

US EPA ARCHIVE DOCUMENT

# *Composition and Behavior of Ethanol Gasolines*

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National Exposure Research Laboratory  
Athens, Georgia

*For*

*Freshwater Spills 2009*

*St. Louis, Missouri*

*April 27-30, 2009*

## Outline

- Historical Usage of Ethanol
- Phase behavior of ethanol/gasoline blends
- Composition of Fuel Ethanol Samples
- Field Examples: Land and Water
- Modeling Examples

## Overview from producers to consumers

- Producers
  - Distill to 190 proof – 95% alcohol
  - Dry to 200 proof (molecular sieve)
  - Denature
- Transport
  - Not to go in U.S. pipelines,
    - 1 exception (Florida)
  - Therefore: barges, trucks, trains
- Distribute
  - (Terminals) Separate Ethanol Tank
  - Splash Blending at rack
- Use
  - Compatible Tanks, Pipes, Dispensers?
  - Water Bottoms?

# Alternate Marketing Strategies



## Terminology

- Following **ASTM D 4806 – 08a Standard Specification for Denatured Fuel Ethanol for Blending with Gasolines for Use as Automotive Spark-Ignition Engine Fuel.**
  - ***Denaturant*** -- a material added to fuel ethanol to make it unsuitable for beverage use, but not for unsuitable for automotive use.
  - ***Fuel Ethanol*** -- ethanol with impurities common to its production including water but not denaturants.
  - ***Denatured Fuel Ethanol*** – fuel ethanol made unfit for beverage use by the addition of denaturants.
  - ***Higher Molecular Weight Alcohols*** – aliphatic alcohols of general formula  $C_nH_{2n+1}OH$  with n from 3 to 8.

## **Ethanol and Gasoline Blends**

- E100: Non-denatured fuel ethanol
- E95: Denatured fuel ethanol
  - ASTM 4806
    - Min 92% ethanol
    - Denaturants: 2% to 5%
      - Natural gasoline
      - Gasoline components
      - Unleaded gasoline
- E85: “Flex Fuel” ASTM D 5798
  - Three ethanol classes: min. 70%, 74%, 79%
- E10: “may contain 10% ethanol”
  - ~90% gasoline

## ASTM SPEC

- Ethanol > 92.1%
- Water < 1%
- Methanol < 0.5%
- Solvent-wasted gum < 5 mg/100 mL
- Denaturant 1.96% to 5%
- Inorganic Chloride < 10 mg/L

## A Word on Taxes

- 2008 Food, Conservation, and Energy Act (Public Law 110-123)
  - January 1, 2009
  - full ethanol production credit only if denaturant content < 2%.
- U.S. Internal Revenue Service
  - temporarily allowing credits for denaturant(s) < 2.5% of the fuel ethanol
  - Notice 2009-06

## Denaturants

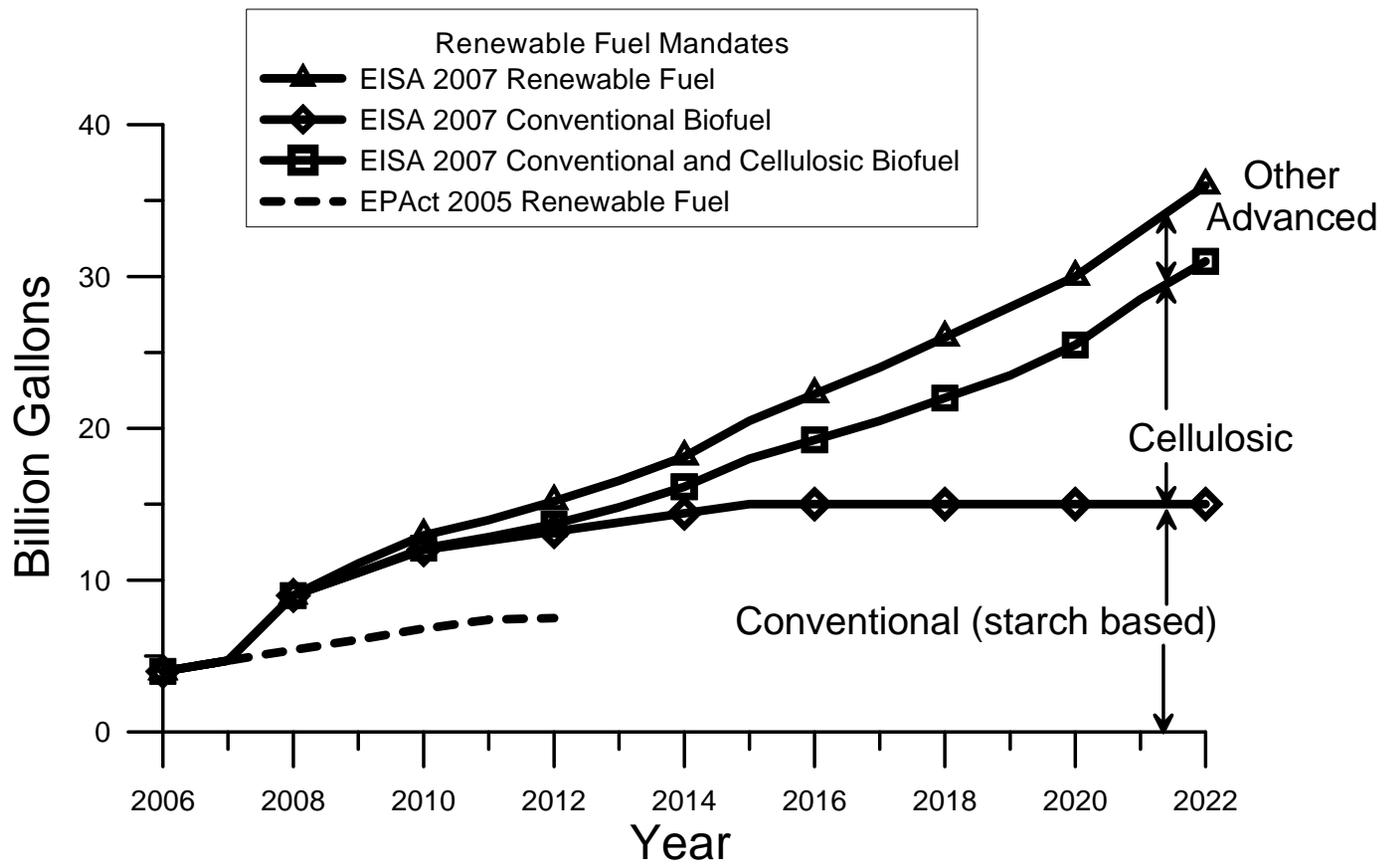
- ASTM also specifies prohibited denaturants:
  - (adverse effects on fuel stability, automotive engines, and fuel systems)
  - hydrocarbons with an end boiling point above 225 °C,
  - methanol not meeting ASTM D1152,
  - pyrroles, turpentine, ketones, and tars.

## Ethanol Usage in Gasoline

- Historic
  - Known as an octane booster since the 1920s
  - Oxygenates required in reformulated gasoline 1995-2006
- Mandated
  - Energy Policy Act 2005
  - Energy Independence and Security Act of 2007
  - Some state rules

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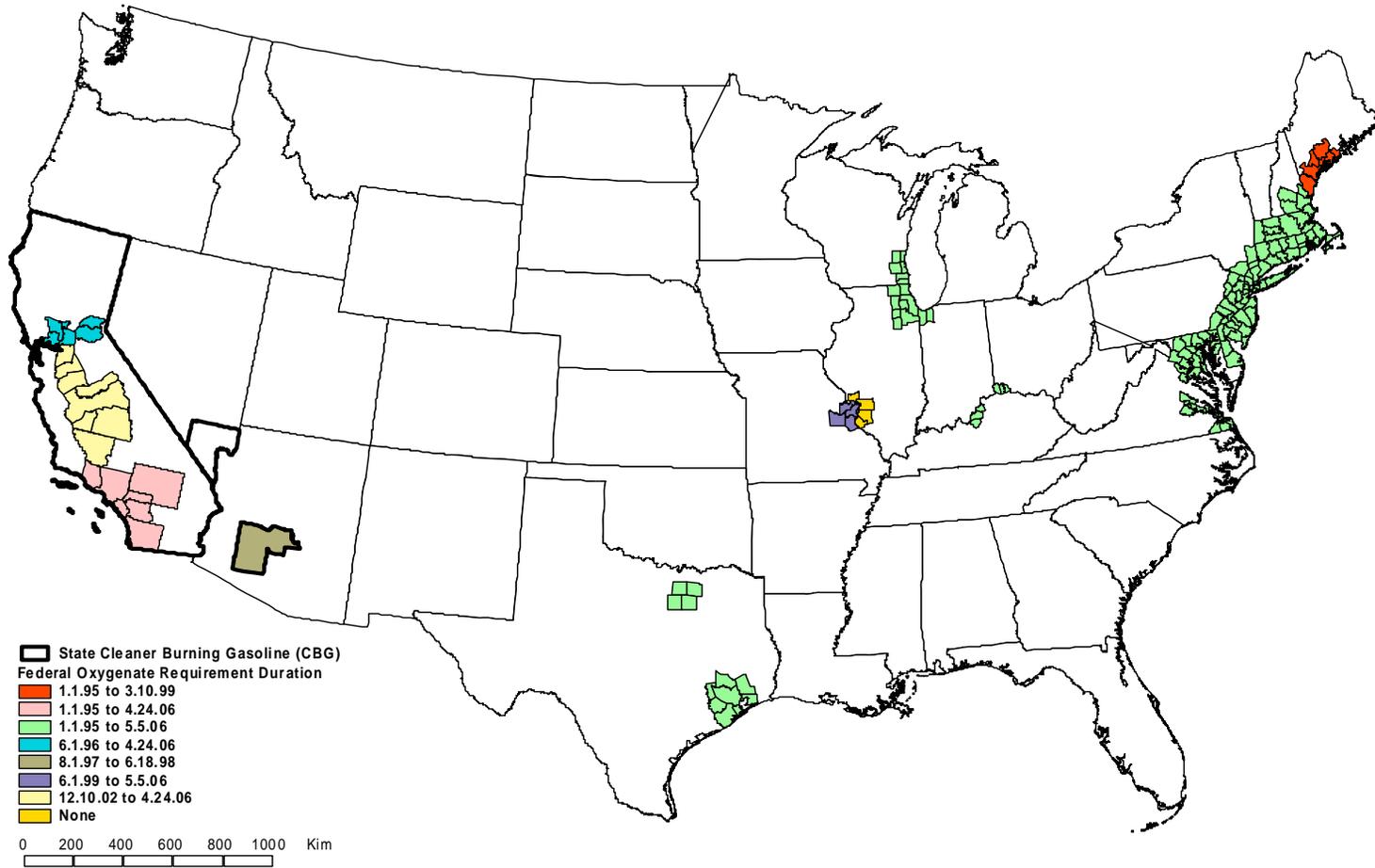
*Building a scientific foundation for sound environmental decisions*



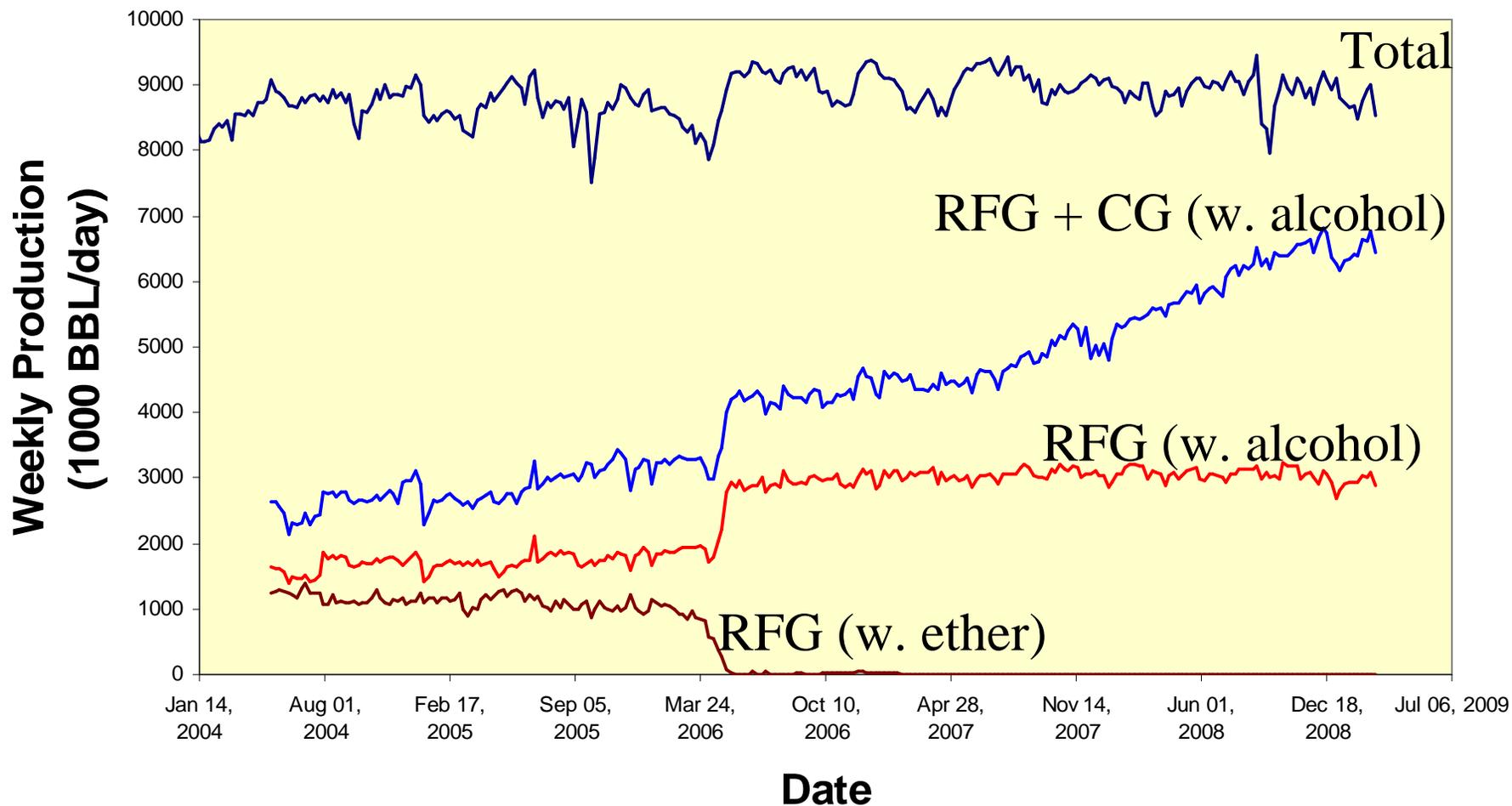
# Terminology

- Following the **Code of Federal Regulations (40 CFR) Part 80 – Regulation of Fuels and Fuel Additives:**
  - **Reformulated Gasoline (RFG)** is any gasoline whose formulation has been certified under 40 CFR § 80.40 and which meets each of the standards and requirements prescribed under 40 CFR § 80.41.
    - From 1995 until 2006, RFG was required to contain 2 % by weight oxygen-containing compounds (“oxygenates”)
    - *Benzene* < 1%
  - **Conventional Gasoline (CG)** is any gasoline which has not been certified under 40 CFR § 80.40.
  - **Oxygenated Gasoline (OG)** is any gasoline which contains a measurable amount of oxygenate.

