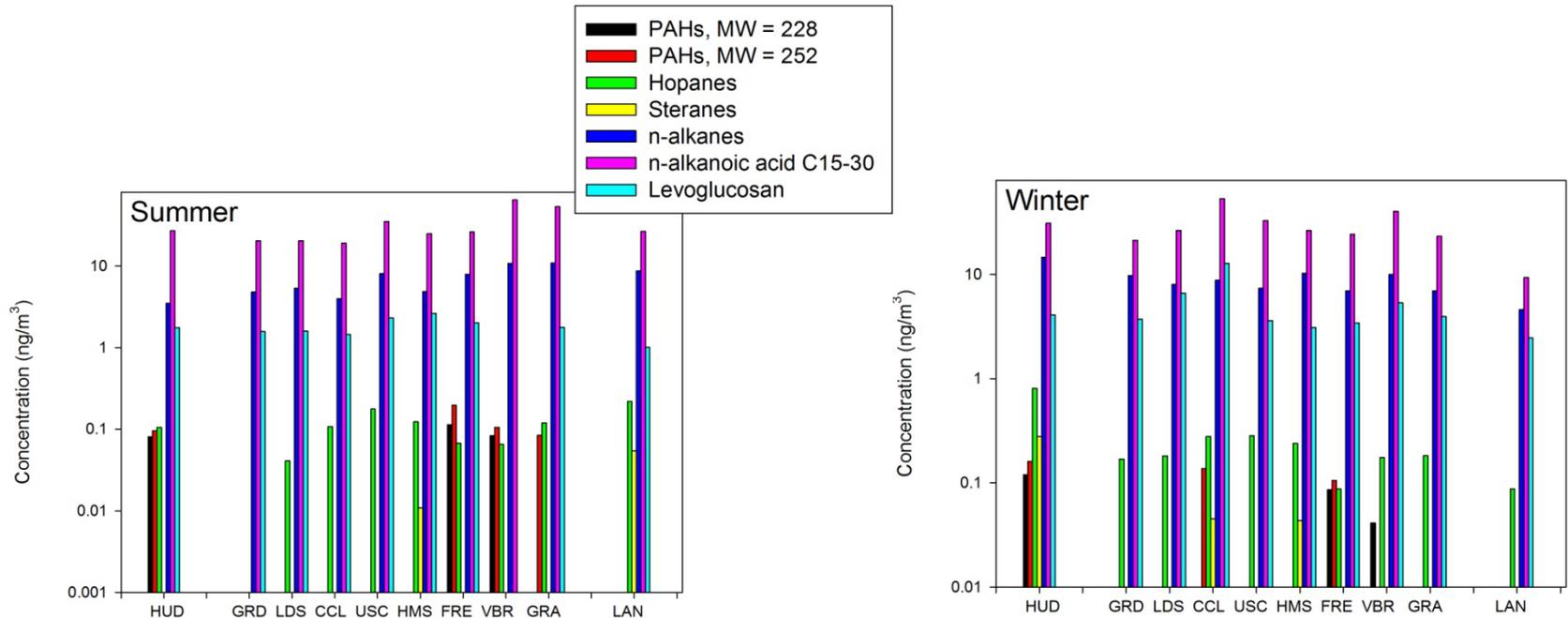
Temporal variation of selected mineral dust tracers (Fe, Al, K, Ti) in different sampling site clusters: (a) Los Angeles cluster, (b) Long Beach, (c) Riverside cluster, and (d) Lancaster.



Pakbin et al, AS&T 2011

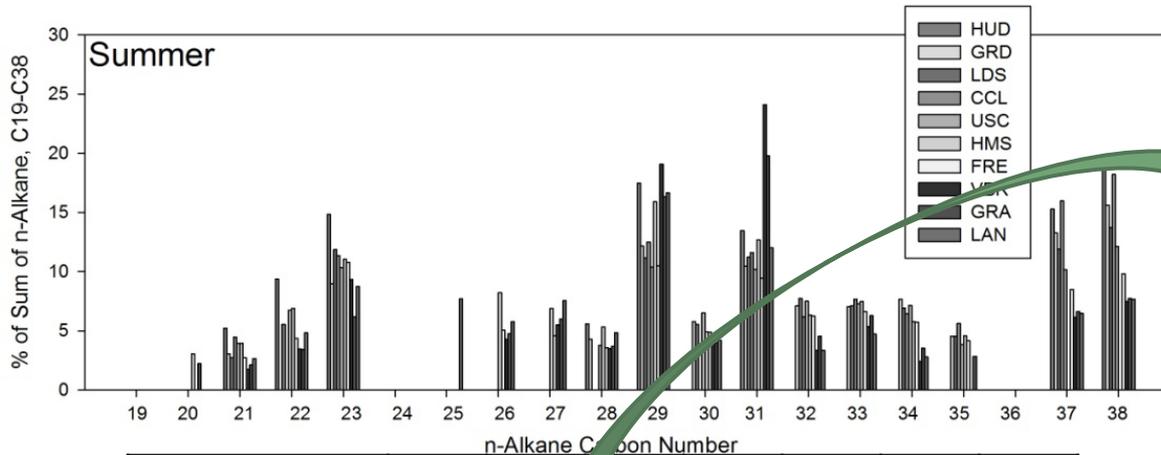


Organics

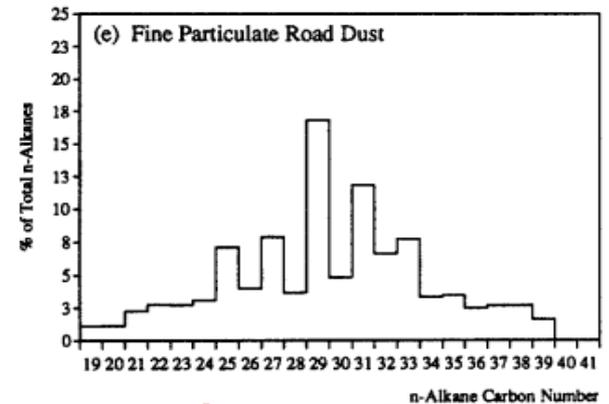
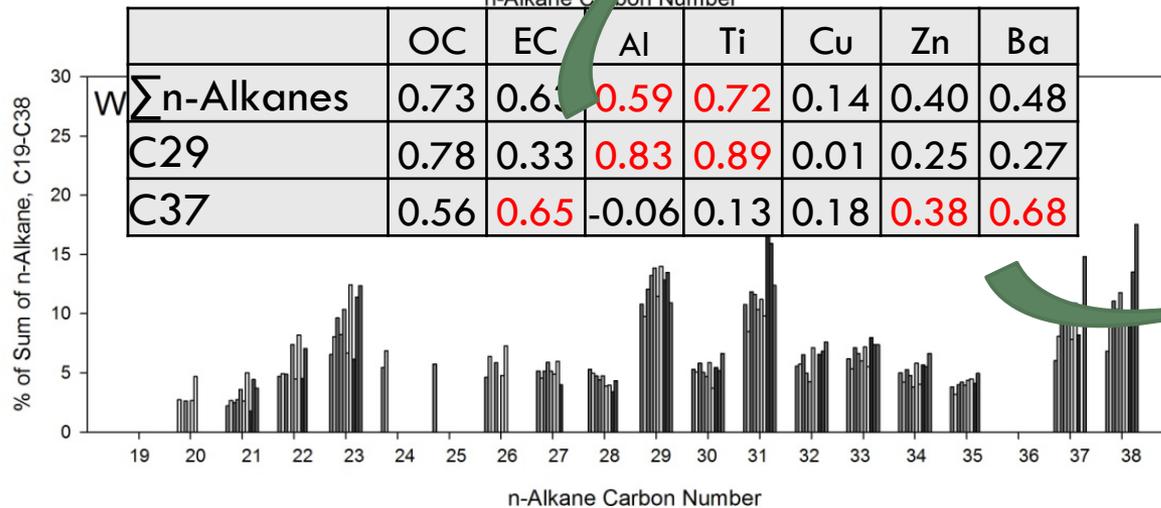
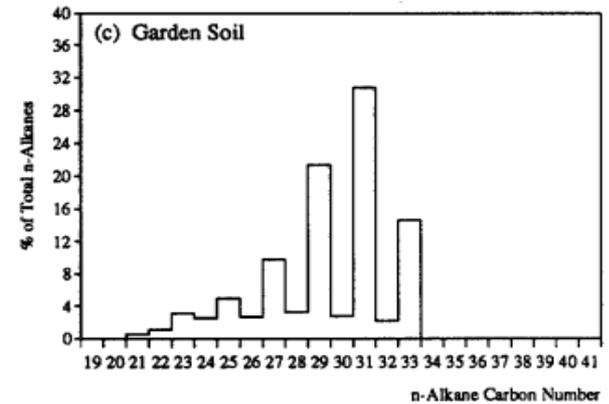