# Reformulated and Conventional Gasoline in the US



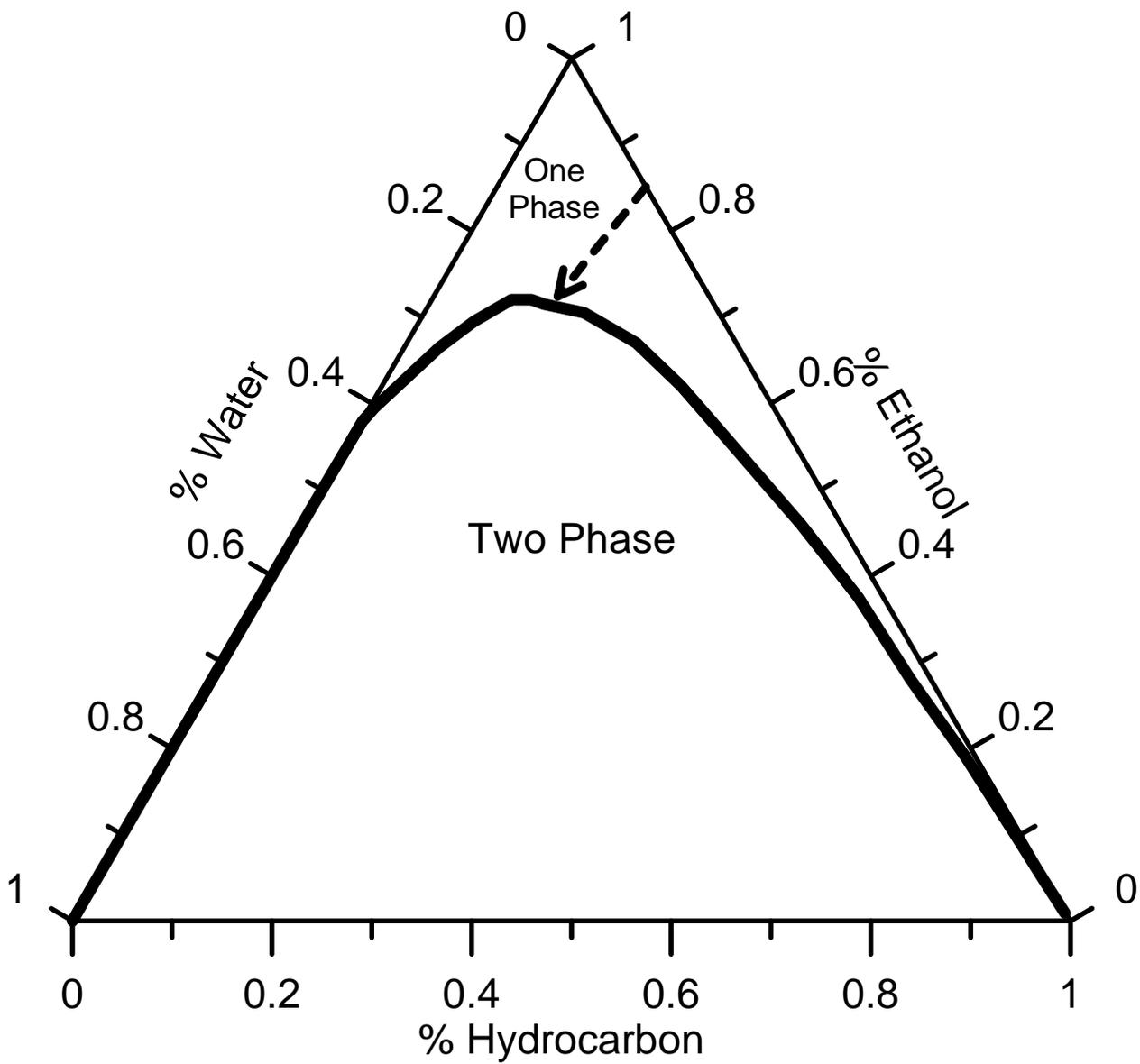
# DOE Production Data



## *Phase Behavior*

- Phase Separation
  - Gasoline adsorbs water up to a point where phase separation occurs
    - Gasoline ~0.1%
- Volume Changes
  - Ethanol/E85 volume reduction with water addition

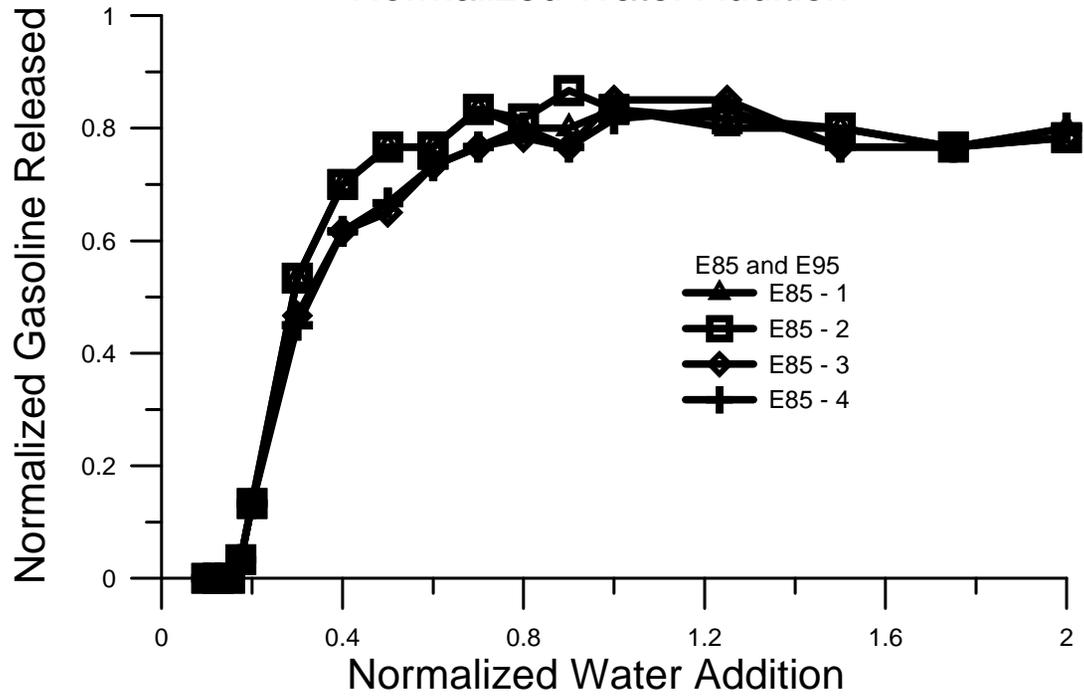
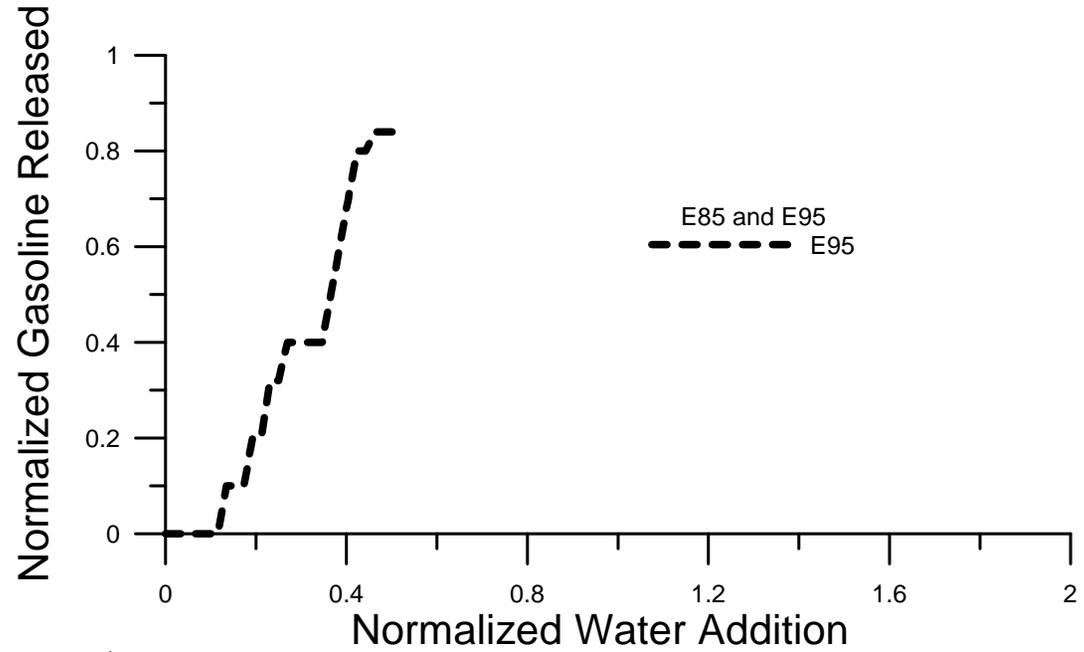






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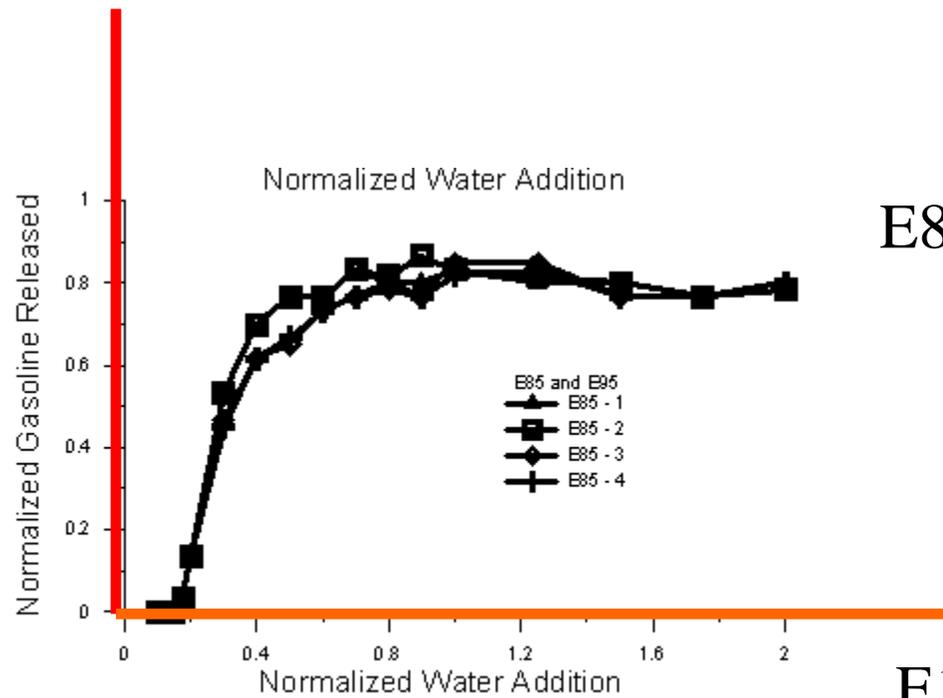
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# E10 Gasoline