- ❑ Hopanes, steranes and PAHs, mostly in trace levels / undetected
- ❑ Alkanes and alkanolic acids are the two most dominant components
- ❑ Levoglucosan is still significant in CPM, with higher levels in winter

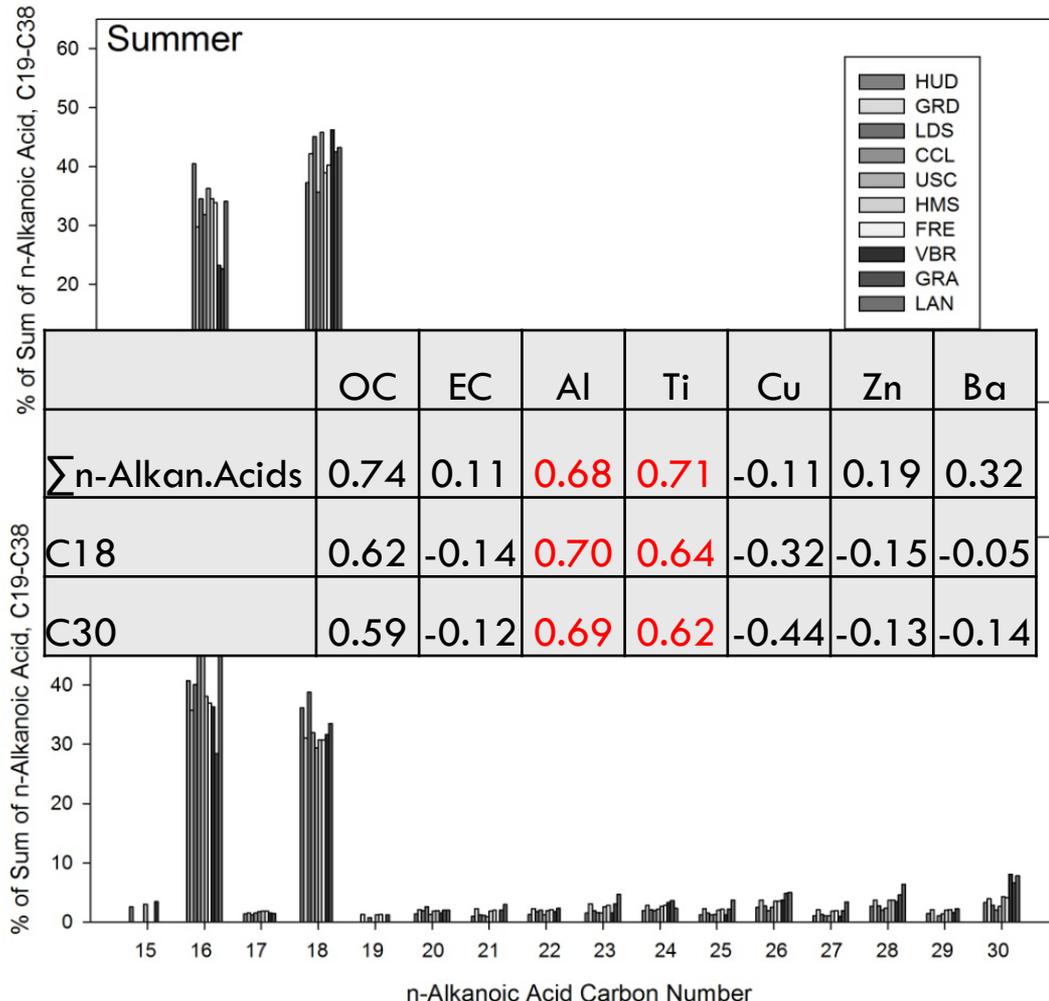
Organics – n-alkanes



Similar to the source profiles of:



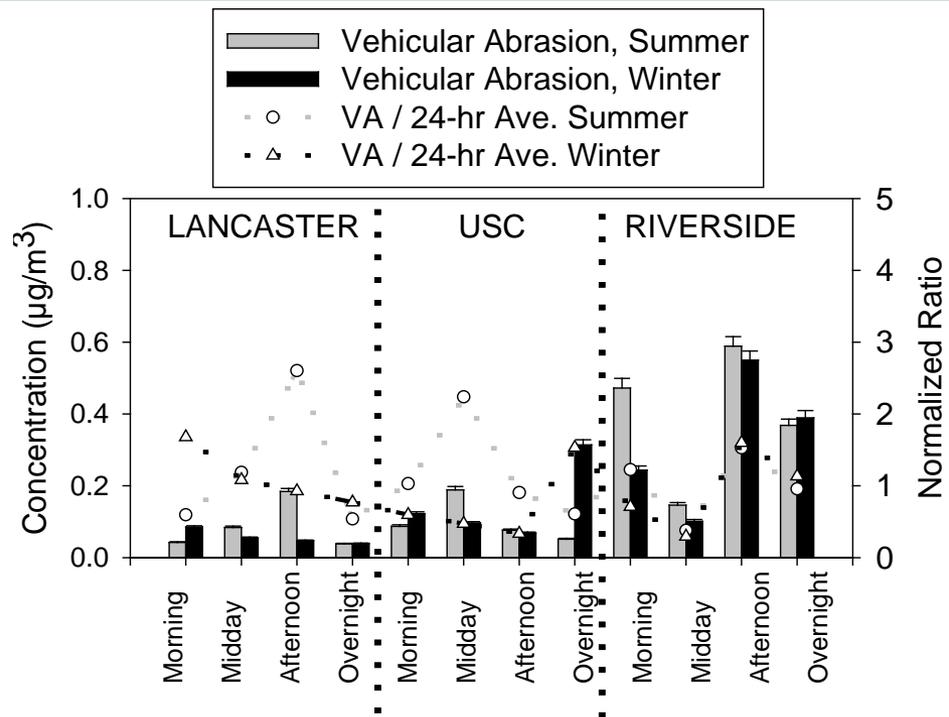
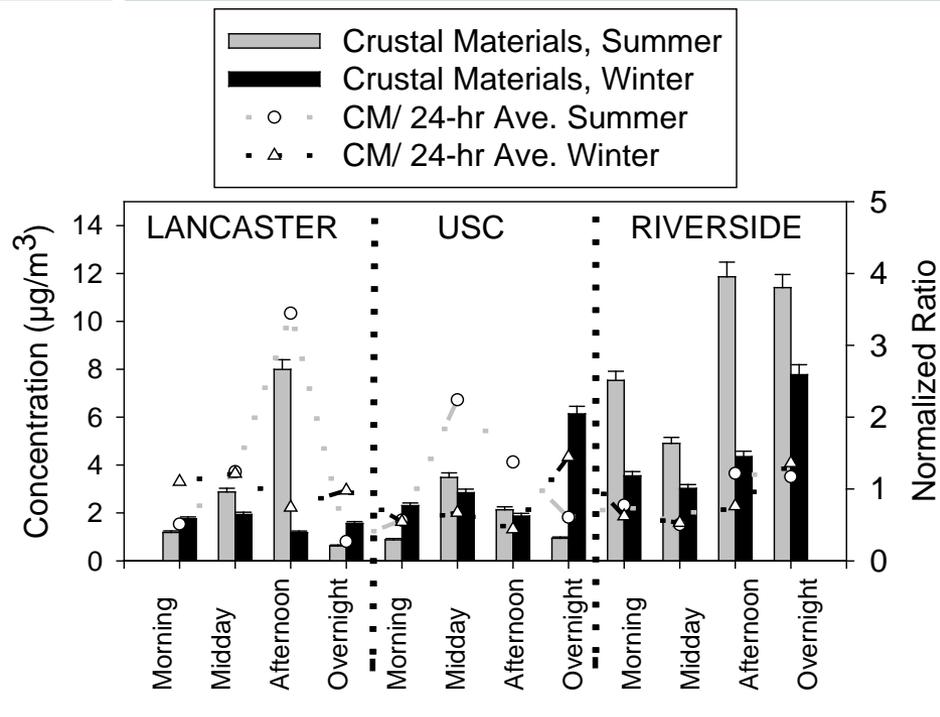
Organics – n-alkanoic acids



- Similar to the source profiles of:
 - Road dust
 - Green leaf debris
 - Tire wear
- Al and Ti – crustal materials
- EC-tailpipe exhaust
- Cu, Zn and Ba- vehicular abrasive emissions

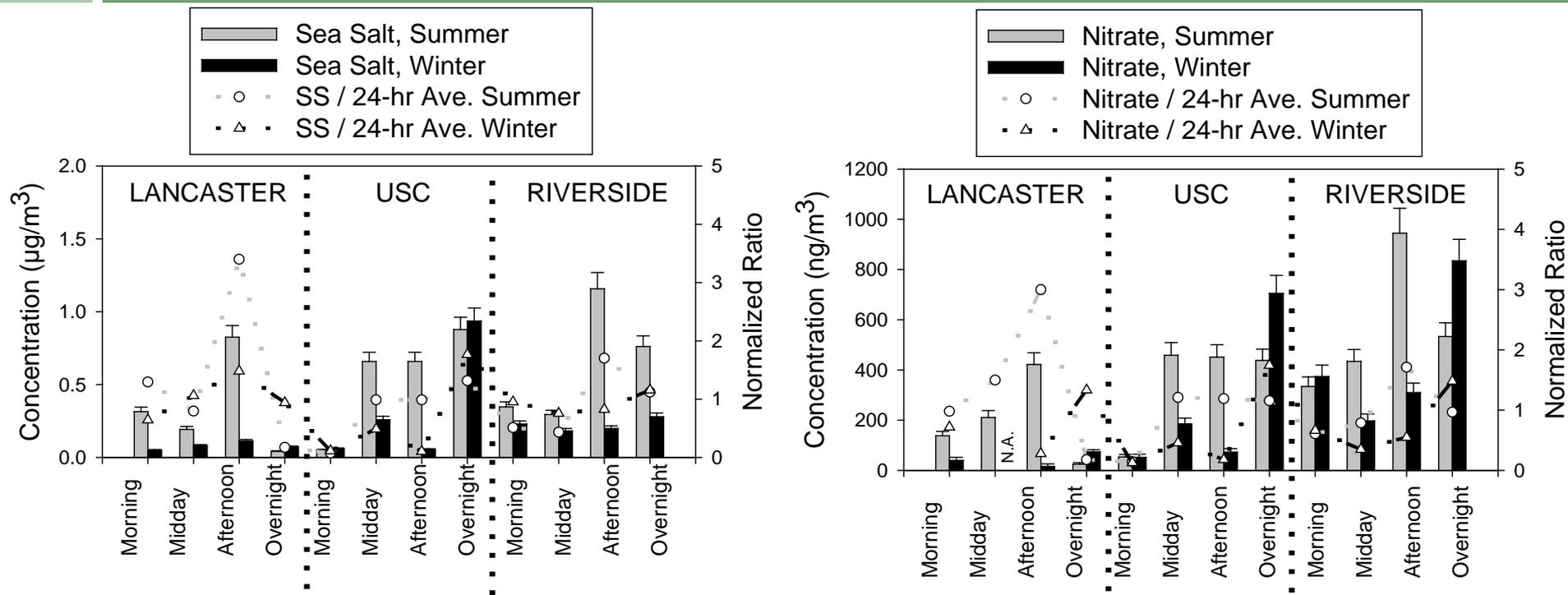
Cheung et al, Atmos Environ 2011

Diurnal Trends



- In summer, higher levels in midday / afternoon
- In winter, overnight peaks, near-freeway re-suspension

Diurnal Trends



- Sea salt and nitrate follows similar diurnal trend
- Nitrate formation
 - In summer, sea salt depletion
 - In winter, depletion in addition to reactions with mineral dust and ammonium

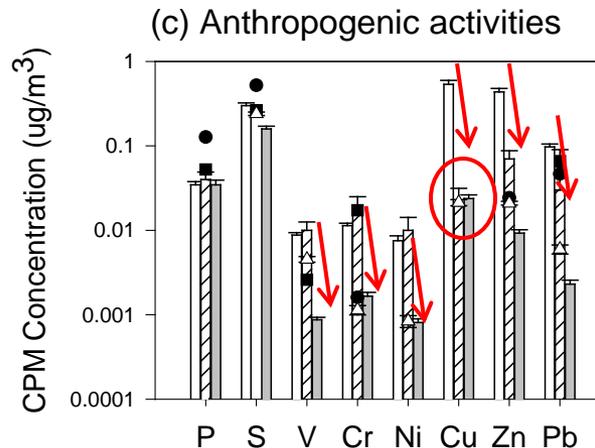
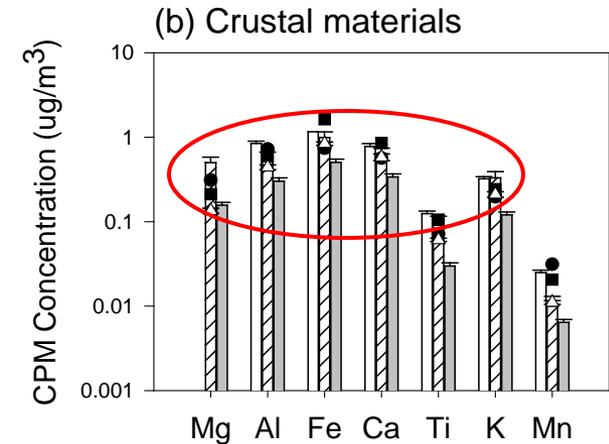
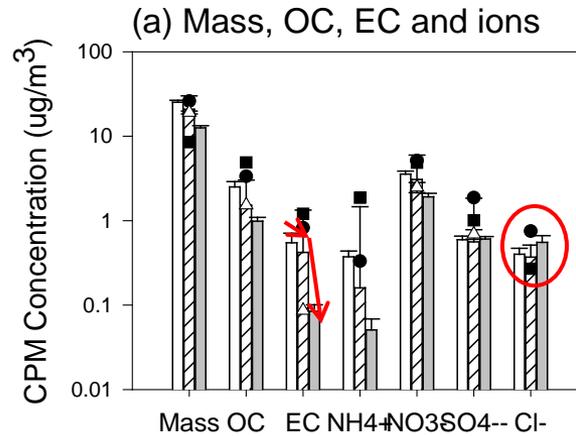
Historical Trends in CPM in the LA Basin