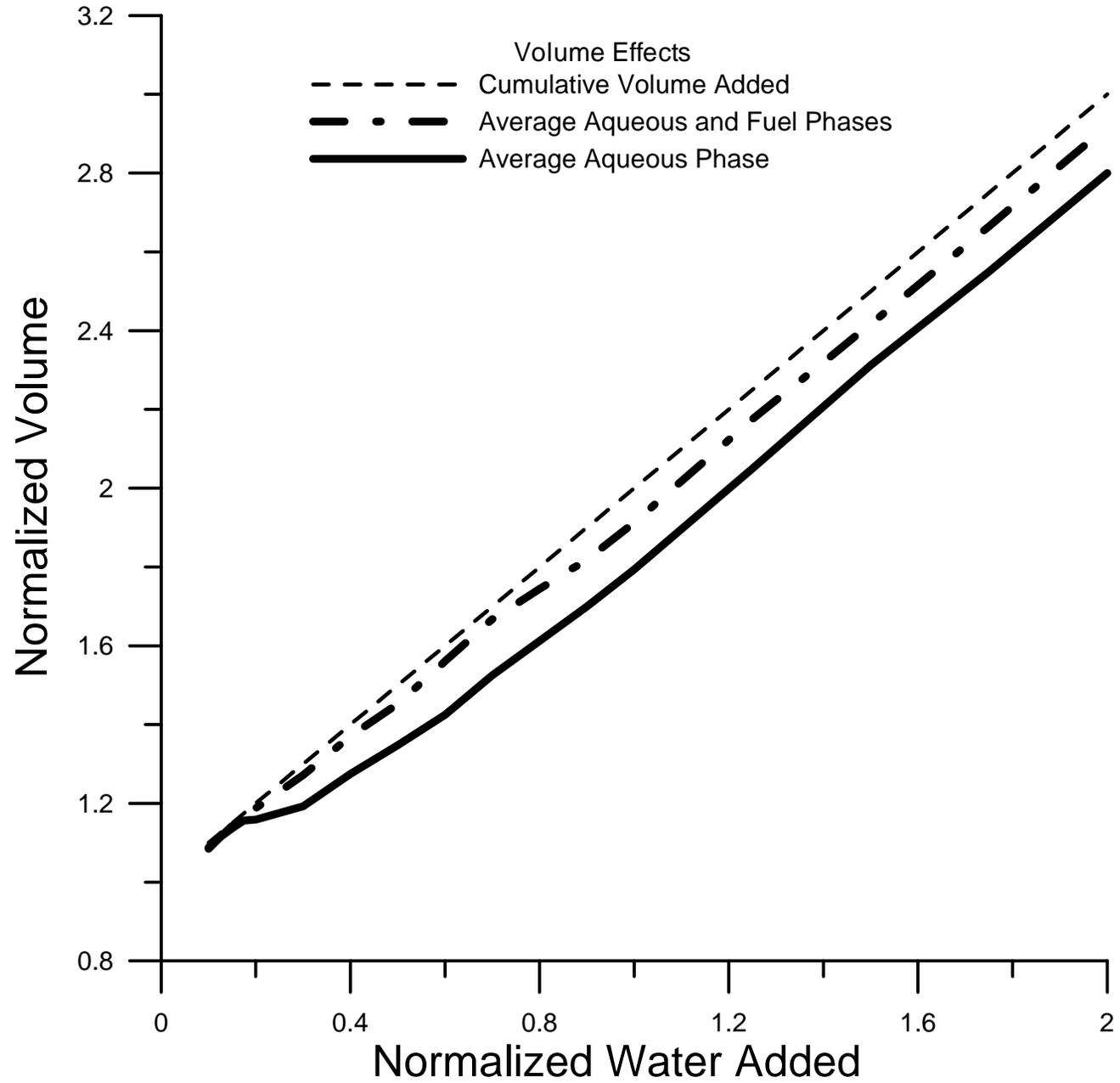


E85 and E95

E100 Ethanol

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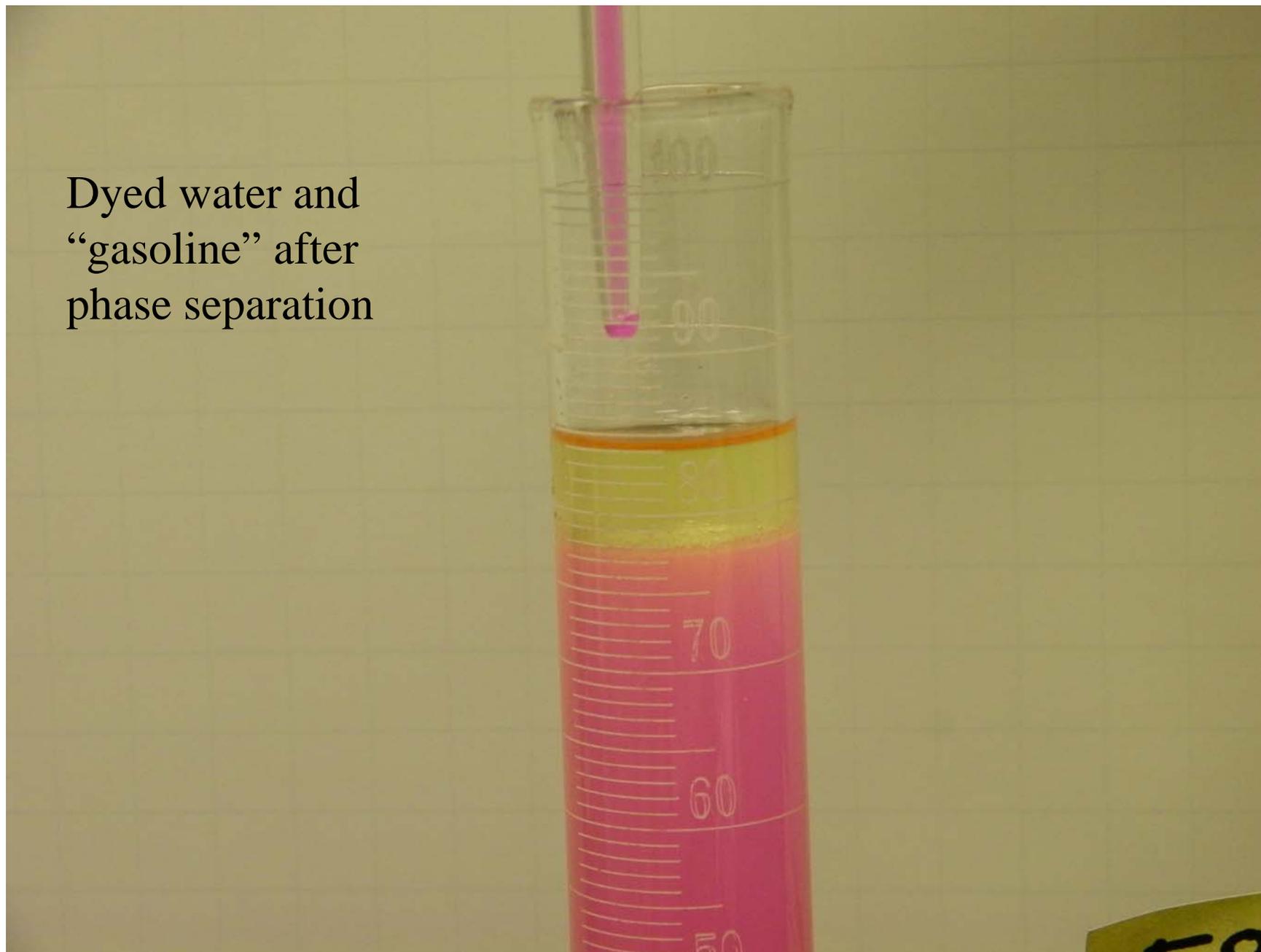
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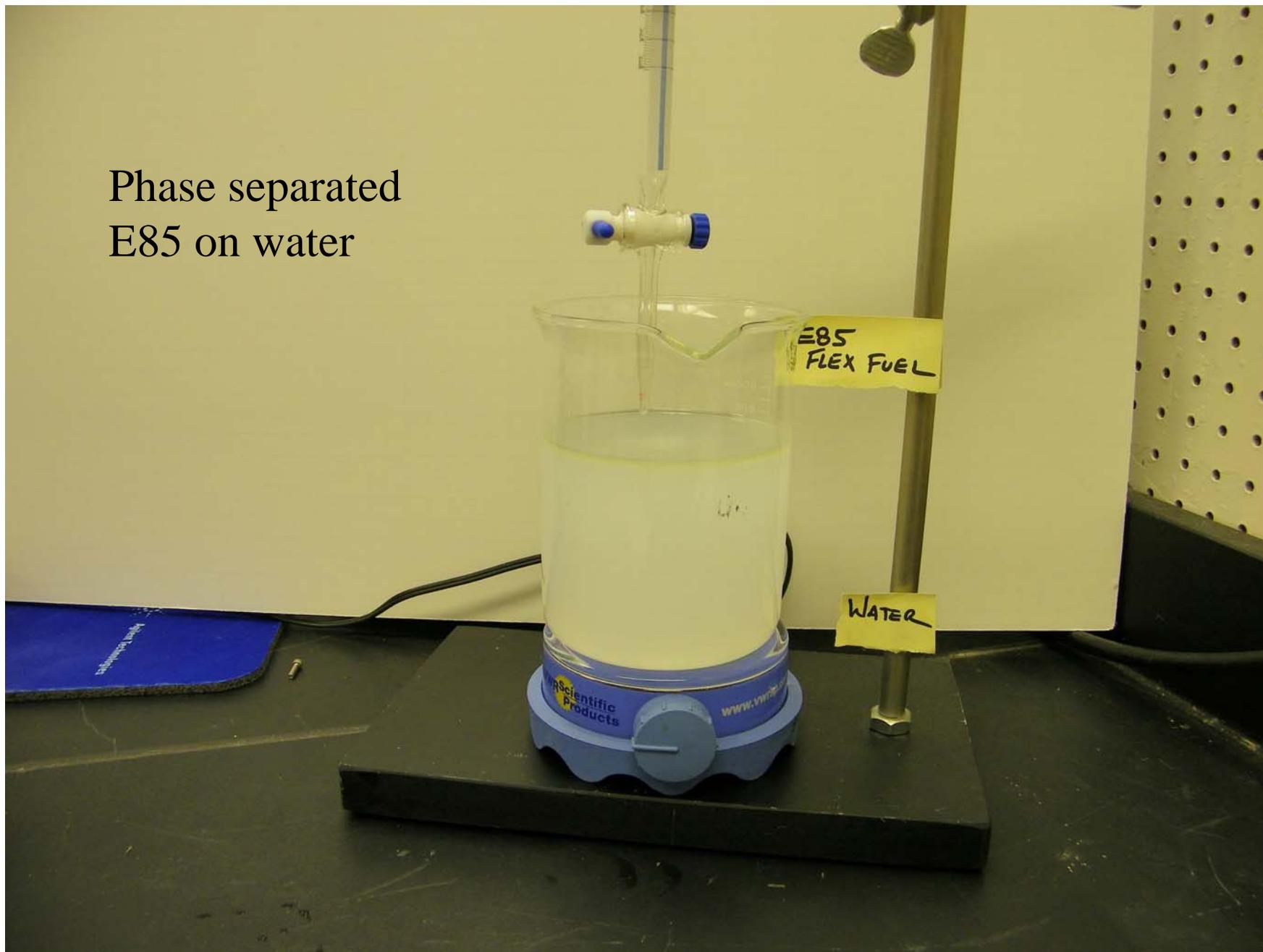
## Videos

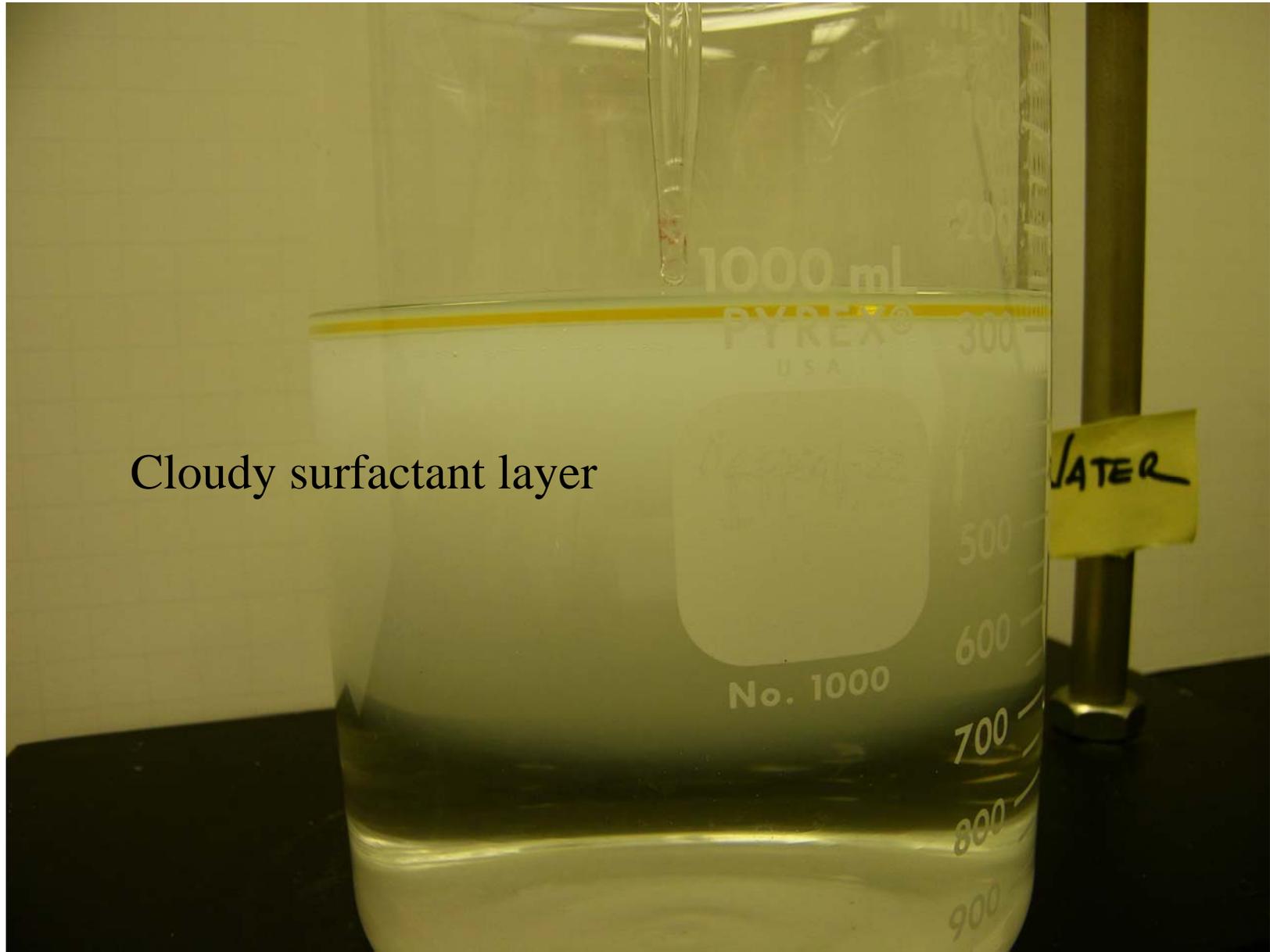
- Dyed Water Added to E85
  - Initially water is absorbed by E85
  - Initially no increase in volume
  - E85 breaks into “gasoline” and aqueous/ethanol phase
- E85 added to water
  - Quiescent E85 jets into water
  - Cloudy surfactant layer over clear water
  - Gasoline accumulates on surface
  - In moving system gasoline “rides” surface