Historical chemical composition

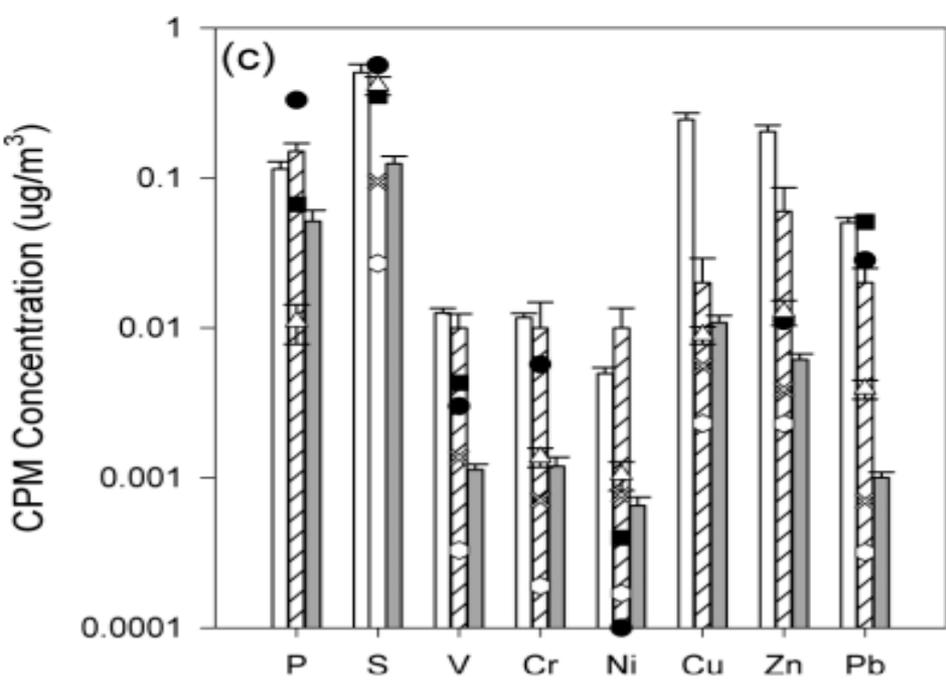
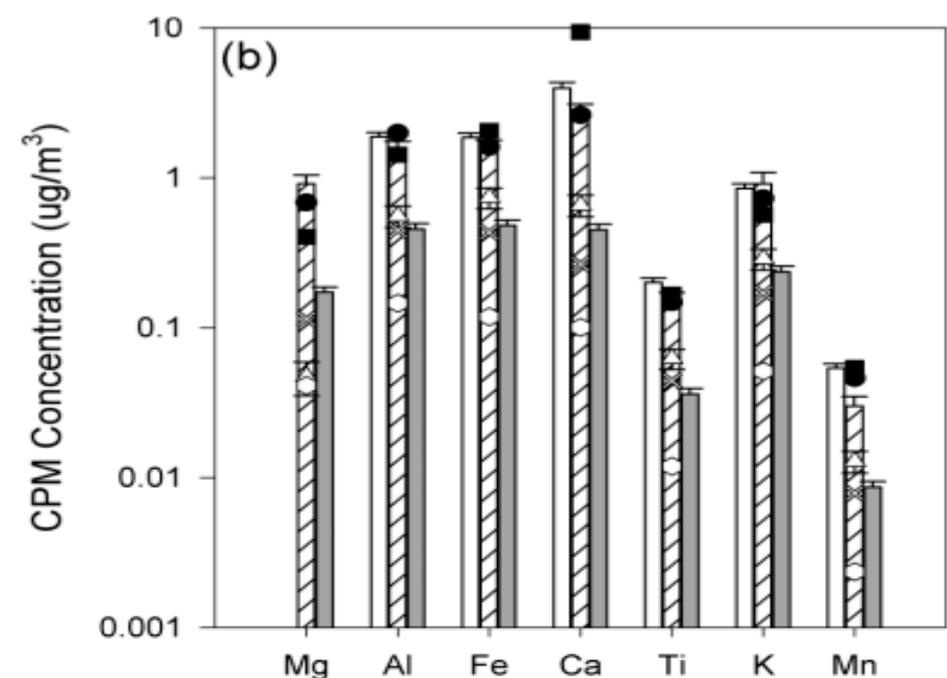
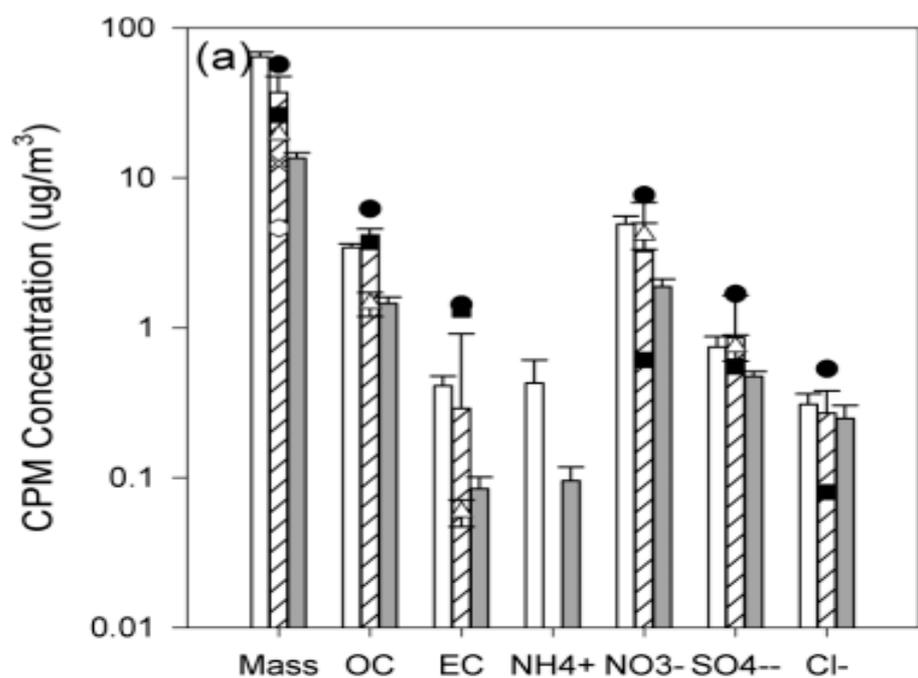
- Peer-reviewed journals
- Government agency database

3 year-long studies

- 1986-87, CARB
- 1995-96, AQMD
- 2008-09, EPA/USC



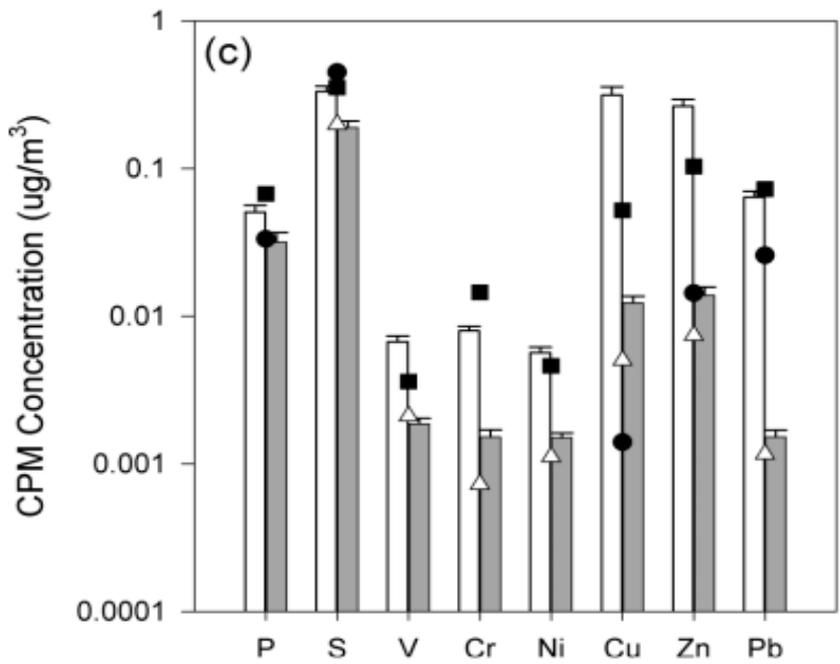
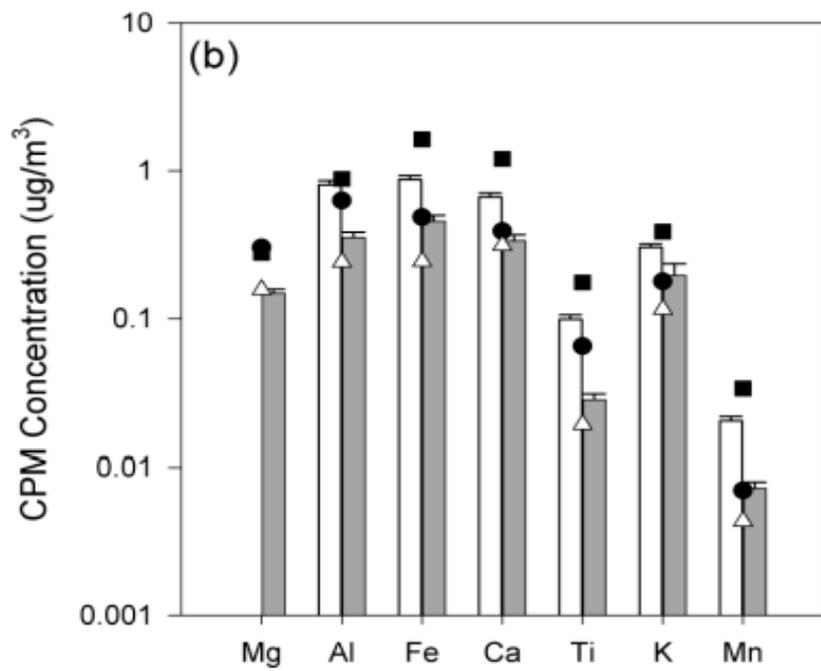
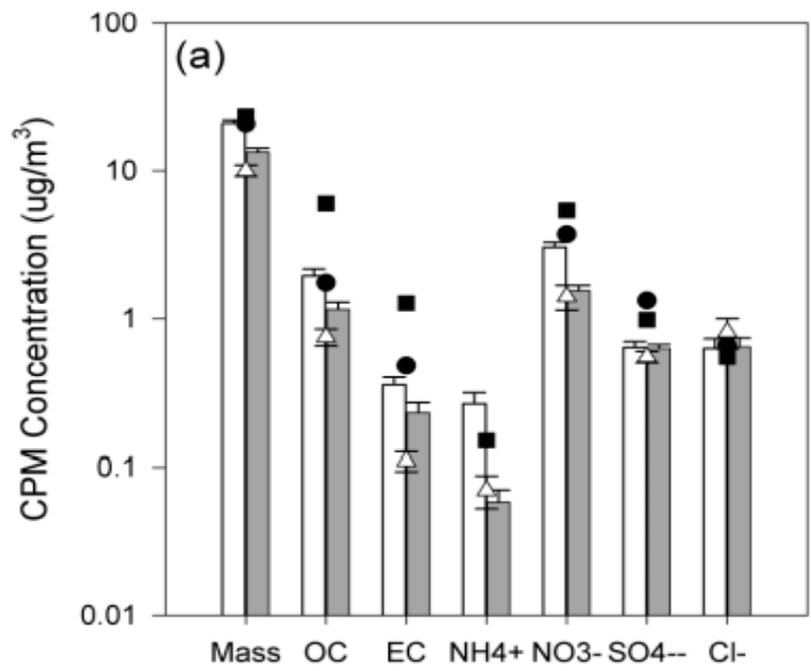
- 1986, CARB
- ▨ 1995-1996, Kim et al., 2000
- ▒ 2008-2009, Cheung et al., 2011
- 1987 Fall, Chow et al., 1994
- 1987 Summer, Chow et al., 1994
- △ 2002-2003, Sardar et al., 2005



- 1986, CARB
- ▨ 1995-1996, Kim et al., 2000
- 2008-2009, Cheung et al., 2011
- 1987 Summer, Chow et al., 1994
- 1987 Fall, Chow et al., 1994
- △ 2001 Mar-Jun, Sardar et al., 2005
- × 2006 Summer, Polidori et al., 2009
- 2007 Winter, Polidori et al., 2009

Riverside

Cheung et al, JAWMA 2012



- 1986, CARB
- 2008-2009, Cheung et al., 2011
- 1987 Summer, Chow et al., 1994
- 1987 Fall, Chow et al., 1994
- △ March-May 2007, Arhami et al., 2009

Long Beach

Cheung et al, JAWMA 2012

Summary

- Mineral dust is the most dominant group, with higher fractions in inland areas
- Organic matter mostly arises from soil and biological materials
- Combustion emissions have decreased considerably even in CPM in the last decade
- Significant diurnal trend and contrasting primary re-suspension mechanism in summer and winter

Physical Characterization

- What causes the spatial and seasonal trends in CPM mass?
- Has the mass concentration changed over the years?
- Is PM_{10} a good surrogate of CPM?

Chemical Characterization

- Is the chemical composition of CPM different from $PM_{2.5}$?
- What are the sources and formation mechanisms of CPM?
- Has the chemical profile of coarse particle changed over the years?

Toxicological Characterization

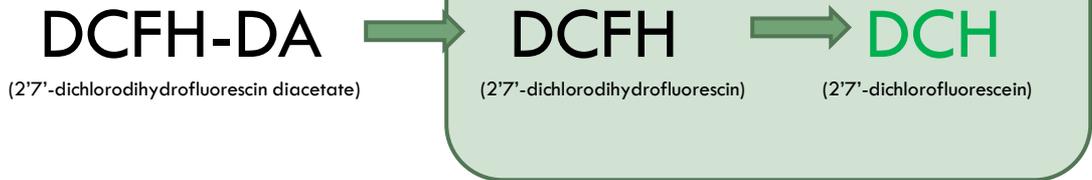
- What are driving the toxicity of coarse particles?
- Is there a link between source, composition and toxicity of CPM?