Dyed water and  
“gasoline” after  
phase separation



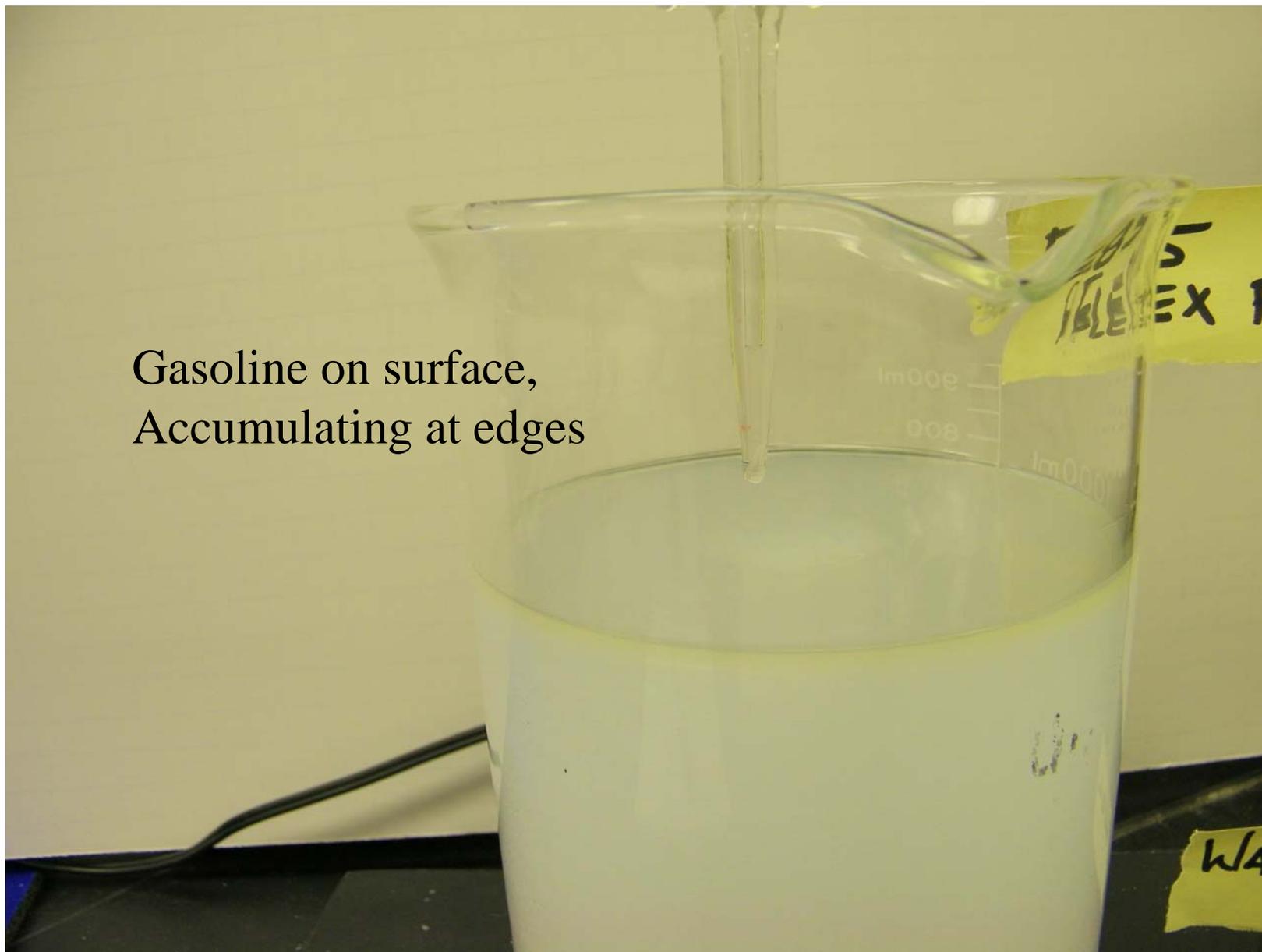
Phase separated  
E85 on water

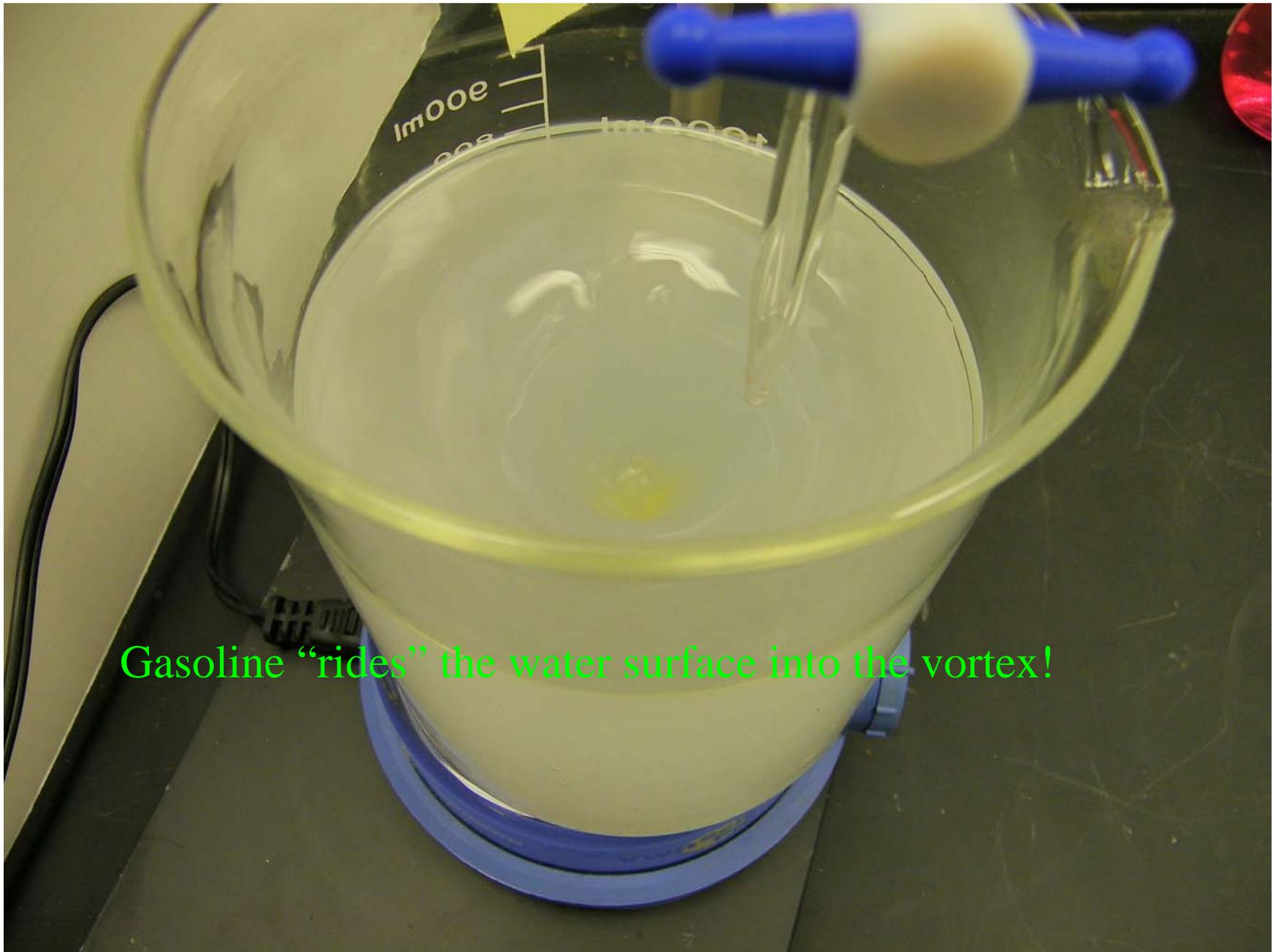




Cloudy surfactant layer

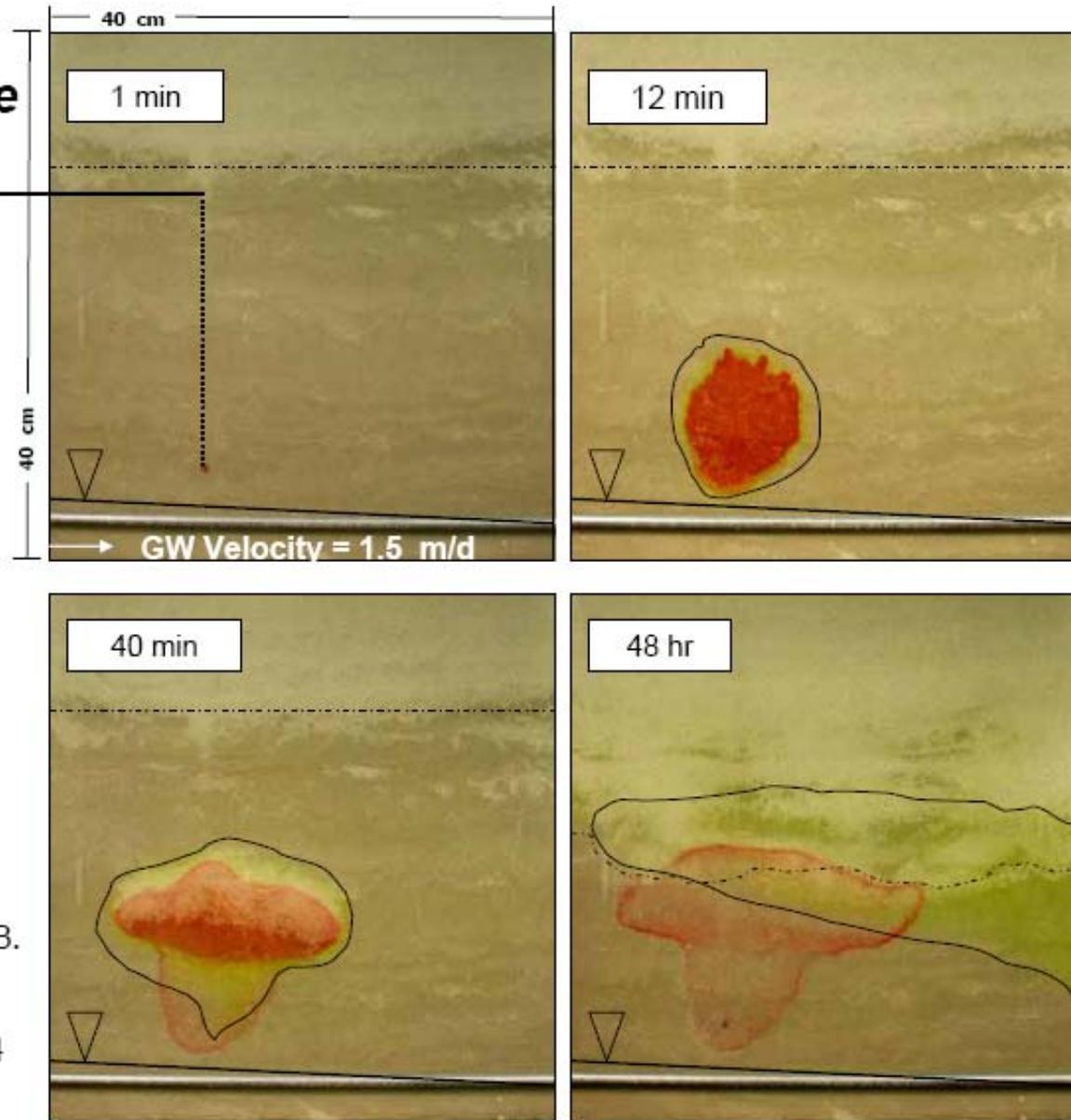
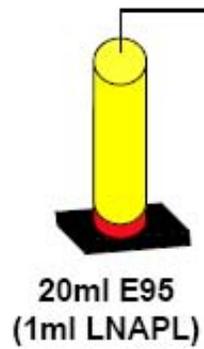
Gasoline on surface,  
Accumulating at edges





Gasoline "rides" the water surface into the vortex!

## 2D Bench-Scale

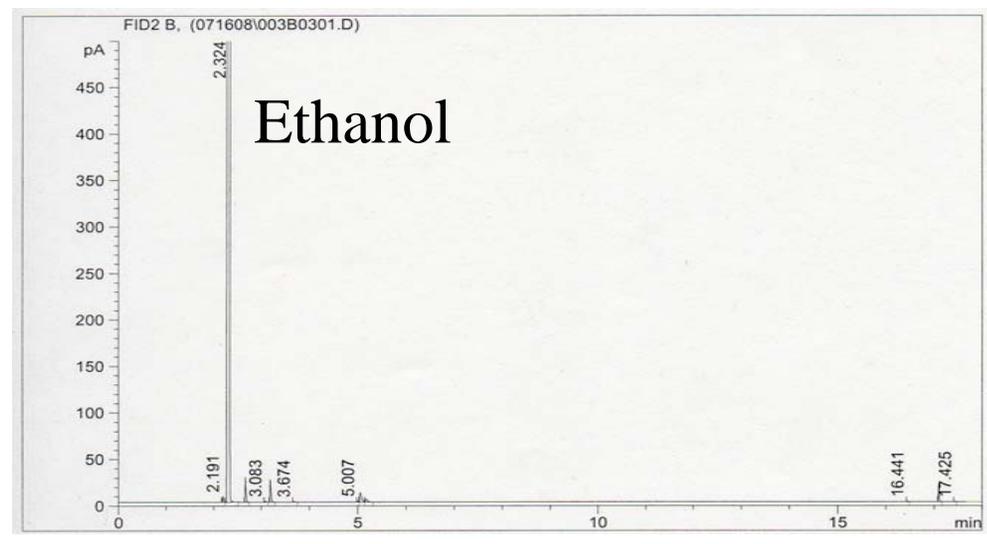
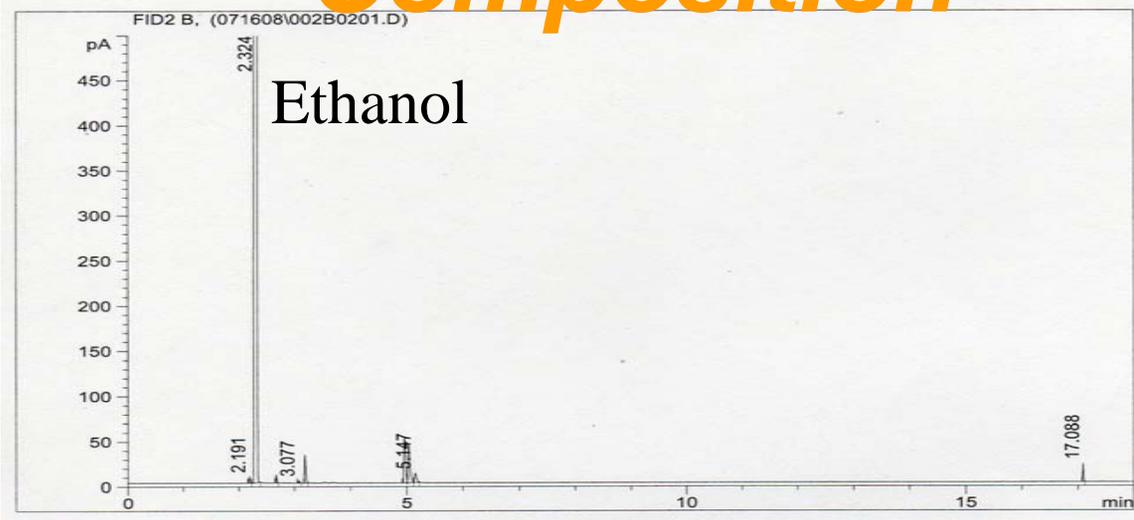