Cellular ROS Assay

- A Cell-based Method
 - ▣ Alveolar macrophage (AM)
 - ▣ Measures total ROS activity

- Method

- 1) Water extraction
- 2) Filtration
- 3) Add DCFH-DA solution of AM cell culture
- 4) Exposure of AM cells to PM extracts
- 5) Measure the increase of fluorescence in treated samples every 30 mins for an incubation time of 150 mins, and compared with untreated controls

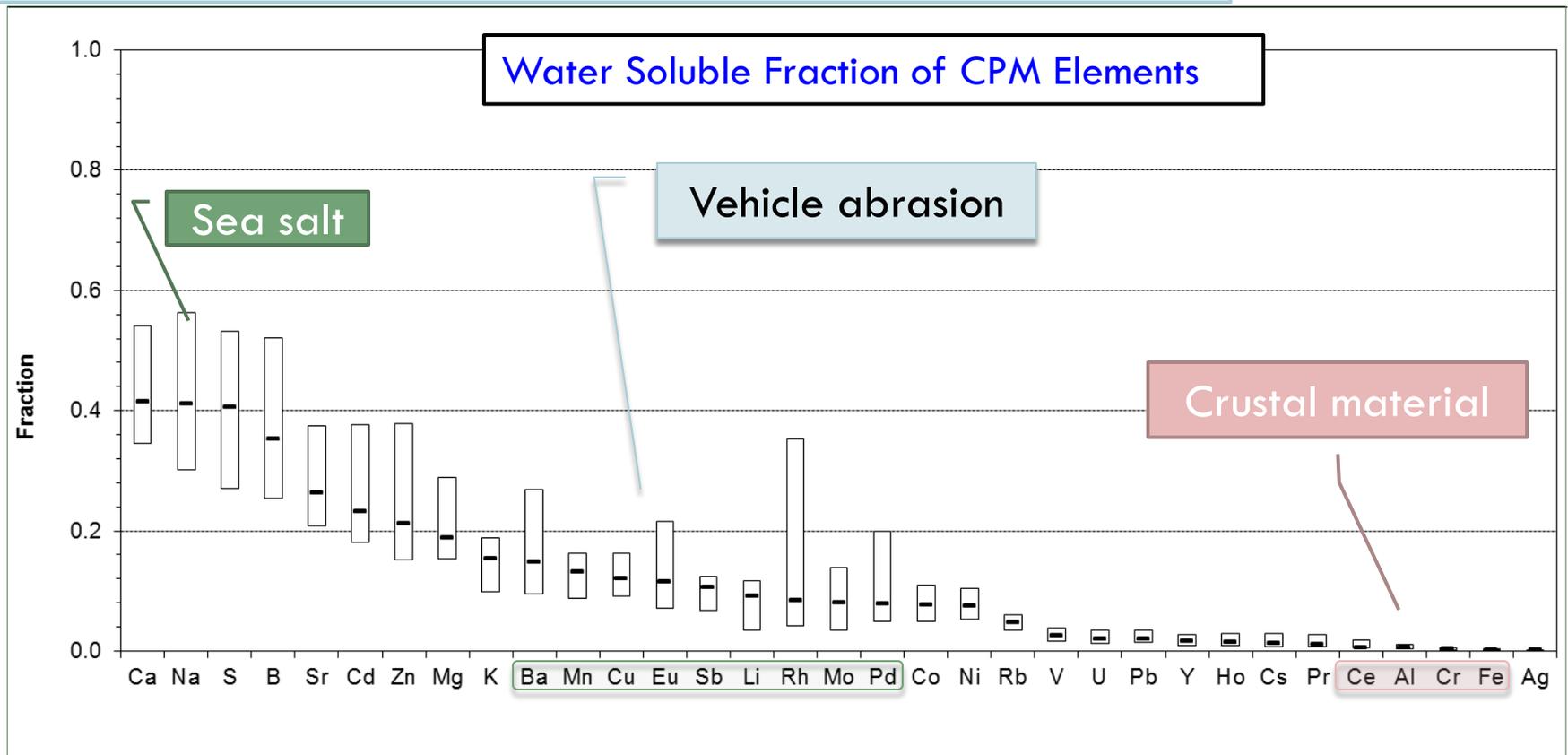


PM Induced-ROS Formation

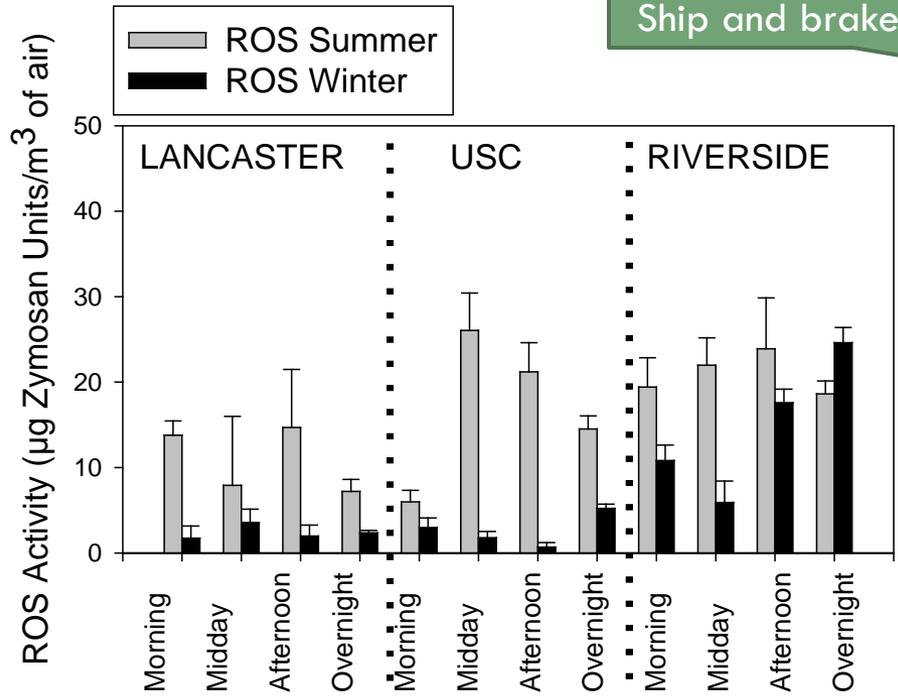
Cheung et al ES&T 2012

□ Particle-cell interaction

1. Fine/Coarse PM enter cells by phagocytosis.
2. Ultrafine PM, organic compounds enter by diffusion.
3. **Water-soluble ions** enter by ion channels, carrier molecules.

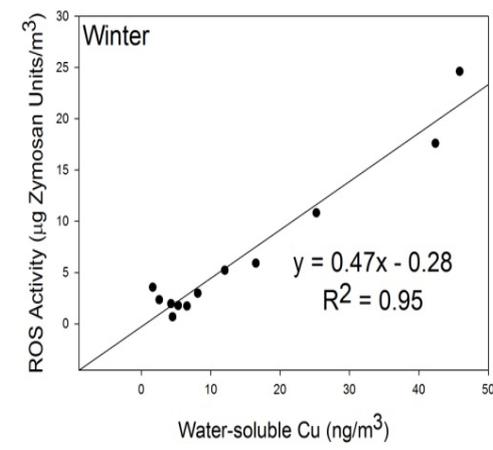
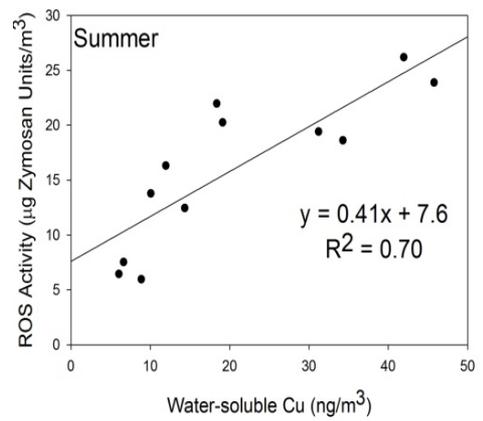


ROS Activity in Coarse Particles



Ship and brake wear Catalytic converter wear Brake wear

	WS-V	WS-Pd	WS-Cu	WS-Rh	WS-Ti	WS-Al
ROS	0.74	0.78	0.73	0.64	0.14	0.16
WS-V		0.86	0.67	0.64	0.15	0.27
WS-Pd			0.85	0.86	0.11	0.34
WS-Cu				0.71	0.12	0.28
WS-Rh					0.07	0.53
WS-Ti						0.06