*In* Cápiro, N., B.P.  
Stafford, W.G. Rixey, P.B.  
Bedient, and P.J.J.  
Alvarez (2007). *Water  
Research* 41(3):656-664

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# Composition



# Ethanol Components

Name	Formula	CAS. Number	Concentration (wt. %)	
			Wet Mill Sample	Dry Mill Sample
Water <sup>(1)</sup>	H <sub>2</sub> O	7732-18-15	0.65	0.08
Methanol	CH <sub>4</sub> O	67-55-1	0.07	0.06
Ethanol <sup>(2)</sup>	C <sub>2</sub> H <sub>6</sub> O	64-17-5	97.89	99.75
1-Propanol	C <sub>3</sub> H <sub>8</sub> O	71-23-8	0.03	0.08
Isobutyl Alcohol	C <sub>4</sub> H <sub>10</sub> O	78-83-1	0.10	0.08
2-Methyl 1-Butanol	C <sub>5</sub> H <sub>12</sub> O	137-32-6	0.06	0.01
3-Methyl 1-Butanol	C <sub>5</sub> H <sub>12</sub> O	123-51-3	0.21	0.02
Ethyl Acetate	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	141-78-6	0.02	--
1,1-Diethoxyethane	C <sub>6</sub> H <sub>14</sub> O <sub>2</sub>	105-57-7	0.28	--

<sup>(1)</sup> Determined by Karl Fischer titration

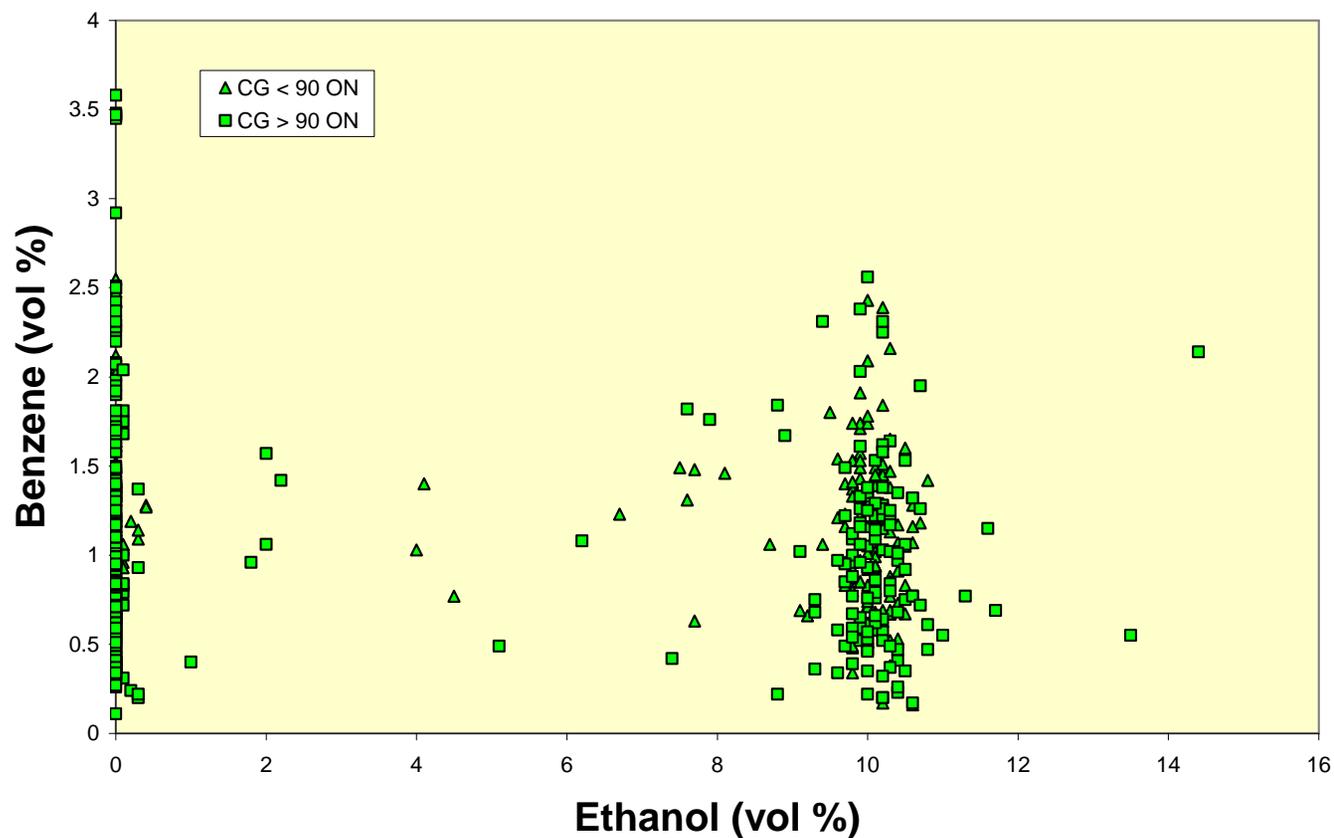
<sup>(2)</sup> Determined by remainder of other compounds

# Observed Ethanol Concentrations in Gasoline



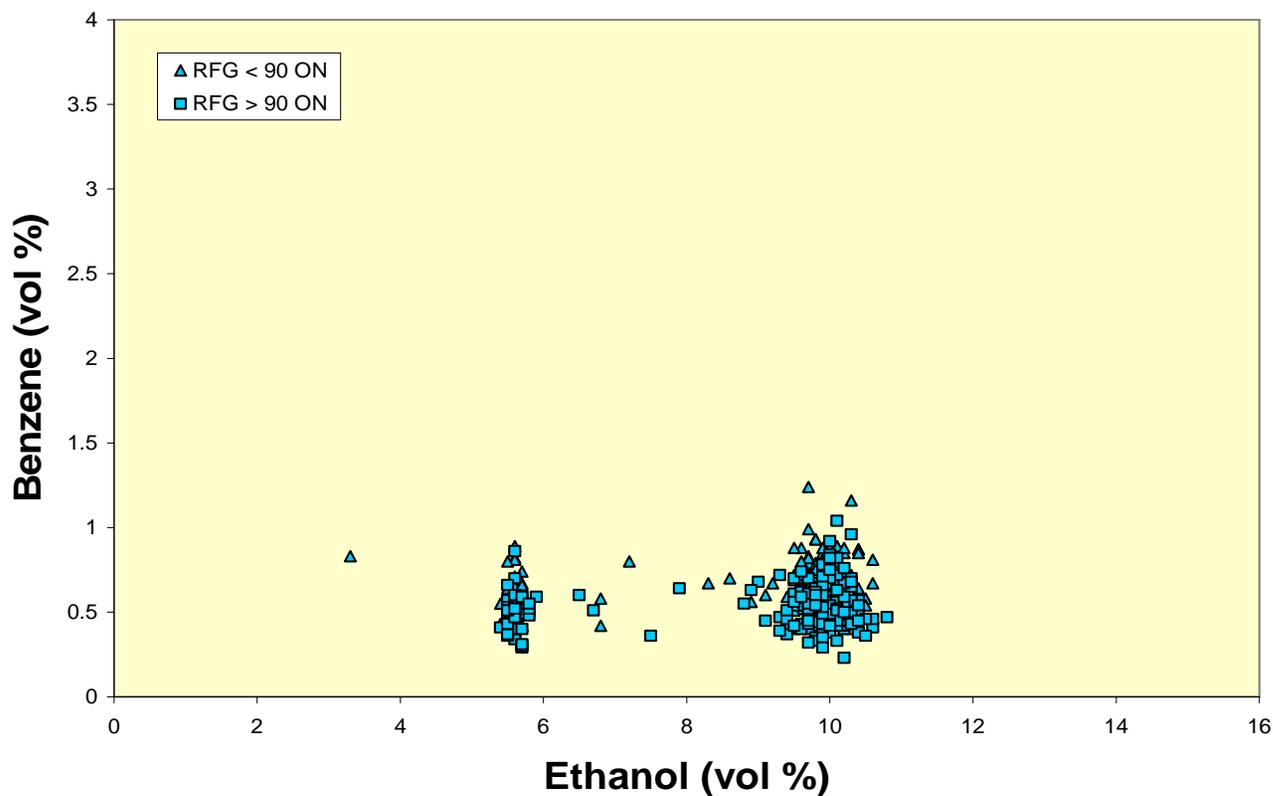
# Northrop-Grumman (successor to NIPER) Bartlesville, Oklahoma Summer 07