Summary

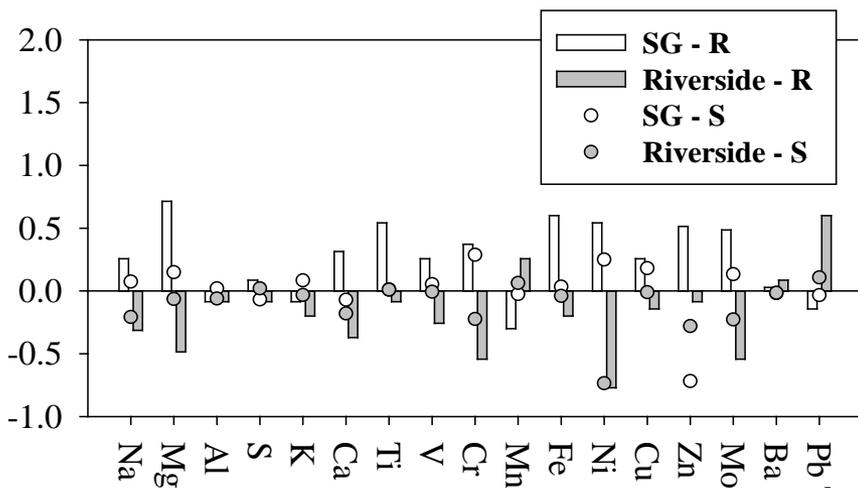
- Water solubility in crustal vs. anthropogenic elements
- Distinct ROS activity and profile in summer vs. winter
- The water-soluble fraction of four elements (V, Pd, Cu and Rh) displayed the highest associations with ROS activity ($R^2 > 0.60$).

Limitations

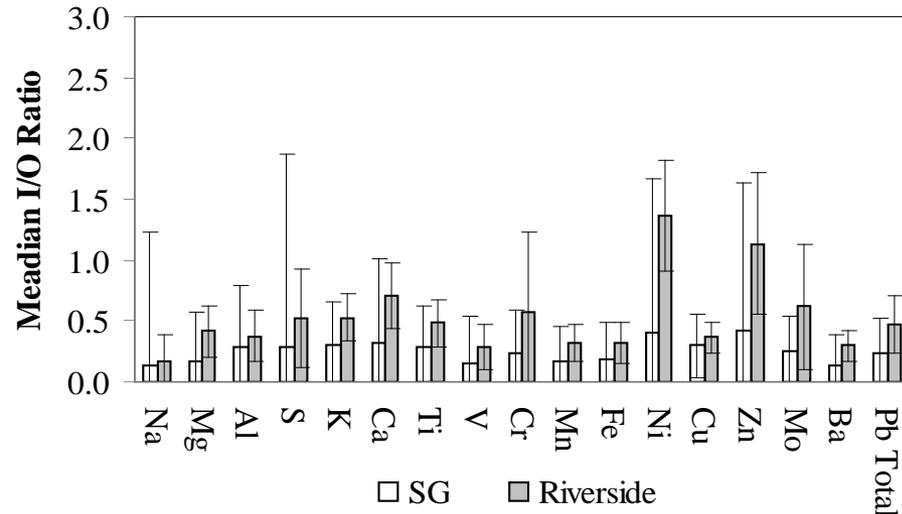
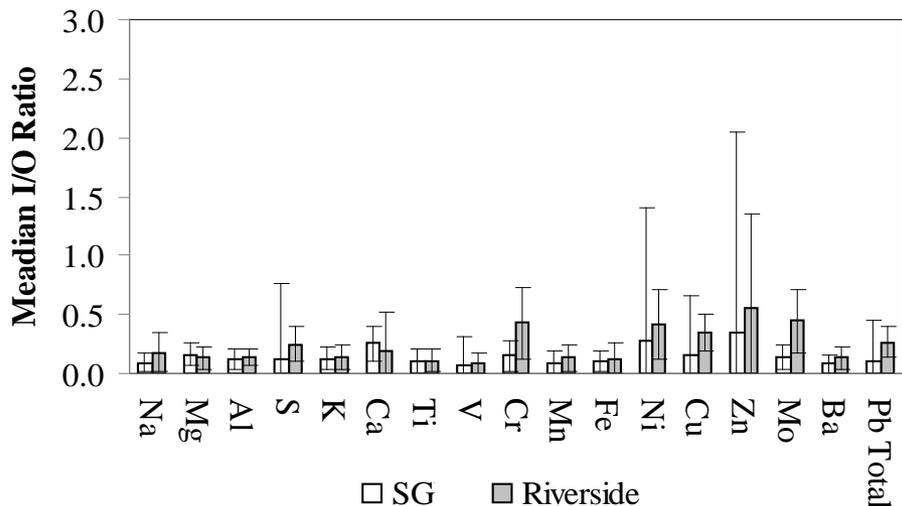
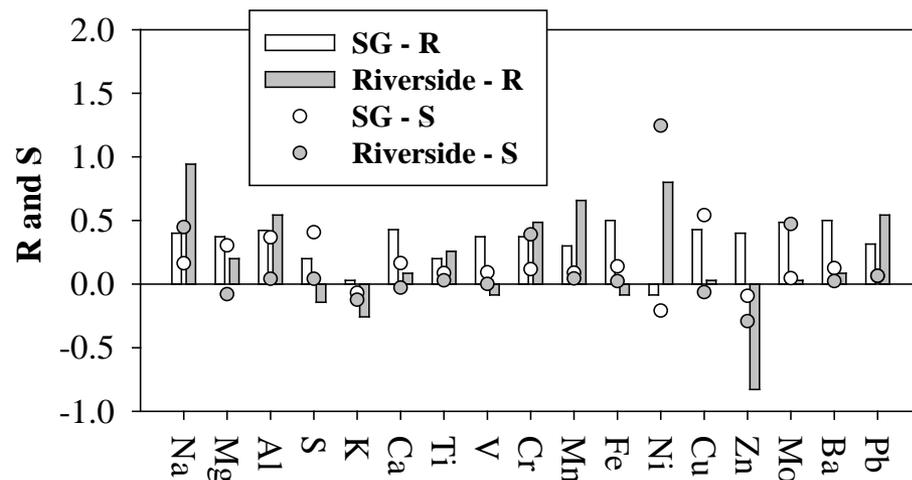
- Generalizability
- Toxicological assessment
- Penetration in Indoor Environments

Indoor vs Outdoor CPM Ratios and Correlation (Polidori et al, 2009)

Coarse - Warmer Season



Coarse - Cooler Season



Coarse PM have very low I/O ratios and low (often negative) I/O correlations (R)