Conventional Gasoline



# Northrop-Grumman (successor to NIPER) Bartlesville, Oklahoma

## Reformulated Gasoline



# E85<sup>TM</sup>

85% Ethanol

11

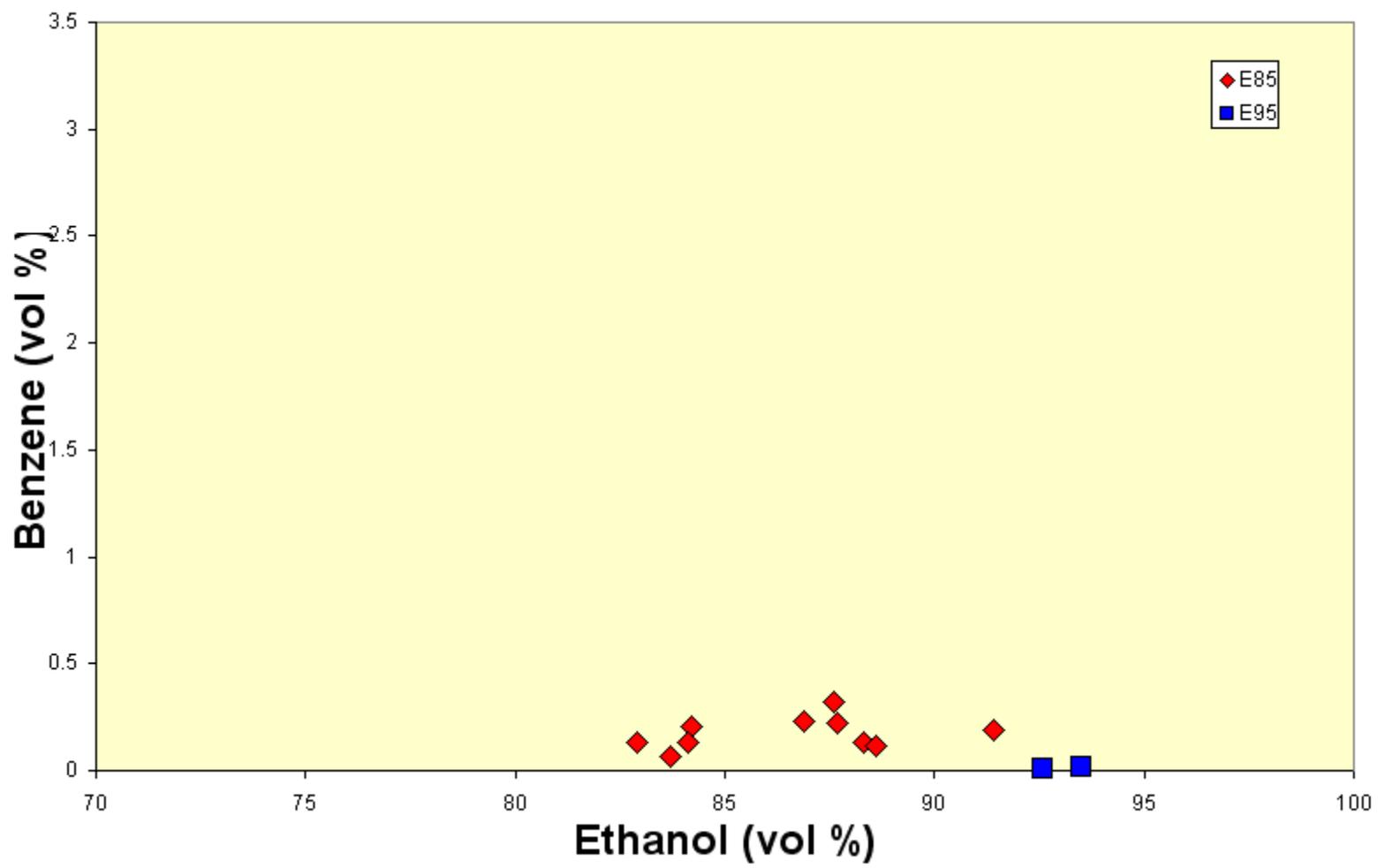
**E-85**  
MINIMUM  
85%  
ETHANOL



2.99

Price per gallon  
All taxes included

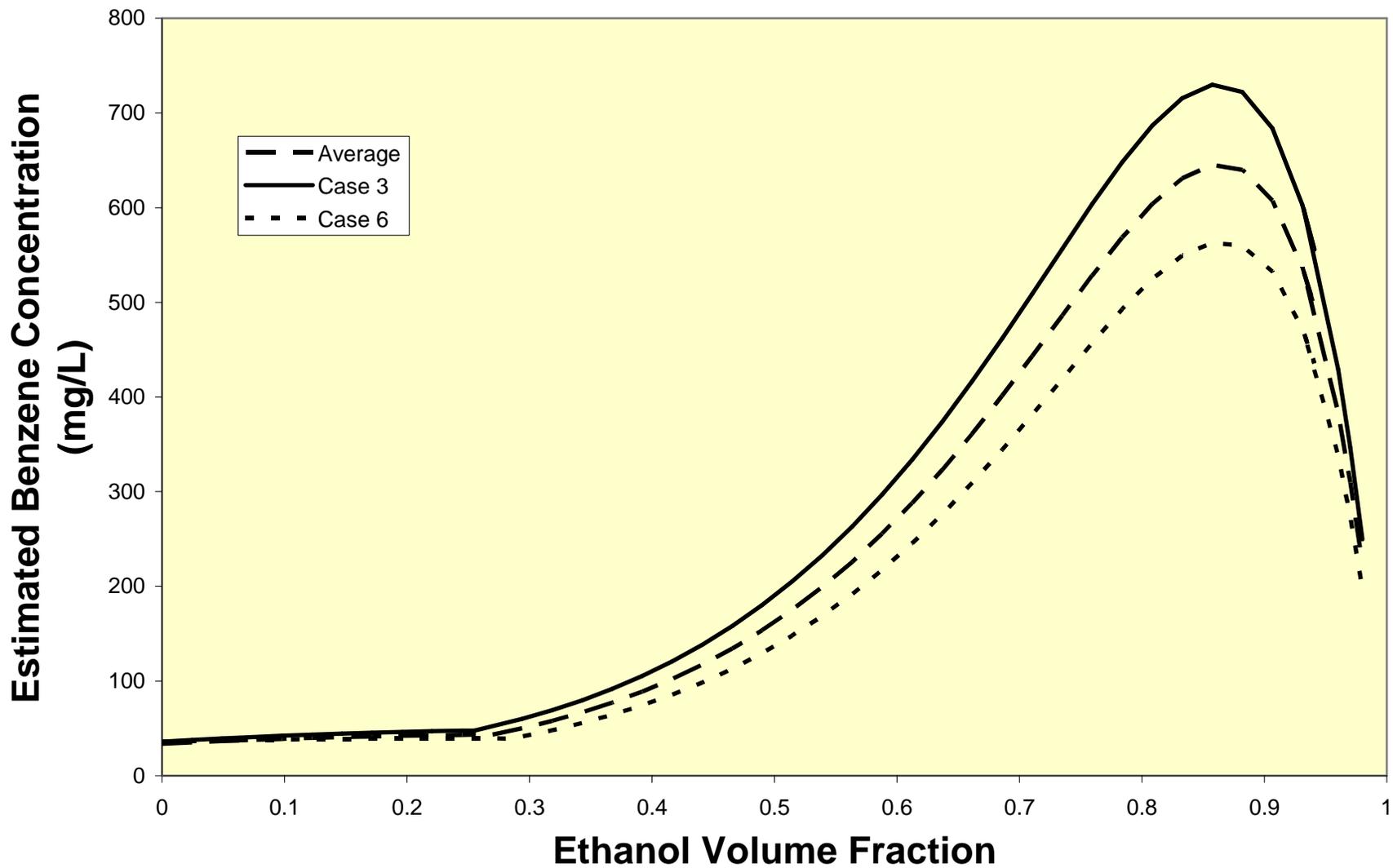
**Ethanol**  
P R E S S



# Cosolubility

- Ethanol increases the aqueous solubility of petroleum hydrocarbons
  - Dependent on
    - Ethanol concentration in water
    - Petroleum hydrocarbon concentration in gasoline
  - Theory developed by Heerman and Powers, 1998
    - Example:
      - Gasoline containing 1% benzene, mixed with denatured alcohol
      - Alcohol denatured with gasoline containing 1% benzene
      - Benzene mass not limiting

# Estimated Benzene Concentration



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decisions*

# **Spill to Land, Info. Courtesy of Dr. Roy Spalding, U of Nebraska**

## **Balaton, Minnesota**



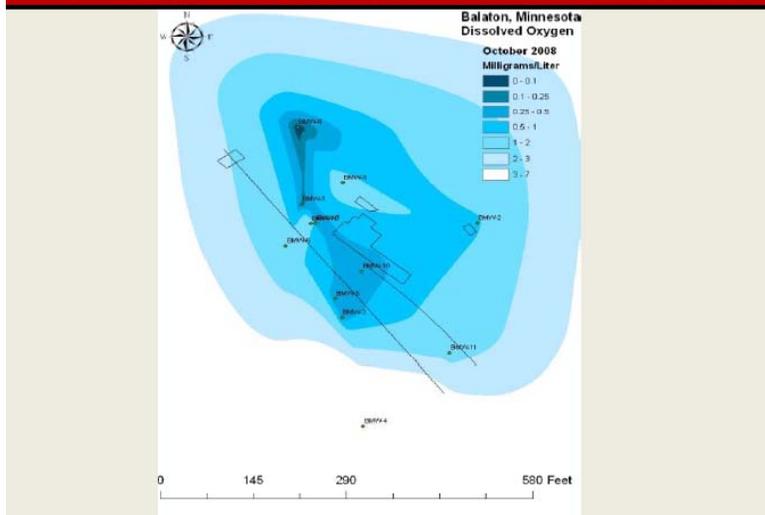
# Balaton, Minnesota

July 28, 2004

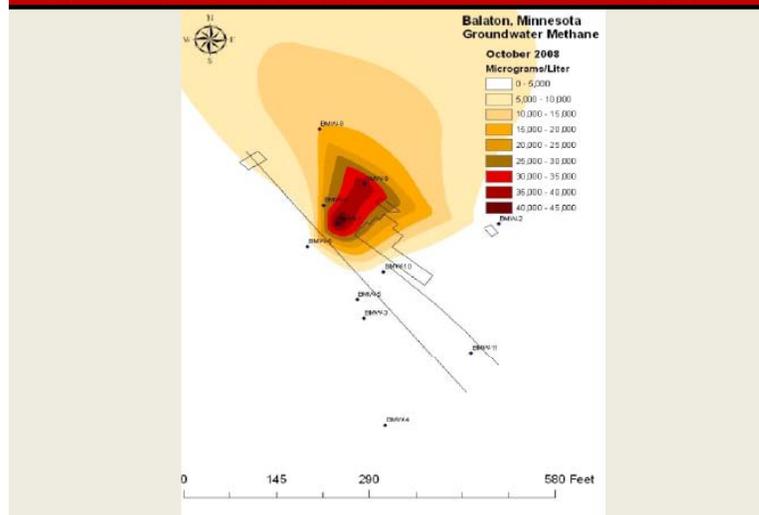
**~90,000 Gs of d-ethanol released**  
**~10,000 Gs residual ethanol after**  
**product removal and soil**  
**excavation**



**D. Oxygen: 4.3 Years After Derailment**

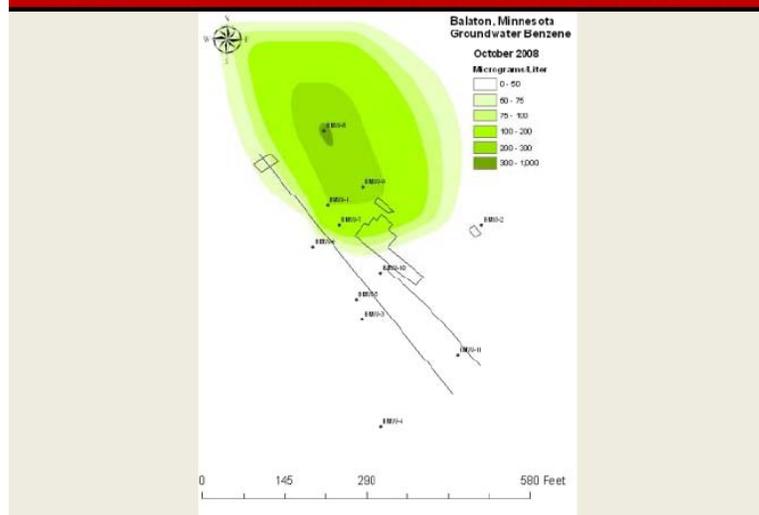


**Methane: 4.3 Years After Derailment**



- Long delay in ground water impact
- Ethanol hanging up in vadose zone for undetermined reasons
- Methane at water solubility
- Similar behavior to two other sites under investigation

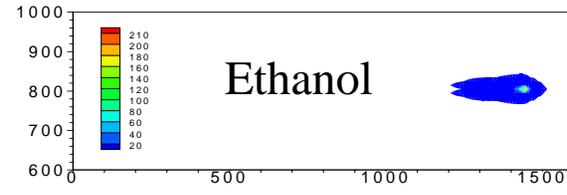
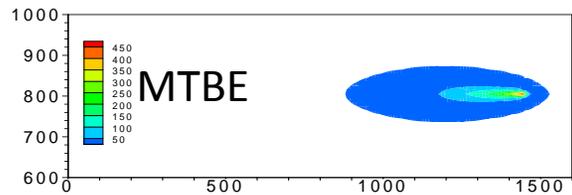
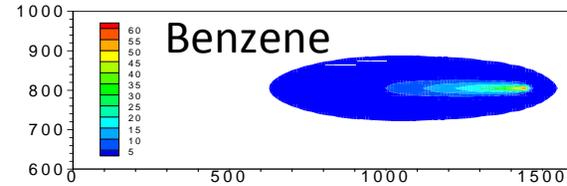
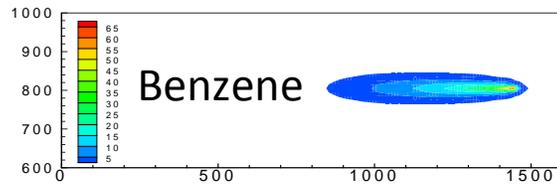
**Benzene: 4.3 Years After Derailment**



## ***E10 Releases to Land***

- Formation of groundwater plumes
- Biotransformation of ethanol causes extension of BTEX plumes
  - Order of 2x length
- Simulation (next slide) requires accounting for electron acceptors/donors: O<sub>2</sub>, NO<sub>2</sub>, Fe<sup>+++</sup>, SO<sub>4</sub>, CO<sub>2</sub>

# Modeled RFG (w MTBE) vs E10



# ***Spill to Broadland Creek***

## ***May 17, 2008***

### ***Info. Courtesy of***

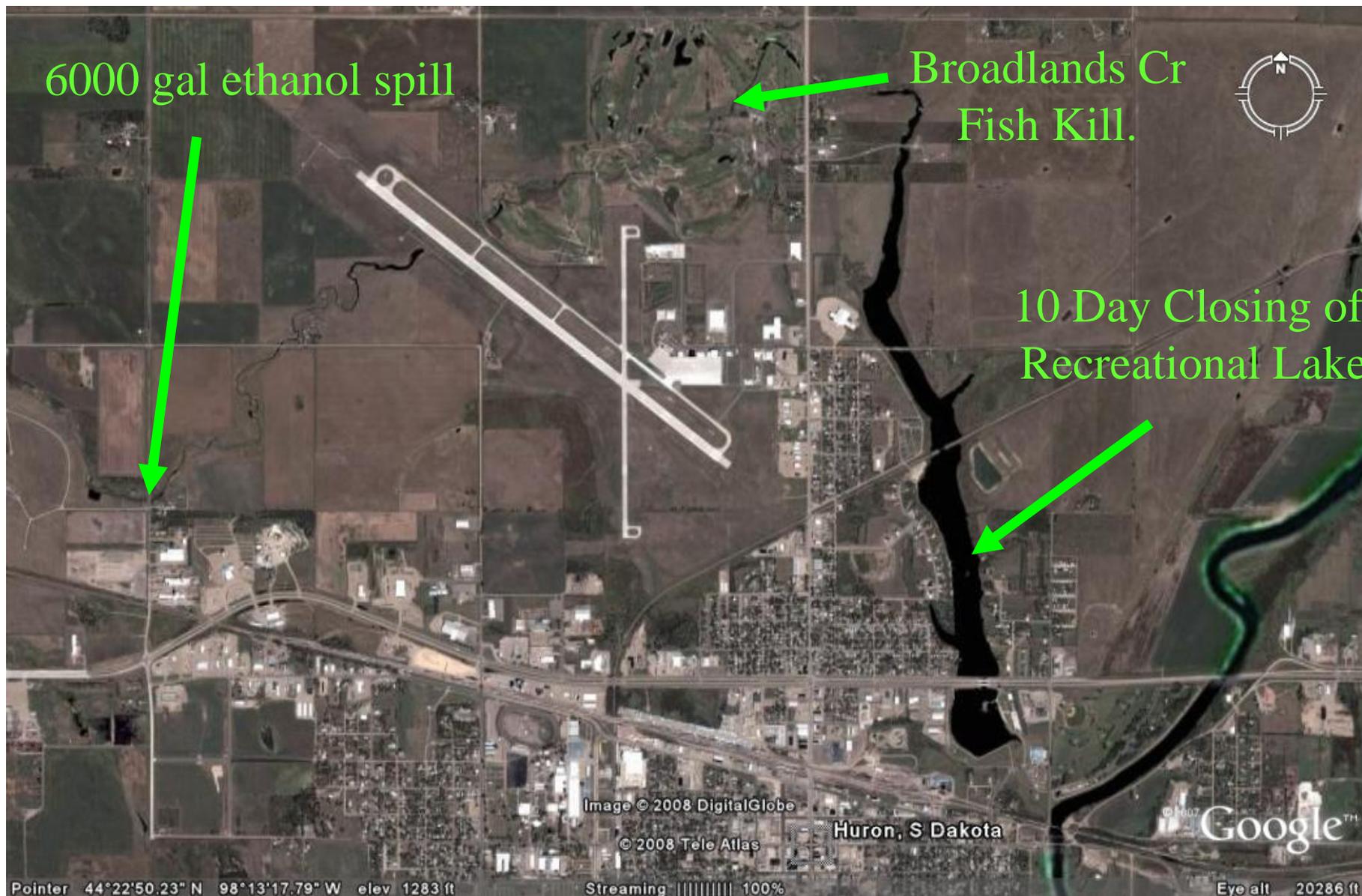
#### ***Kim McIntosh, SD-DENR***

- Dry Mill plant in Huron, South Dakota
- Transfer line hose broke during filling of tank car
- Approx. 6000 gallons of ethanol released
- 100s of fish killed in creek
  - Minnows, bullhead, carp
- Recreational lake closed for 10 days

# Response

- Deploy boom and aerators 3 mi downstream
- Daily check of temperature, dissolved oxygen, pH, conductivity
  - 5/20/2008 am D.O. 0.2 to 10.6 mg/L
  - 5/21/2008 am D.O. 0.3 to 5.4 mg/L
  - 5/21/2008 pm D.O. 1.0 to 7.6 mg/L
  - 5/22/2008 am D.O. 1.0 to 6.0 mg/L
  - 5/22/2008 pm D.O. 1.2 to 5.7 mg/L (downstream)
  - 5/22/2008 pm D.O. 0.4 to 8.0 mg/L
  - 5/23/2008 am D.O. 1.0 to 6.6 mg/L (upstream)
  - 5/23/2008 pm D.O. 0.3 to 5.8 mg/L
  - 5/23/2008 pm D.O. 0.1 to 5.6 mg/L (downstream)
  - 5/24/2008 am D.O. 0.2 to 5.9 mg/L
  - 5/27/2008 am D.O. 1.0 to 8.0 mg/L
- Moved aerators as plume moved downstream

# Huron, S.D., May 17, 2008



## Comments

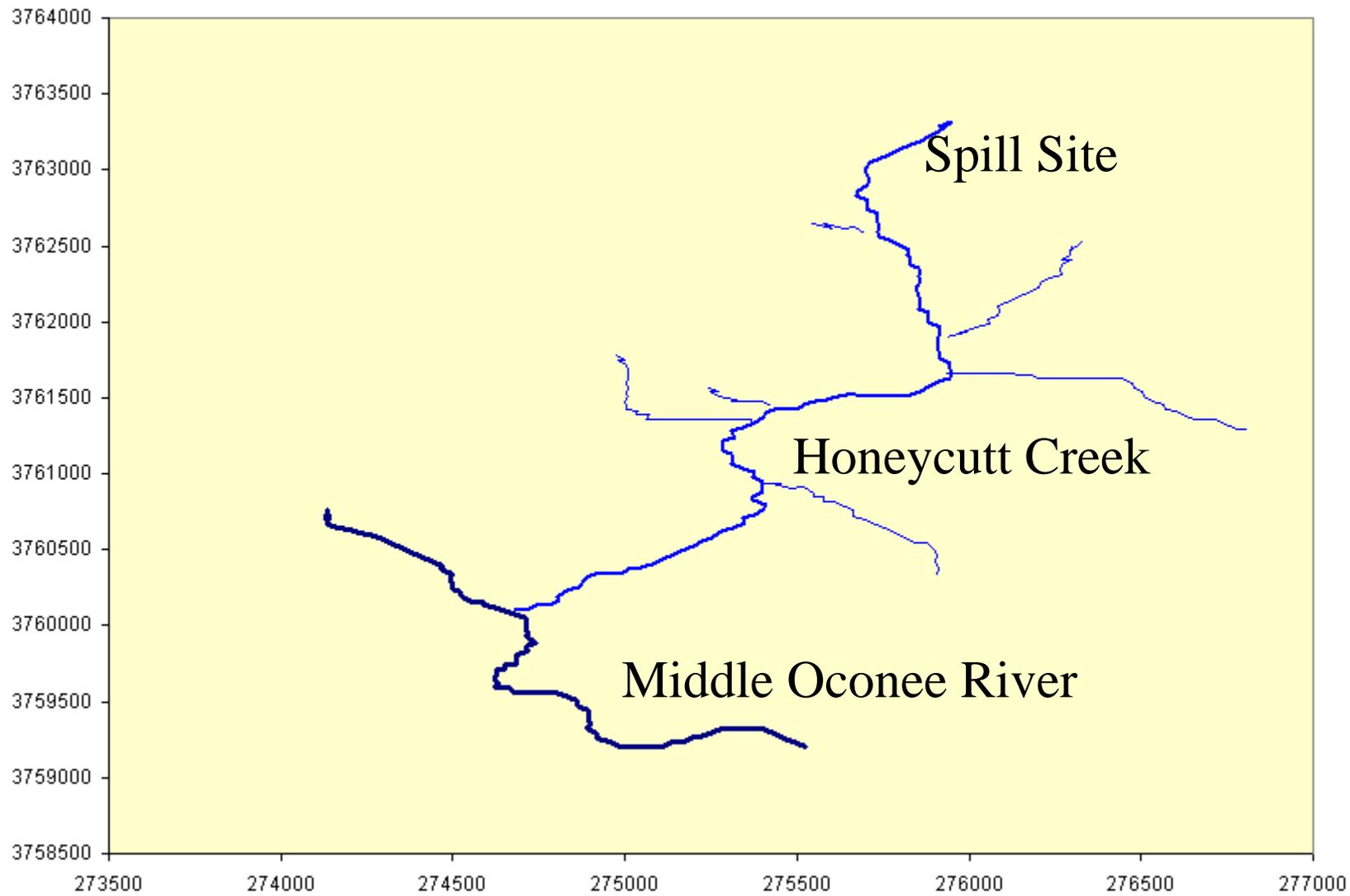
- No gasoline observed
- 4 water samples for VOCs 5/21
  - BTEX, tri-methylbenzenes, ethanol, methanol
  - Most results ND, one sample ethanol at 6800 ppb
- Generally increased D.O. by 5/27/2008

# *Honeycutt Creek/Middle Oconee River Athens, Georgia*

- Scenario:
  - Small creek that drains area near fuel terminal
  - Oil spill in 2003 (14K gallons mixed gasoline, diesel, waste oil)
  - Approx 3 miles downstream intercepts Middle Oconee River
  - Drinking Water Intake for Athens, Georgia

## Data

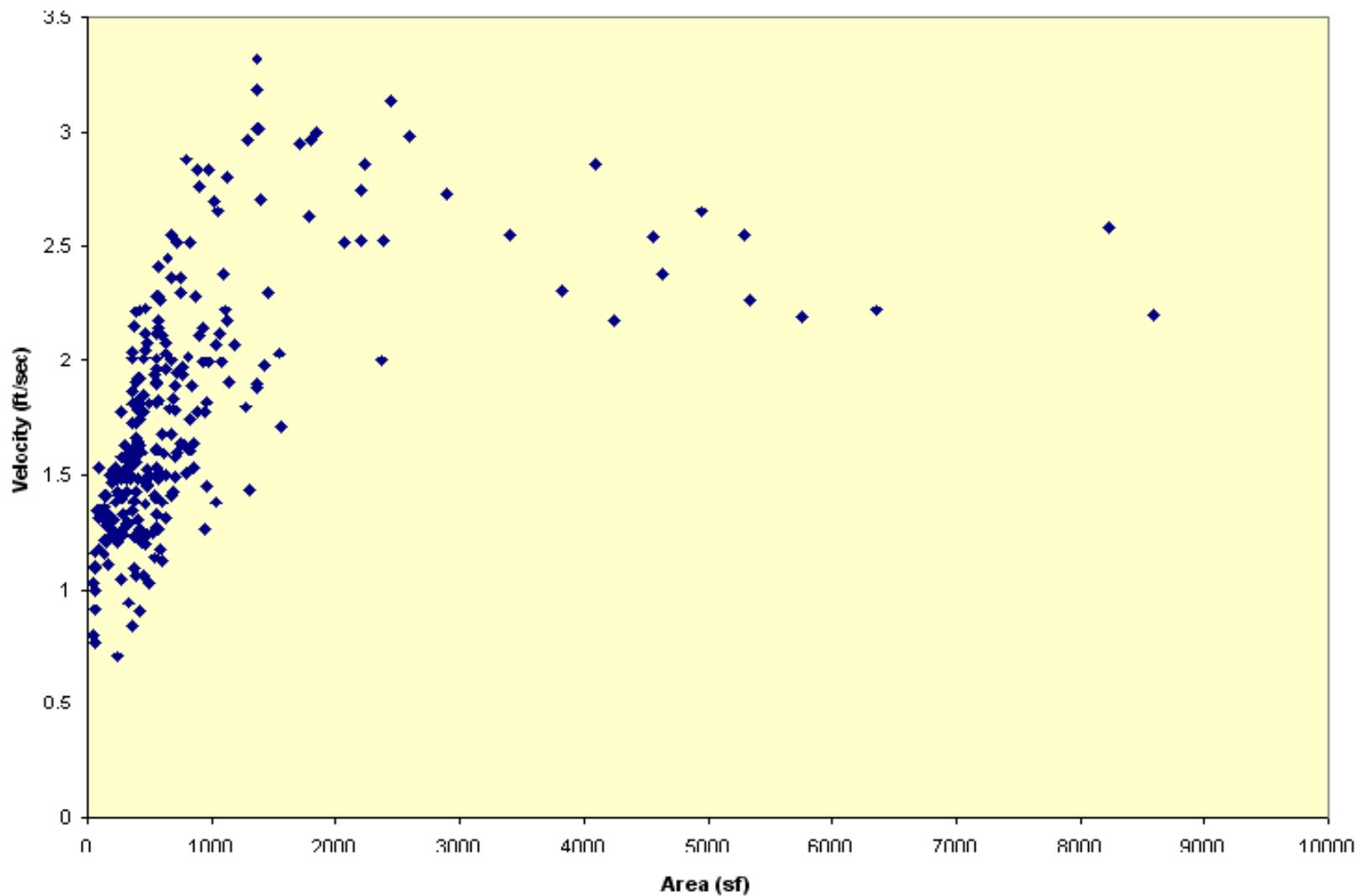
- Hypothetical Release Scenario
- Composition and Phase Separation Data from Laboratory
- Stream network geometry from USGS DEMs
- Flows
  - Honeycutt Creek flow and geometry from a day's work
  - Middle Oconee River flows and discharge stage from USGS gages



## USGS Station Geometry

- Depth and Width of River/Streams are necessary
  - Determines dilution of dissolved chemicals
- Velocity
  - Determines transport rate
- USGS surveys stations each year
  - Continual change due to scouring and sedimentation
  - Our protocol is to use the latest data for a given channel size

# USGS Velocity/Area Data



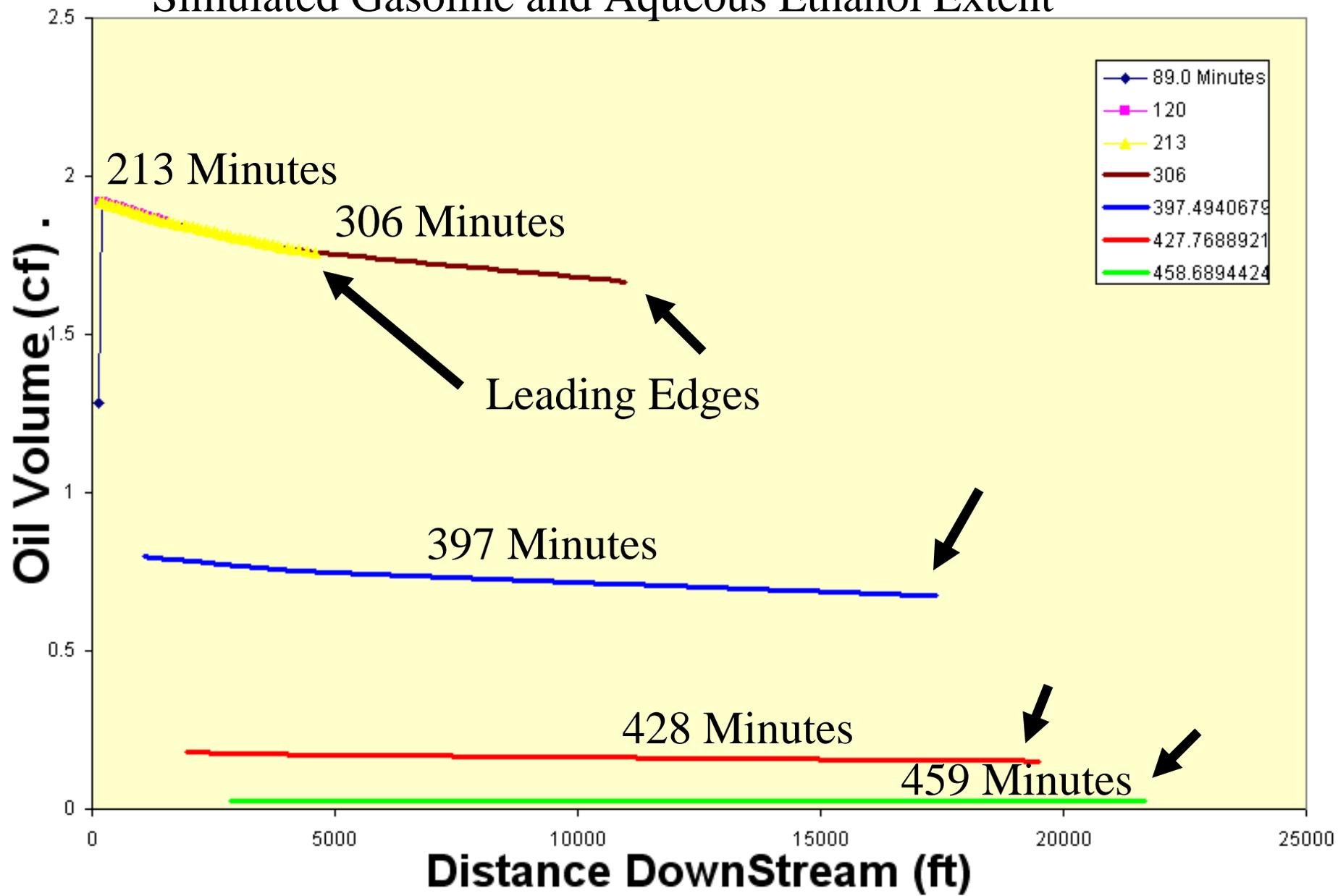
## **Model Approach**

- Use measured flows to drive Lagrangian transport model
  - Flow ~ spill rate
    - Assume complete mixing
    - Adjust velocity via Manning formula
  - Flow  $\gg$  spill rate
    - Mixing zones defined from turbulent diffusion coefficients