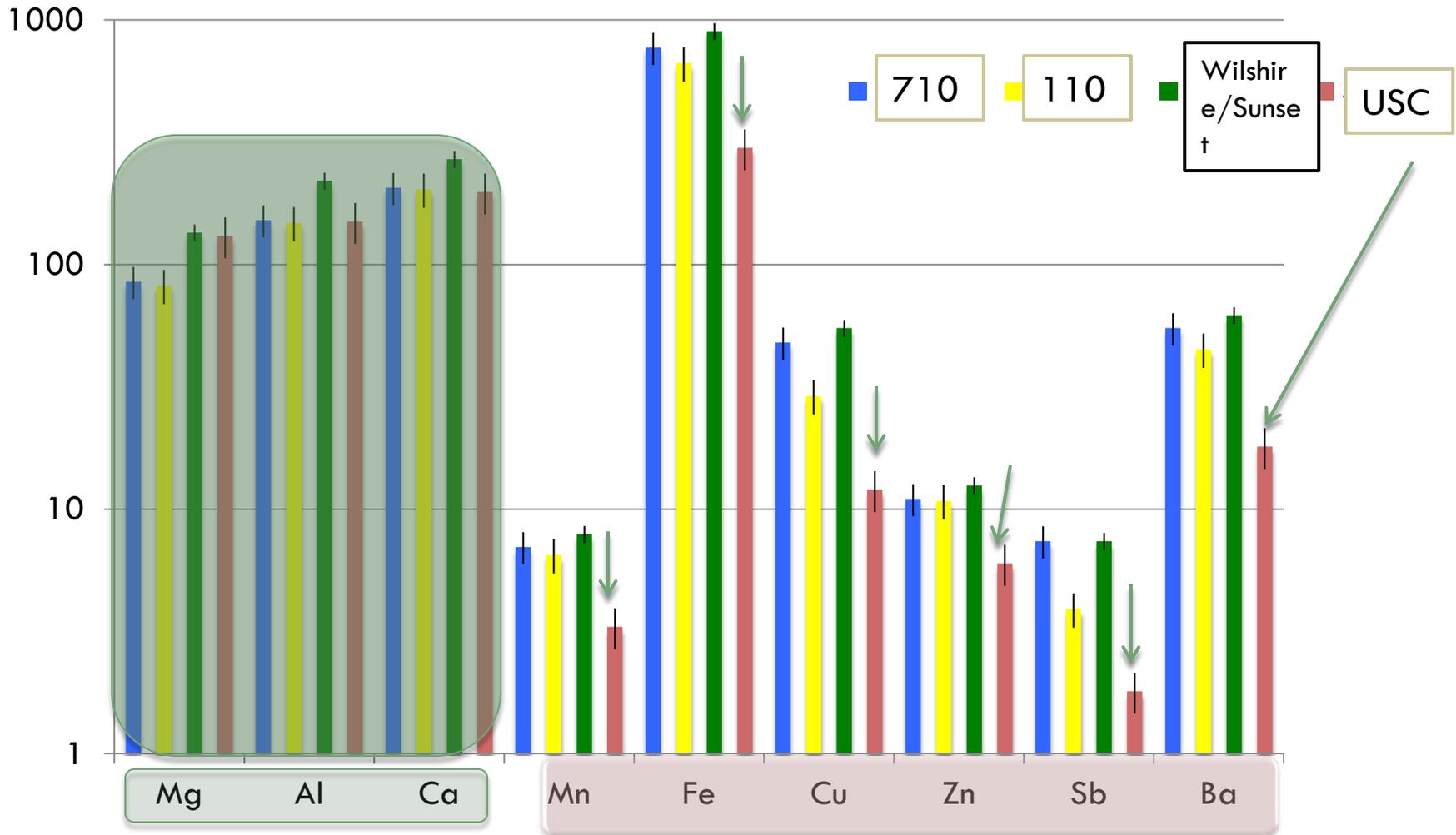
SG: San Gabriel

R; Pearson Coefficient

S; regression slope

Riverside

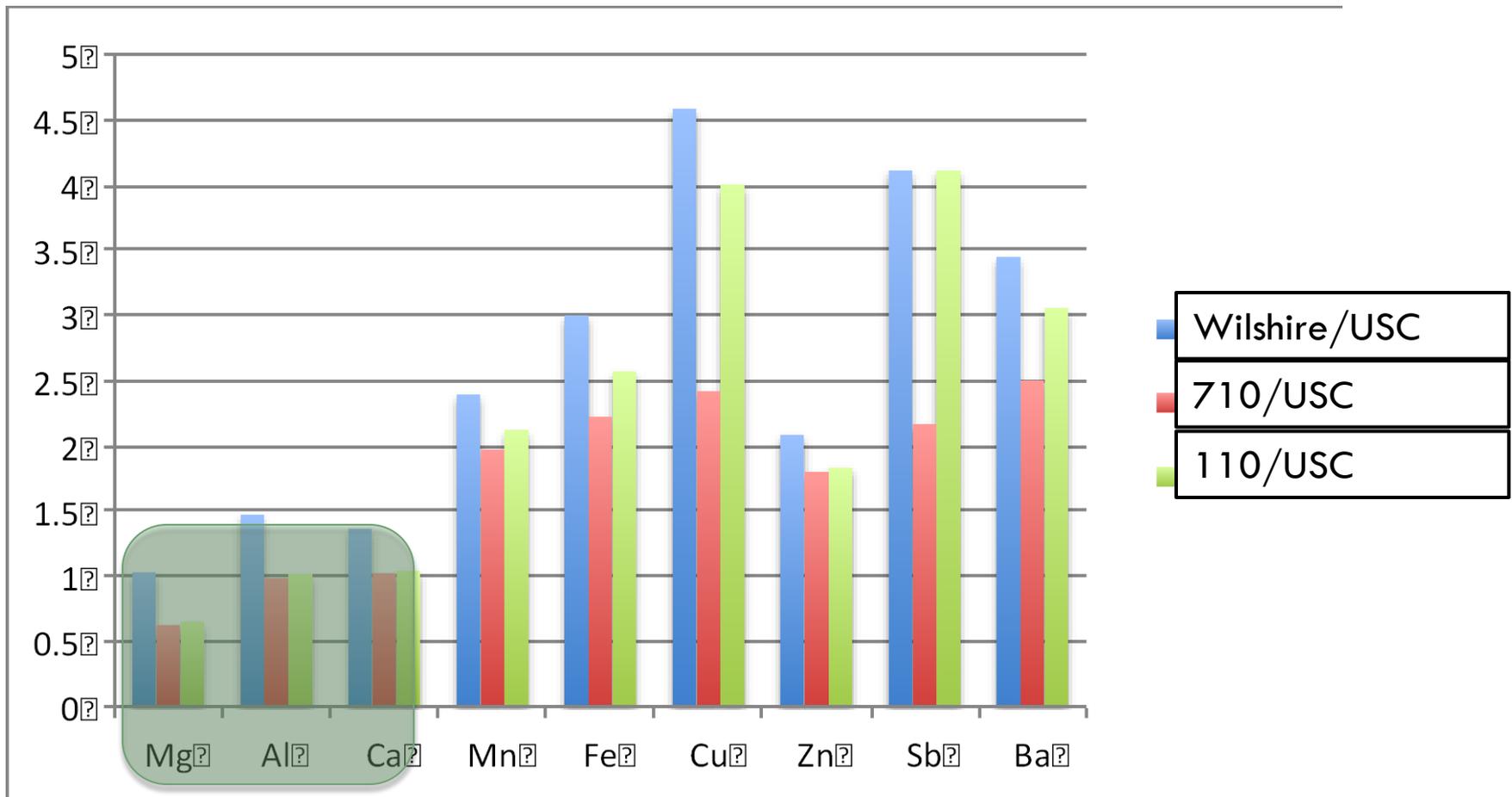
Coarse PM Elements and Metals in Freeways, Busy Streets and Urban Background in LA (Kam et al 2012)



Crustal elements

Very similar concentrations

Road dust, vehicular abrasion, tire and break wear – USC on average 20%-40% of other areas- highest levels at Wilshire and Sunset Blvds.



Ratios of Coarse PM in Various LA Freeways and busy streets vs USC (urban background site)- from [Kam et al Atmos Environ 2012](#)

2-5 fold higher levels of CPM-bound metals from road dust in busy streets compared to urban background

Recommendations on Future Aerosol Research

□ Bioaerosols

- Quantification of fungi, bacteria, plant pollen, and spore materials

□ Road dust

- Focused-characterization
- In-vivo exposures

Increasingly important, given difficulties controlling non-tailpipe exhaust



Journal Papers (peer-reviewed)

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