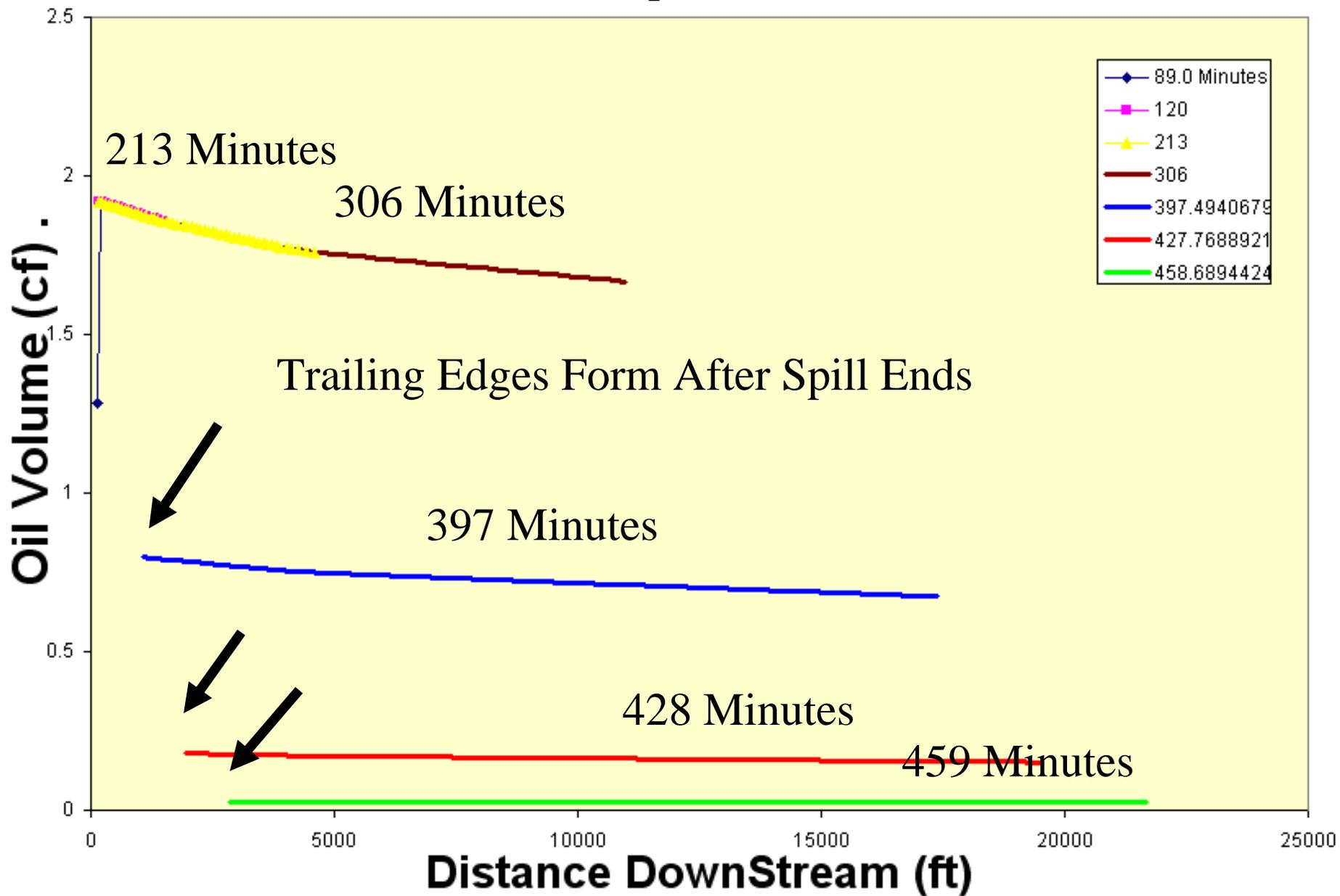
## **Spill Chacteristics**

- 10,000 Gal Spill of E-85
- 5 hour spill duration

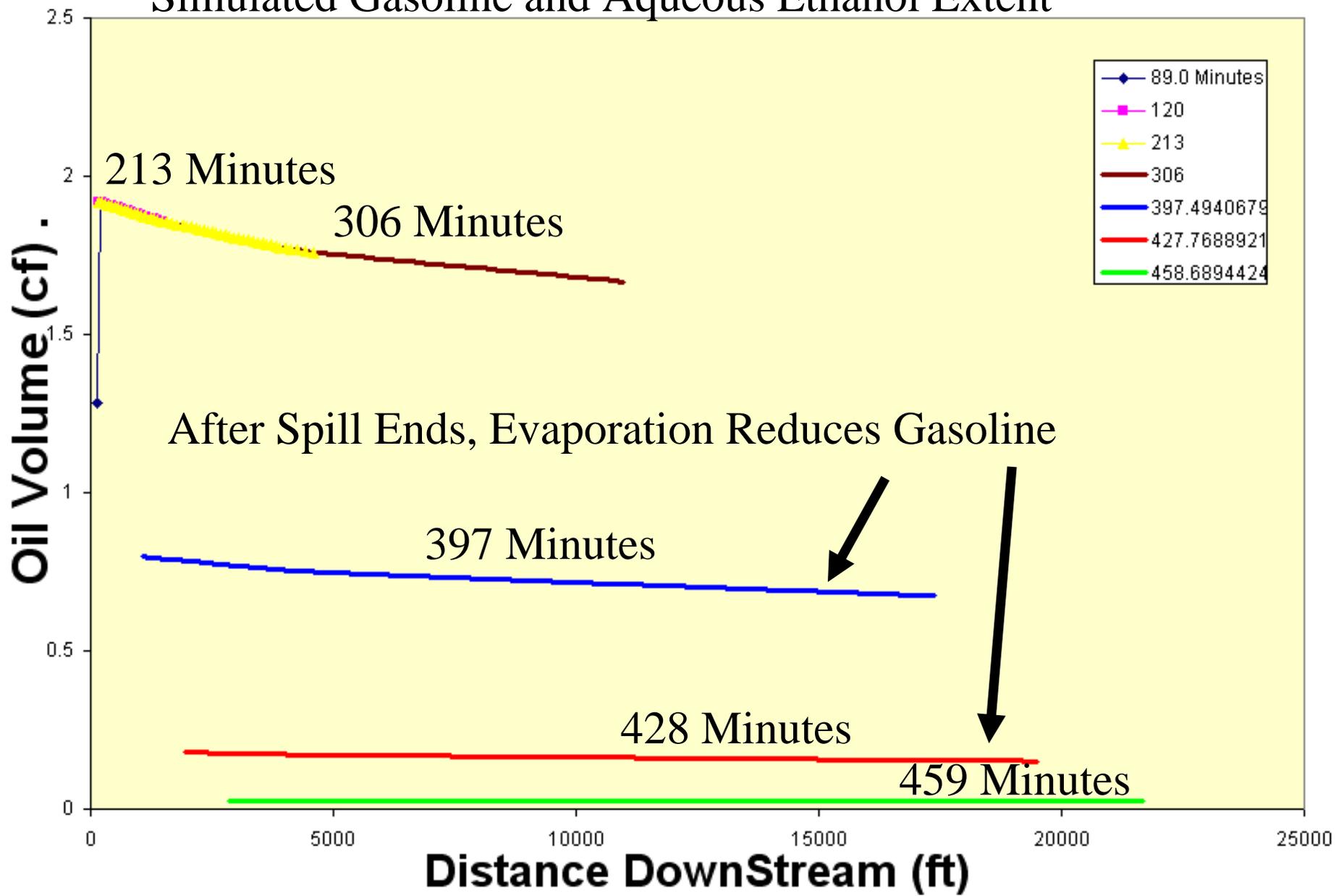
# Simulated Gasoline and Aqueous Ethanol Extent



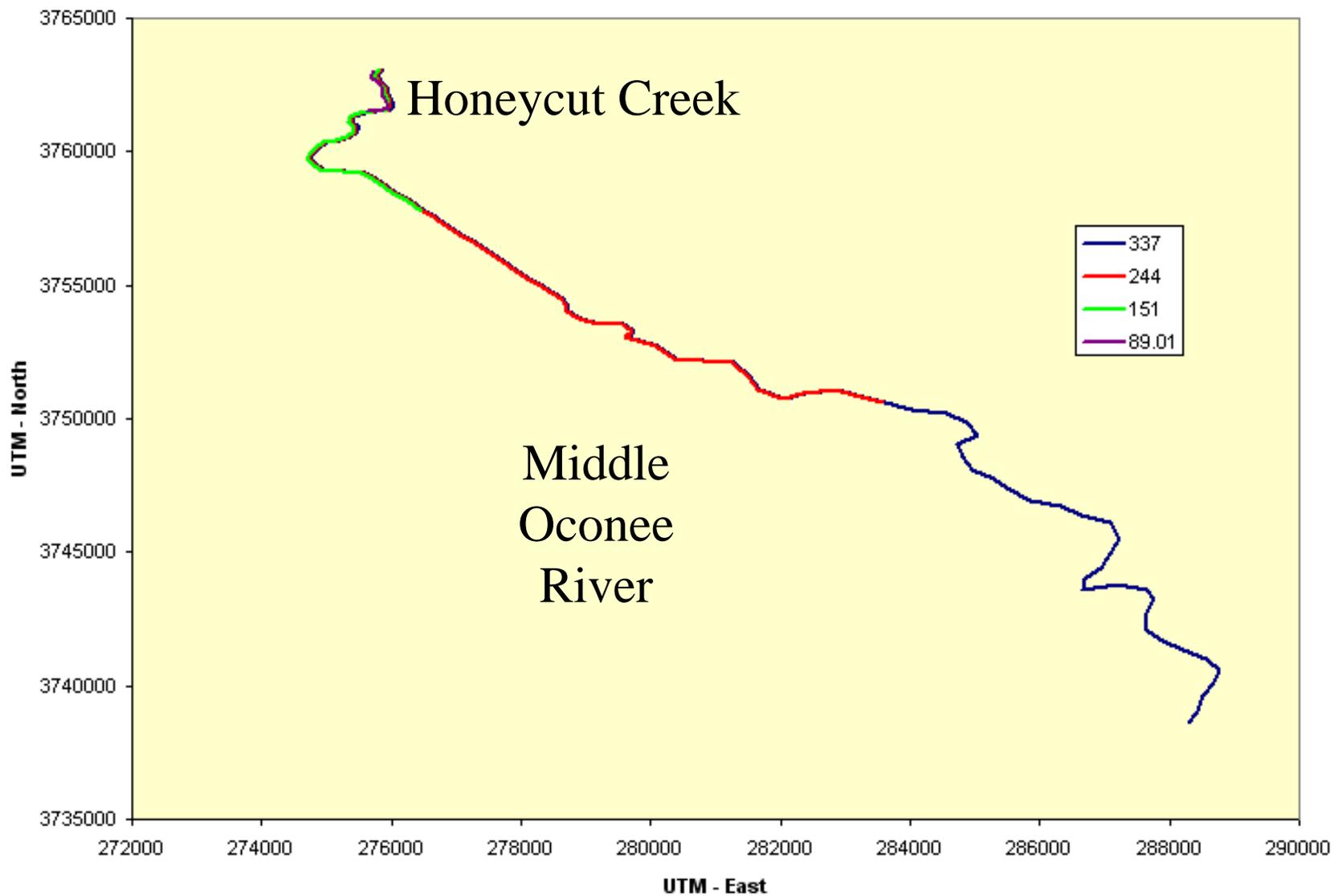
# Simulated Gasoline and Aqueous Ethanol Extent



# Simulated Gasoline and Aqueous Ethanol Extent



# Oil Slick Extent



## Conclusions

- Ethanol rapidly replaced ethers in mid 2006 in reformulated gasoline
- Ethanol is used in about 75% of U.S. gasoline.
- Fuel ethanol contains several impurities including higher molecular weight alcohols
  - 3 to 5 carbon atoms
  - ...But at concentrations < 1/4 percent

## Conclusions

- E85 and E95 adsorb about 20%-30% of own volume in water before phase separating
  - Fuel Ethanol doesn't phase separate
  - E10 phase separates at about 0.5% water

## Conclusions

- Spills of denatured alcohol to land based on three field studies:
  - Hangs up in the vadose zone
  - Methane at max solubility
  - Impacts to ground water delayed
- E10 Releases cause extended BTEX plumes
- Spills to water (Broadlands Ck)
  - No observed gasoline slick from denatured alcohol
  - Loss of dissolved oxygen major impact

# **National Exposure Research Laboratory**

- Although this work was reviewed by EPA and approved for presentation, it may not necessarily reflect official Agency policy.
- Thanks to:
  - Mark Toso, Minnesota PCA,
  - Cheryl Dickson, Northrop-Grumman
  - Kim McIntosh, South Dakota DENR,
  - Dr. Roy Spalding, University of Nebraska,
  - Dr. Illena Rhodes, Shell Global Solutions
- Contact: [weaver.jim@epa.gov](mailto:weaver.jim@epa.gov)
- EPA report, April 2009:
  - ***Composition and Behavior of Fuel Ethanol, EPA 600/R-09/037***
  - ***from [www.epa.gov/athens/publications](http://www.epa.gov/athens/publications)***